



# **Enzyme tools to determine the well-being**

# of biological communities

Martínez I., Packard T.T., Tames-Espinosa M., Romero-Kutzner V.,

### **Bondyale-Juez D., Gómez M.**

Marine Ecophysiology Group (EOMAR). ECOAQUA Institute, Universidad de Las Palmas de Gran Canaria, Canary Islands, Spain





http://eomar.ulpgc.es

The marine environment is subject to numerous human activities, with significant pressures and impacts from fishing, alterations derived from climate change, and discharges (urban and/or industrial), among others, with consequent loss of environmental quality, and the elimination or alteration of habitat and species populations. The possible alterations of biological communities can cause changes in physiological states, species diversity, abundance and biomass. Because of this, we have aligned our research with the objectives of environmental management. Using enzymes as biomarkers helps us understand and correct the effects of anthropogenic stresses on marine ecosystems.

Enzymes are central to every biochemical process. They catalyze the hundreds of stepwise reactions that degrade nutrient molecules, conserve and transform chemical energy. To assay the enzyme activities, the velocity of the specific-enzyme reaction is determined, measuring the generation of product per time using different chemical techniques.



### **ETS Electron transport system**

- A combination of enzymes that control the respiratory oxygen consumption
- Measure of activity: production of formazan, from INT (a tetrazolium salt) reduction

IDH Isocitrate dehydrogenase



• 1. GDH activity measured in the mysid *Leptomysis lingvura* over a month of

• 2. IDH vs Φ (potential respiration from ETS activity), in 200-2000μm and 0.7-50µm plankton samples (Tames-Espinosa et al., 2017)

- Enzyme responsible for respiratory CO<sub>2</sub> production
- Measure of activity: production of NADPH, from NADP<sup>+</sup> reduction

## GDH

#### **Glutamate dehydrogenase**

- The main role of this enzyme consists of glutamate deamination, producing  $\alpha$ -ketoglutarate, feeding the tricarboxylic acid cycle
- Measure of activity: production of NADH, from NAD<sup>+</sup> reduction



metabolism-related Energy biomarkers in Holothuria forskali. Fig 2 in Tonn et al. (2016). Measures of ETS activity, E<sub>a</sub>, CEA and IDH in the muscle tissue.







• 4. Alterations in GDH studied in different tissues of Labeo rohita exposed to sublethal concentration of cypermethrin. Fig 7 in Philip and Rajastree (1996).

Based on the metabolic costs hypothesis (Calow and Sibly, 1990), in the last decades, a biomarker technique to assess the effect of stress on the energy budget of organisms has been developed  $\rightarrow$  the cellular energy allocation (CEA)

$$\checkmark$$
 E<sub>a</sub> : energy reserves available = E<sub>protein</sub> + E<sub>carbohydrate</sub> + E<sub>lipid</sub>

 $\checkmark$  E<sub>c</sub>: energy consumption = ETS activity



5. Effect of lindane on the energy budget (CEA Daphnia values) ot magna. Fig 1 in DeCoen and Janssen (1997



CELLULAR

**ENERGY** 

**ALLOCATION** 



 $CEA = E_a / E_c$ 

**ENERGY METABOLISM** 

(a key goal in defining metabolic stress in marine organisms)



). MI ETS activity	METHODS* s.h. : sample homogenateS activityBased on Owens & King (1975) and Kenner & Ahmed(1975)(1975)100µl s.h. + 300µl substrate solution + 100µl INTRead the absorbance during 8min at 490nm	Proteins	According to Smith et al. (1985) 25μl s.h. + 500μl work reagent / Incubate 30' at 37.5°C Bovine serum albumin as standard Read the absorbance at 562nm
			Based on Bligh and Dyer (1959)
IDH activity	According to Tames-Espinosa et al. (2017) 100μl s.h. + 300μl (MgCl <sub>2</sub> +Isocitrate) solution + 100μl NADP <sup>+</sup> solution Read the absorbance at 340nm during 10min	Lipids	$200\mu$ s.n. + $500\mu$ chloroform + $500\mu$ methanol + $250\mu$ bidistilled water centrifugation $\rightarrow$ 100 $\mu$ l supernatant + $500\mu$ l sulfuric acid $\rightarrow$ Charred $\rightarrow$ diluted 1:6 in bidistilled water Glyceryl tripalmitate as standard Read the absorbance at 340nm
GDH activity	As described in Fernández-Urruzola et al. (2011) 200µl s.h. + 300µl NAD <sup>+</sup> solution + 250µl ADP solution + 500µl Glutamate Read the absorbance at 340nm during 8min	Carbohydrates	According to Nahrgang et al. (2010) 50µl s.h. + 200µl sulfuric acid + 50µl phenol Glycogen as standard Read the absorbance at 490nm

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