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# **Enzyme Assays in Assessing Respiratory Metabolism in Marine Plankton.**

**The development of a new CO<sub>2</sub>-production proxy and further  
applications of respiratory enzymatic assays in oceanography**

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**Doctoral dissertation  
for the Ph.D. degree in Oceanography and Global Change**

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PROGRAMA DE DOCTORADO DE OCEANOGRAFÍA Y CAMBIO  
GLOBAL DE LA UNIVERSIDAD DE LAS PALMAS DE GRAN  
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**INFORMA,**

Que la Comisión Académica del Programa de Doctorado, en su sesión de fecha veinticuatro de mayo de dos mil diecinueve tomó el acuerdo de dar el consentimiento para su tramitación, a la tesis doctoral titulada “Enzyme Assays in Assessing Respiratory Metabolism in Marine Plankton. The development of a new CO<sub>2</sub>-production proxy and further applications of respiratory enzymatic assays in oceanography” presentada por la doctoranda Dª María Teresa Tames Espinosa y dirigida por la Dra. May Gómez y el Dr. Theodore T. Packard.

Asimismo, se acordó el informar favorablemente la solicitud para optar a la Mención Internacional del Título de Doctor, por cumplir los requisitos reglamentarios.

Y para que así conste y a efectos de lo previsto en el Artº 11 del reglamento de Estudios de Doctorado (BOULPGC 7/10/2016) de la Universidad de Las Palmas de Gran Canaria, firmo la presente en Las Palmas de Gran Canaria, a veinticuatro de mayo de dos mil diecinueve.



# Enzyme Assays in Assessing Respiratory Metabolism in Marine Plankton.

The development of a new CO<sub>2</sub>-production proxy and further applications of respiratory enzymatic assays in oceanography



DOCTORAL DISSERTATION  
for the Ph.D. degree in Oceanography and Global Change

Tesis doctoral presentada por María Teresa Tamés Espinosa  
dentro del Programa de Doctorado en Oceanografía y Cambio Global

Dirigida por los Doctores  
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*A mis padres*



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# Abstract

The anthropogenic CO<sub>2</sub> emissions have been increasing for the past 250 years. Oceans are estimated to take up nearly one third of this CO<sub>2</sub>. However, proper assessment and understanding of the processes controlling these carbon fluxes requires, among other factors, quantifying primary production rates (CO<sub>2</sub> fixation) and respiration rates (CO<sub>2</sub> production,  $R_{CO_2}$ ). Direct  $R_{CO_2}$  measurements are not always feasible. Often they have to be calculated from O<sub>2</sub> consumption ( $R_{O_2}$ ). To do so, respiratory quotients (RQ), reported to be variable, are needed. In addition, when assessed from potential respiration ( $\Phi$ ),  $R_{O_2}/\Phi$  ratios are also needed. In order to reduce the variability linked to the use of these ratios and quotients, this thesis is focussed on developing a more direct assessment of  $R_{CO_2}$  throughout the water column. It aims to develop and use an assay for NADP<sup>+</sup>-dependent isocitrate dehydrogenase (NADP-IDH) in marine plankton.

In the development process, we determined the NADP-IDH optimum concentrations of substrate and cofactors, the optimum pH, and the proper treatment conditions needed to attain the NADP-IDH maximum velocity ( $V_{max}$ ). Unfortunately, NADP-IDH is not the only CO<sub>2</sub>-producing enzyme in the cells. Thus, we also assessed the relationship between the NADP-IDH and the other CO<sub>2</sub>-producing enzymes. After normalising their activities per NADP-IDH, we were able to develop a model to predict the potential CO<sub>2</sub> production in marine plankton based on NADP-IDH measurements ( $\Psi_{NADP}$ ). These results were validated by comparison with  $R_{CO_2}$  values calculated from  $\Phi$ , after applying different  $R_{O_2}/\Phi$  ratios and RQs.  $\Psi_{NADP}$  always exceeded  $R_{CO_2}$ . This is consistent with the theoretical arguments that any physiological response should never exceed the potential activity controlled by the underlying enzymology.

As part of our investigation, we also measured both the NADP-IDH and the  $R_{CO_2}$  during incubations of two communities of the mysid *Leptomysis lingvura*. Among all the studied factors, we observed that trophic conditions before sampling affected the enzyme activities. In addition, nutrient availability had significant impacts on the physiological response. However, the  $R_{CO_2}$ /NADP-IDH ratio was similar between the two different communities of *L. lingvura*, which suggests that those values would be representative of *L. lingvura*'s physiology.

Last but not least, we wanted to highlight how valuable respiratory enzyme measurements (NADP-IDH and ETS) are in assessing the catabolic metabolism in

the plankton community. To do so, we measured the  $\Phi/\text{Chl}_a$  ratio in a microplankton community during the KOSMOS GC 2014 experiment (BIOACID Project from GEOMAR, Kiel, Germany). We monitored the changes as the plankton community oscillated from photoautotrophic dominance to heterotrophic dominance following an injection of nutrient-rich deep water. We were simulating an upwelling event. The lowest values were reached in a phytoplankton bloom while the highest were reached in a heterotrophic dinoflagellate bloom.

Taking everything into consideration, we believe that the development of this new enzyme assay avoids the use of the RQ while measuring  $R_{\text{CO}_2}$ , thus minimising the variability of previous methodologies. In addition, the NADP-IDH facilitates development of bisubstrate enzyme kinetic models (EKM) which, based on the measurement of both the NADP-IDH and the in-cell concentration of substrates and cofactors, predict accurately the physiological response. By doing so, the quantification of the vertical carbon flux ( $F_c$ ) and the role of the ocean as a  $\text{CO}_2$  sink can be strengthened. Furthermore, the regular measurements of NADP-IDH and  $\Phi$  in oceanographic campaigns, and their comparison with autotrophic parameters (e.g. Chl<sub>a</sub>), will enhance the understanding of changes in the microplankton community and the assessment of the metabolic state of the ocean.

# Resumen

Debido a las emisiones por combustión de energías fósiles, los niveles de dióxido de carbono ( $\text{CO}_2$ ) atmosféricos han aumentado exponencialmente en los últimos 250 años. El océano puede considerarse un sumidero de este  $\text{CO}_2$ . Sin embargo, para estimar adecuadamente su papel, es necesario conocer tanto las tasas de producción primaria de los organismos autótrofos (fijación de  $\text{CO}_2$ ), como las de respiración (producción de  $\text{CO}_2$ ). La producción de  $\text{CO}_2$  ( $R_{\text{CO}_2}$ ) es particularmente difícil de medir, siendo generalmente calculada a partir de mediciones de consumo de  $\text{O}_2$  ( $R_{\text{O}_2}$ ) con la aplicación de cocientes respiratorios (RQ). Esta dificultad se hace mayor en los estudios del océano profundo, donde la  $R_{\text{O}_2}$  es obtenida tras la medición de la respiración potencial ( $\Phi$ ) y la aplicación de ratios  $R_{\text{O}_2}/\Phi$ . Estos ratios pueden variar significativamente en función de la disponibilidad de nutrientes. A esta variabilidad, hay que añadir la existente en los RQ. Por ello, esta tesis está centrada en el desarrollo de un nuevo método que profundice en una medición más directa de  $R_{\text{CO}_2}$  en toda la columna de agua.

Este nuevo método, específico para el microplacton y mesozooplancton, es una adaptación de la medición de una de las principales enzimas productoras de  $\text{CO}_2$ : la isocitrato deshidrogenasa dependiente del  $\text{NADP}^+$  (NADP-IDH). Durante su desarrollo se han determinado las concentraciones de sus substratos y cofactores, su pH óptimo y las condiciones de tratamiento que permiten alcanzar la máxima actividad en muestras de la comunidad planctónica. Lamentablemente, la NADP-IDH no es la única enzima productora de  $\text{CO}_2$  en la célula. Por ello, estudiamos su relación con las principales enzimas involucradas, de manera que, normalizando sus actividades por la NADP-IDH, desarrollamos un modelo para la predicción de la producción de  $\text{CO}_2$  potencial máxima en la célula, basado en la medición de la NADP-IDH ( $\Psi_{\text{NADP}}$ ). Una vez medida en muestras de mesozooplancton y microplancton, comparamos la  $\Psi_{\text{NADP}}$  con valores de  $R_{\text{CO}_2}$  obtenidos a partir de la  $\Phi$  de las mismas muestras, tras la aplicación de los principales ratios  $R_{\text{O}_2}/\Phi$  y RQs utilizados en la literatura. En todos los casos,  $\Psi_{\text{NADP}}$  fue superior a  $R_{\text{CO}_2}$ , siendo coherente con la idea teórica de que la respuesta fisiológica nunca debe superar a la capacidad metabólica del organismo.

Además, gracias a la experimentación con distintas comunidades de *Leptomyysis lingvura* pudimos comparar medidas directas de  $R_{\text{CO}_2}$  con la actividad de la NADP-IDH. De entre todos los factores analizados, comprobamos que las condi-

ciones tróficas previas al muestreo tienen efectos significativos sobre sus actividades enzimáticas. Asimismo, observamos que la disponibilidad de nutrientes afecta significativamente a la respuesta fisiológica. Sin embargo, las distintas comunidades obtuvieron ratios similares de  $R_{CO_2}$ /NADP-IDH, lo que sugiere que los valores obtenidos pueden ser representativos para *L. lingvura*.

Finalmente, hemos querido destacar el valor de las medidas de NADP-IDH y de  $\Phi$  como indicadores del metabolismo respiratorio de la comunidad planctónica. Para ello, ahondamos en el desarrollo del ratio  $\Phi/\text{Chl}_a$  que permitió la monitorización de los cambios metabólicos en la comunidad microplanctónica durante el experimento KOSMOS GC 2014 dentro del proyecto BIOACID. Los valores más bajos se dieron durante el desarrollo de un bloom fitoplanctónico, mientras que los más altos, durante la proliferación de dinoflagelados heterótrofos.

En conclusión, el desarrollo de un nuevo método enzimático para el análisis de la producción de  $CO_2$  en toda la columna de agua evitar la aplicación de los RQ, reduciendo la variabilidad de los métodos usados hasta ahora. Además, permite el desarrollo de modelos cinéticos bisustrato (EKM) que, en base a la concentración de sustratos en la célula, predicen la respuesta fisiológica. Esto tendrá un impacto muy positivo en la cuantificación de la función del océano como sumidero de  $CO_2$  y en el estudio del flujo de carbono vertical ( $F_c$ ) al poder obtener datos de  $R_{CO_2}$  del océano profundo. Además, consideramos que la medición del metabolismo catabólico en campañas oceanográficas de manera regular, y su comparación con parámetros relacionados con la autotrofia, favorecerá la comprensión de los cambios en la comunidad microplanctónica y la estimación del estado metabólico de las distintas regiones oceanográficas.

# Thesis Preview

The present dissertation entitled “Enzyme Assays in Assessing Respiratory Metabolism in Marine Plankton: the development of a new CO<sub>2</sub>-production proxy and further applications of respiratory enzymatic assays in oceanography” has been conducted within the Doctoral Program of Oceanography and Global Change from the University of Las Palmas de Gran Canaria and has been supervised by Dr. María M. Gómez Cabrera and Dr. Theodore T. Packard. It is comprised of 4 original research articles (one of them already published) developed in the frame of the projects BIOMBA (CTM-2012-32729), granted to Dr. May Gómez. Moreover, this thesis includes samples collected during KOSMOS experiment from research project BIOACID that was granted to Dr. Ulf Riebesell. In addition, a research stay was made at the GEOMAR Helmholtz Centre for Ocean Research Kiel (Kiel, Germany) during this period. This thesis is structured in two parts. The first is divided into six chapters that encompass: a general introduction with the major objectives of the work; four original contributions in the form of scientific research articles that embed discussion sections and specific conclusions; a synthesis chapter that embeds the major conclusions of the thesis; and finally, the outline for future research. This first part is completely written in English (except for the abstracts on chapters 2 to 5, which are written in both English and Spanish), fulfilling the demands to obtain the International Doctor Mention (Real Decreto 99/2011 del 28 de enero por el que se regulan las enseñanzas oficiales de doctorado, Art.15). The second part, in turn, is written in Spanish and consists of a brief summary of the English section, with an introduction to the topic, the general material and methods, the main results and the conclusions derived from the work. This is a requirement of the PhD Thesis Regulation from the Universidad de Las Palmas de Gran Canaria (BOULPG, January 9th, 2013, modified in BOULPGC, November 4th, 2013). During the development of this research, I received financial support from University of Las Palmas de Gran Canaria (PIFULPGC-2013-CIENCIAS-1).



# **Presentación de la Tesis**

La tesis que aquí se presenta bajo el título “Los Ensayos Enzimáticos en el Análisis del Metabolismo Respiratorio del Plancton Marino: el desarrollo de un nuevo método para medir la producción de CO<sub>2</sub> y nuevas aplicaciones de estas metodologías en oceanografía” se ha desarrollado dentro del Programa de Doctorado en Oceanografía y Cambio Global de la Universidad de Las Palmas de Gran Canaria y ha estado dirigida por los doctores May Gómez y Ted Packard. En ella se incluyen 4 contribuciones originales (una de ellas ya publicada) enmarcadas principalmente dentro del proyecto de investigación BIOMBA (CTM 2012-32729) otorgado a la Dr. May Gómez. Además, esta tesis incluye muestras tomadas durante el experimento KOSMOS GC 2014, dentro del proyecto BIOACID dirigido por el doctor Ulf Riebesel. Por último, durante el desarrollo de la misma se realizó una estancia en GEOMAR Helmholtz centre for ocean research en Kiel (Alemania). Esta tesis está estructurada en dos partes. La primera engloba: la Introducción con los principales objetivos del trabajo; las Contribuciones Originales en formato de artículo científico, incluyendo la discusión y conclusiones específicas; las Conclusiones Generales de la tesis; y las Futuras Líneas de Investigación. Estos capítulos están escritos íntegramente en inglés (a excepción de los resúmenes de los capítulos 2 al 5, que se han escrito en ambos idiomas), de acuerdo a la normativa para la obtención de la Mención Internacional del Título de Doctor (Real Decreto 99/2011 del 28 de enero por el que se regulan las enseñanzas oficiales de doctorado, Art.15). La segunda parte de la tesis incluye un resumen en castellano que consta de una breve introducción, el material y los métodos generales, los principales resultados y las conclusiones más relevantes del presente trabajo, cumpliéndose así los requisitos expuestos en el Reglamento de Estudios de Doctorado de la Universidad de Las Palmas de Gran Canaria (BOULPG, January 9th, 2013, and modified in BOULPGC, November 4th, 2013). El desarrollo de esta tesis no habría sido posible sin el apoyo económico recibido de la Universidad de Las Palmas de Gran Canaria en el marco del contrato predoctoral PIFULPGC-2013-CIENCIAS-1.



# Contents

<b>Agradecimientos/Acknowledgements</b>	<b>vii</b>
<b>Abstract</b>	<b>xi</b>
<b>Resumen</b>	<b>xiii</b>
<b>Thesis Preview</b>	<b>xv</b>
<b>Presentación de la Tesis</b>	<b>xvii</b>
<b>List of Figures</b>	<b>xxv</b>
<b>List of Tables</b>	<b>xxix</b>
<b>1 Chapter 1. General Introduction</b>	<b>1</b>
1.1 The role of respiration in the ocean metabolism . . . . .	1
1.1.1 Physiological O <sub>2</sub> consumption rates . . . . .	2
1.1.2 Enzymatic activity: potential O <sub>2</sub> consumption rates . . . . .	2
1.1.3 The $R_{O_2}/\Phi$ ratio . . . . .	3
1.2 The impact of respiration on the carbon flux assessment . . . . .	5
1.2.1 Physiological CO <sub>2</sub> production rates . . . . .	6
1.2.2 Enzymatic potential CO <sub>2</sub> production rates . . . . .	8
1.2.3 The $R_{CO_2}/NADP-IDH$ ratio . . . . .	10
1.3 Further applications of respiratory enzymatic measurements . . . . .	11
1.3.1 Carbon flux and Nutrient regeneration efficiency . . . . .	12
1.3.2 Heterotrophic energy production . . . . .	13
1.3.3 Use of respiratory enzymes as a proxy of the catabolic metabolism in the oceans . . . . .	14
1.4 Thesis objectives and outline . . . . .	15
<b>2 Chapter 2. NADP<sup>+</sup>-dependent isocitrate dehydrogenase activity in marine plankton</b>	<b>17</b>
2.1 Introduction . . . . .	18

2.2	Material and Methods . . . . .	19
2.2.1	Plankton sampling procedures . . . . .	19
2.2.2	Crude homogenate preparations and protein determination . . . . .	20
2.2.3	NADP-IDH assay . . . . .	20
2.2.4	Kinetic properties of NADP-IDH activity . . . . .	21
2.2.5	pH optimum . . . . .	22
2.2.6	Temperature dependence . . . . .	22
2.2.7	Centrifugation studies . . . . .	23
2.2.8	Linearity with enzyme concentration and limit of detection . . . . .	23
2.2.9	Stability of the crude homogenate . . . . .	23
2.2.10	Statistics of the substrate and temperature kinetics . . . . .	24
2.3	Results . . . . .	24
2.3.1	Kinetic properties . . . . .	24
2.3.2	Effect of pH . . . . .	26
2.3.3	Effect of Temperature . . . . .	26
2.3.4	Biomass linearity and limit of detection . . . . .	29
2.3.5	Effect of centrifugation . . . . .	29
2.3.6	Stability of the enzyme preparation from mysid samples . . . . .	30
2.4	Discussion . . . . .	32
2.4.1	What does the development of the assay reveal about plankton NADP-IDH? . . . . .	32
2.4.2	Why should this metabolic assay be measured kinetically? . . . . .	33
2.5	Conclusions . . . . .	34
3	<b>Chapter 3: The Potential Respiratory CO<sub>2</sub> Production from NADP<sup>+</sup>-dependent isocitrate dehydrogenase in marine plankton.</b>	35
3.1	Introduction . . . . .	36
3.2	Material and Methods . . . . .	38
3.2.1	Plankton sampling procedures . . . . .	38
3.2.2	Preparations for kinetic enzyme activity measurements . . . . .	39
3.2.3	Potential respiration ( $\Phi$ ) . . . . .	40
3.2.4	$R_{O_2}$ calculations from Potential respiration ( $\Phi$ ) measurements . . . . .	40
3.2.5	$R_{CO_2}$ calculations from $R_{O_2}$ . . . . .	41
3.2.6	Biomass . . . . .	41
3.2.7	Data analysis and Statistics . . . . .	41
3.3	Results . . . . .	43
3.3.1	Potential CO <sub>2</sub> production from NADP-IDH measurements . . . . .	43
3.3.2	Physiological rates from $\Phi$ measurements . . . . .	47
3.3.3	Variability of $R_{CO_2}/\Psi_{NADP}$ ratios . . . . .	50
3.4	Discussion . . . . .	51

3.4.1	What is the relationship between potential CO <sub>2</sub> production and potential O <sub>2</sub> consumption? . . . . .	51
3.4.2	What is the relationship between $R_{CO_2}$ and $\Psi_{NADP}$ ? . . . . .	53
3.5	Conclusions . . . . .	55
<b>4</b>	<b>Chapter 4: Impact of sex differences and starvation on respiratory metabolism: CO<sub>2</sub> production on marine mysid <i>Leptomysis lingvura</i>.</b>	<b>57</b>
4.1	Introduction . . . . .	59
4.2	Material and Methods . . . . .	60
4.2.1	<i>Leptomysis lingvura</i> sampling and cultures . . . . .	60
4.2.2	NADP-IDH activity: potential CO <sub>2</sub> production . . . . .	62
4.2.3	Electron transport system (ETS) activity: potential O <sub>2</sub> consumption . . . . .	63
4.2.4	Physiological O <sub>2</sub> consumption rates ( $R_{O_2}$ ) in <i>L. lingvura</i> incubations . . . . .	63
4.2.5	Physiological CO <sub>2</sub> consumption rates ( $R_{CO_2}$ ) in <i>L. lingvura</i> incubations . . . . .	64
4.2.6	Respiratory Quotient (RQ) calculations . . . . .	65
4.2.7	Biomass . . . . .	65
4.2.8	Statistical Analysis . . . . .	66
4.3	Results and Discussion . . . . .	66
4.3.1	Impact of trophic conditions on potential respiration ( $\Phi$ and NADP-IDH) and $R_{CO_2}$ /NADP-IDH ratios in ovigerous females of two communities of <i>L. lingvura</i> . . . . .	66
4.3.2	Impact of sex differences and starvation on potential respiration ( $\Phi$ and NADP-IDH) . . . . .	70
4.3.3	Impact of sex differences and starvation on the $R_{O_2}/\Phi$ and $R_{CO_2}$ /NADP-IDH ratios . . . . .	73
4.3.4	Impact of long-term incubations on $R_{CO_2}$ /NADP-IDH ratios in females of <i>L. lingvura</i> . . . . .	76
4.4	Conclusions . . . . .	77
<b>5</b>	<b>Chapter 5: Metabolic responses of subtropical North Atlantic microplankton after a simulated deep-water upwelling event.</b>	<b>79</b>
5.1	Introduction . . . . .	81
5.2	Material and Methods . . . . .	83
5.2.1	Experimental design . . . . .	83
5.2.2	Sampling . . . . .	83
5.2.3	Nutrients, Chlorophyll <i>a</i> (Chl <sub><i>a</i></sub> ) and Water-column community composition . . . . .	84
5.2.4	Potential respiration . . . . .	84
5.2.5	Biomass and Net Productivity . . . . .	85

5.2.6	Data Analysis and Statistics . . . . .	85
5.3	Results . . . . .	86
5.3.1	Responses to CO <sub>2</sub> -treatments under low-nutrient conditions.	86
5.3.2	Responses during the deep-water addition event. . . . .	87
5.3.3	Responses to the CO <sub>2</sub> -treatments during the nutrient depletion phase. . . . .	89
5.4	Discussion . . . . .	90
5.4.1	The use of the Φ/Chl <sub>a</sub> ratio as an index of autotrophic, mixotrophic or heterotrophic dominance in the microplankton community. . . . .	90
5.4.2	Acidification, mixotrophy and the development of <i>Vicicius globosus</i> . . . . .	91
5.5	Conclusion . . . . .	93
<b>6</b>	<b>Chapter 6: Conclusions</b>	<b>95</b>
6.1	Developing the assay for NADP-IDH activity as a tool to detect CO <sub>2</sub> production in cells . . . . .	95
6.2	The assessment of mesozooplankton potential CO <sub>2</sub> production from NADP-IDH measurements. . . . .	96
6.3	The assessment of the relationship between $R_{CO_2}$ and NADP-IDH activity . . . . .	96
6.4	Assessing variations in the microplankton autotrophy-heterotrophy ratio using enzyme analysis . . . . .	97
<b>7</b>	<b>Chapter 7: Future research.</b>	<b>99</b>
<b>8</b>	<b>Chapter 8: Resumen en español.</b>	<b>101</b>
8.1	Introducción . . . . .	101
8.2	Metodología General . . . . .	105
8.2.1	Muestreo de mesozooplancton y conservación de las muestras. . . . .	105
8.2.2	Muestreo microplancton y conservación de las muestras. .	105
8.2.3	Muestreo y cultivos de <i>Leptomyysis lingvura</i> . . . . .	106
8.2.4	Medición de NADP-IDH. . . . .	106
8.2.5	Optimización de parámetros para la medición de la $V_{max}$ de NADP-IDH en la comunidad planctónica. . . . .	107
8.2.6	Medición del consumo máximo de O <sub>2</sub> (Φ) en la comunidad planctónica. . . . .	108
8.2.7	Medición del consumo de O <sub>2</sub> fisiológico en base a la aplicación de los ratios $R_{O_2}/\Phi$ a las mediciones de Φ. . . . .	108
8.2.8	Medición de la producción de CO <sub>2</sub> fisiológica ( $R_{CO_2}$ ) teórica en base a la aplicación de RQ ratios a los resultados en $R_{O_2}$	109

8.2.9	Medición del consumo de O <sub>2</sub> ( $R_{O_2}$ ) fisiológico en incubaciones de <i>L. lingvura</i> . . . . .	109
8.2.10	Medición de la producción de CO <sub>2</sub> ( $R_{CO_2}$ ) fisiológica en incubaciones de <i>L. lingvura</i> . . . . .	110
8.2.11	Medición de la biomasa en términos de contenido proteínico.	110
8.2.12	Medición de la productividad neta . . . . .	111
8.3	Principales resultados y discusión . . . . .	111
8.3.1	Optimización de la medición de la actividad de la NADP-IDH en la comunidad planctónica. . . . .	111
8.3.2	Producción máxima de CO <sub>2</sub> de la comunidad planctónica a partir de la medición de la actividad de la NADP-IDH ( $\Psi_{NADP}$ ). . . . .	112
8.3.3	Desarrollo de los ratios teóricos $R_{CO_2}/\Psi_{NADP}$ . . . . .	114
8.3.4	Estudio de los efectos de la inanición, las diferencias sexuales y el estado trófico de la comunidad en la producción de CO <sub>2</sub> del misidáceo <i>L. lingvura</i> . . . . .	115
8.3.5	Variaciones del valor del ratio $\Phi/\text{Chl}_a$ y su relación con la composición de la comunidad microplanctónica en términos de dominancia de fotoautótrofos, heterótrofos o mixótrofos. . . . .	117
8.4	Conclusiones . . . . .	119
8.4.1	El desarrollo de un método para determinar la actividad de la NADP-IDH en el plancton como una de las principales enzimas responsables de la producción de CO <sub>2</sub> . . . . .	119
8.4.2	El estudio de la producción máxima de CO <sub>2</sub> del mesozooplancton a partir de las mediciones de NADP-IDH . . . . .	120
8.4.3	El estudio de la relación entre $R_{CO_2}$ and la actividad de la NADP-IDH . . . . .	121
8.4.4	La aplicación de una técnica enzimática para el estudio de las variaciones en el estado autotrófico-heterotrófico de la comunidad microplanctónica . . . . .	122
<b>A</b>	<b>Appendix A. NADP-IDH assay extended information</b>	<b>123</b>
A.1	Buffer studies . . . . .	123
A.2	Molar extinction coefficient ( $\epsilon$ ) studies . . . . .	124
A.3	Kinetics . . . . .	126
A.4	pH effects . . . . .	129
A.5	Temperature effects: Development of the Arrhenius equation . . .	129
A.6	Precision studies . . . . .	131
A.7	Understanding changes in the NADP-IDH spectrophotometrical response . . . . .	132

<b>B Appendix B. NADP-IDH thermodynamic factors</b>	<b>137</b>
<b>C Appendix C. <i>Leptomyysis lingyura</i> dataset</b>	<b>141</b>
<b>D Appendix D. KOSMOS GC 2014 Main dataset</b>	<b>143</b>
<b>E Appendix E. Abbreviations</b>	<b>149</b>
<b>Bibliography</b>	<b>151</b>
<b>Declaration</b>	<b>170</b>

# List of Figures

1.1	Velocities (V) through the main pathways in <i>E.coli</i> Krebs cycle (KC) in the bacterium, <i>Escherichia coli</i> , during acetate utilisation (redrawn from Walsh and Koshland (1984)). Rates are expressed in terms of mmol substrate consumed min <sup>-1</sup> L <sup>-1</sup> of cell volume. Abbreviations in the velocity expressions are CS, citrate synthase; Acon, aconitase; IDH <sub>KC</sub> , isocitrate dehydrogenase linked to the Krebs cycle; KGDH, $\alpha$ -ketoglutarate dehydrogenase; SuccDH, succinate dehydrogenase; Fum, fumarase; MDH, malate dehydrogenase; ME, malic enzyme; PEPCK, phosphoenolpyruvate carboxykinase. Rates linked to CO <sub>2</sub> production have been highlighted in red. Not shown is the CO <sub>2</sub> -producing enzyme, pyruvate dehydrogenase, that links glucose metabolism in the Embden-Meyerhof pathway with the KC by decarboxylating pyruvate to acetyl-coenzyme A. . . . .	9
1.2	Diagram of the vertical profile of $R_{CO_2}$ and its relationship with vertical $F_C$ . Light blue illustrates the euphotic zone whereas dark blue represents the aphotic zone. . . . .	12
2.1	Michaelis-Menten plots to determine the NADP-IDH $K_m$ and optimum values for isocitrate, NADP <sup>+</sup> and MgCl <sub>2</sub> for different plankton community size fractions. (a) Affinity between NADP-IDH and isocitrate. (b) Affinity between NADP-IDH and NADP <sup>+</sup> . (c) Affinity between NADP-IDH and Mg <sup>2+</sup> . . . . .	25
2.2	pH effects on mesozooplankton NADP-IDH activity, where $V_{max}$ was 0.027 μmol NADPH min <sup>-1</sup> (mg protein) <sup>-1</sup> . . . . .	26
2.3	Temperature dependence of NADP-IDH activity. (a) Temperature impact on NADP-IDH activity, where $V_{max}$ was 0.018 μmol NADPH min <sup>-1</sup> (mg of protein) <sup>-1</sup> in mesozooplankton and 0.808 μmol NADPH min <sup>-1</sup> (mg of protein) <sup>-1</sup> in pure porcine heart enzyme. (b) Arrhenius plot, where ln( $k_{IDH}$ ) is plotted against 1/T and T is in °K units. . . . .	28

2.4	Biomass linearity of marine-microplankton and marine-mesozooplankton NADP-IDH. Enzyme activity as a function of protein in the homogenate. . . . .	29
2.5	Centrifugation effects on NADP-IDH activity, where $V_{max}=0.0073$ in mesozooplankton and $0.0008$ in microplankton. $V_{max}$ is in $\mu\text{mol NADPH min}^{-1}$ ( $\text{mL homogenate}$ ) $^{-1}$ units. Note that, for our centrifuge, a force of $375\text{g}$ was achieved at $1000\text{rpm}$ ; $750\text{g}$ was achieved at $2000\text{rpm}$ ; and $1500\text{g}$ was achieved at $4000\text{rpm}$ . . . . .	30
2.6	Stability of the enzyme preparations from <i>L. lingvura</i> samples, where $V_{max}=0.0085\mu\text{mol NADPH min}^{-1}$ ( $\text{mL homogenate}$ ) $^{-1}$ . $T_0$ is the end of the centrifugation process. . . . .	31
3.1	Canary Islands Archipelago, surrounded by waters of the North Eastern Subtropical Atlantic Ocean where most of the samples in this paper were taken. . . . .	37
3.2	Sampling in the coastal waters of Gran Canaria using the WP2 net dragged by scuba divers during spring-bloom conditions. . . . .	39
3.3	Physiological rates under different scenarios. (a) $R_{O_2}$ rates obtained from $\Phi$ under four different scenarios. (b) $R_{CO_2}$ rates obtained from $R_{O_2}$ resulting in twelve different physiological conditions. . . . .	49
3.4	Relationship between NADP-IDH and Potential Respiration ( $\Phi$ ) in mesozooplankton ( $200\text{-}2000\mu\text{m}$ ; $y=0.5723x-4.4\cdot10^{-5}$ , $r^2=0.994$ , $p<0.0001$ ) and microplankton ( $0.7\text{-}50\mu\text{m}$ ; $y=0.46x-5\cdot10^{-5}$ , $r^2=0.9749$ , $p=0.0002$ ). . . . .	51
3.5	Relationship between potential $CO_2$ production (NADP-IDH based model) and $O_2$ consumption (ETS-based model) in the mainly crustacean $200\text{-}2000\mu\text{m}$ size-fractionated plankton community. (a) Total $\Psi_{NADP}$ (Eq.3.14) vs $\Phi$ ( $y=1.68x-0.0001$ , $r^2=0.994$ , $p<0.0001$ ). (b) Respiratory $\Psi_{NADP}$ (Eq.3.15) vs $\Phi$ ( $y=1.47x-0.0001$ , $r^2=0.994$ , $p<0.0001$ ). . . . .	53
4.1	Differences in normalised respiratory enzymatic activities in well-fed females of <i>L. lingvura</i> from a bloom and from a post-bloom community. Note that $\Phi$ represents the potential $O_2$ consumption of the mysids and NADP-IDH is the potential $CO_2$ production of the NADP-IDH enzyme in the mysids. . . . .	67
4.2	NADP-IDH to $\Phi$ proportion in well-fed females of <i>L. lingvura</i> belonged to a bloom and to a post-bloom community. Note that $\Phi$ represents the potential $O_2$ consumption of the mysids and NADP-IDH is the potential $CO_2$ production of the NADP-IDH enzyme in the mysids. . . . .	68

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4.3	$R_{CO_2}$ /NADP-IDH ratio in well-fed females of <i>L. lingvura</i> belonged to a bloom and to a post-bloom community. . . . .	69
4.4	Differences in normalised respiratory enzymatic activities in well-fed and nutrient-limited (starved) females (WF F & S F) and well-fed and non-starved males (WF M & NS M) from the post-bloom community. Note that $\Phi$ is the potential O <sub>2</sub> consumption and NADP-IDH is the potential CO <sub>2</sub> production of the NADP-IDH enzyme in the mysid. . . . .	71
4.5	NADP-IDH to $\Phi$ proportion in well-fed and nutrient-limited (starved) females (WF-F & S-F) and well-fed and non-starved males (WF-M, NS-M) from the post-bloom community. . . . .	72
4.6	Effects of starvation and sex differences on the ratios of respiration to enzymatic activity in well-fed and nutrient-limited females (WF-F & S-F) and well-fed and non-starved males (WF-M & NS-M) from the post-bloom community . . . . .	74
4.7	Effects of starvation and sex difference on the respiratory quotient (RQ) in well-fed and nutrient-limited (starved) females (WF-F & S-F) and well-fed and non-starved males (WF-M & NS-M) from the post-bloom community. . . . .	75
4.8	$R_{CO_2}$ /NADP-IDH ratio variability during long-term incubation periods. . . . .	76
5.1	Potential respiration per Chl <sub>a</sub> during Phase I: Oligotrophic phase; Phase II: Bloom phase after deep-water addition; and Phase III: Post-bloom phase after nutrient-depletion. . . . .	86
5.2	$P_N$ of the microplankton community, as the first derivative of biomass, during Phase I: Oligotrophic phase; Phase II: Bloom phase after deep-water addition; and Phase III: Post-bloom phase after nutrient-depletion. Note that negative values reflects the net loss of biomass, when the combination of sinking, remineralization and grazing outnumbers the biomass production by both autotrophs and heterotrophs . . . . .	87
5.3	Integral analysis of Potential respiration, Chl <sub>a</sub> and Biomass by phase of experiment and treatment. Note that every point symbolizes the integral of each parameter by the related phase in the mesocosm with the CO <sub>2</sub> -treatment reflected in the x-axis. Blue colour represents Phase I, green represents Phase II, and Phase III is represented in red. . . . .	88
A.1	Buffer effects on NADP-IDH activity . . . . .	124
A.2	NADPH molar extinction coefficient, in three different buffer combinations. . . . .	125

A.3 Hanes-Woolf, Lineweaver-Burk and Eadie-Hofstee linearisations to determine the affinity ( $K_m$ ) of NADP-IDH for isocitrate, NADP <sup>+</sup> and MgCl <sub>2</sub> in marine-plankton samples. The symbols in the key of panel G apply to all six panels. . . . .	128
A.4 Porcine-heart NADP-IDH activity measured by the marine-plankton NADP-IDH assay at pH:8.2 and pH:7.4 . . . . .	132
A.5 Spectrophotometrical response of NADP-IDH activity, following the $A_{340}$ per time of the substrate and blank buffered cuvettes, under conditions that affect the signal during the measurement . . . . .	134
A.6 Kinetic responses of NADP-IDH to different levels of T and pH. (a) Temperature effects on NADP-IDH activity by the kinetic assay. (b) pH effects on NADP-IDH activity by the kinetic assay. . . . .	135

# List of Tables

3.1	Different scenarios designed to develop the $R_{CO_2}/\Psi_{NADP}$ ratios from the $\Phi$ and $\Psi_{NADP}$ measurements, using conversion ratios reported in the literature. Scenarios 2.2 and 3.2 describe the natural conditions of our 200-2000 $\mu\text{m}$ zooplankton samples, and 4.1 describes the natural conditions in our 0.7-50 $\mu\text{m}$ samples. . . . .	42
3.2	Enzyme activities in crustacean samples, in $\mu\text{mol substrate min}^{-1}$ ( $\text{g fresh weight}^{-1}$ ). IDH <sub>KC</sub> activity was calculated as expressed in Eq.3.9; NADP-IDH <sub>mit</sub> activity was calculated as expressed in Eq.3.7; and NADP-IDH <sub>out</sub> activity was calculated as expressed in Eq.3.10. In each case, the activities of CS, NADP-IDH and NAD-IDH in crustacean samples from Alp et al. (1976) were used. . . . .	46
3.3	Ratios of NAD-IDH, NADP-IDH <sub>mit</sub> and NADP-IDH <sub>out</sub> activities to NADP-IDH activity. These ratios are needed for the calculation of total potential CO <sub>2</sub> production (Eq. 3.12). . . . .	46
3.4	Results in $\Psi_{NADP}$ (in $10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1} (\text{mg protein})^{-1}$ ), in $R_{CO_2}$ (also in $10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1} (\text{mg protein})^{-1}$ units) and in the $R_{CO_2}/\Psi_{NADP}$ ratios for every scenario. Values that likely represent the actual conditions of our mesozooplankton samples (between scenarios B2 and C2) and microzooplankton samples (scenario D1 and D3) have been highlighted in bold text. . . . .	50
A.1	Michaelis-Menten hyperbolic curves for isocitrate, NADP <sup>+</sup> and MgCl <sub>2</sub> affinity in marine plankton samples. $K_m$ is in mM units. For isocitrate, $V_{max}=0.038$ in mesozooplankton; 0.031 in microzooplankton; and 0.002 in microplankton. For NADP <sup>+</sup> , $V_{max}=0.029$ in mesozooplankton; 0.019 in microzooplankton; and 0.001 in microplankton. For MgCl <sub>2</sub> , $V_{max}=0.010$ in mesozooplankton; 0.009 in microzooplankton; and 0.00002 in microplankton. $V_{max}$ is in $\mu\text{mol NADPH min}^{-1} (\text{mL homogenate})^{-1}$ units. . . . .	126
B.1	NADP-IDH thermodynamic factors. . . . .	139

C.1	Respiratory rates measured in a bloom community of <i>L. lingvura</i> . Note that both NADP-IDH and CO <sub>2</sub> production rates ( $R_{CO_2}$ ) are expressed in $\mu\text{molCO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$ units. . . . .	141
C.2	Respiratory rates measured in a post-bloom community of <i>L. lingvura</i> . Note that both NADP-IDH and CO <sub>2</sub> production rates ( $R_{CO_2}$ ) are expressed in $\mu\text{molCO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$ units; and both $\Phi$ and O <sub>2</sub> consumption rates ( $R_{O_2}$ ) are expressed in $\mu\text{molO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$ units. . . . .	142
D.1	Potential respiration ( $\Phi$ ) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that $\Phi$ is expressed in $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$ units. To convert to “per protein” units, use the data on table D.4. . . . .	143
D.2	Potential respiration ( $\Phi$ ) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that $\Phi$ is expressed in $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$ units. To convert to “per protein” units, use the data on table D.5. . . . .	144
D.3	Potential respiration ( $\Phi$ ) during the Post-bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that $\Phi$ is expressed in $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$ units. To convert to “per protein” units, use the data on table D.6. . . . .	144
D.4	Biomass (B) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L <sup>-1</sup> units. . . . .	145
D.5	Biomass (B) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L <sup>-1</sup> units. . . . .	145
D.6	Biomass (B) during the Post-bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L <sup>-1</sup> units. . . . .	146
D.7	Chlorophyll <i>a</i> (Chl <sub><i>a</i></sub> ) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that Chl <sub><i>a</i></sub> is expressed in $\mu\text{g L}^{-1}$ units. To convert to “per protein” units, use the data on table D.4. . . . .	146

D.8	Chlorophyll <i>a</i> (Chl <sub><i>a</i></sub> ) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that Chl <sub><i>a</i></sub> is expressed in $\mu\text{g L}^{-1}$ units. To convert to “per protein” units, use the data on table D.5. . . . .	147
D.9	Chlorophyll <i>a</i> (Chl <sub><i>a</i></sub> ) during the Post-Bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO <sub>2</sub> levels of the treatment. Note that Chl <sub><i>a</i></sub> is expressed in $\mu\text{g L}^{-1}$ units. To convert to “per protein” units, use the data on table D.6. . . . .	147
E.1	Summary of the symbols used to identify the parameters (1/2). . .	149
E.2	Summary of the symbols used to identify the parameters (2/2). . .	150



# Chapter 1

## Chapter 1. General Introduction

*Respiration is a slow combustion of carbon and hydrogen, similar in every way to that which takes place in a lamp or lighted candle.*

*Breathing animals are active combustible bodies that are burning and wasting away.*

Antoine Laurent Lavoisier

### 1.1 The role of respiration in the ocean metabolism

Ocean metabolism is one of the main biological processes that control marine chemistry. Its anabolic facet is responsible for the carbon dioxide ( $\text{CO}_2$ ) fixation and oxygen ( $\text{O}_2$ ) formation. This facet creates new particulate organic matter (POC) through photosynthesis in the euphotic zone, or through widely diverse forms of chemosynthesis throughout the water column. Metabolism's catabolic facet is responsible for  $\text{CO}_2$  production,  $\text{O}_2$  consumption, nutrient remineralization, and in-cell energy production (in the form of adenosine triphosphate (ATP) following degradative reactions that break-down POC (Nelson and Cox (2008)). This degradative facet is mainly driven by respiration in heterotrophs, autotrophs and mixotrophs. All cells share the same catabolic pathway, it is a ubiquitous process (Lane (2006, 2010)). del Giorgio and le B. Williams (2005) described it as the major sink of organic matter in any ecosystem. At this scale, to understand both the marine ecosystems and the biogeochemical relationships that impact them, a deep knowledge of the biological transformation rates becomes mandatory. However, even though marine respiration is ubiquitous throughout the water column, it was not until the 70's when it began to be measured in a regular basis, instead of being estimated indirectly from the primary productivity or from biomass measurements (Packard et al. (1974)). The classic assumption of strong coupling between both facets on a global scale supported the assessment of respiration derived from primary production. However, since Sverdrup et al. (1942), the application of the

dynamic equilibrium concept to the O<sub>2</sub> distribution illustrated that both ecosystem processes often occur separated in time and space. This is because ecosystems can exchange both mass and energy with their surroundings, they are open systems. Thus, an accurate assessment of respiration in the ocean requires direct measurements not derived from primary production. In addition, direct measurements of respiration reflect the activity of organisms at a cellular level. At this scale, respiration is the pathway whereby organisms produce their vital energy in the form of ATP. Meanwhile, CO<sub>2</sub> is produced and O<sub>2</sub> is consumed.

### 1.1.1 Physiological O<sub>2</sub> consumption rates

The most commonly used techniques in respiration assessment are related to measuring O<sub>2</sub> consumption rates ( $R_{O_2}$ ) when organisms or ecosystem samples are incubated in isolation from their surroundings. Early measurements of respiration in marine organisms were carried in the 30's, by observing the decrease of O<sub>2</sub> content in incubations of pure, marine-bacterial cultures (Waksman and Carey (1935); Johnson (1936)) or in bottled seawater (Seiwell (1937); Seiwell and Seiwell (1938)). Since then, in addition to the Winkler method (Winkler (1888)),  $R_{O_2}$  has been commonly assessed by electrodes or optodes during up-to-24h of incubation. Even though all these physiological measurements are direct, they are not flawless. Artefacts linked to the manipulation of the living organisms, to the starvation of the sample when long periods of incubation are needed, and to overcrowding in the bottle, affect the reliability of the measure as an accurate representation of what is happening in the ocean (Fernández-Urruzola et al. (2011); Williams et al. (2013); Bondyale-Juez et al. (2017)). Furthermore, many factors impact the physiological response. A few of these are: nutrient limitation, circadian rhythm, size (Marshall et al. (1935); Christensen et al. (1980); Herrera et al. (2011b); Fernández-Urruzola et al. (2011)), age (Hernández-Leon and Gómez (1996)), the reproductive state (Kiørboe et al. (1985)), sex differences (Isla and Perissinotto (2004)), and oxygen partial pressure (Salvato et al. (2001)). Besides, the respiration measurements are time-and-manpower consuming, so they usually result in low data acquisition rates, which for oceanographic purposes is a disadvantage. These measures are useful in physiology, but oceanography requires a higher data acquisition rate over time and space (think satellite mapping of phytoplankton productivity). For that reason, enzymatic assays to detect metabolism were developed as oceanographic tools in the 70's (Packard (1971); Packard et al. (1971a)).

### 1.1.2 Enzymatic activity: potential O<sub>2</sub> consumption rates

Enzymes in the electron transport system (ETS) are the major players in O<sub>2</sub> consumption. Most ETS enzymes are large complexes. NADH-dehydrogenase, the principle target of the ETS assay used in oceanography, is an enormous, "L"-shaped, 40 unit, superprotein complex of more than 1000 kDa. Diagonally, it is 330 Angstroms long. It is one of the largest enzymes known; containing more

than 7724 amino acids<sup>1</sup> (Fisher (1964); Walker (1992); Kühlbrandt (2015)). Depending on the presence or absence of inhibitors or activators, the ETS enzymes lead the respiratory physiological response of the organism, the O<sub>2</sub> consumption rates ( $R_{O_2}$ ). Thus, Packard assayed the ETS activity to measure potential respiration ( $\Phi$ ) in marine plankton, allowing the assessment of O<sub>2</sub> consumption rates even in the deep ocean (Packard (1971); Packard et al. (1971b)). When the ETS enzymes are saturated with their specific substrates, the ETS works at its maximum velocity ( $V_{max}$ ). Theoretically, this is the maximum capacity of the cell in consuming O<sub>2</sub> (Moran et al. (2012); Nelson and Cox (2008)). This value reflects the respiratory enzymatic structure of the cell, describing the catabolic metabolism of the organism. Thus, the ETS  $V_{max}$  remains stable under different physiological conditions in the short term. This is because the respiratory ETS is a constitutive enzyme system. Except under long-term stress, it is maintained at a constant concentration in its membrane matrix. In contrast,  $R_{O_2}$  is more variable, as it is the result of the *in vivo* ETS activity, the actual, substrate modulated, velocity of the reactions catalysed by the ETS. In detail, this physiological rate is controlled by the in-cell availability of substrates, activators and inhibitors, as well as temperature and pressure (Packard et al. (1975); King and Packard (1975); Packard et al. (1996b)).

### 1.1.3 The $R_{O_2}/\Phi$ ratio

First attempts in proxying  $R_{O_2}$  from ETS measurements involved seeking the relationship between  $R_{O_2}$  and  $\Phi$ . In the beginning, it was thought to be a quotient close to half  $\Phi$  (Cleland (1967); Segel (1975)). Since the 80's, studies were run in different regions to assess the direct measurement of this ratio: from the Antarctic Ocean (Ikeda and Fay (1981)); from the Gulf of Maine (Packard and Williams (1981)); from the North Pacific Ocean (Ikeda et al. (2006)); and from 8 different oceanic regions (Arístegui and Montero (1995b,a, 2005)). In addition, a comparison was made between assessing  $R_{O_2}$  with the He-tritium/AOU method of Jenkins (1982) and assessing  $R_{O_2}$  with the ETS assay and the  $R_{O_2}/\Phi$  ratio (0.26) described in Packard and Williams (1981). As predicted (see section 1.1.2) variability in the ratio was found (from 0.26 to 0.53 in microplankton (Packard and Williams (1981); Gómez et al. (2011)); from 0.4 to >1 in zooplankton (Gómez et al. (2011))).  $R_{O_2}$ , as any physiological rate, depends on different factors (Hernández-Leon and Gómez (1996); Fernández-Urruzola et al. (2011); Herrera et al. (2011b)), while the ETS reflecting the maximum  $V_{max}$  of the enzymatic activity, i.e., the potential O<sub>2</sub> consumption ( $\Phi$ ) is more stable. In consequence, the variations in the  $R_{O_2}/\Phi$  ratios depend on the same factors that impact the physiology (Hernández-Leon and Gómez (1996)). Until recently,  $R_{O_2}$  has been calculated from  $\Phi$  based on different biochemical hypotheses or statistically based ratios. Packard et al. (1971b) based their  $R_{O_2}$  calculations on the respiratory control ratio (0.10) of Chance and

<sup>1</sup>Note the difference with other simple enzymes, such as insulin, with only 51 amino acids and a molecular weight of only 5.8 kilodaltons (kDa)

Williams (1956); Chance (1965). Later, Packard et al. (1977) solved Harmon Craig's advection-model for deep-sea  $R_{O_2}$  (Craig (1971)) to derive an  $R_{O_2}/\Phi$  ratio suitable for the dark-ocean (0.02)<sup>2</sup>. Subsequent research on interpreting the meaning of  $\Phi$  considered the theoretical assumption that: 1) the physiological respiratory response is half the ETS  $V_{max}$  (Cleland (1967); Segel (1975)); 2) the directly measured physiological respiratory response in a region at a certain moment is representative of that region, so the  $R_{O_2}/\Phi$  is developed to be applied to the  $\Phi$  measurements in that region (Packard and Williams (1981); 3) an average  $R_{O_2}/\Phi$  for a plankton size is representative and applied to  $\Phi$  measurements on that size group; and finally, 4) a linear regression with as many measures as possible throughout the whole ocean is used to derive  $R_{O_2}$  from  $\Phi$  (Arístegui and Montero (1995a)).

In the meanwhile, other studies were carried out in the laboratory in order to understand the variability of this ratio, which complemented oceanographic results (Hernández-Leon and Gómez (1996); Martínez et al. (2010); Herrera et al. (2011b); Osma et al. (2016a,b)). Among the variables investigated, starvation and nutrient availability resulted to be the factors which mainly drove changes in the physiological response, and so, changed the  $R_{O_2}/\Phi$  ratio. As enzymes are the main drivers of respiratory reactions, these results, combined with kinetics, fostered consideration of a new approach to this topic. Enzymes obey Michaelis-Menten kinetics, where substrates are the main regulators of their function, and consequently, on the physiological response. How the different environmental and physiological factors impact the intracellular substrate availability to the ETS, will lead to different  $R_{O_2}$ . This conceptual idea fuelled the development of the enzyme kinetic model (EKM) in the 90's Packard et al. (1996b,a). Since Packard et al. (1996b), several studies in bacteria (Aguiar-González et al. (2012)) and in marine zooplankton (Packard and Gómez (2008); Osma et al. (2016b)) have modelled  $R_{O_2}$  by measuring  $\Phi$ , its intracellular substrates and applying the bisubstrate enzyme kinetics. These models were validated with direct  $R_{O_2}$  measurements during the experiments, and resulted in more accurate predictions of  $R_{O_2}$  than the use of simple  $R_{O_2}/\Phi$  ratios. Both Aguiar-González et al. (2012) and Osma et al. (2016b) demonstrated the advantage of using an EKM instead of a constant  $R_{O_2}/\Phi$  ratio, or the Metabolical Theory of Ecology (MTE) based on biomass.

In addition, the use of *in vivo* tetrazolium reduction in whole populations of microplankton was introduced in the last decade by Martínez-García et al. (2009) and Aranguren-Gassis et al. (2012). This method involves adding the tetrazolium salt (INT<sup>3</sup>) directly to bottled seawater samples containing living-organism and incubating them for various time-periods. While respiring, these organisms reduce INT to formazan by a wide variety of enzymatic and non-enzymatic reactions. Res-

<sup>2</sup>The original  $R_{O_2}/\Phi$  ratio of 0.05, reported in Packard et al. (1977), was based on the Packard (1971) ETS assay. Modern deep-sea ETS assays are based on the more sensitive Kenner and Ahmed (1975) ETS assay. According to Table 3 in Christensen and Packard (1979) were the ETS measurements in Packard et al. (1977) made by the Kenner and Ahmed (1975) ETS assay the  $R_{O_2}/\Phi$  ratio would be 0.02 rather than 0.05.

<sup>3</sup>2-(p-iodophenyl)-3-(p-nitrophenyl)-5-phenyl tetrazolium chloride (Nachlas et al. (1960)

piratory INT reduction, directly related to the O<sub>2</sub> consumption, is thought to be the dominant reaction. However, INT is toxic to whole cells (Villegas-Mendoza et al. (2015)) because it uncouples respiratory O<sub>2</sub> consumption from ATP production (Clark et al. (1965)). Starved for energy, the cells die during the incubation, thus the measurement represents the respiration of a dying community. Especially for long-period incubations, this toxicity will generate misleading measurements. In spite of this, a useable correlation between both measurements has been reported (Martínez-García et al. (2009); Aranguren-Gassis et al. (2012)), so this tool may be useful depending on the aim of the study. However, because Φ and in-cell substrates are the causal basis of respiration, we consider the measurement of Φ, the use of the EKM, and research into  $R_{O_2}$  variability a better way to asses respiration in the ocean than *in vivo* INT reduction.

## 1.2 The impact of respiration on the carbon flux assessment

As the global economy is based on the combustion of fossil resources, the anthropogenic CO<sub>2</sub> emissions have been increasing for the past 250 years. This rise, combined with the spread of deforestation and agriculture, has resulted in the highest atmospheric CO<sub>2</sub> levels in the last 800000 years (400ppm) and keeps increasing up to one order of magnitude higher than one million years ago (Doney and Schimel (2007)). The oceans are estimated to take up nearly one third of this anthropogenic CO<sub>2</sub>. Thus, the assessment and understanding of the processes that control the carbon fluxes between atmosphere, marine surface waters, the deep ocean and the benthos, have been the focus of international research programs since the 1980s (e.g. Joint Global Ocean Flux Study, JGOFS). In the ocean, the net carbon flux is controlled by three key-processes : 1) the autotrophic CO<sub>2</sub> fixation into POC which is mainly synthesised by autotrophs in the upper euphotic layer of the water column; 2) the POC remineralisation back to CO<sub>2</sub> through respiration, mainly driven by heterotrophs but also by autotrophs and mixotrophs (Redfield (1934)); and 3) the export of particulate and dissolved organic carbon (POC and DOC) to the sea bottom (Ketchum (1957); Munk (1966); Rivkin and Legendre (2001)). This transfer of organic carbon from the euphotic zone to the seafloor occurs by different processes: a) gravitational sinking of particles (McCave (1975)); b) advection (Seiwell and Seiwell (1938); Redfield (1934); Riley (1951)); c) active transport by vertical migrants (Ketchum (1957); Munk (1966)); and d) isopycnal pumping (Packard et al. (1977)). Among them, the passive sinking of POC results in the main source of organic carbon in the dark ocean (Eppley and Peterson (1979); Osma et al. (2014)). A better understanding of the processes that control vertical carbon flux ( $F_C$ ) requires an improvement in the knowledge of both primary production and respiration in terms of carbon. However, although respiration is the most important process involved in carbon cycling, its measurements are scarce relative to primary production. It was not until the 90's when

the first studies in carbon flow between production and respiration arose. Many of these were focused in the assessment of remineralization from sub-euphotic ETS measurements throughout the water column (Lefévre et al. (1996); Arístegui et al. (2002a); Packard and Codispoti (2007)). Furthermore, the connection between the primary production and the deep ocean carbon demands was also assessed from ETS measurements (Packard et al. (1971b, 1977); Arístegui and Harrison (2002); Arístegui et al. (2002b)). More recently, vertical active  $F_C$  was also studied by the ETS profile measurements in migrants throughout the water-column (Dam et al. (1995); Darnis and Fortier (2012); Packard and Gómez (2013); Ariza et al. (2015); Hernández-León et al. (2019)).

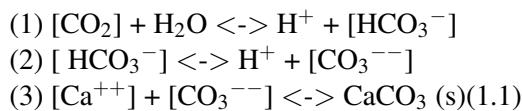
Finally, considering respiration as the major sink of organic matter in any ecosystem (del Giorgio and le B. Williams (2005)), many researchers have studied how respiration impacts the marine vertical passive  $F_C$  (Packard and Christensen (2004); Giering et al. (2014); Osma et al. (2014); Packard et al. (2015)). In these studies, vertical  $F_C$  was calculated assuming that the decrease in sinking POC is caused by respiratory CO<sub>2</sub> production. This means that the first derivative, mathematically speaking, of the  $F_C$ -depth function is the respiration depth function (Suess (1980); Martin et al. (1987); Packard and Christensen (2004)). The development of the ETS assay resulted in metabolism measurements that could be converted to respiration throughout the water-column down to the deep ocean. From these measurements, the production of CO<sub>2</sub> related to respiration was calculated. The CO<sub>2</sub> production from the bottom of the euphotic zone to the deep ocean was calculated as the integral of the CO<sub>2</sub> production rates at different depths (Packard and Christensen (2004); Osma et al. (2014); Packard et al. (2015)). This equalled the  $F_C$  sinking out of the euphotic zone.

### 1.2.1 Physiological CO<sub>2</sub> production rates

Even though during the last decades the CO<sub>2</sub>-producing facet of the respiration ( $R_{CO_2}$ ) received more attention because of its importance in assessing the role of the ocean as an anthropogenic CO<sub>2</sub> sink, respiration has been mainly approached by measuring  $R_{O_2}$  rates. Direct physiological measurement of  $R_{CO_2}$  has not been as feasible as measuring  $R_{O_2}$  in the ocean, mainly because of the complexity and paucity of the available techniques to measure it and because of the buffering effect of the carbonate system.

Despite these difficulties, marine  $R_{CO_2}$  rates have been assessed from different approaches since the 70's. The first marine, direct  $R_{CO_2}$  rates measurements were done by following the CO<sub>2</sub> absorption by KOH (Rakusa-Suszczewski et al. (1976)). Since then, other studies have measured these rates by following the changes in the Total dissolved inorganic carbon (TC) as an end-point measurement after long incubations (Johnson et al. (1987, 1993)). These TC measurements involve a gas extraction of acidified seawater by coulometric titration of the acid formed by the resultant CO<sub>2</sub> and monoethanolamine. In addition, other studies, building on the  $R_{CO_2}$  control, via Henry's Law, of the pCO<sub>2</sub> in the head space

above the respiring plankton, have assessed  $R_{CO_2}$  by monitoring the changes in the pCO<sub>2</sub> by infrared detection (2000 μm of absorption peak). These studies used a Micro-Oxymax respirometer ([www.colinst.com](http://www.colinst.com)) (Berdal et al. (1995a); Packard et al. (1996a); Romero-Kutzner et al. (2015)). More recently, the development of infrared spectrometry has allowed not only the TC measurement but also the δ<sup>13</sup>C and δ<sup>18</sup>O stable isotopic composition determination (Yakir and da SL Sternberg (2000); Braden-Behrens et al. (2017)). This new approach has already facilitated the continuous measurement of these parameters in terrestrial ecosystems during time, assessing the separation of the net CO<sub>2</sub> exchange into photosynthetic and respiration components (Braden-Behrens et al. (2017)). However, its application on marine ecosystems is still under development (Smajgl et al. (2018)). Finally, the improvement of pCO<sub>2</sub> optodes may also become a proper tool to measure  $R_{CO_2}$ . However, even though they have provided accurate and precise measurements, they still require strong calibration and preconditioning to the high salinity of marine ecosystems (Atamanchuk et al. (2014)). In addition, the CO<sub>2</sub> produced by respiration enters the surrounded water and participates in the equilibrium processes as in Eq.1.1 (Millero (2006)):



So this reaction sequence leads to pCO<sub>2</sub> buffering in marine waters, as equilibria is established. Thus, direct measurements of pCO<sub>2</sub> in the water, as those addressed by the pCO<sub>2</sub>-optodes, are not useful to directly assess  $R_{CO_2}$  rates. However, every mol of CO<sub>2</sub> produced by respiration adds one mol to the sum of all the inorganic carbon pool in the marine water. This is the TC, Eq.1.2:

$$TC = [CO_2] + [HCO_3^-] + [CO_3^{--}]$$
(1.2)

Taking this into consideration, the optode pCO<sub>2</sub> measurements require the conversion to TC based on pH and total alkalinity (TA) (Berggren et al. (2012)).

Due to all these constraints in determining CO<sub>2</sub> changes in seawater related to respiration such as: (1) the need for long incubation periods to measure TC as an end-point measurement Mayzaud et al. (2005); (2) the need for strong calibration and preconditioning to high salinity of the pCO<sub>2</sub> optodes (Atamanchuk et al. (2014)) and for the conversion to TC by alkalinity and pH (Berggren et al. (2012)); (3) the not-so-affordable cost of the technology linked to the infrared sensing of pCO<sub>2</sub> (Berdal et al. (1995a); Packard et al. (1996a)) and of the new marine application of stable isotopes (Smajgl et al. (2018)), respiratory rates have mainly been reported as O<sub>2</sub> consumption in oceanography. As a result, Respiratory Quotients (RQs) ratios between mol of CO<sub>2</sub> produced per mol of O<sub>2</sub> consumed, have

been required to calculate respiratory CO<sub>2</sub> production rates from  $R_{O_2}$ . However, these ratios are variable, not only because of different nutrient sources, but also to differences in nutrient availability (Berggren et al. (2012); Romero-Kutzner et al. (2015)). So, when assessing  $R_{CO_2}$  rates, not only does the variability in the physiological  $R_{O_2}$  rates affect the final measurement, but also the variability in the RQ ratios, which would bias the accuracy of the results.

To sum up, all the direct  $R_{CO_2}$ -measurement techniques are currently time-and-manpower consuming and cannot be used on deep ocean communities, so they usually result in low data acquisition rates. As oceanography requires a higher data acquisition rate over time and space, an enzymatic assay related to CO<sub>2</sub> production would have arisen as an interesting improvement in the carbon flux assessment research.

### 1.2.2 Enzymatic potential CO<sub>2</sub> production rates

The biochemistry of respiratory CO<sub>2</sub> production is dominated by NAD(P)<sup>+</sup>-dependent isocitrate dehydrogenase in the Krebs cycle (IDH<sub>KC</sub> in Fig. 1.1) (Nelson and Cox (2008)). In cells, other biochemical reactions produce CO<sub>2</sub>, but their production rates can be predicted from IDH<sub>KC</sub> activity (Packard et al. (1996a); Roy and Packard (2001); Tames-Espinosa et al. (2018)). Of those linked to respiration, pyruvate dehydrogenase (PDH) serves to bridge the metabolism of glucose (glycolysis) to the Krebs cycle. Then, inside this cycle of enzyme reactions, both IDH<sub>KC</sub> and alpha-ketoglutarate dehydrogenase (KGDH in Fig. 1.1), produce CO<sub>2</sub>, being the IDH<sub>KC</sub> activity 1.1 times the KGDH activity (Walsh and Koshland (1984), Fig. 1.1). Finally, linked to, but not in the Krebs cycle, are the malic enzyme (ME in Fig. 1.1) and phosphoenol pyruvate carboxykinase (PEPCK in Fig. 1.1). They also generate CO<sub>2</sub> but at much reduced rates. IDH<sub>KC</sub> activity is 8.4 times greater than these activities.

IDH<sub>KC</sub> encompass two isoenzymes that catalyse the conversion of isocitrate to  $\alpha$ -ketoglutarate in a reaction summarized in Eq. 1.3 (Nelson and Cox (2008)).



where 1 mol of CO<sub>2</sub> is stoichiometrically related to the production of 1 mol of NAD(P)H.

These enzymes can be either NAD<sup>+</sup> or NADP<sup>+</sup>-dependent (NAD-IDH or NADP-IDH respectively) (Kornberg and Pricer (1951); Gálvez and Gadál (1995)). However, unlike NAD-IDH, NADP-IDH is ubiquitous in all types of organisms (from *E.coli*-like bacteria to plants and mammals); its activity has been reported to be up to 10-fold greater than NAD-IDH activity (Alp et al. (1976); Colman (1975); Denton et al. (1978)); and, unlike NAD-IDH, NADP-IDH is stable during the measurement (Plaut (1970, 1962)). Furthermore, NADP-IDH largely controls CO<sub>2</sub>

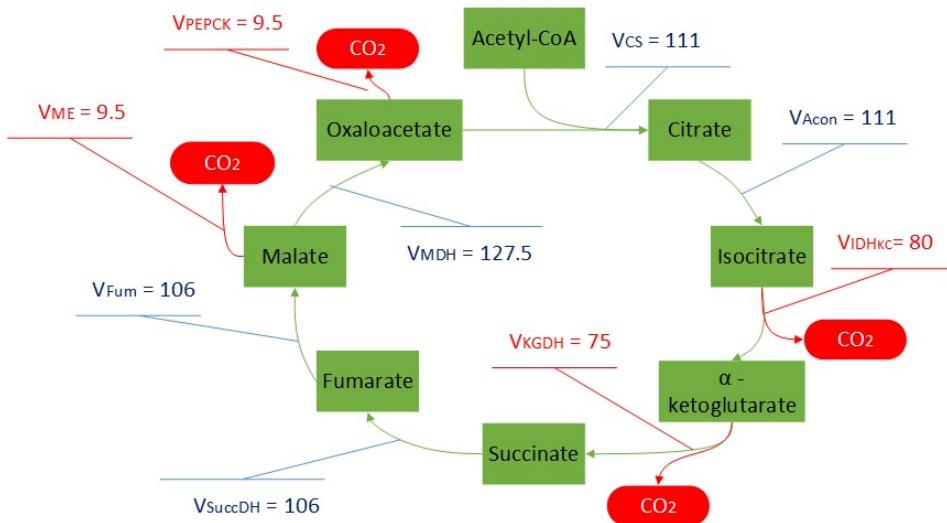


Figure 1.1: Velocities (V) through the main pathways in *E.coli* Krebs cycle (KC) in the bacterium, *Escherichia coli*, during acetate utilisation (redrawn from Walsh and Koshland (1984)). Rates are expressed in terms of mmol substrate consumed min<sup>-1</sup> L<sup>-1</sup> of cell volume. Abbreviations in the velocity expressions are CS, citrate synthase; Acon, aconitase; IDH<sub>KC</sub>, isocitrate dehydrogenase linked to the Krebs cycle; KGDH, α-ketoglutarate dehydrogenase; SuccDH, succinate dehydrogenase; Fum, fumarase; MDH, malate dehydrogenase; ME, malic enzyme; PEPCK, phosphoenolpyruvate carboxykinase. Rates linked to CO<sub>2</sub> production have been highlighted in red. Not shown is the CO<sub>2</sub>-producing enzyme, pyruvate dehydrogenase, that links glucose metabolism in the Embden-Meyerhof pathway with the KC by decarboxylating pyruvate to acetyl-coenzyme A.

production in the Krebs cycle (Nicholls and Garland (1969); Walsh and Koshland (1984); Williamson and Cooper (1980)) not only because it is the more active CO<sub>2</sub>-producing enzyme (Fig. 1.1) but also because it works as a key enzyme in the Sazanov and Jackson substrate cycle in the regulation of Krebs cycle activity (Sazanov and Jackson (1994)). In addition, mitochondrial NADP-IDH produces NADPH that is used to produce both, ATP by the electron transport system (ETS), and reduced glutathione (a reactive oxygen species (ROS) scavenger which serves as the cellular defence against oxidative stress-induced damage (Jo et al. (2001))). Moreover, cytosolic CO<sub>2</sub>-producing NADP-IDH generates NADPH outside the mitochondria, linked to lipid synthesis (Koh et al. (2004))). Increases in its activity have been related to hypothyroidism in mammals (Czyzewska et al. (2012); Kadenbach et al. (1964)). Thus, cytosolic NADP-IDH measurements may improve the assessment of lipid production on the base of the marine food chain. Even though the main goal of this research is the development of the NADP-IDH assay for the plankton community, we firmly believe it will be useful in further research within

these diverse fields.

An early attempt to assess the relationship between CO<sub>2</sub> production rates and the NADP-IDH activity was made in the marine bacterium *Pseudomonas nautica* (Berdal et al. (1995a)). From these measurements, the first steps in modelling the CO<sub>2</sub> production were developed (Packard et al. (1996a)). Later, this model was improved and tested in the same bacterium by making measurements of the substrates, NADP<sup>+</sup> and isocitrate (Roy and Packard (2001)). From these analyses, a factor ( $\chi$ ) was derived to calculate CO<sub>2</sub> production from NADP-IDH activity in the *P. nautica* Krebs cycle (Roy and Packard (2001)). These studies on marine bacteria were the first assessment on the NADP-IDH role in the CO<sub>2</sub> production in marine organisms.

However, they were focused on marine bacteria, whereas oceanography needed a plankton community tool. Thus, the first objective in the present thesis has been the development of an NADP-IDH assay in the marine plankton (Chapter 2). To do so, we considered previous studies of NADP-IDH activity in yeast, marine bacteria, and marine fish (turbot) (Berdal et al. (1995a); Kornberg and Pricer (1951); Munilla-Moran and Stark (1989); Munilla-Moran (1994); Packard et al. (1996a); Roy and Packard (2001)). After carrying out many experiments in the laboratory, this was the first time NADP-IDH was measured in marine plankton. The new assay was published on the journal, Marine Chemistry, in 2018 (Tames-Espinosa et al. (2018)).

Although we were able to measure the NADP-IDH activity, the potential CO<sub>2</sub> production in the cells involves also the CO<sub>2</sub> produced by other respiratory and non-respiratory enzymes. Thus, we work on the second objective of this thesis in Chapter 3: modelling the potential CO<sub>2</sub> production based on NADP-IDH in marine plankton, taking into account the main CO<sub>2</sub>-producing enzymes in the cells.

### 1.2.3 The $R_{CO_2}$ /NADP-IDH ratio

One of the main advantages of the NADP-IDH assay for the plankton community is the possibility of proxying the  $R_{CO_2}$  from these measurements throughout the water-column. The development of these ratios reduces the variability linked to the estimations of  $R_{CO_2}$  from  $R_{O_2}$  by using RQ, particularly in water-column studies, where  $R_{O_2}$  is also derived from  $\Phi$  at depth. However, as in the case of the  $R_{O_2}/\Phi$  ratios, we expect the plankton physiological rate to be variable whereas the enzymatic measurements would remain more stable in the short term. In previous studies on the marine bacterium *Pseudomonas nautica* the carbon-source depletion induced these differences between  $R_{CO_2}$  and NADP-IDH in the short term (Berdal et al. (1995a); Packard et al. (1996a); Roy and Packard (1998, 2001)).

Thus, the use of such an oceanographic tool would require the following research:

- From a physiological point of view, we should improve the current knowledge about the influence of different factors on the performance of the CO<sub>2</sub> producing enzymes, their kinetics and their in-cell substrates availability. Here, in chapter 3, we have attempted for the first time how the sex-differences and starvation may impact this ratio in *Leptomysis lingvura*.
- On a short time-scale, we should measure and develop  $R_{CO_2}$ /NADP-IDH ratios that describe a specific region, at a certain time for a size-fractionated plankton community, so that the ratios can be considered representative for communities under similar environmental conditions in that region. In chapter 3, we measured the *L. lingvura*  $R_{CO_2}$ /NADP-IDH ratio in two communities from different years, leading to similar representative results.
- In the long run, when many ratios were measured, we tried to assess an average ratio (or a linear regression between  $R_{CO_2}$  and NADP-IDH measurements). However, we have to bear in mind that this kind of approach may greatly deviate from the actual measurement at an instantaneous time-scale. Unfortunately, we will have to wait until widespread use of this tool before we can properly define an average relationship between both parameters.
- Finally, an accurate model for proxying  $R_{CO_2}$  from NADP-IDH, isocitrate and NADP<sup>+</sup> measurements in plankton communities should be developed. To do so, the same theoretical bases of the bisubstrate kinetics models would be applied, as done on *P. nautica* (Packard et al. (1996a); Roy and Packard (2001)).

### 1.3 Further applications of respiratory enzymatic measurements

In the previous sections, we presented the valuable role of the respiratory enzymatic measurements to assess  $R_{O_2}$  and  $R_{CO_2}$  and the factors that affected their variability. However, we should keep in mind that they are the reflection of an inherent characteristic of the community: its metabolism. Several studies took this into consideration and, after applying not-so-difficult mathematical models, they were able to assess other parameters in the ecosystem that were not previously studied.

The next subsections describe several examples of how the combination of metabolic measurements, mathematics, and the relationship with other oceanographic parameters may improve the marine research and the study of the aquatic ecosystems.

### 1.3.1 Carbon flux and Nutrient regeneration efficiency

Vertical carbon flux ( $F_C$ ) in the ocean can be assessed from the vertical profiles of  $R_{CO_2}$ . Conceptually, “planktonic  $R_{CO_2}$  in a seawater cube is considered equivalent to the difference between the total  $F_{C1}$  through the top of the cube ( $z_0$ ) and total  $F_{C2}$  through the bottom of the cube ( $z$ )” (Packard et al. (2015)).  $F_C$  changes in that depth layer are related to the loss of carbon by respiration between the boundaries of that layer. This is the integral of  $R_{CO_2}$  between them. So, the first step in this assessment requires the development of the vertical profile of  $R_{CO_2}$  (see Fig.1.2).

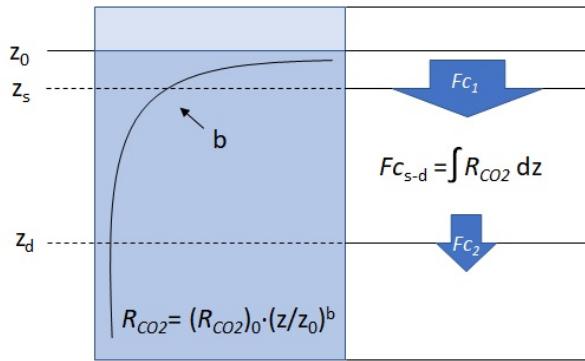


Figure 1.2: Diagram of the vertical profile of  $R_{CO_2}$  and its relationship with vertical  $F_C$ . Light blue illustrates the euphotic zone whereas dark blue represents the aphotic zone.

In Packard et al. (2015), the respiration-depth profile was obtained by modelling  $R_{CO_2}$  from microplankton  $\Phi$ . Then, the power function that describes these depth profile (Eq. 1.4) was developed (see Fig.1.2).

$$R_{CO_2} = (R_{CO_2})_0 \cdot (z/z_0)^b \quad (1.4)$$

where  $(R_{CO_2})_0$  is the  $R_{CO_2}$  at  $z_0$  and  $b$  (always negative) is the exponent that defines the curvature of the vertical profile. The larger  $|b|$  is, the lower  $F_C$  is from the epipelagic ocean. It is the opposite when  $|b|$  is low.

Then, to calculate  $F_C$  from any depth in the water column ( $z_s$ ) down to a deeper one ( $z_d$ ), the mathematical vertical profile of  $R_{CO_2}$  between those two depth was integrated (Eq. 1.5, Fig.1.2) (Packard et al. (2015))).

$$F_{Cs-d} = \int_{Z_s}^{Z_d} R_{CO_2} dz$$

$$F_{Cs-d} = \int_{Z_s}^{Z_d} (R_{CO2})_0 \cdot (z/z_0)^b dz$$

$$F_{Cs-d} = \{(R_{CO2})_0 / [(b+1)z_0^b]\} (z_s^{b+1} - z_d^{b+1}) \quad (1.5)$$

Note that  $z_s$  is a depth between  $z_0$  and  $z_d$  ( $z_0 \leq z_s \leq z_d$ ).

In addition, the nutrient retention efficiency (NRE), described as the nutrient remineralization rate within an ocean layer (Packard and Gómez (2013); Osma et al. (2014); Packard et al. (2015)), can be also calculated from both a profile of plankton respiration, or as the inverse of the  $F_C$  transfer efficiency below the euphotic zone(Buesseler and Boyd (2009)). In Packard et al. (2015), the NRE was assessed as the  $R_{CO2}$  within an ocean layer per the  $F_C$  into that layer, expressed as a percentage.

$$NRE = R_{CO2} \cdot \Delta z / F_C \quad (1.6)$$

Note that  $R_{CO2}$  was reported in mol CO<sub>2</sub> d<sup>-1</sup> m<sup>-3</sup> and  $F_C$  in mol C d<sup>-1</sup> m<sup>-2</sup>. Since the Redfield N=C or P=C ratio is applied for both parts of the ratio in Eq.1.6, the NRE is unitless and can be expressed as a percent.

Taking all into consideration, plankton respiration depth-profiles from ETS measurements allowed the assessment of the nutrient retention capacity of a plankton community and its variability throughout a geographical section (Packard et al. (2015)). In addition, the use of mathematical analyses highlighted that both the water column NRE, and the  $F_C$  are controlled by the curvature of the vertical  $R_{CO2}$  profile (labeled in the depth-profile power function).

### 1.3.2 Heterotrophic energy production

The energy and carbon flux are terms commonly used in ecology to describe the main processes in the ecosystem (solar energy capture, organic matter production or energy dissipation through the food chain in form of heat). However, despite the fact the carbon flux has been widely addressed, the measurement of the energy flow has been rarely assessed (Karl). Surprisingly, even though all life forms live on a complex network of energy flow, the quantification of its capture, transformation and dissipation is not entirely well-known. In 2014, Karl pointed out this lack of knowledge, highlighting the already-existing pioneering methods that could have also been calibrating to assess these energy transformations, instead of just estimating living carbon and oxygen consumption. He was referring, among others, to the Packard's ETS method, the measurement of potential respiration ( $\Phi$ ) (Packard (1971)).

As it has been said before,  $\Phi$  is a measurement of the mitochondrial electron transport system (ETS) in terms in O<sub>2</sub> consumption. Thus,  $R_{O2}$  is coupling with the oxidative phosphorylation of the adenosine diphosphate (ADP). This process results in energy production in the form of ATP (Ochoa (1943)). While measuring

$\Phi$  (or  $R_{O_2}$ ), the potential capacity (or actual capacity, respectively) of producing heterotrophic energy by the community can be assessed.

Considering this theoretical basis, Packard et al. (2015) addressed the heterotrophic energy production (HEP) in the microplankton community from  $\Phi$  measurements, after converting them to  $R_{O_2}$ ). To do so, they related the  $R_{O_2}$  to energy as expressed in Eq. 1.7:

$$\text{HEP} = R_{O_2} \cdot 2 \cdot 2.5 \cdot 48 \quad (1.7)$$

where 2 are the number of pairs of electrons ( $2e^-$ ) needed to reduce  $O_2$  to  $H_2O$  ( $O_2 + 2e^- + 2H^+ \leftrightarrow H_2O$ ); 2.5 is the number of molecules of ATP generated by the ATP synthase per pair of electrons (ATP/ $2e^-$ , Ferguson (2010)); and 48 is the Gibbs Free Energy ( $\Delta G$ ) in J per mmol of ATP (Alberty and Goldberg (1992)).

So, with this equation, after considering Karl's suggestion in 2014, Packard calculated the heterotrophic energy transformation from existint ETS data in 2015. In oxic waters, when  $R_{O_2}$  was expressed in mmol  $O_2$  per time and volume, it was directly converted to J per time and volume (Eq.1.7).

### 1.3.3 Use of respiratory enzymes as a proxy of the catabolic metabolism in the oceans

The measurement of the enzymes that drive respiration, i.e., their activity (Fruton and Simmonds (1958); Kornberg (1991)), reflects the catabolic metabolism in the plankton community that is used to obtain energy from organic matter. It is the biochemical basis of Harmon Craigâs Deep Metabolism (Craig (1971)) and the driving force for Walter Munk's Abyssal Recipes (Munk (1966)) that was recognized and used by Packard et al. (1971b). Comparison with other oceanographic parameters, linked to anabolic processes such as Chl<sub>a</sub> and <sup>14</sup>C uptake in photosynthesis, facilitates the assessment of the level of autotrophy or heterotrophy of the community at a certain moment.

However, despite the fact that many oceanographic studies have measured both Chl<sub>a</sub> and  $\Phi$  (Packard et al. (1974); Packard (1979); Packard and Williams (1981); Packard et al. (2000); Packard and Christensen (2004)), only a few assessed the relationship between both parameters (Packard et al. (1974); Martinez et al. (1990); Martinez (1991); Bangqin et al. (2005)). After analysing the reported values and its relationship we observed a  $\Phi/\text{Chl}_a$  ratio variability ranging from  $0.3 \mu\text{LO}_2\text{h}^{-1}(\mu\text{g Chl}_a)^{-1}$  in eutrophic waters (Packard et al. (2000)) or in the maximum Chl<sub>a</sub> depth (Martinez et al. (1990)), to  $5 \mu\text{LO}_2\text{h}^{-1}(\mu\text{g Chl}_a)^{-1}$  in oligotrophic regions (Martinez et al. (1990)). This variability seems to be related to the dominance of autotrophs or heterotrophs in the community in different regions during certain moments.

In 2014 we participated in a mesocosm experiment where the oligotrophic community was monitored before, during and after a simulated upwelling of en-

riched deep-water. After measuring  $\Phi$  and having access to the Chl<sub>a</sub> measurements we assessed the  $\Phi/\text{Chl}_a$  ratio and monitored the changes in the community metabolism towards a pure-autotrophic community during the phytoplankton bloom, its shift towards a more heterotrophic one due to the proliferation of heterotrophic dinoflagellates or the moments where mixotrophs (or a mixed community) dominated. Here, we present our results in Chapter 5.

The definition of the values related to a more autotrophic or more heterotrophic community makes the  $\Phi/\text{Chl}_a$  an useful oceanographic tool to determine the metabolic state of the ocean at a certain moment. The measurement of both parameters in a regular basis in oceanography may lead to a further knowledge about the ecosystem, specially in the open ocean.

## 1.4 Thesis objectives and outline

Taking all into consideration, the present dissertation is, firstly, focused on the development of a new enzymatic assay for NADP<sup>+</sup>-dependent isocitrate dehydrogenase (NADP-IDH) to facilitate the assessment of the potential CO<sub>2</sub> production in marine plankton. To do so, the following objectives were addressed.

- Objective 1: The development of an assay for marine NADP-IDH activity because this activity is one of the main reactions for the CO<sub>2</sub> production in cells. From existing NADP-IDH assays in yeast, bacteria and fish, a new, composit NADP-IDH assay was adapted to the microplankton and mesozooplankton (Chapter 2).
- Objective 2: The assessment of mesozooplankton potential CO<sub>2</sub> production from NADP-IDH measurements. The relationships among NADP-IDH and all the others CO<sub>2</sub>-producing enzymes in the Krebs cycle was assessed in planktonic cells taking into account the studies from Walsh and Koshland (1984) and Alp et al. (1976) (Chapter 3). This approach led to the normalisation of every enzyme activity by the NADP-IDH. Total CO<sub>2</sub> production is the sum of all the CO<sub>2</sub>-producing enzymes.
- Objective 3: The assessment of the relationship between  $R_{CO_2}$  and NADP-IDH activity (Chapter 4).  $R_{CO_2}$  was measured by following the TC increase in *L. lingvura* incubations. With the collaboration of members of the QUIMA research group. NADP-IDH was determined with the marine plankton NADP-IDH assay.

These three objectives led to a methodological improvement in the assessment of respiration and its impact on the vertical carbon flux in the water column. Furthermore, it allowed a direct measurement of the CO<sub>2</sub> production facet of respiration. This is particularly important in the deep-water studies. It facilitates an improvement in the knowledge of metabolism in bathypelagic organisms and ecosystems.

In addition, a fourth objective was also pursued. It was as an example of how metabolic assays, as oceanographic tools, can improve marine research.

- Objective 4: Use of an enzymatic tool to assess variations in the autotrophy versus heterotrophy state of the microplankton community in the oligotrophic Atlantic ocean during time (Chapter 5). Here, the assessment of the metabolic state is done by measuring  $\Phi$  per mg of Chl<sub>a</sub> during 55 days under the effects of a simulated enriched-deep-water upwelling. Furthermore, the effects of acidification on the former community and its metabolic state, were also studied.

## Chapter 2

# Chapter 2. NADP<sup>+</sup>-dependent isocitrate dehydrogenase activity in marine plankton

*The existence of atoms and molecules is dominated by 'attractions', 'repulsions', 'wants' and 'discharges', to the point that it becomes virtually impossible to write about chemistry without giving in to some sort of randy anthropomorphism.*

Nick Lane

**RESUMEN:** En este capítulo se recoge cómo se ha medido la NADP<sup>+</sup>-isocitrato deshidrogenasa, una de las principales enzimas productoras de CO<sub>2</sub>, en el plancton marino. Para ello, se ha adaptado la metodología a las fracciones de 0,7-50μm, 50-200μm, y 200-2000μm de la comunidad planctónica de las Islas Canarias. Durante el proceso se estudió la variabilidad de la actividad de la NADP-IDH en relación con el pH, temperatura, dilución de la muestra, centrifugación o la concentración de sustratos. En nuestras condiciones de trabajo, la velocidad máxima de reacción ( $V_{max}$ ) se alcanzó en un buffer fosfato 0,1M, con un pH 8,2 ( $pK_a_1=7,6$  y  $pK_a_2=8,8$ ), añadiendo 6mM de MgCl<sub>2</sub>, 0,3mM de NADP<sup>+</sup> y 2mM de isocitrato DL-trisódico. La temperatura óptima de reacción para el mesozooplancton subtropical fue 28°C, con una energía de activación de Arrhenius ( $Ea$ ) de 20,4 Kcal mol<sup>-1</sup> (85,4J mol<sup>-1</sup>), y un factor de frecuencia de colisión ( $A$ ) de 2,9·10<sup>11</sup> mol NADPH s<sup>-1</sup> (Kg of protein)<sup>-1</sup>. Las constantes de Michaelis-Menten ( $Kms$ ), que muestran la afinidad de la enzima con sus sustratos a pH 8,5 y 18°C fueron 271±63 μM para el isocitrato y 18±3 μM para el NADP<sup>+</sup>. Esta enzima, aparte de ser una de las principales responsables de la producción de CO<sub>2</sub>, puede utilizarse para el estudio de: (1) la respiración en muestras marinas; (2) el flujo de carbono en la columna de agua; (3) las adaptaciones metabólicas de los organismos ante cambios medioambientales; (4) el pa-

pel de los distintos componentes de la comunidad planctónica en la cadena trófica; y (5) la capacidad de mitigación de las especies de oxígeno reactivo (ROS) de los recursos planctónicos marinos.

**ABSTRACT:** NADP<sup>+</sup>-isocitrate dehydrogenase (NADP-IDH), as one of the most active intracellular CO<sub>2</sub>-producing enzymes, was measured in marine plankton by adapting an enzyme assay to the 0.7-50μm, 50-200μm, and 200-2000μm size fraction of the Canary Islands coastal plankton community. The variability of NADP-IDH activity in relation to pH, temperature, dilution of the sample, centrifugation or substrates was measured. In our hands, the maximum NADP-IDH activity ( $V_{max}$ ) in marine plankton samples was attained in 0.1M phosphate buffer at pH 8.2 ( $pK_a_1=7.6$  and  $pK_a_2=8.8$ ), by adding 6mM MgCl<sub>2</sub>, 0.3mM NADP<sup>+</sup> and 2mM DL-trisodium isocitrate. The optimum temperature in these subtropic mesozooplankton samples was 28°C, with an Arrhenius energy of activation ( $E_a$ ) of 20.4 Kcal mol<sup>-1</sup> (85.4J mol<sup>-1</sup>), and an Arrhenius collision frequency factor ( $A$ ) of  $2.9 \cdot 10^{11}$  mol NADPH s<sup>-1</sup> (Kg of protein)<sup>-1</sup>. The apparent Michaelis-Menten  $K_m$  values for the substrates in crude homogenates at pH 8.5 and 18°C were  $271 \pm 63$  μM for isocitrate and  $18 \pm 3$  μM for NADP<sup>+</sup>. This enzyme, in addition to its high CO<sub>2</sub>-producing activity, can be used to calculate: (1) respiration in marine samples; (2) carbon flux in the water column; (3) metabolic adaptations to environmental changes; (4) roles of the plankton components of the food chain; and (5) the reactive oxygen species (ROS) scavenging capacity of the marine plankton resources.

## 2.1 Introduction

Here, we propose the first step of a further enzymatic assessment of the CO<sub>2</sub> production related to the marine plankton respiration. To do so, we have developed an oceanographic tool to accurately measure the activity of NADP<sup>+</sup>- dependent isocitrate dehydrogenase (NADP-IDH, Eq. 1.3) and characterize the properties of this enzyme in the marine plankton. We have considered previous studies of NADP-IDH activity in yeast, marine bacteria, and marine fish (turbot) (Berdal et al. (1995a); Kornberg and Pricer (1951); Munilla-Moran and Stark (1989); Munilla-Moran (1994); Packard et al. (1996a); Roy and Packard (2001)). This is the first time NADP-IDH has been measured in marine plankton. Thus, in order to make this assay sensitive to all its components, we fractionated plankton samples into three separated size-fractions. We measured NADP-IDH activity in the Canarian microplankton (0.7 to 50μm), microzooplankton (50 to 200μm,), and mesozooplankton (200-2000μm). These microplankton samples are mixed communities with heterotrophic organisms (procaryotes, nanoflagellates and also small dinoflagellates) and also autotrophs (cyanobacteria, other eukaryotic picoplankton, autotrophic nanoflagellates and some diatoms) (Schmoker et al. (2013)). The microzooplankton community around the Canary Islands is dominated by heterotrophs

(large ciliates and dinoflagellates). Finally, Canarian mesozooplankton is also dominated by heterotrophs, largely crustacean copepods (Hernández-León et al. (2007)).

In summary, here we propose a new method to assess NAPD-IDH activity in the marine plankton community. This tool may be useful in further studies on CO<sub>2</sub> production in the water column, lipid synthesis in the base of the food chain, and in studies on the ROS scavenging capacity of the marine plankton resources.

## 2.2 Material and Methods

Assessing NADP-IDH activity, accurately and reproducibly, requires determinations of the optimum substrate concentration, optimum cofactor concentration (sections 2.2.4 and 2.3.1), and optimum pH (sections 2.2.5 and 2.3.2). Also, the nature of the temperature dependence (sections 2.2.6 and 2.3.3) and the biomass linearity (sections 2.2.8 and 2.3.4) need to be determined (Suelter et al. (1985)). In addition, other factors such as the optimum time and speed of centrifugation (sections 2.2.7 and 2.3.5); as well as the stability of the enzyme preparation (sections 2.2.9 and 2.3.6), were assessed. Quantifying these factors not only makes the assay accurate and reproducible, but it stimulates the reaction to emit its maximum signal. Other factors that enhance the assays accuracy have been included in the Appendix A. These include: buffer sensitivity (Appendix A.1); extinction coefficient (Appendix A.2); precision (Appendix A.6); and spectrophotometric response (Appendix A.7).

### 2.2.1 Plankton sampling procedures

Zooplankton were collected off the north coast of Gran Canaria Island (GCI) from the upper 10m of the euphotic zone, stored alive in plastic containers until fractionated through 2000µm, 200µm, and 50µm nets in the laboratory, and finally stored for less than 3 months in 2mL microcentrifuge tubes at -80°C until enzyme analysis (Ahmed et al. (1976)). Microplankton were sampled from Alcaravaneras Beach, GCI. Seawater samples were stored in 25L high-density polyethylene containers until they were filtered through a 50µm net and then through a 0.7µm glass fiber filter (GF/F) in the laboratory. Note that the use of a 0.7µm filter instead of a 0.2µm, means a large percentage of the bacterial and archaeal community were not included in this measurement. After being blotted to remove excess seawater, all samples were stored, for less than 3 months, on the GF/F filters, in 2mL microcentrifuge tubes, at -80°C until analysis (Ahmed et al. (1976)).

### 2.2.2 Crude homogenate preparations and protein determination

The zooplankton samples from above, were homogenized in 0.1M phosphate buffer (0.1M Na<sub>2</sub>HPO<sub>4</sub>, 0.1mM KH<sub>2</sub>PO<sub>4</sub>, 75μM MgSO<sub>4</sub>· 7H<sub>2</sub>O, PVP (1.5mg mL<sup>-1</sup>), TRITON X-100) (Appendix A.1), by sonication for 45sec at 70% amplitude in a Vibracell VCX 130 Sonics® ultrasonic processor maintained at 0-4°C. Zooplankton homogenates ranged from 1.2 to 15mL. Microplankton samples were homogenized at 0-4°C in the same 0.1M phosphate buffer, with a Teflon® 2mL-pestle PYREX® Potter-Elvehjem tissue grinder at 2600rpm for 2 min. Microplankton homogenate volumes ranged from 1.0 to 1.2mL. Homogenate volume ( $V_t$ ) was measured to 50μL. Crude homogenates were centrifuged in 2mL microcentrifuge tubes at 0-4°C and 4000rpm in an 8.4cm-radius centrifuge rotor (giving 1500 g) for 10 min. Biomass, as mg of protein, was determined from subsamples of the crude homogenate. The protein method of Lowry et al. (1951) modified by Rutter (1967) was used. Bovine serum albumin was the standard.

### 2.2.3 NADP-IDH assay

NADP-IDH activity was measured kinetically by following the time dependent production of NADPH ( $\Delta_{NADPH}$ ) at 340nm ( $\Delta_{340}$ ) and 18°C in a 1cm plastic cuvette ( $l$ ) in a Cary 100 U-V visible spectrophotometer for 1000sec. To be clear,  $\Delta_{340}$  is the time-dependent absorbance increase at 340nm of the reaction in the spectrophotometer. [NADPH], at any time, was determined by the Beer-Lambert law (2.1) from the absorbance ( $A_{340}$ ), the molar absorption coefficient for our reactive solution ( $\varepsilon$ ), and the optical pathlength of the cuvette ( $l$ ) (see Appendix A.2).

$$A_{340}=l \cdot \varepsilon \cdot [NADPH] \quad (2.1)$$

So, transposing and using 5.42 absorbance units mL μmol<sup>-1</sup>cm<sup>-1</sup> for  $\varepsilon$ , we obtained Eq.2.2 for the calculation of NADPH in mmol<sup>-1</sup> mL<sup>-1</sup>:

$$[NADPH]=A_{340}/(5.42 \cdot l) \quad (2.2)$$

If  $A_{340}$  is followed kinetically over time (t), then the slope ( $A_{340}$ ) becomes  $\Delta_{340}$  and Eq.2.2 becomes:  $[\Delta_{NADPH}]=\Delta_{340}/(5.42 \cdot l)$ . The reaction was carried out in 0.5mL of 0.1M phosphate buffer (pH 8.5), with 6 mM MgCl<sub>2</sub> (PANREAC #131396), 2.1 mM DL-trisodium-isocitrate (SIGMA #I1252; which, from here on, we will call, isocitrate) and 0.3 mM β-NADP<sup>+</sup> (SIGMA #N0505; which, from here on, we will call, NADP<sup>+</sup>) in the cuvette. The blank assay ( $B_{340}$ ) controlled for the background activity of other enzymes and the non-enzymatic reduction of NADP<sup>+</sup>. It lacked isocitrate. Occasionally, to check for contamination and to assess any non-enzymatic reaction between NADP<sup>+</sup> and isocitrate, an assay was run without homogenate. Normally, contamination and any non-enzymatic reaction

between NADP<sup>+</sup> and isocitrate are negligible. For all assays and blanks, the reaction volume in the cuvette ( $V_c$ ) was maintained at 0.5 mL with buffer. The assay consisted of sequential additions of 0.3mL of a solution containing 10mM MgCl<sub>2</sub> and 3.5mM isocitrate, 0.1 mL homogenate ( $V_h$ ), and 0.1mL of 1mM NADP<sup>+</sup> to a 1mL cuvette. NADP<sup>+</sup> started the reaction. After 3 min, the linear signal ( $\Delta_{340}$ ) represented the NADP-IDH activity.  $\Delta_{340}$  depended on  $V_c$ . To relate  $\Delta_{340}$  per mL of the solution to the original homogenate which represents the sample ( $V_s$ ),  $\Delta_{340}$  must be multiplied by the factor, ( $V_s \cdot V_c/V_h$ ) (Eq.2.3). Since the original units of  $\Delta_{340}$  are cm<sup>-1</sup> sec<sup>-1</sup>, it must be multiplied by 60 sec min<sup>-1</sup>, multiplied by 1 cm (pathlength), and divided by 5.42 absorbance units mL μmol<sup>-1</sup> cm<sup>-1</sup> ( $\epsilon$ ) (Eq.2.3) (Appendix A.2). This calculation is for the production of NADPH per time, per sample:

$$\Delta_{NADPH} = ((\Delta_{340} - B_{340}) \cdot 60 \cdot V_s \cdot V_c) / (5.42 \cdot V_h) \quad (2.3)$$

A unit analysis of Eq.2.3 is:

$$\Delta_{NADPH} [\mu\text{mol min}^{-1} \text{ mL}^{-1}] = (\Delta_{340} - B_{340}) [\text{absorbance units cm}^{-1} \text{ sec}^{-1}] \cdot 60 [\text{sec min}^{-1}] \cdot V_s [\text{mL}] \cdot V_c [\text{mL}] / (5.42 [\text{absorbance units cm}^{-1} (\mu\text{mol} \cdot \text{mL}^{-1})^{-1}] \cdot V_h [\text{mL}])$$

Since CO<sub>2</sub> is produced stoichiometrically 1:1 with NADPH, this results in NADP-IDH activity in units of mM CO<sub>2</sub> min<sup>-1</sup> per sample. Note that, to promote oceanographic significance, we do not follow the biochemical convention of enzyme units. The NADP-IDH activity could be reported for microplankton in terms of seawater volume; for zooplankton field samples in terms of net-haul volume, or for physiological/biochemical experiments in terms of protein.

#### 2.2.4 Kinetic properties of NADP-IDH activity

The affinity between substrates and the mineral cofactor (Mg<sup>2+</sup>) and NADP-IDH was determined in the three plankton size-fractions (Section 2.3.1). Their optimum concentrations, their effective Michaelis-Menten constants ( $K_m$ s) and the NADP-IDH maximum velocity ( $V_{max}$ ) were calculated from Michaelis-Menten and Hanes-Woolf plots (Suelter et al. (1985)). First, using the same homogenate at pH 8.5 and 18°C, NADP-IDH activity was measured, in triplicate, for ten different concentrations of isocitrate. These ranged from 0.002 to 2.1mM in the cuvette, while MgCl<sub>2</sub> and NADP<sup>+</sup> were held constant at 2mM and 0.3mM, respectively. Second, NADP-IDH activity was measured in the same homogenate, at the same conditions as above, but with isocitrate being held at its optimum, 2.1mM, while 10 concentrations of NADP<sup>+</sup> from 0.4 to 400μM in the cuvette, were tested. Finally, to determine the optimum concentration of MgCl<sub>2</sub>, the same procedure was followed. NADP-IDH activity was measured in the same homogenate, at the same conditions as above, with 2.1mM isocitrate, and 0.3mM NADP<sup>+</sup> (its optimum), while 9 concentrations of MgCl<sub>2</sub> from 0.05 to 12mM, were tested. Before each experiment, pilot experiments were performed in order to determine the concentra-

tion ranges needed to minimize errors in the linearisation of the Michaelis-Menten equations.

We tried three linearisation techniques to evaluate the  $K_m$ s (Appendix A.3). Hanes-Woolf linearisation was the most precise for our data base ( $p<0.001$  in the whole community). In Hanes-Woolf plots, substrate concentration [S] was plotted against the ratio of concentration to enzymatic activity ([S]/V). The  $V_{max}$  was the inverse of the slope (a) (i. e.,  $V_{max} = 1/a$ ), whereas  $K_m$ s were equal to the ratio of the Y intercept ( $y_0$ ) and a ( $K_m = y_0/a$ ) (Suelter et al. (1985)).

## 2.2.5 pH optimum

The effect of pH on NADP-IDH activity was determined on mesozooplankton samples at 18°C. Assays were run using phosphate buffer at eleven levels of pH from pH 5.5 to pH 11 (Section 2.3.2 and Appendix A.4). The buffer solution was the same as used for homogenization, reagent solutions, and the kinetic assay. Assays were done kinetically, in quintuplicate, with 0.2mM NADP<sup>+</sup>, 2.1mM isocitrate and 2mM MgCl<sub>2</sub> in the cuvette.

## 2.2.6 Temperature dependence

The temperature (T) dependence of activity in NADP-IDH in crude mesozooplankton homogenates and in pure NADP-IDH from porcine heart (Sigma Aldrich #I2002) was determined at pH 8.5 (Section 2.3.3). Assays were conducted, in triplicate, under optimum conditions, at nine temperatures ranging from 5 to 60°C. The Arrhenius equation describes the temperature dependence of chemical reaction rates (Eq.2.4 (Arrhenius (1889)):

$$k_{IDH}=A \cdot \exp(-E_a/(R \cdot T)) \quad (2.4)$$

where:  $k_{IDH}$  is NADP-IDH activity;  $A$  is the preexponential frequency factor in the same units as  $k_{IDH}$ . It is a unique value for each chemical reaction and is related to the collision frequency when reactants are in the correct orientation.  $E_a$  is the energy of activation in Kcal mol<sup>-1</sup>;  $T$  is the absolute temperature in °K and  $R$  is the universal molar gas constant in Kcal mol<sup>-1</sup> °K<sup>-1</sup> (1.9872 · 10<sup>-3</sup> Kcal mol<sup>-1</sup> °K<sup>-1</sup>). The  $E_a$  and  $A$  were obtained from the Arrhenius plot (Fig. 2.3). Eq.2.5 describes the linear regression from this Arrhenius plot ( $y=a \cdot x + y_0$ ), where  $x$  is 1/T, the slope (a) is  $-E_a/R$ , and y intercept ( $y_0$ ) is  $\ln A$ :

$$\ln(k_{IDH})=\ln(A)-(E_a/R) \cdot (1/T) \quad (2.5)$$

The enthalpy of activation ( $\Delta H^*$ ), in Kcal mol<sup>-1</sup>, for 25°C was calculated from the  $E_a$  according to Eq.2.6 (Low et al. (1973)):

$$\Delta H^*=E_a-R \cdot T \quad (2.6)$$

Here, T is expressed in °K (298°K).

The effect of T on NADP-IDH was also assessed by measuring the  $Q_{10}$  temperature coefficient, a unitless factor by which the activity increases for every 10-degree rise in T (Przibram (1923)).  $Q_{10}$  was measured according to Eq.2.7, where NADP-IDH<sub>T2</sub> was the NADP-IDH activity at T<sub>2</sub> (T<sub>2</sub>=25°C), and NADP-IDH<sub>T1</sub> was the NADP-IDH activity at T<sub>1</sub> (T<sub>1</sub>=15°C).

$$Q_{10} = (\text{NADP-IDH}_{T_2}/\text{NADP-IDH}_{T_1})^{10/(T_2-T_1)} \quad (2.7)$$

### 2.2.7 Centrifugation studies

Homogenate centrifugation separates cell membranes and organelles from the cytosol, it clarifies the homogenate, and increases assay sensitivity. Effects of centrifugation on NADP-IDH activity were analysed at pH 8.5 and 18°C. Five experimental conditions were tested: (1) a blank without centrifugation, where activity was measured after 8min of gravitational sinking; (2) 2min at 1000 rpm (375 g); (3) 2min at 2000rpm (750 g); (4) 2min at 4000rpm (1500 g); (5) 10min at 4000 rpm (1500 g). Measurements were made in quintuplicate. NADP-IDH activity was reported in units of  $\mu\text{mol NADPH min}^{-1}(\text{mL of homogenate})^{-1}$  units and normalized by V<sub>max</sub> to give a percentage.

### 2.2.8 Linearity with enzyme concentration and limit of detection

Establishing linearity with enzyme concentration in crude homogenates is essential for any enzyme assay. Here, it was tested for microplankton and mesozooplankton by diluting original homogenates (Section 2.3.4). Protinaceous biomass was simultaneously diluted and proxied for NADP-IDH concentration. Seven dilutions from 5% ( $4\mu\text{g protein mL}^{-1}$ ) to 100% ( $83\mu\text{g protein mL}^{-1}$ ) of the original were made for microplankton and nine dilutions, from 0.39% ( $3\mu\text{g protein mL}^{-1}$ ) to 100% ( $684\mu\text{g protein mL}^{-1}$ ) were made for the mesozooplankton. Activity was measured at 18°C, with the optimised assay. The limit of detection was determined by plotting the NADP-IDH activity against the homogenate protein (Fig.2.4), as the lowest biomass with a standard deviation lower than 50% of the NADP-IDH activity.

### 2.2.9 Stability of the crude homogenate

Stable enzyme activity in crude homogenates promotes ease, flexibility and dependability of analysis. We assessed NADP-IDH activity stability in Mysids (*Lepidomysis lingvura*). Samples were collected off Taliarte, GCI by scuba using an 8" net, transferred to the lab as were the zooplankton (above), gently pipetted into 2mL microcentrifuge tubes, and stored at -80°C until analysis. Mysids were homogenized by sonication as explained above. The homogenate was always main-

tained on ice. Loss of NADP-IDH activity per time after homogenisation was measured at 18°C, under optimum assay conditions (above). Activity was measured in triplicate, every 90min, for 8h; and was reported in  $\mu\text{mol NADPH min}^{-1}$ (mL of homogenate) $^{-1}$ . Then it was normalized by  $V_{max}$  to give a percentage of maximum activity (Section 2.3.6).

### 2.2.10 Statistics of the substrate and temperature kinetics

In substrate kinetic studies (Section 2.3.1), *F* tests (Pentecost (1999)) quantified the differences between the logarithmically transformed Michaelis-Menten curves ( $\ln V$  vs  $\ln [S]$ ). *F* tests were also used with the Arrhenius plots (Section 2.3.3). The *t* test (Sachs (2012)) quantified differences between the regression coefficients of the Arrhenius plots.

## 2.3 Results

### 2.3.1 Kinetic properties

Substrate, cofactor and effector concentrations were determined to optimize the conditions of the enzyme assay. As standard practice in developing an enzyme assay, NADP-IDH kinetic constants were calculated from Michaelis-Menten plots (Fig.2.1 and Appendix A.3) and Hanes-Woolf linearisation. The average  $K_m$ s were  $271 \pm 63 \mu\text{M}$  for isocitrate, and  $18 \pm 3 \mu\text{M}$  for NADP<sup>+</sup>. For the metal cofactor MgCl<sub>2</sub>, that stimulated NADP-IDH activity following a rectangular-hyperbola, its concentration at 1/2  $V_{max}$  was  $3060 \pm 643 \mu\text{M}$  (Appendix A.3). For isocitrate,  $V_{max}$  was 0.038 in mesozooplankton; 0.031 in microzooplankton; and 0.002 in micropaplankton. For NADP<sup>+</sup>,  $V_{max}$  was 0.029 in mesozooplankton; 0.019 in microzooplankton; and 0.001 in micropaplankton. For MgCl<sub>2</sub>,  $V_{max}$  was 0.010 in mesozooplankton; 0.009 in microzooplankton; and 0.00002 in micropaplankton.  $V_{max}$  is in  $\mu\text{mol NADPH min}^{-1}$  (mL homogenate) $^{-1}$  units.

The optimal values to measure NADP-IDH activity (100%  $V_{max}$  in Fig.2.1) in this community are 2mM for isocitrate and 0.3mM for NADP<sup>+</sup> (Fig.2.1). By inspection of Fig.2.1c, we conclude that, although 4mM of MgCl<sub>2</sub> would be enough to reach the  $V_{max}$  in the zooplankton fractions, an optimal value of 6mM for MgCl<sub>2</sub>, is needed to ensure that all the community reaches 100% $V_{max}$ .

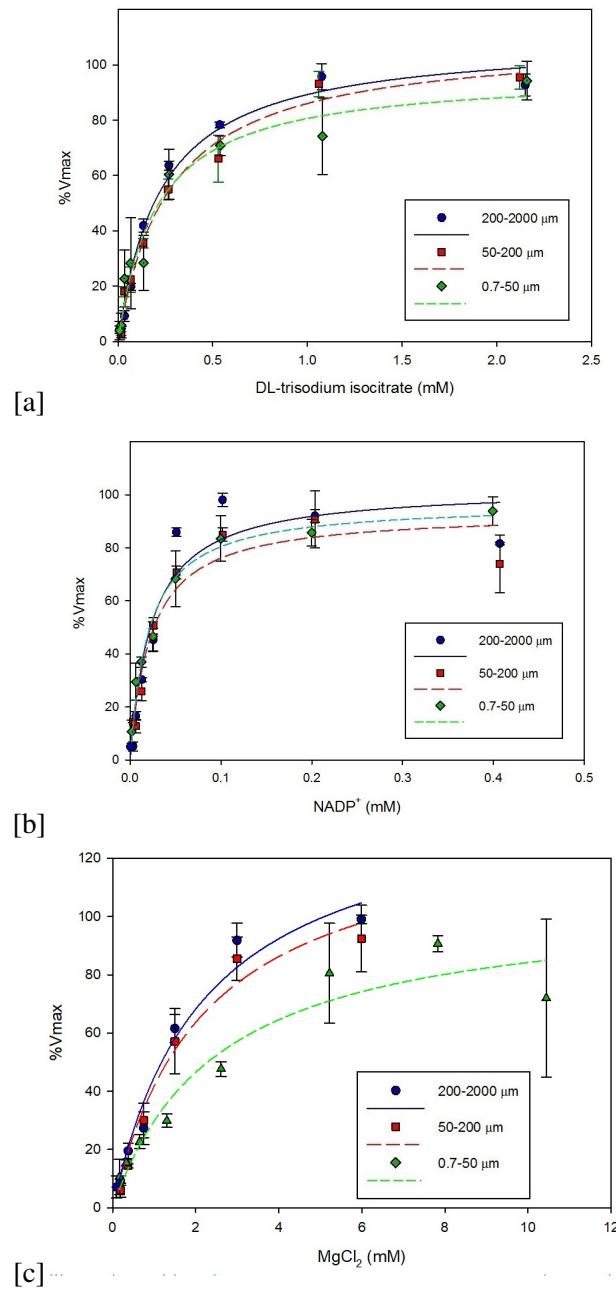


Figure 2.1: Michaelis-Menten plots to determine the NADP-IDH  $K_m$  and optimum values for isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub> for different plankton community size fractions. (a) Affinity between NADP-IDH and isocitrate. (b) Affinity between NADP-IDH and NADP<sup>+</sup>. (c) Affinity between NADP-IDH and Mg<sup>2+</sup>.

### 2.3.2 Effect of pH

The effect of pH on the catalytic activity of the NADP-IDH was measured on phosphate buffer at 18°C, with the same concentration of ligands. In our hands, the maximum activity was obtained at pH 8.19 (Fig.2.2), whereas measurements below pH 7.62 and above pH 8.75 led to activities lower than 50% of its  $V_{max}$  ( $pK_{as}$ ) (Appendix A.4).

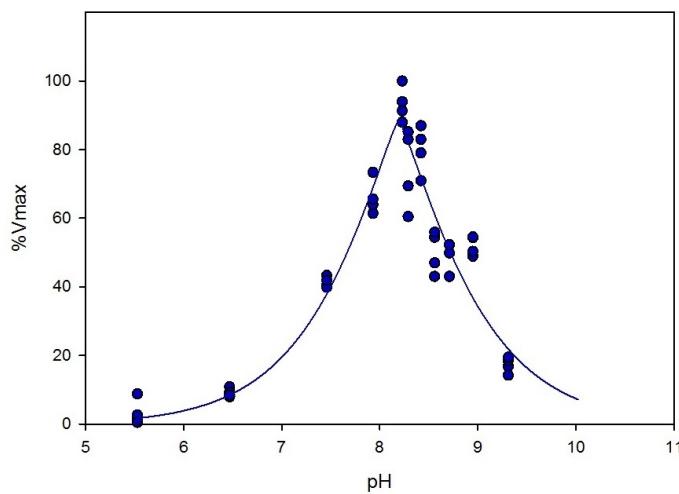


Figure 2.2: pH effects on mesozooplankton NADP-IDH activity, where  $V_{max}$  was  $0.027 \mu\text{mol NADPH min}^{-1} (\text{mg protein})^{-1}$ .

### 2.3.3 Effect of Temperature

The relationship between NADP-IDH activity and temperature was studied on mesozooplankton samples and compared to the results from pure porcine-heart NADP-IDH (Fig.2.3). Both activities increased with temperature before reaching an optimum temperature, 27.6°C on the subtropical mesozooplankton samples, and 38.6°C on the pure porcine-heart NADP-IDH ones. At higher temperatures, the activity decreased. Besides,  $Q_{10}$  between 15°C and 25°C was 1.98 for mesozooplankton samples. Although this value is higher than the NADP-IDH in turbot larvae between 20-35°C (1.74) (Munilla-Moran (1994)), it is consistent with the  $Q_{10}$  for respiratory rates in epipelagic marine copepods (1.8-2.1) (Ikeda et al. (2001)). From this data set, the Arrhenius energy of activation ( $E_a$ ) had a value of 20.4 Kcal mol<sup>-1</sup> for the subtropical mesozooplankton samples and 21.40 Kcal mol<sup>-1</sup> for the pure enzyme from porcine heart; the Arrhenius frequency factor ( $A$ ) was  $2.19 \cdot 10^{11} \text{ mol NADPH s}^{-1} (\text{kg of protein})^{-1}$  for the mesozooplankton samples and  $1.60 \cdot 10^{13}$  for the pure porcine heart NADP-IDH; and finally, the Arrhenius enthalpy of activation at 25°C ( $\Delta H^*$ ) was 19.81 Kcal mol<sup>-1</sup> for the mesozooplankton

samples and 20.76 Kcal mol<sup>-1</sup> for the porcine heart NADP-IDH. These results suggested that, for NADP-IDH, although the frequency factor (*A*) was different (*F* test,  $F_{(2,37)}=4.64$ ,  $p=0.05$ ), *Ea* and  $\Delta H^*$  were not significantly different between the two types of organisms (*t* test,  $t(39)=0.0773$ ,  $p=0.05$ ).

This permits the development of a common Arrhenius equation for both types of NADP-IDH. When the in-situ T ( $T_i$ ) is different from T during the assay ( $T_m$ ), the measured NADP-IDH activity needs to be corrected for the difference. The in-situ activity (NADP-IDH( $T_i$ )) is calculated using the Arrhenius equation (Atkin and de Paula (2006), Appendix A.5), based on Arrhenius *Ea* (Eq.2.8):

$$\text{NADP-IDH}(T_i) = \text{NADP-IDH}(T_m) \cdot \exp^{((20.39/R) \cdot (1/T_m - 1/T_i))} \quad (2.8)$$

where: NADP-IDH( $T_i$ ) is NADP-IDH activity at  $T_i$ ; NADP-IDH( $T_m$ ) is NADP-IDH activity at  $T_m$ ; and  $R$  is  $1.9872 \cdot 10^{-3}$  Kcal mol<sup>-1</sup> °K<sup>-1</sup>.

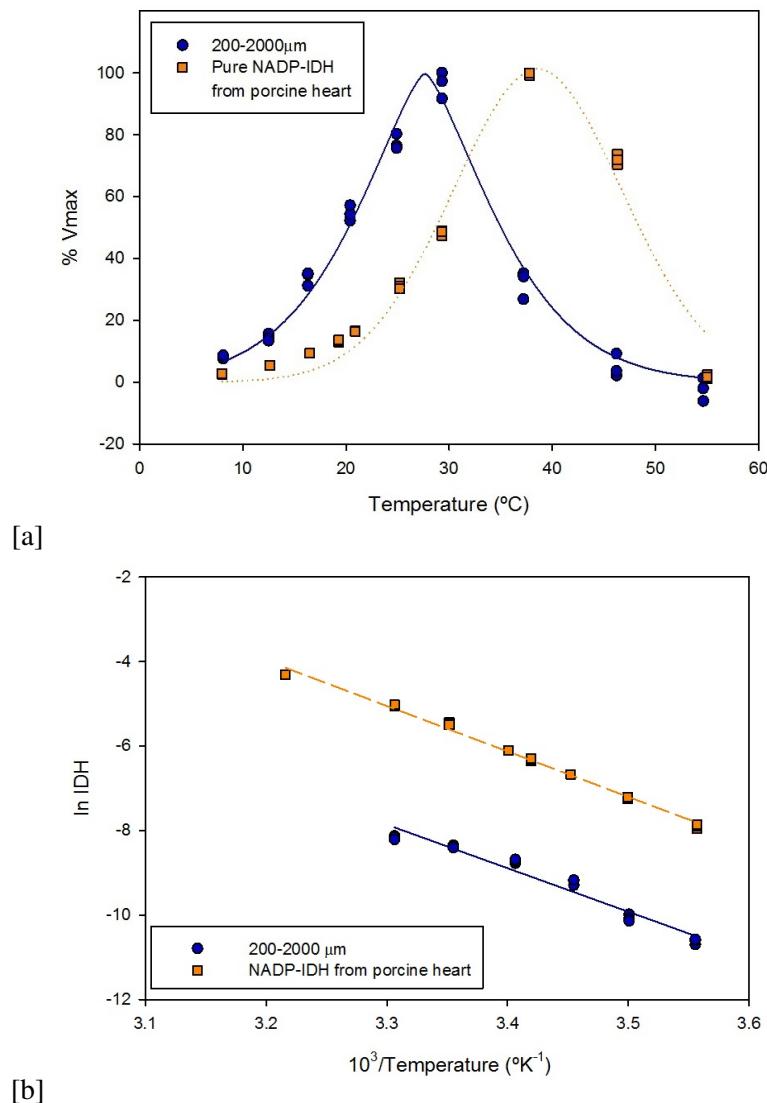


Figure 2.3: Temperature dependence of NADP-IDH activity. (a) Temperature impact on NADP-IDH activity, where  $V_{max}$  was  $0.018\mu\text{mol NADPH min}^{-1}$  ( $\text{mg of protein})^{-1}$  in mesozooplankton and  $0.808\mu\text{mol NADPH min}^{-1}$  ( $\text{mg of protein})^{-1}$  in pure porcine heart enzyme. (b) Arrhenius plot, where  $\ln(k_{IDH})$  is plotted against  $1/T$  and  $T$  is in  $^{\circ}\text{K}$  units.

### 2.3.4 Biomass linearity and limit of detection

Once optimum conditions were applied, the NADP-IDH could be measured with a standard deviation less than 3.5% on every ten duplicates (Appendix A.6). To determine the limit of detection, NADP-IDH was measured in seven different dilutions using both the mesozooplankton and the microplankton homogenates. The linearity in Fig.2.4, shows that the relationship between NADP-IDH and biomass holds even when samples are diluted to 1% of the original strength. The limit of detection for mesozooplankton samples was  $10 \pm 2 \mu\text{g protein mL}^{-1}$  and for microplankton samples, it was  $8 \pm 2 \mu\text{g protein mL}^{-1}$ .

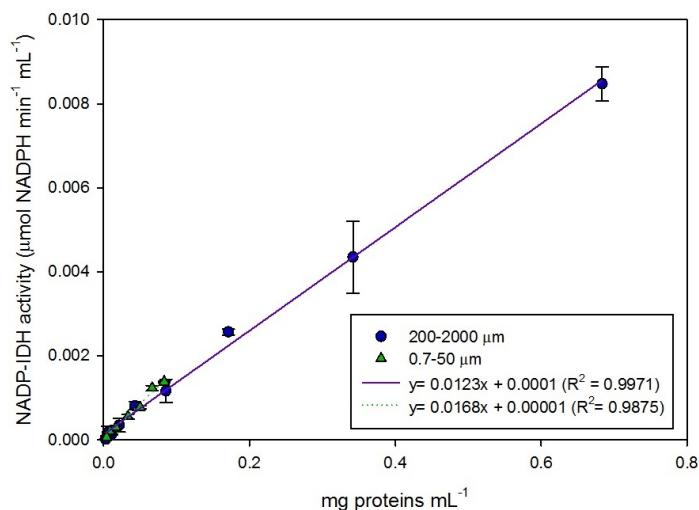


Figure 2.4: Biomass linearity of marine-microplankton and marine-mesozooplankton NADP-IDH. Enzyme activity as a function of protein in the homogenate.

For microplankton samples under oligotrophic conditions, this biomass can be attained by filtering 4L through 0.7 GF/F glass fiber filters, whereas during eutrophic conditions, 2.5L were enough. In deep water samples, more volume will be required (10L or even 20L) to stay above the limit of detection. For unknown conditions, and also for mesozooplankton samples, pilot measurements to determine the optimum seawater volume needed should be made (see Section 2.2).

### 2.3.5 Effect of centrifugation

GF/F filtered samples (microplankton, green columns in Fig.2.5) need centrifugation greater than 1500g for 2min to achieve the maximum NADP-IDH activity and a low standard deviation (12.6 % of the activity). Note that without centrifugation the signal is obscured by the turbidity of the filter fibers.

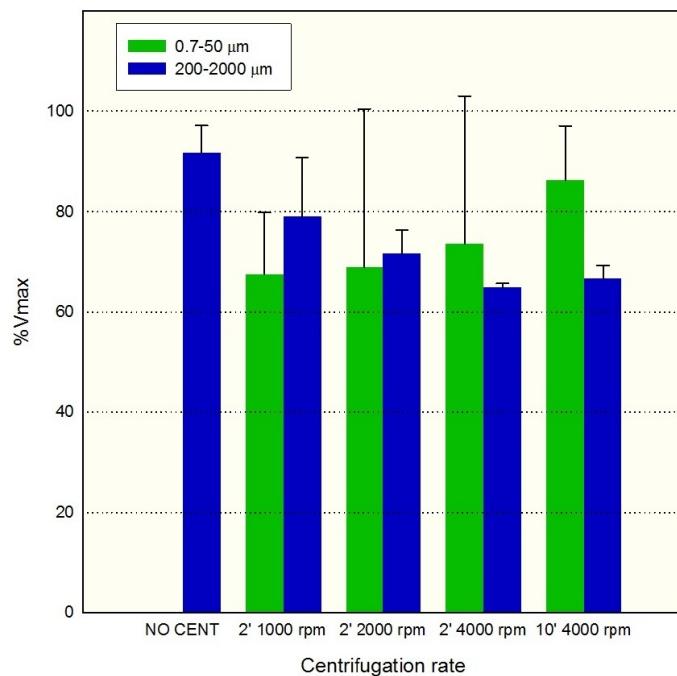


Figure 2.5: Centrifugation effects on NADP-IDH activity, where  $V_{max}=0.0073$  in mesozooplankton and  $0.0008$  in microplankton.  $V_{max}$  is in  $\mu\text{mol NADPH min}^{-1}$  ( $\text{mL homogenate}^{-1}$ ) units. Note that, for our centrifuge, a force of 375g was achieved at 1000rpm; 750g was achieved at 2000rpm; and 1500g was achieved at 4000rpm.

Sonicated samples (mesozooplankton, blue columns in Fig.2.5), on the other hand, are not improved by centrifugation. However, part of the activity reached without centrifugation is related to a negative blank, which may indicate that either other enzymes are consuming NADPH or that particle sinking has interfered with the signal. As both problems are solved by centrifugation, we argue that sonicated samples should be centrifuged at a minimum of 375g for 2min.

### 2.3.6 Stability of the enzyme preparation from mysid samples

NADP-IDH activity decreased continuously (Fig.2.6) during the 500min experiment. However, the regression coefficient of 0.109 ( $r^2=0.9771$ ,  $p<0.0001$ ) indicates an average loss of 1.1% of NADP-IDH activity per 10min. This means that only 3% of the activity is lost if analyses are run within half-hour after centrifugation.

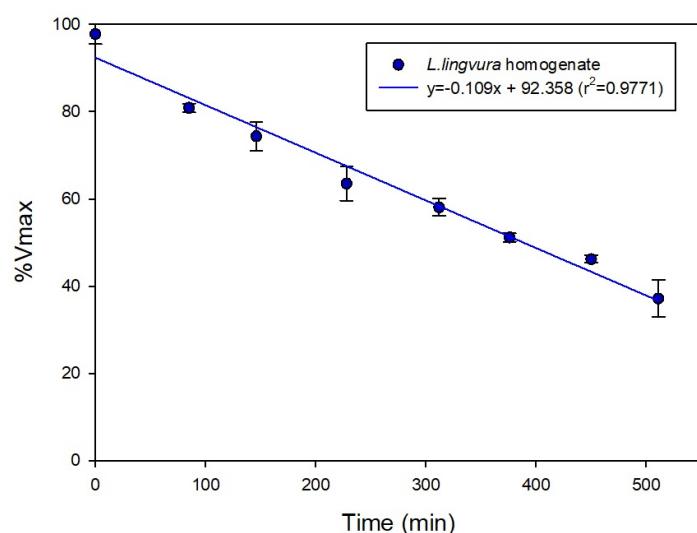


Figure 2.6: Stability of the enzyme preparations from *L. lingvura* samples, where  $V_{max}=0.0085\mu\text{mol NADPH min}^{-1} (\text{mL homogenate})^{-1}$ .  $T_0$  is the end of the centrifugation process.

## 2.4 Discussion

### 2.4.1 What does the development of the assay reveal about plankton NADP-IDH?

While determining the conditions for assaying the  $V_{max}$  of NADP-IDH, we also determined the  $K_m$  for isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub> (Fig. 2.1, Appendix A.3). Earlier research on IDH in marine bacteria cultures developed mathematical models to determine the physiological  $R_{CO_2}$  (Roy and Packard (2001); Packard et al. (1996a)), but the research was not extended to ocean microplankton and zooplankton. Here, the older IDH assay (Berdalet et al. (1995b)) has been revised, modified, and updated for ocean plankton samples. As always in science, additional research is still needed. However, we predict that, given this updated assay, new models based on measurements of NADP-IDH, isocitrate and NADP<sup>+</sup> in marine plankton will result in improved assessment of marine  $R_{CO_2}$ . In addition, being able to measure NADP-IDH opens up new research avenues for plankton research. For example, cytosolic NADP-IDH measurements may facilitate the assessment of lipid synthesis at the base of the marine food chain (Koh et al. (2004)). Also, because Wieckowski et al. (2009) have greatly improved the isolation of mitochondria from cytosol, mitochondrial NADP-IDH measurements may proxy the ROS scavenging capacity of cells (Jo et al. (2001)).

On marine plankton samples, NADP-IDH measurements should be done maintaining the buffer, pH levels, and the Arrhenius relationship developed in this study. If these conditions are changed, the  $K_m$ s would need to be reassessed (Olano et al. (1995)). In our hands, isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub>  $K_m$ s were up to one order of magnitude higher than other reported values (Munilla-Moran (1994)). These differences can be attributed to our specific chemical conditions (buffer, pH, T) that are likely responsible for the variations in the affinity between enzymes and their substrates (Olano et al. (1995)).

Michaelis-Menten constants are affected by pH levels that impact ionisation around the enzyme active site. Our results, with an optimum pH of 8.2 (section 2.3.2), are consistent with those reported in turbot larvae (optimum pH between 7.8 and 9.0) (Munilla-Moran (1994)). The p*K<sub>a</sub>*s of 7.62 and 8.75, which are close to the optimum value, indicate a high enzyme sensitivity to pH changes. As the pH involved in the reaction is intracellular, it is controlled by cellular membranes so slight environmental pH changes are unlikely to affect the intracellular enzymatic activity (Simčič and Brancelj (2006)). However, the influence of pH on the metabolic response (in the short, medium and long term) of the plankton community should be assessed in further research, particularly in the current scenario of increasing ocean acidification.

Another key factor that impacts enzyme kinetics is temperature (T). The NADP-IDH activity in the subtropical mesozooplankton drops as T rises above 27.6°C, whereas in pure NADP-IDH from porcine heart, this occurs above 38.6°C. This

decreasing activity can be related: (1) to denaturation of the enzyme; (2) to a direct effect of T on the collision history at any stage of the reaction (formation of the enzyme-substrate complex; conversion of the enzyme-product complex; or dissociation of the product); (3) or to the ionisation of the different groups comprising the active site. This will change the substrates'  $K_m$ s even under the same pH conditions (Dixon and Webb (1979)). Under a global warming scenario, further research on the T effects on plankton metabolism and the biochemical response will improve our knowledge about the magnitude and reversibility of the negative high-temperature effects on respiratory metabolism and physiology. These future studies will be relevant for poikilotherms, taking into account that environmental temperatures higher than the optimum (25-30°C for subtropical mesozooplankton) will affect the enzymatic structure that supports the metabolism of these organisms. High-latitude plankton communities, likely with a lower optimum T, should be particularly studied. In the case of homoeothermic samples (from porcine heart) the optimum T is around ten degrees higher than in poikilothermic samples. However,  $E_a$  and  $\Delta H^*$  are not significantly different, being also consistent with  $E_a$  reported by Munilla-Moran for NADP-IDH in turbot (around 23.5 Kcal mol<sup>-1</sup> on turbot liver and 23.3 Kcal mol<sup>-1</sup> for turbot larvae) and by Moon and Hochachaka for cold-adapted and warm-adapted NADP-IDH in rainbow-trout liver (18 Kcal mol<sup>-1</sup>) (Munilla-Moran (1994); Munilla-Moran and Stark (1989); Moon and Hochachaka (1971)). Thus, although there seems to be an adaptation to homoeothermic conditions allowing high activity at higher temperatures, a common basic structure seems to have been maintained even after differences in evolution.

#### **2.4.2 Why should this metabolic assay be measured kinetically?**

While studying enzyme activities, two approaches can be used: (1) Two point measurements (endpoint); and (2) time-course monitoring (kinetics). Endpoint measurements are only feasible, with reliable results, when the reaction is linear, and this condition must be frequently validated (Škorjanc and Pette (1998)) unless the behaviour of the sample is well-known. Here, all measurements were made kinetically, and even though the linear relationship was well maintained for more than 1000s, we realise that some conditions, such as low and high biomass samples, the lag until thermal equilibrium in the cuvette, or the blank signalling may change the linearity of the reaction (see Appendix A.7), rendering endpoint detection unsuitable. From these different observations we conclude that by measuring the  $A_{340}$  kinetics for 1000 seconds, the NAPD-IDH activity can be determined with confidence. This time-course analysis demonstrates the impact of pH, T, dilution, substrate concentration, biomass and centrifugation (see Appendix A.7). This visualisation, not detected by end-point approach, lends itself to modifying assay conditions when needed. Thus, we recommend using kinetics for measuring NADP-IDH in oceanographic research.

## 2.5 Conclusions

- An enzyme assay for IDH (NADP-IDH) in zooplankton and microplankton, based on the reduction of NADP<sup>+</sup> has been developed.
- The apparent Michaelis-Menten constants for the substrates were  $271 \pm 63 \mu\text{M}$  for isocitrate and  $18 \pm 3 \mu\text{M}$  for NADP<sup>+</sup>.
- The rates of NADPH formation ( $\mu\text{mol NADPH min}^{-1} \text{ mL}^{-1}$ ) are obtained kinetically at 340nm. Activities in  $\mu\text{mol}$  of NADPH were based on a molar extinction coefficient ( $\epsilon$ ) of  $5.42 \text{ mL } \mu\text{mol}^{-1} \text{ cm}^{-1}$ .
- Each laboratory should verify  $\epsilon$  for their solutions and laboratory conditions to reduce error.
- Arrhenius plots for NADP-IDH in mesozooplankton samples and porcine heart yielded similar Energies of activation,  $20.4 \text{ Kcal mol}^{-1}$  and  $21.4 \text{ Kcal mol}^{-1}$ , respectively.

## Chapter 3

# Chapter 3: The Potential Respiratory CO<sub>2</sub> Production from NADP<sup>+</sup>-dependent isocitrate dehydrogenase in marine plankton.

*Biochemistry asks how the remarkable properties of living organisms arise from the thousands of different biomolecules.*

Albert L.Lehninger

**RESUMEN:** En este capítulo se ha medido la producción de CO<sub>2</sub> potencial máxima de las fracciones 0,7-50μm y 200-2000μm del plancton en la costa de las Islas Canarias, en base a la actividad de la NADP-IDH. Esta enzima es una de las principales responsables de la producción de CO<sub>2</sub> en la célula, pero no es la única. Por ello, tras considerar todas las enzimas productoras de CO<sub>2</sub> y normalizar su actividad por la de la NADP-IDH, la producción potencial máxima de CO<sub>2</sub> ( $\Psi_{NADP}$ ) del organismo resultó 2.94·NADP-IDH. Además, se definieron doce escenarios teóricos para el análisis de la relación entre las tasas fisiológicas de producción de CO<sub>2</sub> ( $R_{CO2}$ ) y  $\Psi_{NADP}$ . Los ratios  $R_{CO2}/\Psi_{NADP}$  obtenidos variaron entre 0,24 y 0,66 para el mesozooplancton y entre 0,15 y 0,28 para el microplancton. En base a nuestros resultados del Atlántico Nordeste Subtropical, predecimos que los ratios  $R_{CO2}/\Psi_{NADP}$  para el mesozooplancton en el océano abierto variarán entre 0,33 y 0,44, mientras que en el microplancton variarán entre 0,15 y 0,27. Consideramos que el uso de los ratios  $R_{CO2}/\Psi_{NADP}$  puede ser una herramienta muy útil en oceanografía. Podrían incrementar la eficiencia de las campañas aument-

tando la tasa de obtención de datos, y con ello, el conocimiento asociado. Estos ratios pueden ser especialmente interesantes en los estudios del flujo de carbono vertical, el estudio de la capacidad de secuestro de carbono del océano, la respiración celular, las adaptaciones metabólicas de los organismos ante cambios ambientales, y el papel de la comunidad planctónica en la cadena trófica.

**ABSTRACT:** Potential CO<sub>2</sub> production of the 0.7-50μm and of the 200-2000μm fraction from the Canary-Island coastal-plankton community was measured by an enzyme assay based on the NADP<sup>+</sup>-dependent isocitrate dehydrogenase (NADP-IDH). This enzyme plays a key role in the metabolism as one of the most active CO<sub>2</sub>-producing enzymes in the cell. Thus, considering all the CO<sub>2</sub>-producing enzymes and normalising their activities by NADP-IDH activity, the potential CO<sub>2</sub> production ( $\Psi_{NADP}$ ) was calculated as 2.94·NADP-IDH. The development of  $R_{CO_2}/\Psi_{NADP}$  ratios obtained from twelve different theoretical scenarios, led to a range of  $R_{CO_2}/\Psi_{NADP}$  ratios which spans from 0.24 to 0.66 for the mesozooplankton community and from 0.15 to 0.27 for the microplankton community. We predict that, from our NADP-IDH measurements in the North Eastern Subtropical Atlantic, the  $R_{CO_2}/\Psi_{NADP}$  ratios for open ocean mesozooplankton will fall between 0.33 and 0.44, and will be between 0.15 and 0.27 in the case of the mixed microplankton community. We propose the use of  $R_{CO_2}/\Psi_{NADP}$  ratios as a powerful tool in oceanographic research. They can be used to improve the efficiency of oceanographic field experiments and the knowledge that they yield. Specifically, NADP-IDH measurements can be used in studies of vertical carbon flux, sequestering capacity for anthropogenic CO<sub>2</sub> in plankton communities, cellular respiration in marine samples, metabolic adaptations to changes in environmental conditions, and the role of plankton community in the food chain.

### 3.1 Introduction

Global climate change and marine acidification linked to the emission of greenhouse gases are leading the marine scientific community to develop new methodologies that advance our knowledge of carbon flux (Cflux), its relationship with biogeochemical processes and, particularly, the importance of the ocean,s role as a carbon sink (Packard et al. (2015); Giering et al. (2014); Ducklow and Doney).

Among biogeochemical processes, technical limitations on the assessment of marine respiration in terms of carbon dioxide production ( $R_{CO_2}$ ), have restricted its direct measurement to the upper water column. Furthermore, these respiration measurements are derived largely from the product of respiratory oxygen consumption rates ( $R_{O_2}$ ) and a theoretical Respiratory Quotient (RQ, Mayzaud et al.

(2005); Berggren et al. (2012); Romero-Kutzner et al. (2015)). Thus, the calculation of  $R_{CO_2}$  is based firstly, on the circumstantially variable  $R_{O_2}$ . It depends on, e.g., the physiological, nutritional and life-cycle state of the organism, and also on the effects of sampling and incubation (Herrera et al. (2011b); Fernández-Urruzola et al. (2011); Osma et al. (2016a); Ferguson et al. (1984)). Secondly, it also depends on the RQ. Even though RQ has been widely used so far, not only does the type of consumed organic matter effect on its variability, but the nutritional state of the organism also affects it (Romero-Kutzner et al. (2015)). Here, we propose new methodology which avoids the variability associated with the RQ by measuring the enzymatic maximum velocity ( $V_{max}$ ) of a key  $CO_2$ -producing enzyme, the  $NADP^+$ -dependent isocitrate dehydrogenase (NADP-IDH) (Tames-Espinosa et al. (2018)), inherent to the plankton community metabolic apparatus. This kind of measurements, once optimized, describes the organisms' maximum capacity which, as a stable characteristic, remains so, even when environmental or circumstantial events modify the physiological response (Herrera et al. (2011b); Fernández-Urruzola et al. (2011); Osma et al. (2016a)). Although this methodology is based on the complex enzymology of the underlying biochemical processes that control the physiological  $CO_2$  production ( $R_{CO_2}$ ), we have developed a functional linear predictive model based on the measurement of the activity of NADP-IDH, one of the key enzymes involved.

Here, we assess the relationships among different  $CO_2$ -producing enzymes in the marine plankton samples, to develop a basic model to evaluate the potential  $CO_2$  production of the plankton community.

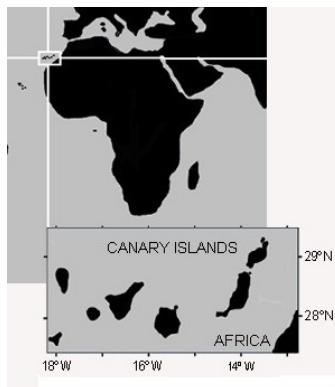


Figure 3.1: Canary Islands Archipelago, surrounded by waters of the North Eastern Subtropical Atlantic Ocean where most of the samples in this paper were taken.

To do this, we collected plankton samples from the coastal waters around Gran Canaria in the Canary Islands archipelago. In the Canary Current System that flows South through the islands (Fig.3.1), microplankton from 0.7 to 50 $\mu m$  is composed of heterotrophic prokaryotes, heterotrophic and autotrophic nanoflagellates, and autotrophic picoplankton (eukaryotic and the cyanobacteria *Prochlorococcus* and *Synechococcus*). All share the same Krebs cycle (Fig.1.1). In addition, diatoms of

the genera *Rhizosolenia*, *Chaetoceros* and *Navicula* and small dinoflagellates of the genera *Gymnodinium*, *Gyrodinium*, *Amphidinium*, *Cochlodinium* and *Torodinium* are also present within this fraction (Schmoker et al. (2013)). The 200-to-2000μm size-fraction of zooplankton is dominated by minute crustaceans, specially copepods (Hernández-León et al. (2007)). The most abundant species reported at our sampling depths (less than 20m) were *Oithona plumifera*, *Nannocephalus minor* and *Temora stylifera* throughout the year, and also *Acartia negligens* during spring and summer, the seasons when our sampling was accomplished (Corral (1970)).

To calculate these communities' CO<sub>2</sub> production from NADP-IDH activity, we derive a factor,  $\chi$ , obtained from the analysis of the rates of different CO<sub>2</sub> production enzymes in the Krebs cycle (Walsh and Koshland (1984)). However, the potential CO<sub>2</sub> production in the cells also involve non-respiratory CO<sub>2</sub> production. To ensure that our modelled CO<sub>2</sub> production predicts the potential, we have to take into account the non-respiratory CO<sub>2</sub> producing enzymes as well as the respiratory CO<sub>2</sub> producing enzymes. The potential CO<sub>2</sub> production model is based principally on NADP-IDH activity ( $\Psi_{NADP}$ ). However, it is developed by combining the CO<sub>2</sub> production rate analysis with the relationship between maximum activities of citrate synthase (CS), NAD-IDH and NADP-IDH in crustacean organisms. These relationships are based on enzyme analysis in crustaceans (Alp et al. (1976)) (Tables 3.2 and 3.3).

Finally, we develop an experimental base for a CO<sub>2</sub> production proxy (Eq.3.14, Table 3.4). We have assessed the  $R_{CO_2}$  theoretically to develop  $R_{CO_2}/\Psi_{NADP}$  ratios that range from 0 to 1. Thus,  $\Psi_{NADP}$  resulted comparable to measurements of the potential CO<sub>2</sub> production of the plankton community. This approach is the second step in developing a chemically-based proxy to rapidly detect the rate of respiratory CO<sub>2</sub> production on oceanographically relevant time and space scales (sensu Sverdrup et al. (1942)).

## 3.2 Material and Methods

### 3.2.1 Plankton sampling procedures

Mesozooplankton size fraction samples (200-2000μm) were collected from the shallow waters of Sardina del Norte (Gáldar, Gran Canaria island) by three scuba divers dragging a WP2 net horizontally in the upper 10m of the euphotic zone (Fig.3.2). After being stored in plastic containers, the samples were fractionated through 2000μm and 200μm nets at laboratory. Pilot measurements to determine the optimum seawater volume needed were made to obtain biomass higher than  $10 \pm 2 \mu\text{g protein mL}^{-1}$  (Tames-Espinosa et al. (2018)). Phytoplankton size fractions (from 0.7 to 50μm) were sampled from Alcaravaneras Beach (Las Palmas de Gran Canaria, Gran Canaria island). Seawater samples (4L per filter) were stored in 25L plastic containers and later passed through a 50μm net and 0.7μm non-combusted 25mm Glass Fiber Filter (GF/F) in the laboratory. Then, all samples



Figure 3.2: Sampling in the coastal waters of Gran Canaria using the WP2 net dragged by scuba divers during spring-bloom conditions.

were stored in 2mL microcentrifuge tubes at -80°C until analysis.

### 3.2.2 Preparations for kinetic enzyme activity measurements

The 200-2000 $\mu$ m plankton samples were homogenated in 0.1M phosphate buffer (0.1M Na<sub>2</sub>HPO<sub>4</sub>, 0.1mM KH<sub>2</sub>PO<sub>4</sub>, 75 $\mu$ M MgSO<sub>4</sub> 7H<sub>2</sub>O, PVP (1.5mg mL<sup>-1</sup>), TRITON X-100 (2mL L<sup>-1</sup>)), by sonication for 45sec at 70% amplitude in a VibraCell VCX 130 Sonics<sup>®</sup> ultrasonic processor. On the other hand, the 0.7-50 $\mu$ m plankton samples were homogenized at 0-4°C in the same 0.1M phosphate buffer, by a Teflon<sup>®</sup> 2mL-pestle PYREX<sup>®</sup> Potter-Elvehjem tissue grinder (homogenizer) at 2600rpm for 2 min. Then, crude homogenates were centrifuged (0-4°C) at 4000rpm for 10 min.

The supernatant aliquot was assayed for NADP-IDH activity following the Tames-Espinosa et al. (2018) method (Tames-Espinosa et al. (2018), see Chapter 2). NADP-IDH activity was measured kinetically by following the time-dependent production of NADPH ( $\Delta_{NADPH}$ ) at 340nm ( $\Delta_{340}$ ) and 18°C in a 1cm UV-transparent plastic cuvette in a Cary 100 U-V visible spectrophotometer for 1200 sec.  $\beta$ -NADP<sup>+</sup> (SIGMA #N0505; from here on we will call NADP<sup>+</sup>) was reduced by NADP-IDH to NADPH while decarboxylyed the DL-trisodium-isocitrate (SIGMA #I1252; from here on we will call isocitrate) to  $\alpha$ -ketoglutarate. Control assays (blanks) lacked isocitrate. The reaction was started by adding the NADP<sup>+</sup> solution. After 3 min, the slope of the linear time-course signal was considered the IDH activity. A molar extinction coefficient ( $\epsilon$ ) of 5.42 absorptivity units mL  $\mu$ mol<sup>-1</sup>cm<sup>-1</sup> was obtained for NADPH under these conditions (Tames-Espinosa et al. (2018)). NADP-IDH activity was reported in  $\mu$ mol NADPH min<sup>-1</sup> (mL of homogenate)<sup>-1</sup> units.

### 3.2.3 Potential respiration ( $\Phi$ )

For 0.7-50μm samples, potential respiration ( $\Phi$ ) was determined from ETS activity (Packard (1971)) incorporating the modifications described in Kenner and Ahmed (1975) and Gómez et al. (1996) on the same centrifuged supernatant aliquotes on which NADP-IDH was also measured. For the 200-2000 μm size-fractionated plankton samples, after centrifugation, the supernatant aliquot was assayed for ETS activity following Owens and King (1975).

The absorbance increase of the INT-formazan at 490nm was continuously monitored at 18°C in 1cm path-length cuvettes for 8min. The regression line of absorption versus time was used to estimate the ETS activity and potential respiration rates ( $\Phi$ ) after Packard and Williams (1981). The tetrazolium salt, INT (p-Iodonitrotetrazolium Violet, Sigma #I8377) was reduced by the respiratory ETS enzymes, replacing O<sub>2</sub> as the electron acceptor. The INT accepts two electrons whereas the O<sub>2</sub> would accept four. Thus, the INT-formazan production rate is stoichiometrically related by a factor of 2 to ETS activity and by a factor of 0.5 to  $\Phi$ . Blanks were run without the ETS substrates in order to subtract the contribution of the background non-enzymatic and enzymatic reduction of the INT (Maldonado et al. (2012)).  $\Phi$  was reported in μmol O<sub>2</sub> min<sup>-1</sup>(mL of homogenate)<sup>-1</sup> units. To convert to μLO<sub>2</sub> min<sup>-1</sup>(mL of homogenate)<sup>-1</sup>, multiply by 22.4 μL μmol<sup>-1</sup>.

### 3.2.4 $R_{O_2}$ calculations from Potential respiration ( $\Phi$ ) measurements

When direct measurements of  $R_{O_2}$  are not available, reported ratios are commonly used to calculate  $R_{O_2}$  from  $\Phi$  in the micro- and mesozooplankton community. Calculations of this type (Cleland (1967); Hernández-Leon and Gómez (1996); Packard and Codispoti (2007); Osma et al. (2016c)) have been done before by applying a  $R_{O_2}/\Phi$  ratio for the determination of  $R_{O_2}$  (Eq.3.1). Here, we have applied three different zooplankton  $R_{O_2}/\Phi$  ratios from the literature (Tables 3.1) in Eq.3.1 (Section 3.3.2, Fig.3.3), developing three scenarios: (A) 0.50, the ratio when standard physiological rates, governed by enzyme activities, operate at one-half their potential capacity (Cleland (1967)); (B) 0.57, the ratio characteristic for the mesotrophic ocean conditions around the Canary Islands (Osma et al. (2016c)); and (C) 0.77, the ratio that exemplifying the late-winter bloom conditions in the waters around the Canary Islands (Hernández-Leon and Gómez (1996)). In the case of the 0.7-50μm size-fractionated plankton community, a fourth scenario (D) was developed using a  $R_{O_2}/\Phi$  ratio of 0.26 (Packard and Christensen (2004); Osma et al. (2014)) (Tables 3.1). In all of them, the equation for calculating  $R_{O_2}$  from  $\Phi$  is:

$$R_{O_2} = \Phi \cdot R_{O_2}/\Phi \quad (3.1)$$

where  $R_{O_2}$  is the physiological O<sub>2</sub> consumption rate, and  $\Phi$  is the potential O<sub>2</sub> consumption, both are in in μmol of O<sub>2</sub> min<sup>-1</sup> mL<sup>-1</sup>.

### 3.2.5 $R_{CO_2}$ calculations from $R_{O_2}$

From the  $R_{O_2}$  values obtained as above,  $R_{CO_2}$  was calculated by Eq.3.2, applying Redfield ratios (Packard and Gómez (2013)) or different Respiratory Quotients (RQ) (Romero-Kutzner et al. (2015); Berggren et al. (2012)), describing a wide range of different marine scenarios (Section 3.3.2, Fig. 3.3). Such calculations have been done in the literature when physiological measurements are not available. In Packard and Gómez (2013) (Packard and Gómez (2013)) the physiological measurements were not available so the Redfield ratio, C/O in the cells (122/172), was used with natural samples to convert  $R_{O_2}$  into  $R_{CO_2}$ . In other studies, this type of respiration measurements have been also derived from the product of  $R_{O_2}$  and the theoretical RQ (Mayzaud et al. (2005)). Here, the plankton community was sampled during the spring-bloom when the organisms were likely to have been well-fed, a condition described by normal RQ ranging from 0.7 to 1.2 (Romero-Kutzner et al. (2015); Berggren et al. (2012)). Thus, in Table 3.1 we have applied in Eq.3.2 three different conversion factors ( $R_{CO_2:O_2}$ ) for every physiological condition described in Section 3.2.4. These factors were: (1) the Redfield ratio (0.71), as a standard value related to an unknown nutrient composition of the zooplanktonic diet (Packard and Gómez (2013))(note that this value is close to the RQ = 0.7 linked to lipid consumption (Berggren et al. (2012))); (2) an average value of 0.87 obtained for copepods (Mayzaud et al. (2005)); (3) a mean value of RQ = 1.3 linked to a net-heterotrophic system (Berggren et al. (2012)) (note that this value is close to the RQ = 1.2 linked to a carboxylic acid nutritional source (Berggren et al. (2012)), Table 3.1). An RQ value of 0.7 is also used as a low theoretical RQ for phytoplankton, assuming that it is constituted mainly by oxygen-poor components (lipids and proteins) (Takahashi et al. (1985); Berggren et al. (2012)).

$$R_{CO_2} = R_{O_2} \cdot R_{CO_2:O_2} \quad (3.2)$$

where  $R_{CO_2:O_2}$  is the applied conversion in every scenario from Table 3.1.

### 3.2.6 Biomass

Biomass content was determined from samples kept for enzyme activities by the protein method of Lowry *et al.* (1951) (Lowry *et al.* (1951)) modified by Rutter (1967) (Rutter (1967)) using bovine serum albumin (BSA) as standard. Biomass was reported in mg of protein.

### 3.2.7 Data analysis and Statistics

ANOVA and *t* tests were used to analyse the differences between the linear regressions from the NADP-IDH activity and  $\Phi$ , in the two different size-fractionated plankton samples (Section 3.4.1) (Pentecost (1999)) in an attempt to assess the differences between a pure heterotrophic plankton community and a mixed plankton community.

Table 3.1: Different scenarios designed to develop the  $R_{CO_2}/\Psi_{NADP}$  ratios from the  $\Phi$  and  $\Psi_{NADP}$  measurements, using conversion ratios reported in the literature. Scenarios 2.2 and 3.2 describe the natural conditions of our 200-2000  $\mu\text{m}$  zooplankton samples, and 4.1 describes the natural conditions in our 0.7-50  $\mu\text{m}$  samples.

Scenario	$R_{O_2}/\Phi$	Conditions	$R_{CO_2:O_2}$	Conditions
<b>A1</b>	0.5	Ratio for standard physiological rates	0.71	Standard natural sample conditions (Redfield)
<b>A2</b>	0.5	Ratio for standard physiological rates	0.87	RQ in samples dominated by copepods
<b>A3</b>	0.5	Ratio for standard physiological rates	1.3	RQ in net-heterotrophic microplankton samples
<b>B1</b>	0.57	Ratio for mesotrophic conditions in waters around the Canary islands	0.71	Standard natural sample conditions (Redfield)
<b>B2</b>	<b>0.57</b>	<b>Ratio for mesotrophic conditions in waters around the Canary islands</b>	<b>0.87</b>	<b>RQ in samples dominated by copepods</b>
<b>B3</b>	0.57	Ratio for mesotrophic conditions in waters around the Canary islands	1.3	RQ in net-heterotrophic microplankton samples
<b>C1</b>	0.77	Ratio for late winter-bloom conditions in waters around the Canary islands	0.71	Standard natural sample conditions (Redfield)
<b>C2</b>	<b>0.77</b>	<b>Ratio for late winter-bloom conditions in waters around the Canary islands</b>	<b>0.87</b>	<b>RQ in samples dominated by copepods</b>
<b>C3</b>	0.77	Ratio for late winter-bloom conditions in waters around the Canary islands	1.3	RQ in net-heterotrophic microplankton samples
<b>D1</b>	<b>0.26</b>	<b>Ratio for a mixed microplanktonic community</b>	<b>0.71</b>	<b>Standard natural sample conditions (Redfield)</b>
<b>D2</b>	0.26	Ratio for a mixed microplanktonic community	0.87	RQ in samples dominated by copepods
<b>D3</b>	0.26	Ratio for a mixed microplanktonic community	1.3	RQ in net-heterotrophic microplankton samples

### 3.3 Results

#### 3.3.1 Potential CO<sub>2</sub> production from NADP-IDH measurements

Although we propose NADP-IDH activity as a proxy for plankton respiratory CO<sub>2</sub> production, the total CO<sub>2</sub> production of the cell is closer to the CO<sub>2</sub> production of all the enzymes associated with the Krebs cycle (KC<sub>CO2</sub>). This is the sum of the activities of five enzyme activities, pyruvate dehydrogenase (PDH<sub>CO2</sub>);  $\alpha$ -ketoglutarate dehydrogenase (KGDH<sub>CO2</sub>); malic enzyme (ME<sub>CO2</sub>); phosphoenolpyruvate carboxykinase (PEPCK<sub>CO2</sub>); and all the isocitrate dehydrogenase isoenzymes that work within the Krebs cycle (IDH<sub>KC</sub><sub>CO2</sub>) (Packard et al. (1996a)). This summation is represented by Eq.3.3.

$$KC_{CO2} = PDH_{CO2} + IDH_{KC}CO2 + KGDH_{CO2} + ME_{CO2} + PEPCK_{CO2} \quad (3.3)$$

To develop a model based on the NADP-IDH measurements, the CO<sub>2</sub> production of these other enzymes needs to be related to NADP-IDH activity. The contribution of every enzyme of the Krebs cycle linked to the CO<sub>2</sub> production, normalized by NADP-IDH activity, can be obtained from the reported rates in *E.coli* (Fig.1.1) (Walsh and Koshland (1984)), assuming that the *E.coli* Krebs cycle is equivalent to the Krebs cycle in other prokaryotes and also to the mitochondrial Krebs cycle in eukaryotes. Accordingly, the relationship between the CO<sub>2</sub> production of the above-mentioned five-enzymes and the IDH activity in the Krebs cycle (IDH<sub>KC</sub>) can be obtained as follows:

- Pyruvate dehydrogenase (EC 1.2.4.1, PDH). For plankton on a carbohydrate-based diet, PDH bridges glycolysis and the Krebs cycle. In the transfer process between these two metabolic pathways, one mole of CO<sub>2</sub> is produced for every mole of acetyl-CoA generated (El-Mansi and Holms (1989)). Then, in the beginning of the Krebs cycle, citrate synthase (EC 2.3.3.1, CS) takes this acetyl-CoA, and generates the six-carbon tricarboxylic acid, citric acid. These acids are normally found in the cell in their sodium or potassium salt forms (Walsh and Koshland (1984)). According to Walsh and Koshland (1984) in Fig.1.1, in order to obtain an IDH<sub>KC</sub> activity of 80moles isocitrate min<sup>-1</sup> L<sup>-1</sup>, CS must react at an activity of 111moles min<sup>-1</sup> L<sup>-1</sup> of acetyl-CoA. As this acetyl-CoA is being produced by PDH, one mole of CO<sub>2</sub> per mole of acetyl-CoA is also being produced. This means that, according to Fig.1.1, 111 moles of CO<sub>2</sub>, related to the PDH activity, will be produced when 80 moles of CO<sub>2</sub> are produced by IDH<sub>KC</sub>. Thus, the ratio of CO<sub>2</sub> produced by PDH to CO<sub>2</sub> produced by IDH<sub>KC</sub> is 1.39. Thus,  $PDH_{CO2} = 1.39 \cdot IDH_{KC}CO2$ .
- Isocitrate dehydrogenase (both EC 1.1.1.41, NAD-IDH; and the mitochondrial EC 1.1.1.42, NADP-IDH<sub>mit</sub>) in the Krebs cycle (IDH<sub>KC</sub>). This enzyme is the third Krebs-Cycle reaction. It is the oxidation and decarboxylation of isocitrate, producing  $\alpha$ -ketoglutarate, NAD(P)H and CO<sub>2</sub> (Eq.1.3). As we

are normalizing by this rate in our model, the CO<sub>2</sub> production ratio here is 1 in the Krebs cycle (IDH<sub>KC</sub>).

- $\alpha$ -ketoglutarate dehydrogenase (EC 1.2.4.2, KGDH). This enzyme is the 4<sup>th</sup> Krebs-Cycle reaction. In this reaction,  $\alpha$ -ketoglutarate undergoes oxidative decarboxylation, producing succinyl-CoA, NADH and CO<sub>2</sub>. As explained above, according to the rates in Fig.1.1, 75moles of CO<sub>2</sub> were produced by KGDH for every 80 moles of CO<sub>2</sub> produced by IDH<sub>KC</sub>; so, the ratio between the two enzyme activities is 0.94. Thus, KGDH<sub>CO2</sub>=0.94·IDH<sub>KC</sub><sub>CO2</sub>.
- Cytosolic malate dehydrogenase (EC 1.1.1.40, ME). Malate dehydrogenase (MDH, EC 1.1.1.37) in the Krebs-Cycle proper, does not produce CO<sub>2</sub>. However, part of the malate, generated by fumarase (EC 4.2.1.2) in the Krebs cycle, is transported outside to the decarboxylating malate dehydrogenase (EC 1.1.140) in the cytosol (MacDonald (2002)). There, it is decarboxylated to pyruvate, producing NADPH and CO<sub>2</sub>. In this case, from Fig.1.1, 9.5 moles of CO<sub>2</sub> are produced by this enzyme per 80 moles produced by IDH<sub>KC</sub>; so, the ratio between ME and IDH<sub>KC</sub> is: 0.12. Thus, ME<sub>CO2</sub>=0.12·IDH<sub>KC</sub><sub>CO2</sub>.
- Phosphoenolpyruvate carboxykinase (EC 4.1.1.49, PEPCK). This enzyme shunts part of the oxaloacetate produced in the Krebs cycle to the cytosol. There, it is decarboxylated to phosphoenolpyruvate and CO<sub>2</sub>. Here, according to Fig.1.1, 9.5 moles of CO<sub>2</sub> are produced by PEPCK per 80 moles produced by IDH<sub>KC</sub>; so, the ratio between both enzyme activities is 0.12. Thus, PEPCK<sub>CO2</sub>=0.12·IDH<sub>KC</sub><sub>CO2</sub>.

So, assuming all the IDH activity is related to the Krebs cycle, the Krebs cycle CO<sub>2</sub> production, normalized by IDH<sub>KC</sub> would be:

$$KC_{CO2}=1.39\cdot IDH_{KC}CO2+1\cdot IDH_{KC}CO2+0.94\cdot IDH_{KC}CO2+0.12\cdot IDH_{KC}CO2+0.12\cdot IDH_{KC}CO2 \quad (3.4)$$

Or, after combining terms algebraically:

$$KC_{CO2}=3.57\cdot IDH_{KC}CO2 \quad (3.5)$$

However, depending on the type of organism, the IDH<sub>KC</sub> activity is not directly equal to NADP-IDH measurement as explained below. While applying the NADP-IDH activity to calculate the potential CO<sub>2</sub> production, two facts must be taken in account:

1. In prokaryotes, the IDH<sub>KC</sub> is only NADP<sup>+</sup>-dependent:

$$IDH_{KC} = NADP-IDH \quad (3.6)$$

There is no NAD<sup>+</sup>-dependent IDH. However, in the case of eukaryotes, not only is there a mitochondrial-based NADP<sup>+</sup>-dependent IDH reaction, but there is, also inside the mitochondria, a NAD<sup>+</sup>-dependent IDH reaction (Gálvez and Gadal (1995)). There is an isoenzyme of IDH using the other

pyridine nucleotide! So, in the case of eukaryotes, IDH<sub>KC</sub> must be defined as:

$$\text{IDH}_{\text{KC}} = \text{NADP-IDH}_{\text{mit}} + \text{NAD-IDH}_{\text{mit}} \quad (3.7)$$

- 2 Secondly, in eukaryotes, not only is the mitochondrial enzyme being measured, but the isoenzymes located outside the mitochondria are also being measured. These include the isoenzymes in the cytosol, and even the NADP-IDH located in the chloroplasts and the peroxysomes of photosynthetic organisms (Gálvez and Gadal (1995)). So, our NADP-IDH measurements included NADP-IDH isoenzymes from both the inside and outside of the mitochondria. This is expressed in Eq.3.8:

$$\text{NADP-IDH} = \text{NADP-IDH}_{\text{out}} + \text{NADP-IDH}_{\text{mit}} \quad (3.8)$$

Returning to prokaryotes, the specific IDH isoenzymes related to the Krebs cycle, will be purely NADP-IDH, as in Eq.3.6.

Zooplankton are eukaryotes, mainly crustaceans. In a study of muscles from six different crustaceans, Alp et al. (1976) reported maximum activities of CS (a purely Krebs cycle enzyme), NAD-IDH (specifically located in mitochondria), and NADP-IDH (located inside and outside the mitochondria) (Table 3.2). Combining these rates with the rates of CS and Krebs cycle IDH as reported by Walsh and Koshland (1984) (Fig. 1.1), we determined the proportions of NADP-IDH related to the Krebs cycle and related to other metabolic pathways.

Accordingly, we know that, for 111 moles of substrate consumed by CS, IDH<sub>KC</sub> consumes 80 moles of isocitrate (Fig.1.1). So, the activities of CS and IDH<sub>KC</sub> are related as in Eq.3.9.

$$\text{CS} = 1.39 \cdot \text{IDH}_{\text{KC}} \quad (3.9)$$

Rearranging and solving for IDH<sub>KC</sub>, Eq.3.9 is transformed into: IDH<sub>KC</sub>=0.72·CS. Thus, from the CS, NADP-IDH and NAD-IDH activities in crustacean samples reported by Alp et al. (1976) (Table 3.2) we can again calculate IDH<sub>KC</sub>. Then, to obtain the NADP-IDH inside and outside the mitochondria we transformed and combined Eq.3.7 and Eq.3.8 as follows:

$\text{NADP-IDH}_{\text{mit}} = \text{IDH}_{\text{KC}} - \text{NAD-IDH}$ , from Eq.3.7; and  $\text{NADP-IDH}_{\text{out}} = \text{NADP-IDH} - \text{NADP-IDH}_{\text{mit}}$  from Eq.3.8. So, combining these transformations, we obtained Eq.3.10:

$$\text{NADP-IDH}_{\text{out}} = \text{NADP-IDH} - (\text{IDH}_{\text{KC}} - \text{NAD-IDH}) \quad (3.10)$$

The results of these calculations are shown in Table 3.2.

Table 3.2: Enzyme activities in crustacean samples, in  $\mu\text{mol substrate min}^{-1}$  (g fresh weight)<sup>-1</sup>. IDH<sub>KC</sub> activity was calculated as expressed in Eq.3.9; NADP-IDH<sub>mit</sub> activity was calculated as expressed in Eq.3.7; and NADP-IDH<sub>out</sub> activity was calculated as expressed in Eq.3.10. In each case, the activities of CS, NADP-IDH and NAD-IDH in crustacean samples from Alp et al. (1976) were used.

Organism	CS	IDH <sub>KC</sub>	NADP-IDH	NAD-IDH	NADP-IDH <sub>mit</sub>	NADP-IDH <sub>out</sub>
<b>Lobster</b>	1	0.72	1.1	0.1	0.62	0.48
<b>Lobster</b>	1.1	0.79	1.1	0.1	0.69	0.41
<b>Squad lobster</b>	3.4	2.45	3.5	0.1	2.35	1.15
<b>Swimming crab</b>	7	5.04	6.5	1	4.04	2.46
<b>Edible crab</b>	2.4	1.73	3	0.1	1.63	1.37
<b>Pedunculate barnacle</b>	1.2	0.86	2.3	0.1	0.76	1.54
<b>Horse-shoe crab</b>	17.4	12.53	10.2	0.6	11.93	0
<b>Average</b>	4.79	3.45	3.96	0.3	3.15	1.06
<b>Standard deviation</b>	5.95	4.28	3.31	0.36	4.06	0.83

From the enzyme activity in Table 3.2, the ratios of NAD-IDH, NADP-IDH<sub>mit</sub>, NADP-IDH<sub>out</sub> activities were calculated as a fraction of NADP-IDH activity. Table 3.3 was then constructed from these ratios.

Table 3.3: Ratios of NAD-IDH, NADP-IDH<sub>mit</sub> and NADP-IDH<sub>out</sub> activities to NADP-IDH activity. These ratios are needed for the calculation of total potential CO<sub>2</sub> production (Eq. 3.12).

Organism	NAD-IDH/ NADP-IDH	NADP-IDH <sub>mit</sub> / NADP-IDH	NADP-IDH <sub>out</sub> / NADP-IDH
<b>Lobster</b>	0.09	0.56	0.44
<b>Lobster</b>	0.09	0.63	0.37
<b>Squad lobster</b>	0.03	0.67	0.33
<b>Swimming crab</b>	0.15	0.62	0.38
<b>Edible crab</b>	0.03	0.54	0.46
<b>Pedunculate barnacle</b>	0.04	0.33	0.67
<b>Horse-shoe crab</b>	0.06	1.17	0
<b>Average</b>	<b>0.07</b>	<b>0.65</b>	<b>0.38</b>
<b>Standard deviation</b>	0.04	0.26	0.20

Eq.3.5 describes the CO<sub>2</sub> production related to the Krebs cycle (KC<sub>CO<sub>2</sub></sub>), however, the total potential CO<sub>2</sub> production of the cell ( $\Psi$ ), from both the mitochondria and the cytosol, will be:

$$\Psi = \text{KC}_{CO_2} + \text{NADP-IDH}_{out,CO_2} \quad (3.11)$$

Combining Eq.3.11 with Eq.3.5 yields Eq.3.12

$$\Psi = 3.57 \cdot (\text{NADP-IDH}_{mit} + \text{NAD-IDH}) + \text{NADP-IDH}_{out} \quad (3.12)$$

Where all terms are defined in Table 3.2. From the average activity ratios in Table 3.3 (in bold), applied to Eq.3.12, we calculated the total potential CO<sub>2</sub> production normalized by NADP-IDH ( $\Psi_{NADP}$ ), as in Eq.3.13:

$$\Psi_{NADP} = 3.57 \cdot (0.65 \cdot NADP-IDH + 0.07 \cdot NADP-IDH) + 0.38 \cdot NADP-IDH \quad (3.13)$$

Condensing Eq.3.13, yields Eq.3.14

$$\Psi_{NADP} = 2.94 \cdot (NADP-IDH) \quad (3.14)$$

Note that the NADP-IDH activity should be in molar CO<sub>2</sub> units rather than molar NADPH units. This is because the stoichiometry of the reaction (Eq.1.3) indicates that 1 mole of CO<sub>2</sub> is produced for every mole of NADPH produced. To change to volume units we simply need to remember that 1 mole of CO<sub>2</sub> is equivalent to 22.4L of CO<sub>2</sub>.

Eq.3.14 is equivalent to the Roy and Packard (2001) equation ( $R_{CO_2} = \chi \cdot V_{NADP-IDH}$ , where  $\chi=3$ ) that they used to calculate  $R_{CO_2}$  production from modelled *in vivo* NADP-IDH activity in bacterial cultures ( $V_{NADP-IDH}$ ). In bacteria the physiological  $R_{CO_2}$  was equal to 3 times modelled *in vivo* NADP-IDH activity when the nutrient source was based on carbohydrates (Roy and Packard (2001)), so the activity of the PDH was being considered. In our case, it is not the *in vivo* NADP-IDH activity but the  $V_{max}$  what is being measured. Thus, using our  $V_{max}$  in the Roy equation, would yield the potential  $R_{CO_2}$ .

Here, we will use Eq.3.14 to calculate potential CO<sub>2</sub> in crustacean-dominated zooplankton samples. However, in the case of a more heterogeneous community, for example, the 0.7-50μmm size-fractionated plankton, other types of organisms such as prokaryotes and photoautotrophs may dominate the community. In a prokaryotic community, all the NADP-IDH is related to the Krebs cycle, so the total CO<sub>2</sub> production can be calculated directly from Eq.3.5, where  $\chi$  is 3.57 (instead of 2.94, the value calculated for crustaceans). In the case of photoautotrophs, where the NADP-IDH fraction that is not related to the Krebs cycle, the NADP-IDH<sub>out</sub> could be greater than it is in crustaceans. This is because the NADP-IDH in photoautotrophs is also located in the cytosol, peroxisomes and chloroplasts (Gálvez and Gadal (1995)). Values of  $\chi$  lower than 2.94 are expected to be found if additional experiments were performed on photoautotrophs. If the kind of experiment, similar to the one performed by Alp et al. (1976), were conducted on photoautotrophs, it would facilitate an analysis similar to the one we made on crustacean enzyme activities (Tables 3.2, 3.3). This would lead to a better relationship between  $\Psi$  and NADP-IDH activity, in photoautotrophs.

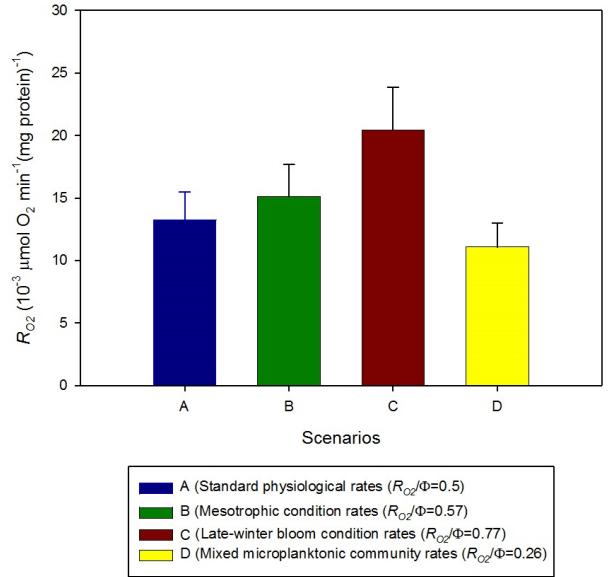
As the 0.7-50μm samples are from a heterogeneous community, the prokaryotic part will have  $\chi$  values higher than 2.94, whereas the photoautotrophic part will have lower  $\chi$  values. Accordingly, we suggest using  $\chi = 2.94$ , as an average, being aware that this will underestimate potential CO<sub>2</sub> production by prokaryotes, and will overestimate potential CO<sub>2</sub> production by photoautotrophs.

### 3.3.2 Physiological rates from Φ measurements

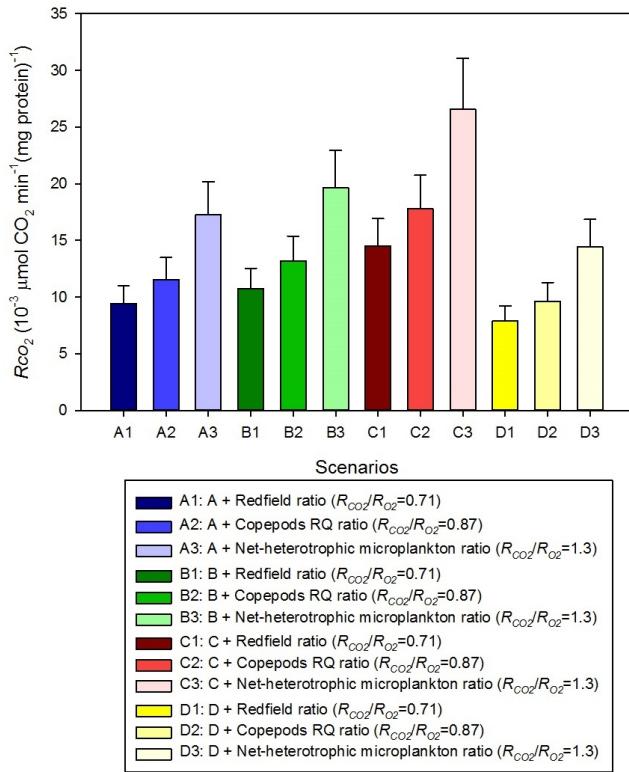
After being normalized by protein, Φ was used to calculate  $R_{O_2}$  for different physiological conditions as described in section 3.2.4 (Table 3.1).  $R_{O_2}/\Phi$  ratios from sce-

narios A-C were applied to the mesozooplankton samples, where  $\Phi=26.6\pm4.8\cdot10^{-3}$   $\mu\text{mol O}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ . Besides, the ratio for condition D is specific for microplankton, where  $\Phi=42.7\pm9.1\cdot10^{-3}$   $\mu\text{mol O}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ .

Results were plotted in Fig.3.3a, where a high  $R_{O_2}$  occurred related to the late-winter bloom conditions in the Canary Islands (scenario C in Fig.3.3a, and Table 3.1). Values between scenario C and B (from  $15.2\cdot10^{-3}$  to  $20.5\cdot10^{-3}$   $\mu\text{mol O}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$  respectively) were likely to reflect the physiological state of our Canary-Island mesozooplankton samples, as they were taken during the spring bloom. Scenario D ( $R_{O_2}=11.1\pm2.4\cdot10^{-3}$   $\mu\text{mol O}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ ) reflected the  $R_{O_2}$  related to mixed microplankton. Besides,  $R_{CO_2}$  was calculated applying different Respiratory Quotients (RQ) to  $R_{O_2}$ , as described in section 3.2.5 (Table 3.1). This assessment results in twelve different scenarios, nine related to mesozooplankton communities (from A1 to C3 in Fig.3.3b and Table 3.1), and three related to mixed microplankton communities (from D1 to D3 in Fig.3.3b and Table 3.1). As a result, the highest  $R_{CO_2}$  production rate occurred in scenario C3 ( $26.6\pm4.8\cdot10^{-3}$   $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ , Fig.3.3b), during a late-winter bloom condition, considering that these mesozooplankton samples were similar to a net-heterotrophic microzooplankton community (Section 3.2.5, Table 3.1). However, as our mesozooplankton samples were composed mainly of copepods during the spring, the  $R_{CO_2}$  rates were likely to be between scenarios B2 and C2 (from  $13.2\pm2.4\cdot10^{-3}$   $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$  to  $17.8\pm3.2\cdot10^{-3}$   $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ ) (Fig. 3.3). This corresponds to copepods under mesotrophic conditions or under late-winter bloom conditions respectively, as represented in Fig. 3.3b and Table 3.1. In the case of microzooplankton samples, a value between scenario D1 and D3 (between  $7.9\pm1.7\cdot10^{-3}$   $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ ) related to the phytoplankton community (Takahashi et al. (1985); Berggren et al. (2012)), and  $14.4\pm3.1\cdot10^{-3}$   $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mg protein})^{-1}$ , related to a net-heterotrophic community is likely representative of the mixed-microzooplankton sample (Fig.3.3).



[a]



[b]

Figure 3.3: Physiological rates under different scenarios. (a)  $R_{O_2}$  rates obtained from  $\Phi$  under four different scenarios. (b)  $R_{CO_2}$  rates obtained from  $R_{O_2}$  resulting in twelve different physiological conditions.

### 3.3.3 Variability of $R_{CO_2}/\Psi_{NADP}$ ratios

Taking into account Section 3.3.1, we applied Eq.3.14 to the NADP-IDH results obtained from both the mesozooplankton and the microzooplankton samples (Fig.3.4) in order to calculate  $\Psi_{NADP}$ . After normalizing the  $\Psi_{NADP}$  results by biomass, we obtained the average values of  $\Psi_{NADP}$  for both types of zooplankton samples in  $10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1} (\text{mg protein})^{-1}$  (Table 3.4).

Table 3.4: Results in  $\Psi_{NADP}$  (in  $10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1} (\text{mg protein})^{-1}$ ), in  $R_{CO_2}$  (also in  $10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1} (\text{mg protein})^{-1}$  units) and in the  $R_{CO_2}/\Psi_{NADP}$  ratios for every scenario. Values that likely represent the actual conditions of our mesozooplankton samples (between scenarios B2 and C2) and microzooplankton samples (scenario D1 and D3) have been highlighted in bold text.

		$\Psi_{NADP}$	$R_{CO_2}$	$R_{CO_2}/\Psi_{NADP}$
12[2]*Scenarios	<b>A1</b>	42.1±10.9	9.4±1.7	0.24±0.07
	<b>A2</b>	42.1±10.9	11.6±2.1	0.29±0.03
	<b>A3</b>	42.1±10.9	17.3±3.1	0.43±0.13
	<b>B1</b>	42.1±10.9	10.8±1.9	0.27±0.08
	<b>B2</b>	42.1±10.9	13.2±2.4	<b>0.33±0.10</b>
	<b>B3</b>	42.1±10.9	19.7±3.5	0.49±0.15
	<b>C1</b>	42.1±10.9	14.5±2.6	0.36±0.11
	<b>C2</b>	42.1±10.9	17.8±3.2	<b>0.44±0.14</b>
	<b>C3</b>	42.1±10.9	26.6±4.8	0.66±0.20
	<b>D1</b>	53.5±9.5	7.9±1.7	<b>0.15±0.03</b>
	<b>D2</b>	53.5±9.5	9.7±2.1	0.18±0.03
	<b>D3</b>	53.5±9.5	14.4±3.1	<b>0.27±0.05</b>

The highest  $R_{CO_2}/\Psi_{NADP}$  ratio was 0.7, attained under C3 conditions. In this case, a late-winter bloom scenario in which the RQ was close to a net-heterotrophic microplankton community (RQ = 1.3) was represented. This RQ was also close to that based on carboxylic acid nutritional source (RQ=1.2) (see Section 3.2.5, Table 3.1) (Berggren et al. (2012)). However, our mesozooplankton samples were composed mainly by copepods, and were sampled during spring. Thus, a value between 0.3 and 0.4 (B2 and C2 in Table 3.4 respectively) would be expected to be more representative for these specific conditions in the Canary Islands.

In the case of the microplankton community, values between 0.2 and 0.3 (scenario D1, related to the phytoplankton community, and D3, related to a net-heterotroph community respectively), would be representative for a mixed community.

## 3.4 Discussion

### 3.4.1 What is the relationship between potential CO<sub>2</sub> production and potential O<sub>2</sub> consumption?

Cellular respiration has four components: (1) CO<sub>2</sub> production; (2) organic carbon oxidation; (3) O<sub>2</sub> consumption; (4) ATP production. Here, CO<sub>2</sub> production was assessed from NADP-IDH measurements, considering them as a proxy of the CO<sub>2</sub>-producing apparatus in the cells. Besides, potential respiration,  $\Phi$ , was considered also a proxy of the O<sub>2</sub>-consuming apparatus in the cells.

#### Relationship between NAP-IDH activity and $\Phi$

In an attempt to assess variations between autotrophic and heterotrophic metabolic pathways, we analysed both NADP-IDH and  $\Phi$  (Fig.3.4) on 0.7-50μm and 200-2000μm samples.

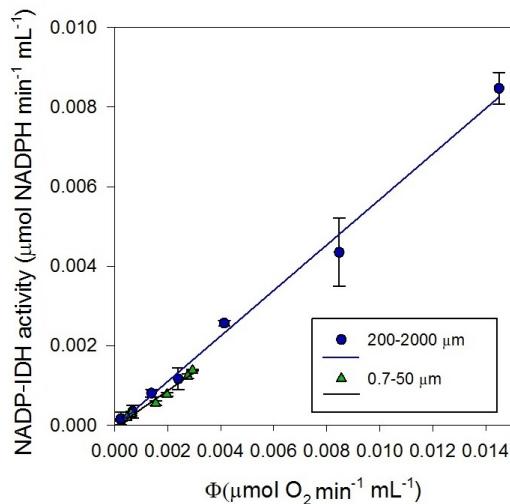


Figure 3.4: Relationship between NADP-IDH and Potential Respiration ( $\Phi$ ) in mesozooplankton (200-2000μm;  $y=0.5723x-4.4\cdot10^{-5}$ ,  $r^2=0.994$ ,  $p<0.0001$ ) and microplankton (0.7-50μm;  $y=0.46x-5\cdot10^{-5}$ ,  $r^2=0.9749$ ,  $p=0.0002$ ).

The slope of the linear regressions in Fig.3.4 reflects a zooplankton NADP-IDH activity of 57.23% of  $\Phi$ . In the microplankton community (Fig.3.4), this activity reached 46.00% of  $\Phi$ . ANOVA and a *t* test demonstrated that there was no significant difference between the two samples. Specifically, neither the linear regressions (ANOVA,  $F_{(2,38)}=2.15$ ,  $p=0.01$ ), nor the regression coefficients (*t* test,  $t(40)=1.34$ ,  $p=0.01$ ), were significantly different.

### Relationship between $\Psi_{NADP}$ and $\Phi$

In eukaryotes, the respiratory CO<sub>2</sub> production occurs both in the cytosol, associated with PDH, and in the mitochondrial matrix, associated directly with IDH-KC and KGDH, and indirectly with ME and PEPCK. In prokaryotes, the respiratory CO<sub>2</sub> production also occurs in the cytosol. However, the O<sub>2</sub> consumption and ATP production occur in the inner mitochondrial membrane of eukaryotes and in the plasmalemma membrane of prokaryotes (Nelson and Cox (2008)).

Total CO<sub>2</sub> production,  $\Psi_{NADP}$ , (Eq.3.13), not only involves the respiratory CO<sub>2</sub> production (as above and in Eq.3.15) to be considered, but also the CO<sub>2</sub> production due to other metabolic pathways, such as fatty acid synthesis (biosynthetic CO<sub>2</sub> production, Eq.3.16). When compared to  $\Phi$ , total potential CO<sub>2</sub> production was up to 1.7 times potential O<sub>2</sub> consumption in the mesozooplankton community (Fig.3.5(a), based on the slope of the linear regression ( $r^2=0.994$ ,  $p<0.0001$ )).

However, being able to calculate separately the respiratory potential CO<sub>2</sub> production facilitates a new estimation of the plankton carbon cycling related to the ATP production. The other part of the total potential CO<sub>2</sub> production is linked to biosynthesis.

For the zooplankton community, composed mainly of crustaceans, we can estimate these values using Eq.3.13. By just extracting the CO<sub>2</sub> production related to the NADP-IDH activity outside the mitochondria, we can express the respiratory potential CO<sub>2</sub> production, normalized per NADP-IDH, as Eq.3.15. The biosynthetic CO<sub>2</sub> production can be expressed as Eq.3.16:

$$\text{Respiratory } \Psi_{NADP} = 2.57 \cdot \text{NADP-IDH} \quad (3.15)$$

$$\text{Biosynthetic } \Psi_{NADP} = 0.38 \cdot \text{NADP-IDH} \quad (3.16)$$

Thus, respiratory  $\Psi_{NADP}$  is 1.47 greater than  $\Phi$  in the crustacean community (Fig.3.5(b), based on the slope of the linear regression ( $r^2=0.9940$ ,  $p<0.0001$ )). In effect, this means that the RQ for the potential respiratory metabolism would be 1.47. However, assuming total potential CO<sub>2</sub> production ( $\Psi_{NADP}$ ) and O<sub>2</sub> consumption ( $\Phi$ ), we would expect an RQ even higher when biosynthesis and growth occur, because of the CO<sub>2</sub> production related to the pentose phosphate pathway, as well as to the fatty acid synthesis repeating reaction sequence (Nelson and Cox (2008)), which could not be estimated in this study. Note that physiological RQ could still reach higher values than 1.47 if physiological CO<sub>2</sub> production rates were close to the  $\Psi_{NADP}$  or the O<sub>2</sub> consumption rates were far from  $\Phi$ . If compared with the cellular respiration reaction (Eq.3.17), this value of 1.47 is higher than the average as calculated from the stoichiometry of the cellular respiration equation.



This may mean that the enzymatic pathways that supply the ETS are prepared to produce more NAD(P)H than is needed. This may be advantageous when, eventually, the organism has a high energy requirement. Once this requirement

is over, NAD(P)H excess may be redirected to biosynthesis by the IDH substrate cycle of Sazanov and Jackson (1994).

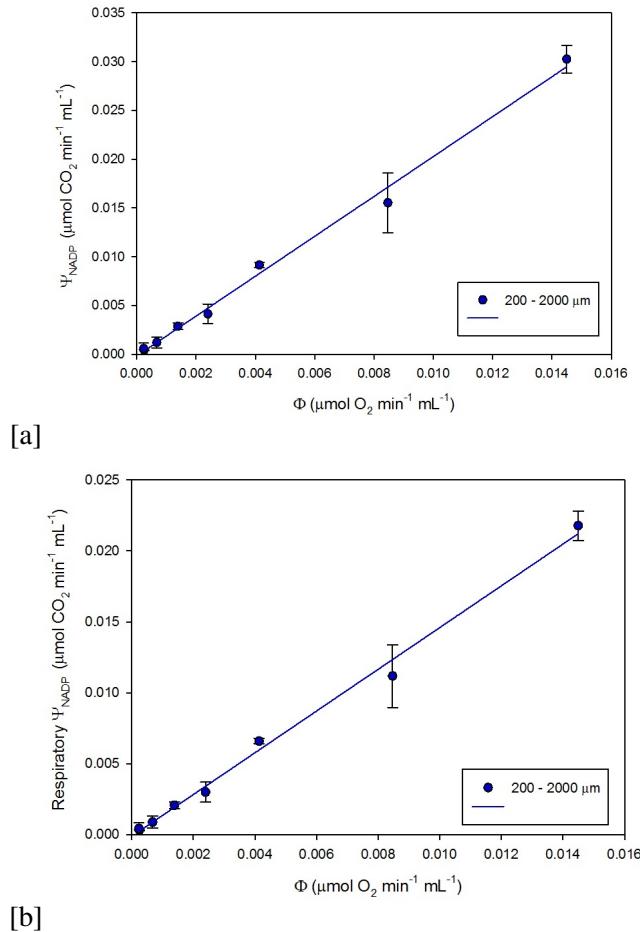


Figure 3.5: Relationship between potential  $\text{CO}_2$  production (NADP-IDH based model) and  $\text{O}_2$  consumption (ETS-based model) in the mainly crustacean 200-2000  $\mu\text{m}$  size-fractionated plankton community. (a) Total  $\Psi_{\text{NADP}}$  (Eq.3.14) vs  $\Phi$  ( $y=1.68x-0.0001$ ,  $r^2=0.994$ ,  $p<0.0001$ ). (b) Respiratory  $\Psi_{\text{NADP}}$  (Eq.3.15) vs  $\Phi$  ( $y=1.47x-0.0001$ ,  $r^2=0.994$ ,  $p<0.0001$ )

### 3.4.2 What is the relationship between $R_{\text{CO}_2}$ and $\Psi_{\text{NADP}}$ ?

Carbon flux, when not measured directly by sediment traps, can be indirectly calculated from ETS activity or  $R_{\text{O}_2}$  in the ocean water column (Osma et al. (2014); Packard et al. (2015)). However, using measurements of respiratory oxygen consumption entails the use of an RQ, and  $R_{\text{O}_2}/\Phi$  ratio particularly in the deep water samples, in the calculation of  $R_{\text{CO}_2}$  (Osma et al. (2014)). Here, we propose the

measurement of  $\Psi_{NADP}$  as a direct proxy for  $R_{CO_2}$  and, by doing so, we eliminate the need for the RQ and all its variability (Romero-Kutzner et al. (2015); Berggren et al. (2012); Mayzaud et al. (2005)). With physiological measurements of CO<sub>2</sub> production ( $R_{CO_2}$ ), its ratio to potential CO<sub>2</sub> production can be determined ( $R_{CO_2}/\Psi_{NADP}$ ). Once this ratio is calculated for a station in an oceanographic region, it may be used to obtain the actual  $R_{CO_2}$  for other samples of the NADP-IDH activity from the same region. Integrating  $R_{CO_2}$  will yield a mathematical model of carbon flux in carbon units. Furthermore, measuring  $\Psi_{NADP}$  as a  $R_{CO_2}$  proxy will improve data acquisition on oceanographic expeditions, and it will increase the sensitivity in detecting deep water CO<sub>2</sub> production. Thus,  $\Psi_{NADP}$  may be a powerful tool for calculating vertical carbon flux in the ocean in the short term. This approach requires calibration of  $\Psi_{NADP}$  with another measure of  $R_{CO_2}$  which, currently, is extremely challenging in all but the most eutrophic conditions.

Thus, we recommend the direct measurement of  $R_{CO_2}$  if possible. However, as we have just noted, this is inconvenient, impractical or impossible in the open ocean (Mayzaud et al. (2005)). That is why we have modelled the  $\Psi_{NADP}$  from the NADP-IDH measurements and the other CO<sub>2</sub>-producer enzyme activities in the literature (Section 3.3.1). Here, the physiological  $R_{CO_2}$  were not available, so we made a first step in determining  $R_{CO_2}/\Psi_{NADP}$  ratio from the  $R_{CO_2}$  obtained by ETS activity ( $\Phi$ ) (Fig. 3.3). We found that in all cases the results were between 0 and 1 (Table 3.4, Section 3.3.3). This means that in any scenario  $\Psi_{NADP}$  exceeded physiological  $R_{CO_2}$  rate, validating the reliability of the  $\Psi_{NADP}$  measurements as a measurement the potential CO<sub>2</sub> production in the cells. Even under the scenario with highest  $R_{CO_2}$  (C3 for zooplankton samples and D3 for microplankton in Table 3.4), suggesting a plankton community grown up under eutrophic or late winter-bloom conditions, the value  $R_{CO_2}/\Psi_{NADP}$  is still below 1 ( $R_{CO_2}/\Psi_{NADP}=0.7$  in zooplankton communities, 0.3 in heterogeneous microplankton communities, Table 3.4). This argues that  $\Psi_{NADP}$  accurately measures potential CO<sub>2</sub> production for these communities. As no physiological  $R_{CO_2}$  measurements are available in the case of our zooplankton samples, and assuming they are composed mainly by copepods grown during the spring bloom around the Canary island, we suggest that scenarios B2 and C2 (Table 3.1) would be the most representative. Accordingly, an  $R_{CO_2}/\Psi_{NADP}$  ratio between 0.3 and 0.4 approximates the range of the actual ratio (Table 3.4). In the case of the microplankton community, a value between 0.2 and 0.3, related to the scenarios D1 and D3 (Tables 3.1, 3.4), is a likely approximation for a mixed microplankton community.

However, we suggest to use these ratios as a guideline when  $R_{CO_2}$  measurements are not available. To develop an accurate  $R_{CO_2}/\Psi_{NADP}$  ratio, further experiments using physiological  $R_{CO_2}$  methods (Mayzaud et al. (2005)), need to be done for every case. As RQ has been demonstrated to be variable depending on the conditions of the community growth phase, composition and starvation level (Berggren et al. (2012); Romero-Kutzner et al. (2015)), we suggest avoiding the use of these values, particularly when a more accurate measurement of  $R_{CO_2}$  is available.

### 3.5 Conclusions

- A linear model for potential CO<sub>2</sub> production in zooplankton was developed from a measurement of NADP-IDH activity and related CO<sub>2</sub>-producing enzymes:  
 $\Psi_{NADP} = 2.94 \cdot \text{NADP-IDH}$  in  $\mu\text{mol CO}_2 \text{ min}^{-1} \text{ mL}^{-1}$  units.
- The ratio between the purely respiratory CO<sub>2</sub> production (Respiratory  $\Psi_{NADP}$ ) and the potential O<sub>2</sub> consumption is 1.47 (Fig.3.5), indicating that the enzymatic pathways related to the organic carbon oxidation phase of cellular respiration is prepared to exceed the requirements of the ETS. This may serve as an advantage when, eventually, the organism has a high energy requirement. Once this requirement is met, NAD(P)H excess may be redirected to biosynthesis by the IDH substrate cycle of Sazanov and Jackson.
- This also means that RQ between the potential respiratory metabolism responses would be 1.47. However, assuming potential CO<sub>2</sub> production and O<sub>2</sub> consumption, we would expect an RQ even higher when biosynthesis and growth occur. This is because the CO<sub>2</sub> production is related to the pentose phosphate pathway, as well as to the fatty acid synthesis, which could not be completely estimated in this study. Note that the physiological RQ could still reach higher values than 1.47 if physiological CO<sub>2</sub> production rates were close to the  $\Psi_{NADP}$  or if the O<sub>2</sub> consumption rates were far lower than  $\Phi$ .
- $\Psi_{NADP}$  is a useful tool to assess  $R_{CO_2}$  after applying  $R_{CO_2}/\Psi_{NADP}$  ratios. Its utility will be amplified when used to calculate vertical carbon flux in the open ocean. In both cases it eliminates the use of the RQ.
- We predict that, from our measurements in the North Eastern Subtropical Atlantic, the  $R_{CO_2}/\Psi_{NADP}$  ratios for open ocean zooplankton samples will fall between 0.3 and 0.4, whereas the ratios for a mixed microplankton community will fall between 0.2 and 0.3. However, we recommend the use of physiological  $R_{CO_2}$ , if available, to develop accurate  $R_{CO_2}/\Psi_{NADP}$  ratios for every region, experiment or case.
- From our calculations,  $\Psi_{NADP}$  always exceeded the physiological  $R_{CO_2}$  rate. This is in harmony with theory because the physiological rate should not exceed the potential. Thus, it reinforces the concept of using  $\Psi_{NADP}$  measurements as a tool to assess potential CO<sub>2</sub> production in the marine plankton community.



## Chapter 4

# Chapter 4: Impact of sex differences and starvation on respiratory metabolism: CO<sub>2</sub> production on marine mysid *Leptomysis lingvura*.

*Scientific physiology has the task of determining the functions of the animal body and deriving them as a necessary consequence from its elementary conditions.*

Carl Ludwig

**RESUMEN:** En este capítulo se ha medido la tasa de producción de CO<sub>2</sub> ( $R_{CO_2}$ ) en *Leptomysis lingvura* para compararla con la producción máxima de CO<sub>2</sub> asociada a la actividad de la NADP<sup>+</sup>-dependiente isocitrato deshidrogenasa (NADP-IDH). Tanto las medidas de respiración fisiológicas ( $R_{CO_2}$ ,  $R_O_2$ ), su ratio (RQ) como la actividad enzimática (NADP-IDH y Φ) se midieron en dos comunidades de *L. lingvura*: una en estado de florecimiento (bloom), y otra es un estado posterior al florecimiento (post-bloom). En ellas pudimos analizar la variabilidad de la respiración dada por diferencias en el sexo, por la inanición, y por el estado trófico original de la comunidad. Las hembras de la comunidad en "bloom" mostraron actividades enzimáticas significativamente mayores que las hembras de la comunidad en "post-bloom" ( $\Phi=1.04\pm0.14$  versus  $0.58\pm0.12 \mu\text{mol O}_2 \text{ h}^{-1}$  ( $\text{mg of protein})^{-1}$ , respectivamente; y NADP-IDH= $1.96\pm0.09$  versus  $0.97\pm0.19 \mu\text{mol CO}_2 \text{ h}^{-1}$  ( $\text{mg of protein})^{-1}$ , respectivamente).

protein)<sup>-1</sup>, respectivamente). Por el contrario, no se observaron diferencias entre ambas comunidades en términos de producción de CO<sub>2</sub>. Por lo tanto, un ratio  $R_{CO_2}/\text{NADP-IDH}$  de  $0.12 \pm 0.05$  representaría con precisión la respuesta fisiológica de una comunidad de *L. lingvura* bien alimentada. El cociente entre NADP-IDH y  $\Phi$  fue  $1.8 \pm 0.4$  en las hembras y  $1.6 \pm 0.4$  en los machos. A parte de esto, no encontramos otras diferencias asociadas al sexo en  $R_{O_2}$ ,  $R_{CO_2}$ , RQ,  $\Phi$  o NADP-IDH. Por el contrario, la inanición fue el factor con mayor impacto en la respuesta fisiológica. La limitación en la disponibilidad de alimento hizo que el ratio  $R_{O_2}/\Phi$  descendiera de  $0.35 \pm 0.02$  a  $0.27 \pm 0.08$ . Sin embargo hizo que el ratio  $R_{CO_2}/\text{NADP-IDH}$  aumentara de  $0.12 \pm 0.04$  a  $0.16 \pm 0.06$ . Por ello, el RQ fue casi el doble en los misidáceos en condiciones de inanición que en aquellos que habían sido bien alimentados. Esto puede reflejar que, en condiciones de inanición, los nutrientes se utilizan para el mantenimiento de la estructura celular en vez de para la generación de energía. Finalmente, las tasas de  $R_{CO_2}/\text{NADP-IDH}$  fueron más bajas en aquellas incubaciones que duraron más de 12h, lo que sugiere que los ritmos circadianos de *L. lingvura* afectan en las incubaciones de larga duración. Por ello, recomendamos incubaciones menores a las 6h para futuros estudios de *L. lingvura*.

**ABSTRACT:** CO<sub>2</sub> production rates ( $R_{CO_2}$ ) were measured in *Leptomysis lingvura* to compare with the potential CO<sub>2</sub> production rates, proxied here for the first time by the NADP<sup>+</sup>-dependent isocitrate dehydrogenase activity (NADP-IDH). Both respiratory physiological rates ( $R_{CO_2}$ ,  $R_{O_2}$ ), their ratio (RQ), and enzymatic potential respiration rates (NADP-IDH and  $\Phi$ ) were measured in a bloom and in a post-bloom community of *L. lingvura*. This way, we were able to assess the respiratory variability linked to sex differences, to starvation (nutrient-limitation), and to differences in the previous trophic state of the community. Bloom-community females showed significantly higher respiratory enzyme activity than post-bloom-community females ( $\Phi = 1.04 \pm 0.14$  versus  $0.58 \pm 0.12 \mu\text{mol O}_2 \text{ h}^{-1}$  (mg of protein)<sup>-1</sup>, respectively; NADP-IDH =  $1.96 \pm 0.09$  versus  $0.97 \pm 0.19 \mu\text{mol CO}_2 \text{ h}^{-1}$  (mg of protein)<sup>-1</sup>, respectively). In contrast, the physiological CO<sub>2</sub> production was the same in the two communities. Thus, an average ratio,  $R_{CO_2}/\text{NADP-IDH}$ , of  $0.12 \pm 0.05$  accurately characterized the well-fed *L. lingvura*, being only slightly higher in nutrient-limited mysids ( $0.16 \pm 0.06$ ). The NADP-IDH/ $\Phi$  ratio was  $1.8 \pm 0.4$  in females and  $1.6 \pm 0.4$  in males. Besides, we did not find significant differences in  $R_{O_2}$ ,  $R_{CO_2}$ , RQ,  $\Phi$  or NADP-IDH between sexes. In contrast, starvation was the most impacting factor on the physiological response. It reduced the  $R_{O_2}/\Phi$  ratio from  $0.35 \pm 0.02$  to  $0.27 \pm 0.08$ , but increased the  $R_{CO_2}/\text{NADP-IDH}$  ratio from  $0.12 \pm 0.04$  to  $0.16 \pm 0.06$ . The RQ was almost double in nutrient-limited

mysids than in well-fed mysids. This may reflect the diversion of nutrients to cell maintenance and anti-oxidation from energy generation. Finally, lower physiological rates on samples incubated longer than 12h suggested circadian-rhythm which argues for incubation periods shorter than 6h in future *L. lingvura* studies.

## 4.1 Introduction

As it was said in the General Introduction (Chapter 1.1), even though during the last decades the CO<sub>2</sub>-producing facet of the respiratory process ( $R_{CO_2}$ ) arose in assessing the role of the ocean as an anthropogenic CO<sub>2</sub> sink, respiration has been mainly approached by assessing  $R_{O_2}$  rates. Direct physiological measurement of  $R_{CO_2}$  is not as feasible as measuring  $R_{O_2}$  in the ocean, mainly because of the buffering effect of the carbonate system and the complexity and paucity of the available techniques to measure it. Thus, these rates have usually been derived from  $R_{O_2}$  measurements by using the Respiratory Quotient ratios between moles of CO<sub>2</sub> produced per moles of O<sub>2</sub> consumed (RQ). These ratios are variable, not only because of different nutrient sources, but also to differences in nutrient availability (Berggren et al. (2012); Romero-Kutzner et al. (2015)). So, when assessing  $R_{CO_2}$  rates, not only affects the variability in the physiological  $R_{O_2}$  rates the final measurement, but the variability of RQ ratios would also bias the accuracy of the results. Here, we propose assessing the  $R_{CO_2}$  production rates in the marine peracardian mysid *Leptomysis lingvura* by the development of physiological versus enzymatic ratios. *L. lingvura* is a mesozooplankton species found on the sandy bottoms of the East Coast of Gran Canaria (Spain). It has been proved to adapt well to lab cultures, even completing its life cycle in captivity (Herrera et al. (2011a)). Thus, it has been considered suitable for several enzymatic and physiological studies (Herrera et al. (2011b); Fernández-Urruzola et al. (2011)). Mysids are abundant in coastal regions, although they can also be found in the pelagic waters throughout the water column (Jumars (2007)). Being omnivorous, they usually feed on phytoplankton, zooplankton (copepods, tintinnids) seagrasses, macroalgae or detritus (Jumars (2007); Herrera et al. (2011a)), although they may switch to cannibalism in cultures when there is not enough food available (Laboratory observations; Domingues et al. (1999); Herrera et al. (2011a)). As they are part of the food chain and are hunted by many benthic, coastal and pelagic fishes, mammals, cephalopods and decapods (Mauchline (1980); Castro (1995); Jumars (2007); Kitsos et al. (2008)), strengthening the knowledge about their metabolism and the physiological responses may improve the assessment of both their roles and impacts on the ecosystem and their use as prey in aquaculture (Woods (2005); Otero-Ferrer et al. (2012)). In this study, we worked

with two communities of *L. lingvura*. The first one was sampled in Taliarte (Telde, Gran Canaria) at the beginning of November 2017. Even though autumn is the most oligotrophic season in the Canaries (Barton et al. (1998)), mysids were forming 300-500 individuals per group at the moment, as if they were under bloom conditions (where "bloom" refers to a surge in the mysid population), and were found together with the mysid *Siriella armata* (SCUBA observations). On the other hand, the individuals sampled at the end of November 2018, 48h after a storm, were found in 20-50 individuals per group, as if they were under post-bloom conditions (SCUBA observations). The impact of the previous trophic conditions, sex differences, starvation, and the potential effect of a dial rhythm on the CO<sub>2</sub>-producing facet of respiration was assessed as follows:

- First, we set out to measure the relationship between the changes in total inorganic carbon ( $\delta\text{TC}$ ) over up-to-24h incubations (to assess  $R_{\text{CO}_2}$ ), and the activity of the NADP<sup>+</sup>-dependent isocitrate dehydrogenase (NADP-IDH), one of the most important cellular CO<sub>2</sub>-producing enzymes.
- Secondly, in order to understand the variability in this relationship:
  1. We assessed the enzymatic activities ( $\Phi$  and NADP-IDH) directly related to respiratory metabolism in bloom and post-bloom communities of *L. lingvura*.
  2. We studied the differences in the physiological response and the  $R_{\text{CO}_2}/\text{NADP-IDH}$  ratios between females and males in the same community.
  3. We studied the differences between well-fed and nutrient-limited females.
  4. Finally, we studied the potential effect of a circadian rhythm on these ratios in up-to-24h incubations.

In addition, all these data were compared with the results on the  $R_{\text{O}_2}$  and  $R_{\text{O}_2}/\Phi$  in the same organisms. Thus, we assess both the viability of the development of  $R_{\text{CO}_2}/\text{NADP-IDH}$  as a proxy of  $R_{\text{CO}_2}$ , and the differences between these two physiological responses of respiration to starvation, sex differences or the circadian rhythm.

## 4.2 Material and Methods

### 4.2.1 *Leptomysis lingvura* sampling and cultures

This work was conducted on the mysid *L. lingvura*, sampled in two different life-cycle modes. The first occurred during the middle of a mysid bloom in early November 2017. The second occurred during the post-bloom phase at

the end of November 2018. Samples were collected off the port of Taliarte (Telde, East coast of Gran Canaria) by scuba-diving at a 5-10m depth. The mysids were swarming over shallow sandy bottoms closed to the rocky edge, in warm seawater at  $20^{\circ}\pm1^{\circ}\text{C}$ . Plankton nets with a mesh size of  $500\mu\text{m}$  were used to capture them, and plastic containers with a mesh cover (to ensure the sea-water flow during sampling time) were used to store them. The samples were taken directly to the laboratory in the plastic containers with an air pump to avoid  $\text{O}_2$  depletion during transport. Two species were identified: *S. armata* and *L. lingvura*. The experiments were carried out only on *L. lingvura* because its survival rate, in culture, is higher (Herrera et al. (2011a)) and because other physiological and enzymatic studies, related to  $\text{O}_2$  consumption and  $\text{NH}_4^+$  excretion (Fernández-Urruzola et al. (2011); Herrera et al. (2011b); Fernández-Urruzola et al. (2016); Osma et al. (2016b)), have been already done. Afterwards, the mysids were maintained at  $19.5\pm1.5^{\circ}\text{C}$  (in situ temperature) and on a 11:13h light:dark cycle, in a culture system according to Herrera et al. (2011a). The system consisted of four 40L-glass aquariums connected to a temperature-controller, 70 L biological filtration system (BIO DIGEST ©). The  $19.5^{\circ}$  seawater flowed through the system at a velocity of  $6\text{L h}^{-1}$ . The levels of  $\text{NH}_4^+$ ,  $\text{NO}_2^-$  and  $\text{NO}_3^-$  are kept below 0.1, 0.02 and 0.2 mg  $\text{L}^{-1}$  by both a biological filtration system and a mechanical one (mini-filters in every aquarium) (Lussier et al. (1988)). In addition, every week, 25% of the volume of sea-water was replaced by fresh filtered seawater. Mysids were fed twice a day on *Artemia sp.* >72h nauplii, at a daily dose of  $100\text{ Artemia mysid}^{-1}\text{ d}^{-1}$  from secondary cultures (Domingues et al. (1999); Fernández-Urruzola et al. (2011); Herrera et al. (2011b); Osma et al. (2016b)). After an acclimation period of one week, the healthiest mature males and females were selected for experimentation:

- On the samples from 2017, the influence of the incubation time on the physiology of well-fed females was assessed. Incubation for  $\text{CO}_2$  production ranged from 4h to 24h (4h, 8h, 12h, 16h and 24h). Every incubation was carried out in triplicate with samples of 10 mysids.
- On the samples from 2018, the influence of sex and starvation on the physiology of well-fed and 12h-starved females (well-fed: WF-F, nutrient-limited (starved): S-F) and well-fed males (well-fed: WF-M) was assessed. Every 6h-incubation was carried out in triplicate with samples of 10 mysids for  $\text{CO}_2$  production and with 3 mysids for  $\text{O}_2$  consumption.

For the well-fed mysid experiments, before incubations, the mysids were placed, for two hours, in filtered sea water containing 72h artemia-nauplii at a concentration of  $50\text{ artemia mysid}^{-1}$ . After 2h of feeding, they were sieved out and placed in filtered seawater for the two types of respiration measurements. For the nutrient-limited-mysid experiments, before incubation,

tions, the mysids were placed, for two hours, in filtered sea water containing 72h artemia-nauplii at a concentration of 50 artemia mysid<sup>-1</sup> (exactly as were the well-fed mysids). After 2h of feeding, they were sieved out and placed in filtered seawater for 12 h. Then they were pipeted out, with a basting pipet, and placed in new filtered seawater for the two types of respiration measurements.

A parallel experiment with 12h-starved males that were fed on 50 artemia mysid<sup>-1</sup> for 2h just before a 4h-incubation was also carried out on the 2018 community (non-starved males: NS-M). After every incubation, mysids were sieved through a 50μm mesh and frozen at -80°C with a freezing container Nalgene© Mr Frosty. Samples were stored in Microvials® 2mL-tubes at -80°C until enzyme and protein analyses.

#### 4.2.2 NADP-IDH activity: potential CO<sub>2</sub> production

Samples were homogenised in 0.1M phosphate buffer (0.1M Na<sub>2</sub>HPO<sub>4</sub>, 0.1mM KH<sub>2</sub>PO<sub>4</sub>, 75μM MgSO<sub>4</sub>·7H<sub>2</sub>O, PVP (1.5mg mL<sup>-1</sup>), TRITON X-100 (2mL L<sup>-1</sup>)), by sonication for 45sec at 70% amplitude in a Vibracell VCX 130 Sonics® ultrasonic processor. Then, crude homogenates were centrifuged (0-4°C) at 4000rpm (1500 g) for 10 min. An aliquot of the supernatant fluid was assayed for NADP-IDH activity following Tames-Espinosa et al. (2018). The activity was measured kinetically by following the time-dependent production of NADPH at 340nm and 20°C in a 1cm UV-transparent plastic cuvette in a Cary 100 U-V visible spectrophotometer for 1200 sec. One mol of β-NADP<sup>+</sup> (SIGMA #N0505; from here on we will call NADP<sup>+</sup>) was reduced by NADP-IDH to NADPH while decarboxylating the DL-trisodium-isocitrate (SIGMA #I1252; from here on we will call isocitrate) to α-ketoglutarate and producing 1 mol of CO<sub>2</sub> (Eq.4.1).



Control assays (blanks) lacked isocitrate. Occasionally, to check for contamination and to assess any nonenzymatic reaction between NADP<sup>+</sup> and isocitrate, an assay was run without homogenate. Normally, contamination and any non-enzymatic reaction between NADP<sup>+</sup> and isocitrate are negligible. The reaction was started by adding the NADP<sup>+</sup> solution. After 3 min, the slope of the linear time-course absorbance signal was considered the IDH activity. A molar extinction coefficient ( $\epsilon$ ) of 5.42 absorptivity units mL μmol<sup>-1</sup>cm<sup>-1</sup> was obtained for NADPH under these conditions (Tames-Espinosa et al. (2018)). As 1 mol of NADPH is produced, 1 mol of CO<sub>2</sub> is generated, NADP-IDH activity was reported in μmol CO<sub>2</sub> h<sup>-1</sup> (mg of protein)<sup>-1</sup> units. To convert to volume units (μLCO<sub>2</sub> h<sup>-1</sup>(mg of proteins)<sup>-1</sup>), multiply by 22.4 μL μmol<sup>-1</sup>.

### 4.2.3 Electron transport system (ETS) activity: potential O<sub>2</sub> consumption

Potential respiration in terms of O<sub>2</sub> consumption ( $\Phi$ ), was determined from the electron transport system activity (ETS) by the in vitro tetrazolium reduction method (Packard (1971)). After centrifugation, the supernatant aliquot was assayed for ETS activity following Owens and King (1975) and Gómez et al. (1996). The absorbance increase of the INT-formazan at 490nm was continuously monitored at 18°C in 1cm path-length cuvettes for 8min. A molar extinction coefficient ( $\epsilon$ ) of 12.8 absorptivity units mL  $\mu\text{mol}^{-1}\text{cm}^{-1}$  was obtained for tetrazolium salt, INT (p-Iodonitrotetrazolium Violet, Across #122400010) under these conditions. The regression line (slope) of absorption versus time was used to estimate the ETS activity and potential respiration rates ( $\Phi$ ) after Packard and Williams (1981). The INT was reduced by the respiratory ETS enzymes, replacing O<sub>2</sub> as the electron acceptor. The INT accepts two electrons whereas O<sub>2</sub> would accept four. Thus, the INT-formazan production rate (i.e., 1 $\mu\text{mol}$  formazan mL<sup>-1</sup> h<sup>-1</sup>) is stoichiometrically related by a factor of 2 to ETS activity (i.e., 2 $\mu\text{mol e}^{-}$  mL<sup>-1</sup> h<sup>-1</sup>) and by a factor of 0.5 to  $\Phi$  (i.e., 0.5 $\mu\text{mol O}_2$  mL<sup>-1</sup> h<sup>-1</sup>). Note that in terms of electrons transferred, the three rates are equivalent. Blanks were run without the ETS substrates in order to subtract the contribution of the background non-enzymatic and enzymatic reduction of the INT (Maldonado et al. (2012)). Occasionally, to check for contamination and to assess any nonenzymatic reaction between the ETS substrates, an assay was run without homogenate. Normally, contamination and any non-enzymatic reaction between the ETS substrates are negligible.  $\Phi$  was reported in  $\mu\text{mol O}_2$  h<sup>-1</sup>(mg of protein)<sup>-1</sup> units. To convert  $\Phi$  to  $\mu\text{LO}_2$  h<sup>-1</sup>(mg of proteins)<sup>-1</sup>, multiply it by 22.4  $\mu\text{L } \mu\text{mol}^{-1}$ .

### 4.2.4 Physiological O<sub>2</sub> consumption rates ( $R_{O_2}$ ) in *L. lingvura* incubations

O<sub>2</sub> consumption rates ( $R_{O_2}$ ) were measured by incubating three mysids in a 60mL bottle per triplicate with a 1-Channel Fibox 4 optode (Presens<sup>®</sup>) with four PSt3 sensors following Bondyale-Juez et al. (2017). Three sensors measured the O<sub>2</sub> concentration [O<sub>2</sub>] in three bottles containing 3 mysids per bottle in filtered seawater, whereas the fourth was used in the blank bottle, filled only with filtered seawater. As this system was limited to one measurement at a time, to ensure the measurement every 15 min, we moved the probe from bottle to bottle, for 120min. [O<sub>2</sub>] was expressed in  $\mu\text{mol O}_2$  L<sup>-1</sup> units. These measurements, according to Bondyale-Juez et al. (2017), are in agreement with (and comparable to) the results obtained from the Winkler method and from the Strathkelvin 928 O<sub>2</sub>-electrode system. The negative slope ( $\alpha$ )

of the regression line of [O<sub>2</sub>] per time, after the first 30min, is directly related to the O<sub>2</sub> consumption per time in the bottle (which is expressed in μmol O<sub>2</sub> L<sup>-1</sup> min<sup>-1</sup> units). To convert this result to the mysids' respiration rates in μmol O<sub>2</sub> h<sup>-1</sup> units, we used the incubation bottle volume (V) in L units, as expressed in Eq.4.2:

$$R_{O2} = -(\alpha) \cdot V \cdot 60 \quad (4.2)$$

Where α is the slope of the obtained linear regression of [O<sub>2</sub>] per min, V the incubation bottle volume in L units, and 60 the minutes in one hour. Thus,  $R_{O2}$ , per bottle, is expressed as μmol O<sub>2</sub> h<sup>-1</sup>. Combined with the biomass measurements, per bottle, for the same samples,  $R_{O2}$  may also be expressed in μmol O<sub>2</sub> h<sup>-1</sup> (mg of protein)<sup>-1</sup>. This facilitates the direct comparison of Φ to  $R_{CO2}$ . For further comparison, to convert to μLO<sub>2</sub> h<sup>-1</sup>(mg of proteins)<sup>-1</sup>, multiply by 22.4 μL μmol<sup>-1</sup>.

#### 4.2.5 Physiological CO<sub>2</sub> consumption rates ( $R_{CO2}$ ) in *L. lingvura* incubations

Total dissolved inorganic Carbon (TC) measurements require bigger volumes (250mL per measurement) than the O<sub>2</sub> optode method. Thus, we assessed the  $R_{CO2}$  as an end point measurement for the increase in TC ( $\delta_{TC}$ ) of ten-mysids, incubated (in triplicate) in 1L-bottles, for an incubation period that ranged from 4h to 24h. The length of this period depended on the aim of the experiment. Initial TC was measured, in triplicate, in bottles filled with filtered seawater at the beginning of the incubation (TC<sub>0</sub>). For the respiration measurements, each bottle contained ten mysids in filtered seawater. Then, after the incubation, the TC was measured (TC<sub>e</sub>). Finally, TC was measured in the blank bottles incubated with filtered seawater, over the same period, (TC<sub>e-blank</sub>). TC was determined in every case together with the total alkalinity (TA, μmol L<sup>-1</sup>) by the VINDTA 3C coulometric detection method (Mintrop et al. (2000)). The titration of certified reference material for oceanic CO<sub>2</sub>, CRMs 117, was used to test the performance of the equipment. Measurements of CRMs were within ±1.0 μmol kg<sup>-1</sup> of the certified value for TC and ±2 μmol kg<sup>-1</sup> for TA. Thus, it had a precision of ±1 μmol TC L<sup>-1</sup> and ±2 μmol kg<sup>-1</sup> for TA ([www.MARIANDA.com](http://www.MARIANDA.com)). Once the TC concentration in every bottle was measured, the rate of TC increase ( $\delta_{TC}$ ) was calculated as in Eq.4.3:

$$\delta_{TC} = ((TC_e - TC_0) - (TC_{e-blank} - TC_0)) \cdot V / t \quad (4.3)$$

Where V is the volume of the incubation bottle in liters (which was 1L during the entire study) and t the incubation time expressed in h. Thus,  $\delta_{TC}$  is expressed in μmol TC h<sup>-1</sup>.

Note that TC does not include the inorganic carbon that has been converted into CaCO<sub>3</sub> (carbonate sequestration, Eq.1.1). This value can be measured by the assessment of the increase in TA. In our hands, TA remained constant within the average standard deviation of the method ( $\pm 2 \mu\text{mol kg}^{-1}$ ). Thus, we assume the carbonate sequestration was not significant in this experiment during the incubation periods.

Therefore, as the production of one mol of CO<sub>2</sub> impacts directly on the TC by adding one mol to the total inorganic carbon in sea water, the increase in TC is directly related to the CO<sub>2</sub> production (Eq. 4.4).

$$R_{CO_2} = \delta_{TC} \quad (4.4)$$

Where  $R_{CO_2}$  is expressed in  $\mu\text{mol CO}_2 \text{ h}^{-1}$ .  $R_{CO_2}$  may be also expressed in  $\mu\text{mol CO}_2 \text{ h}^{-1} (\text{mg of protein})^{-1}$  when normalized by the biomass measurements in the same sample. This normalization facilitates the direct comparison of IDH to  $R_{CO_2}$ . For further comparison, using volume units (i.e.,  $(\mu\text{L CO}_2 \text{ h}^{-1} (\text{mg of proteins})^{-1})$ , one simply multiplies by  $22.4 \mu\text{L } \mu\text{mol}^{-1}$ .

#### 4.2.6 Respiratory Quotient (RQ) calculations

The respiratory quotient (RQ) is a dimensionless ratio of the CO<sub>2</sub> produced to the O<sub>2</sub> consumed. Here, in Eq. 4.5, we assessed the RQ as the ratio between  $R_{CO_2}$  (expressed in  $\mu\text{mol CO}_2 \text{ h}^{-1} (\text{mg of protein})^{-1}$ ) and the  $R_{O_2}$  (expressed in  $\mu\text{mol O}_2 \text{ h}^{-1} (\text{mg of protein})^{-1}$ ) in samples related to the same incubation conditions (e.g. well-fed females, well-fed males, nutrient-limited females, and non-nutrient-limited males).

$$RQ = R_{CO_2}/R_{O_2} \quad (4.5)$$

#### 4.2.7 Biomass

Proteinaceous biomass was determined on aliquots from sample homogenates preserved for enzyme activities. We assayed the protein with the method of Lowry et al. (1951) as modified by Rutter (1967) and Markwell et al. (1981) using bovine serum albumin (BSA) as the standard. Biomass was reported in mg of protein ( $\text{mL of homogenate})^{-1}$ . It may also be reported in mg of protein  $\text{L}^{-1}$ , knowing the homogenate volume and the sea-water volume of the sample. To facilitate further comparison, protein (P) can be converted into dry weight (DW), by using the ratio from Herrera et al. (2011a), where 1mg of DW corresponds to  $0.74 \pm 0.05$  mg protein. Taking this into account, the relation between DW from P would be expressed as Eq.4.6.

$$DW = P \cdot 1.35 \quad (4.6)$$

#### **4.2.8 Statistical Analysis**

In every experiment, the Student's t-test (Sachs (2012)) was applied to quantify the differences between pairs of datasets (e.g. females and males, well-fed and nutrient-limited females, bloom and post-bloom conditions) by the t-test function of the CRAN R Stats package (v.3.4.2, Eddelbuettel et al. (2011)).

### **4.3 Results and Discussion**

#### **4.3.1 Impact of trophic conditions on potential respiration ( $\Phi$ and NADP-IDH) and $R_{CO_2}$ /NADP-IDH ratios in ovigerous females of two communities of *L. lingvura***

The trophic conditions under which the community had been raised had an impact on the development of the catabolic metabolism enzymes (Fig.4.1), as the respiratory enzymatic activities normalised per biomass were significantly higher ( $p<0.001$ ) in those females that belonged to a bloom community than those that belonged to a post-bloom community.

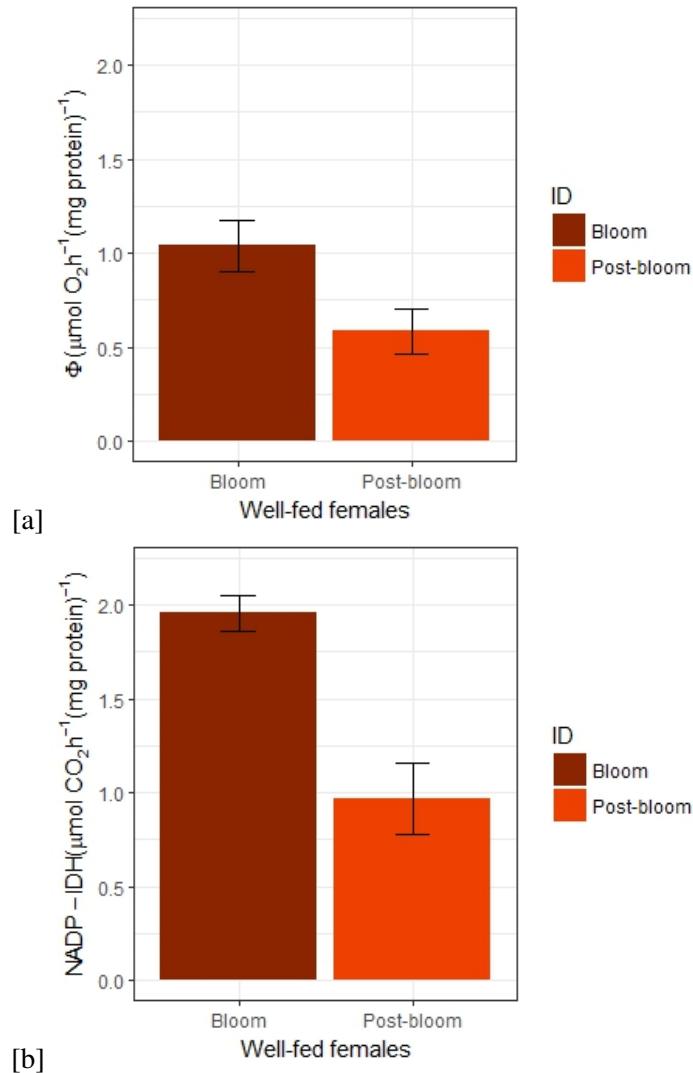


Figure 4.1: Differences in normalised respiratory enzymatic activities in well-fed females of *L. lingvura* from a bloom and from a post-bloom community. Note that  $\Phi$  represents the potential  $O_2$  consumption of the mysids and NADP-IDH is the potential  $CO_2$  production of the NADP-IDH enzyme in the mysids.

The variability of  $\Phi$  between bloom females ( $1.04 \pm 0.14 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ), and post-bloom females ( $0.58 \pm 0.12 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ) is consistent with the variability of  $\Phi$  in other studies of *L. lingvura*. They are from 30 to 60% lower than those reported by Herrera et al. (2011b) ( $1.53 \pm 0.43 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ), which were sampled under mesotrophic conditions during spring. On the other hand, they are 2 to 4-fold higher than the values obtained on samples from October 2015 ( $0.25 \pm 0.06 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ) (Bondyale-Juez et al. (2017)), during the warmest and most oligotrophic season in the year (Jurado-Ruzafa et al. (2019)). Furthermore, our results fall within the range ( $1.47 \pm 0.76 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$ ) of the Canarian mesozooplankton (500-1000 μm) reported by Osma et al. (2016c).

The variability of NADP-IDH, based on the CO<sub>2</sub>-producing enzyme NADP-IDH, was similar. It was significantly higher ( $p < 0.001$ ) in female mysids raised under bloom conditions ( $1.96 \pm 0.09 \mu\text{mol CO}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ) than in the post-bloom females ( $0.97 \pm 0.19 \mu\text{mol CO}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ). These are the first NADP-IDH measurements on two different communities of *L. lingvura*.

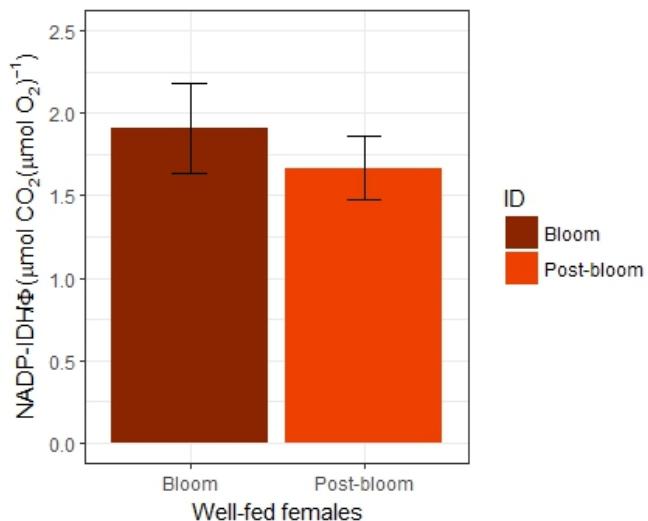


Figure 4.2: NADP-IDH to  $\Phi$  proportion in well-fed females of *L. lingvura* belonged to a bloom and to a post-bloom community. Note that  $\Phi$  represents the potential O<sub>2</sub> consumption of the mysids and NADP-IDH is the potential CO<sub>2</sub> production of the NADP-IDH enzyme in the mysids.

Besides, the ratio between NADP-IDH and  $\Phi$  allows the assessment of the relationship between the decarboxylating facet and the ATP-producing-O<sub>2</sub>-consuming facet in respiration. Both communities followed a similar pattern (Fig. 4.2), where NADP-IDH ranged between 1.5 and 2.2 times  $\Phi$  activ-

ity. These results suggest that the enzymatic pathways related to the organic carbon oxidation phase of cellular respiration are prepared to exceed the demands of the ETS. This may serve as an advantage when, eventually, the mysid has a high energy requirement. Once this requirement is met, NADPH excess may be redirected to lipid synthesis by the NADP-IDH substrate cycle (Sazanov and Jackson (1994)). There was still a slight difference between both communities ( $p=0.1098$ ) that may be related to the differences in the intensity of the reproductive activity that females were experiencing during the moment of collection. Higher results in the NADP-IDH/ $\Phi$  ratio in the bloom females may be linked to higher lipid demands during gametogenesis, egg production, and breeding (Lawrence (1976); Lee et al. (2006)).

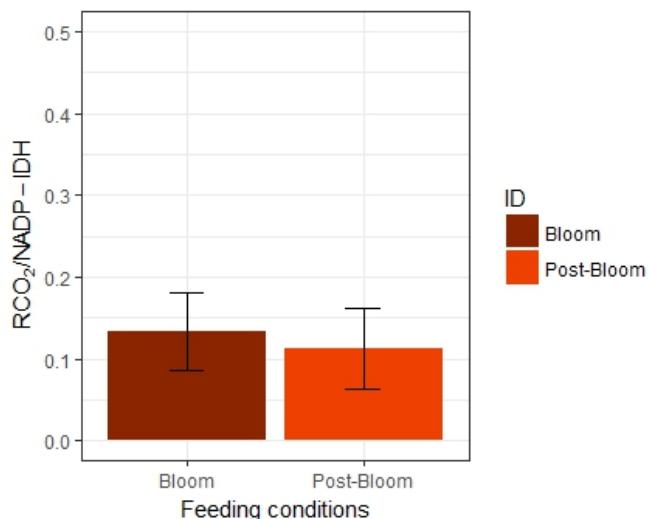


Figure 4.3:  $R_{CO_2}$ /NADP-IDH ratio in well-fed females of *L. lingvura* belonged to a bloom and to a post-bloom community.

Whereas the enzymatic activities were higher in bloom females than in post-bloom females, the  $R_{CO_2}$ /NADP-IDH ratio was not significantly different between the two communities ( $p=0.5623$ ) (Fig. 4.3). Thus, we suggest that an average ratio of  $0.12 \pm 0.05$  would accurately characterize the physiological response for CO<sub>2</sub> production in well-fed females in *L. lingvura* communities (Fig.4.3). Bloom females had slightly, but not significantly, higher values than did post-bloom females ( $0.13 \pm 0.05$  and  $0.11 \pm 0.05$  respectively). These results are lower than expected in comparison with the average O<sub>2</sub> consumption ratios in the literature (0.81 in Bondyale-Juez et al. (2017); from 0.66 to 1 in Herrera et al. (2011b) for those mysids assessed close to dawn). However, they were close to the obtained ratios related to the excretion metabolism during the first 18h of starvation ( $0.11 \pm 0.05$  in Fernández-Urruzola et al. (2011)).

#### **4.3.2 Impact of sex differences and starvation on potential respiration ( $\Phi$ and NADP-IDH)**

Neither the O<sub>2</sub> consumption, the CO<sub>2</sub> production, nor the enzymatic activities showed significant differences between sexes (p=0.1299 in  $\Phi$  between well-fed females and males; and p=0.6126 in NADP-IDH between well-fed females and males), resulting in an average  $\Phi$  of 0.6±0.12  $\mu\text{mol O}_2 \text{ h}^{-1} \text{ mg}$  of protein<sup>-1</sup> and an average NADP-IDH of 0.95±0.17  $\mu\text{mol CO}_2 \text{ h}^{-1} \text{ mg}$  of protein<sup>-1</sup> (Fig. 4.4).

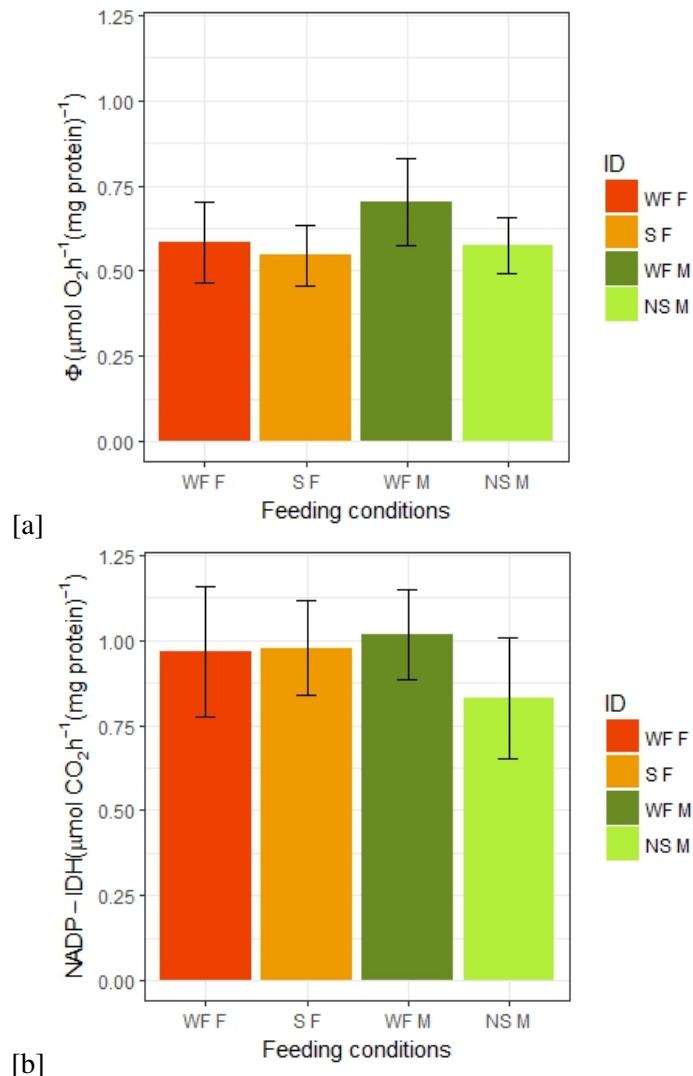


Figure 4.4: Differences in normalised respiratory enzymatic activities in well-fed and nutrient-limited (starved) females (WF F & S F) and well-fed and non-starved males (WF M & NS M) from the post-bloom community. Note that  $\Phi$  is the potential  $O_2$  consumption and NADP-IDH is the potential  $CO_2$  production of the NADP-IDH enzyme in the mysid.

These similarities also occurred between nutrient-limited females and the non-starved males which, after 12h of starvation were fed for two hours ( $p=0.5409$  in  $\Phi$  between nutrient-limited females and non-starved males, and  $p=0.1398$  in NADP-IDH between nutrient-limited females and non-starved males). However, starvation seemed to have a slightly higher impact on males than on females, as the reduction of the metabolic activities in both  $\Phi$  and NADP-IDH after starvation was slightly greater ( $p=0.070382$  in  $\Phi$  between well-fed and non-starved males whereas  $p=0.5333$  in  $\Phi$  between well-fed and nutrient-limited females).

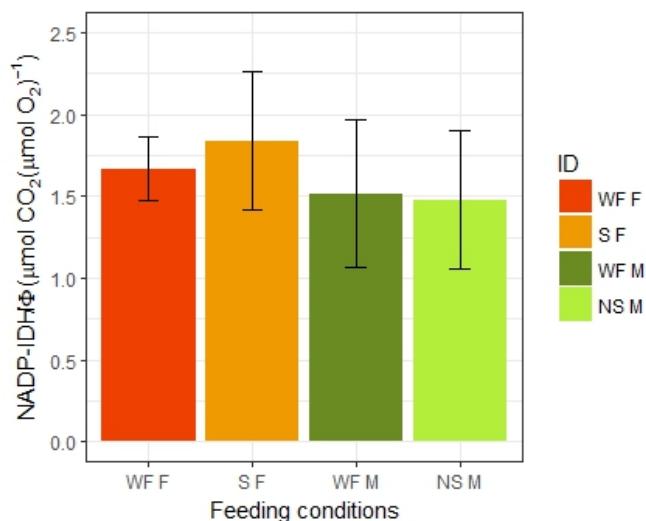


Figure 4.5: NADP-IDH to  $\Phi$  proportion in well-fed and nutrient-limited (starved) females (WF-F & S-F) and well-fed and non-starved males (WF-M, NS-M) from the post-bloom community.

These slight differences between sexes were even smaller in relation to the NADP-IDH/ $\Phi$  proportion ( $p=0.4784$  between sexes in well-fed mysids,  $p=0.1688$  between sexes in nutrient-limited mysids), resulting in NADP-IDH being  $1.63 \pm 0.4$  times  $\Phi$  (Fig.4.5). However, starvation produced a higher difference in this ratio between sexes (NADP-IDH/ $\Phi$ = $1.48 \pm 0.42$  in males, whereas NADP-IDH/ $\Phi$ = $1.84 \pm 0.42$  in females). These results may imply that ovigerous nutrient-limited females tend to protect their lipid synthesis pathways for gametogenesis, egg production, and breeding (Lawrence (1976); Lee et al. (2006)).

On the other hand, the difference between well-fed and nutrient-limited mysids was not significant ( $p=0.4006$  in females, and  $p=0.8762$  in males).

### 4.3.3 Impact of sex differences and starvation on the $R_{O_2}/\Phi$ and $R_{CO_2}/\text{NADP-IDH}$ ratios

Results with the  $R_{O_2}/\Phi$  ratio were not significantly different between sexes in the same post-bloom community of *L. lingvura* ( $p=0.7435$  between well-fed females and males and  $p=0.9466$  between nutrient-limited females and males, Fig.4.6). In contrast, the  $R_{O_2}/\Phi$  ratio in well-fed mysids was  $0.35 \pm 0.07$ , whereas in nutrient-limited mysids was slightly lower ( $0.27 \pm 0.08$ ). For the  $R_{CO_2}/\text{NADP-IDH}$  ratio, lower values were obtained. They ranged from 0.06 up to 0.25, without significant differences between sexes ( $p=0.7592$  between well-fed females and males). As in the case of the enzymatic activity, the physiological response of the non-starved males was similar to those females that were in starvation ( $p=0.7564$ ). This is coherent with the results in  $R_{O_2}$  reported by Osma et al. (2016b), where after having fed the nutrient-limited mysids, the  $R_{O_2}$  slightly increased during the first hours. The highest  $R_{O_2}$  values were attained 6 hours later, being still lower than the well-fed mysids' respiratory rates. Thus, we analysed these results assuming that non-starved males are behaving more likely nutrient-limited mysids than well-fed mysids. The  $R_{CO_2}/\text{NADP-IDH}$  ratio in well-fed mysids was  $0.12 \pm 0.04$ , whereas in nutrient-limited mysids it was only slightly higher,  $0.16 \pm 0.06$ . The  $R_{CO_2}/\text{NADP-IDH}$  ratio in the well-fed females from the bloom community was also around 0.12. Thus, this value seems to accurately describe the  $R_{CO_2}/\text{NADP-IDH}$  ratio in a well-fed community of *L. lingvura*.

Starvation differently affected the physiological O<sub>2</sub> consumption and CO<sub>2</sub> production facets of respiration. Although the differences between well-fed and nutrient-limited mysids cannot be considered statistically significant ( $p=0.3480$  in females ( $n=6$ ) and  $p=0.4176$  in males ( $n=6$ )), there is a clear trend in both cases (Fig.4.6). Whereas the decrease in food availability reduced both the  $R_{O_2}$  and the  $R_{O_2}/\Phi$  ratio, the  $R_{CO_2}$  and  $R_{CO_2}/\text{NADP-IDH}$  increased in both males and females. Note that this experiment was carried out in the same circadian time, as all the incubations began around three hours after dawn on mysids which had been adapted to culture conditions for more than one week.

Due to this imbalance between the physiological responses, the RQ was almost double in nutrient-limited mysids than in well-fed individuals (Fig.4.7). The RQ tended to increase during the first 12–18h of starvation, although the difference could not be considered statistically significant ( $p=0.1592$  between well-fed and nutrient-limited females, and  $p=0.1526$  between well-fed and non-starved males). The increase in these results during starvation is consistent with Romero-Kutzner et al. (2015)'s findings in bacteria, where the highest RQ resulted during the first hours of nutrient depletion. In addition, Mayzaud et al. (2005) reported lower RQ in Euphausiids *Meganyctiphanes norvegica* sampled during the late winter (March 2001) in

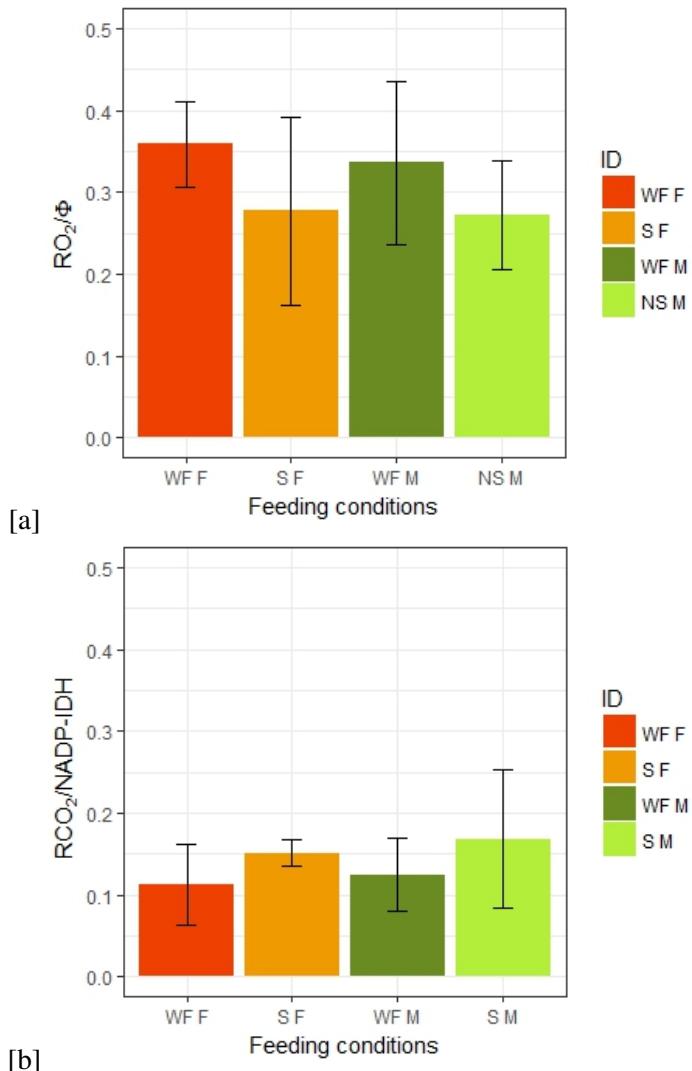


Figure 4.6: Effects of starvation and sex differences on the ratios of respiration to enzymatic activity in well-fed and nutrient-limited females (WF-F & S-F) and well-fed and non-starved males (WF-M & NS-M) from the post-bloom community

the North Atlantic than in those sampled during the more oligotrophic summer (September 2001) ( $1.29 \pm 0.1$  versus  $1.62 \pm 0.23$  respectively). Besides, after converting their Euphasiids  $R_{\text{CO}_2}$  from  $\mu\text{L CO}_2\text{h}^{-1}(\text{mg DW})^{-1}$  to  $\mu\text{mol CO}_2\text{h}^{-1}(\text{mg protein})^{-1}$  units, we found that the average *L. lingvura*  $R_{\text{CO}_2}$  of  $0.13 \pm 0.04 \mu\text{mol CO}_2\text{h}^{-1}(\text{mg protein})^{-1}$  was similar to the average  $R_{\text{CO}_2}$  in *M. norvegica* ( $0.13 \pm 0.03 \mu\text{mol CO}_2\text{h}^{-1}(\text{mg protein})^{-1}$ ). Furthermore, in the terrestrial Arthropoda Lepidoptera, an averaged RQ of 1.5-1.6 under starvation was reported, with a maximum of 2.09 (Kozhantshikov (1938)). This author also reported a peculiar reduction of O<sub>2</sub> consumption related to an

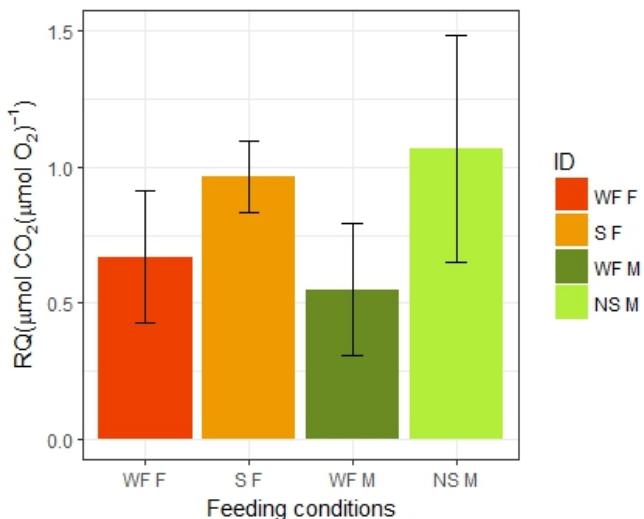


Figure 4.7: Effects of starvation and sex difference on the respiratory quotient (RQ) in well-fed and nutrient-limited (starved) females (WF-F & S-F) and well-fed and non-starved males (WF-M & NS-M) from the post-bloom community.

increase of fat synthesis from carbohydrates. As  $O_2$  is produced during this process, the  $O_2$  consumption related to respiration is reduced.

Besides, the enzyme NADP-IDH is also linked to lipid synthesis as one of the cellular sources of NADPH. An increase in  $CO_2$  production during starvation is consistent with other findings in the literature that assess lipogenesis during starvation. Lipid synthesis was not reduced in mice during the first 18h of starvation (Jansen et al. (1966)). In nutrient-limited eels an increase occurs in the activity of the tricarboxylate carrier that exports citrate from the mitochondria to the cytosol (Zara et al. (2000)). In the liver of these nutrient-limited fish, the cytosolic IDH is thought to be the main source of NADPH, as it is the receptor of carbon flowing from this extramitochondrial citrate towards isocitrate (Aster and Moon (1981)). This increase in cytosolic isocitrate during starvation would drive the enzyme NADP-IDH to generate both NADPH and  $CO_2$  while increasing lipid synthesis.

Taking all this into consideration, we hypothesized that during starvation in *L. lingvura*, nutrients are diverted to the maintenance of cell structures instead of to energy generation. This is consistent with the upregulation of metabolic marks linked to gluconeogenesis and the use of stored lipids reported by Teruya et al. (2019) during fasting in humans. In addition, as NADP-IDH is also related to the production of reduced glutathione, involved in the reactive oxygen species (ROS) scavenging (Jo et al. (2001)), the slight increase in the  $R_{CO_2}$  (from  $0.11 \pm 0.01 \mu\text{mol CO}_2\text{h}^{-1}(\text{mg protein})^{-1}$  in well-fed mysids to  $0.14 \pm 0.04 \mu\text{mol CO}_2\text{h}^{-1}(\text{mg protein})^{-1}$  in nutrient-limited

mysids) may reflect that anti-oxidation is a principal response to fasting Teruya et al. (2019).

Thus, we suggest that the increase in RQ during starvation is related to:

- The  $\text{CO}_2$  production during carbohydrate catabolism to produce NADPH, which is diverted from energy generation to the structural lipid synthesis and the production of the ROS scavenger reduced glutathione.
- The  $\text{O}_2$  consumption linked to the ATP production is limited because the in-cell ATP requirements are minimised.
- Finally, an  $\text{O}_2$  production linked to the lipid synthesis reduced the  $R_{\text{O}_2}$  during starvation.

#### 4.3.4 Impact of long-term incubations on $R_{\text{CO}_2}/\text{NADP-IDH}$ ratios in females of *L. lingvura*

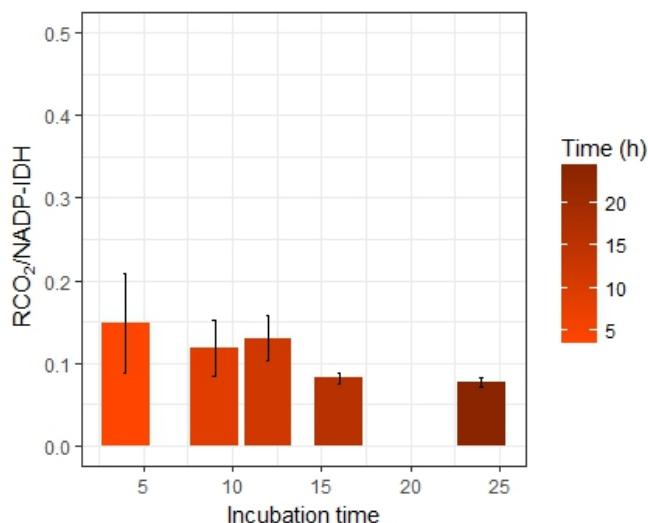


Figure 4.8:  $R_{\text{CO}_2}/\text{NADP-IDH}$  ratio variability during long-term incubation periods.

We assessed the impact of long-term incubations on well-fed females. In previous subsections,  $R_{\text{CO}_2}/\text{NADP-IDH}$  ratios were measured at the same circadian time and, in the first hours after feeding, were lower than those measured after 12h of starvation. Here, we measured the average ratio during the whole incubation period (from 4h to 24h) beginning after feeding.  $R_{\text{CO}_2}/\text{NADP-IDH}$  remained around  $0.13 \pm 0.04$  during the first 12 hours of incubation but, unlike other experiments made at the same circadian time, here the measurements dropped to less than 0.1 in longer incubations ( $0.08 \pm 0.01$ ) (Fig.4.8). As *L. lingvura*'s metabolism has a circadian rhythm (Herrera et al.

(2011b)), we hypothesized that the lower values in those periods longer than 12h were driven by lower activity conditions linked to those circadian rhythms. Thus, we recommend incubation periods shorter than 6h in future *L. lingvura* studies.

#### 4.4 Conclusions

- Nutrition prior to sampling, has a significant impact on the respiratory enzymatic activity of *L. lingvura*. Higher activities normalised per biomass were found in bloom communities than in post-bloom communities.
- An average  $R_{CO_2}$ /NADP-IDH ratio of 0.12 was obtained on well-fed mysids from both communities. This result was only slightly higher in nutrient-limited mysids (0.16). This value accurately characterized the physiological response of *L. lingvura*.
- Sex difference did not significantly impact enzymatic activities nor the ratios of respiratory physiology to enzyme activity.
- Starvation for longer than 12h decreased the  $R_{O_2}/\Phi$  ratios, whereas it increased the  $R_{CO_2}$ /NADP-IDH ratios, leading to higher RQs. Our interpretation is that energy production is diminished whereas lipid synthesis and anti-oxidation are strengthened.
- Incubation of mysids longer than 12h mixes starvation and circadian effects. We recommend incubations between 4 and 6 hours in future ecophysiological studies.



## Chapter 5

# Chapter 5: Metabolic responses of subtropical North Atlantic microplankton after a simulated deep-water upwelling event.

*We cannot solve our problems with the same thinking we used when we created them.*

Albert Einstein

**RESUMEN:** Durante el otoño de 2014 se instalaron nueve mesocosmos en las aguas oligotróficas del Nordeste del Atlántico subtropical, en Gran Canaria, con el objetivo de estudiar el impacto de la acidificación en una comunidad planctónica bajo la influencia de un afloramiento simulado de aguas profundas ricas en nutrientes. Entre otros parámetros, se midieron la Clorofila *a* (Chl<sub>*a*</sub>), la Respiración Potencial ( $\Phi$ ) y la biomasa (B) en el microplancton (0.7-50 $\mu$ m) tanto durante la fase oligotrófica (Fase I), como durante el bloom fitoplantónico (Fase II) y durante la fase posterior al bloom (Fase III). En este capítulo hemos analizado el ratio  $\Phi/\text{Chl}_a$  para monitorizar los cambios en el metabolismo de la comunidad microplanctónica. Cuando la comunidad estuvo dominada por los fotoautótrofos se obtuvieron valores  $< 2,5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  (siendo incluso  $< 1,7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  durante el bloom fitoplantónico). Cuando los valores se situaron entre  $2,5-7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  reflejaron una comunidad mixta o dominada por mixótrofos. Además, los ratios  $\Phi/\text{Chl}_a$  fueron  $> 7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  cuando proliferó el microzooplancton (llegando a  $10 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  en el momento del bloom de dinoflagelados heterótrofos). Utilizando la

primera derivada de la biomasa se ha observado una productividad neta máxima entre uno y dos días antes del pico de Chl<sub>a</sub><sup>-1</sup>, mientras que la mayor pérdida de biomasa se dió en las 24h siguientes. Siguiendo este mismo análisis, no hubo una productividad neta significativa ni durante la fase I ni durante la fase III. El análisis de la integral de  $\Phi$ , Chl<sub>a</sub> y B, por fase y mesocosmos, mostró una relación positiva entre  $\Phi$  y los niveles de pCO<sub>2</sub> durante la fase II, y una relación positiva entre Chl<sub>a</sub> y pCO<sub>2</sub> durante la fase III. Como a partir del final de la fase II proliferó una especie de alga tóxica (*Vicicitus globosus*) en los mesocosmos con mayor nivel de acidificación, la comunidad fue afectada: el microzoopláncton no proliferó, y el tiempo de retención de la Chl<sub>a</sub> aumentó. En estas comunidades los ratios  $\Phi/\text{Chl}_a$  ( $4.1 \pm 1.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ) fueron similares a los de una comunidad mixta o mixótrofa. Esto, unido a que *V. globosus* tiene un sistema fotosintético ineficiente a los niveles de pCO<sub>2</sub> actuales (carece de pirenoide) y la capacidad de cambiar de una forma flagelada a una forma ameboide, parece indicar que *V. globosus* es más un mixótrofo que un fotoautótrofo, y que su crecimiento se refuerza cuando los niveles de pCO<sub>2</sub> son altos. Esto favorece la relación positiva entre Chl<sub>a</sub><sup>-1</sup> y pCO<sub>2</sub> observada en el experimento. Teniendo en cuenta todo podemos concluir que los ratios  $\Phi/\text{Chl}_a$  refuerzan el conocimiento sobre el metabolismo de la comunidad, no sólo a nivel del ecosistema, sino también a nivel de especies.

**ABSTRACT:** In the autumn of 2014, as part of the German KOSMOS research program, nine large mesocosms were deployed in the oligotrophic sub-tropical North-eastern Atlantic close to the island of Gran Canaria. Their deployment was designed to address the acidification effects on a plankton community experiencing upwelling of nutrient-rich deep water. Among other parameters, chlorophyll *a* (Chl<sub>a</sub>), potential respiration ( $\Phi$ ), and biomass (B) were measured in the microplankton community (0.7-50  $\mu\text{m}$ ) during the oligotrophic phase (Phase I), the phytoplankton-bloom phase (Phase II) and the post bloom phase (Phase III). Here, we explored using the  $\Phi/\text{Chl}_a$  ratio to monitor shifts in the microplankton community metabolism.  $\Phi/\text{Chl}_a$  values below  $2.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  indicated a community dominated by photoautotrophs. The ratio reached lows below  $1.7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  during the phytoplankton bloom. When the  $\Phi/\text{Chl}_a$  ratios ranged between  $2.5-7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , they indicated a mixed community or a mixotroph dominance. When these ratios rose above  $7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , microzooplankton proliferated. In a heterotrophic dinoflagellate bloom, the  $\Phi/\text{Chl}_a$  ratios rose as high as  $10 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ . Here, in our data analysis, we explored using Leibnizian differential and Newtonian integral analysis to elucidate plankton dynamics from time-courses of static biomass parameters. Derivative analysis of B revealed maximum net productivity occurring between one and two days before the Chl<sub>a</sub> peak, whereas net maximum loss occurred 24h after this peak. According to this analysis, no significant net productivity occurred either on Phase I nor during Phase III. The integral analysis of  $\Phi$ , Chl<sub>a</sub> and B, per phase and mesocosms, reflected a positive rela-

tionship between  $\Phi$  and  $p\text{CO}_2$  during Phase II and between  $\text{Chl}_a$  and  $p\text{CO}_2$  during Phase III. At the end of Phase II, a harmful algal species, *Vicicius globosus* bloomed in the high  $p\text{CO}_2$  mesocosms and the community was disrupted. In these mesocosms, microzooplankton did not proliferate, and  $\text{Chl}_a$  retention time in the water column increased. In these communities, the  $\Phi/\text{Chl}_a$  ratios ( $4.1 \pm 1.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ) were similar to those in a mixed plankton community. These values, combined with the knowledge of *V. globosus*' inefficient photosynthetic system at current  $p\text{CO}_2$  levels (it lacks pyrenoid bodies) and the knowledge of its ability to quickly shape-shift from a flagellate-shaped microorganism to an amoeboid-shaped one, suggest that *V. globosus* is more a mixotroph than a photoautotroph. Its proliferation is enhanced when high  $p\text{CO}_2$  levels occur, resulting in positive  $\text{Chl}_a$  to  $p\text{CO}_2$  relationships. All in all, the combined use of  $\Phi/\text{Chl}_a$  ratios, Leibnizian differential, and Newtonian integral analysis strengthens our understanding of plankton community dynamics, not only at an ecosystem scale, but also at a species scale.

## 5.1 Introduction

The global economy, based on the combustion of fossil resources, has caused the increase in anthropogenic carbon dioxide ( $\text{CO}_2$ ) emissions over the past 250 years. This rise, combined with the spread of deforestation and agriculture, has resulted in the highest atmospheric  $\text{CO}_2$  levels (400ppm) in the last 800,000 years and keeps increasing up to one order of magnitude faster than it increased over the previous millions of years (Doney and Schimel (2007)). Oceans are taking up nearly one third of the anthropogenically produced  $\text{CO}_2$ , leading to effects on seawater carbonate chemistry, now denominated, the ocean acidification (OA) (Doney et al. (2009)). Even when positive impacts on photosynthetic carbon fixation have been reported when nutrients are available (Riebesell et al. (2007)), the resulting organic matter with low nutritional value may affect the rest of the community (Doney et al. (2009)). Under eutrophic conditions, these changes had different effects. In the most acidified environments, picophytoplankton growth is stimulated, and excreted dissolved organic carbon (DOC) becomes an important fraction of the organic matter. The shift to more DOC reduces the carbon flux, meaning that there is less sinking material in the water column (Paul et al. (2015)).

For 55-days, a KOSMOS mesocosm experiment (Taucher et al. (2017)) was conducted, in the subtropical North-Atlantic coastal waters off Gran Canaria, in order to reveal the effects increasing  $p\text{CO}_2$  would have on the oligotrophic plankton community that was simultaneously experiencing an upwelling event. After 24 days of Oligotrophy (Phase I), an upwelling event

was simulated by enriching nine mesocosms with nutrient-rich deep-water. The pCO<sub>2</sub> levels in these mesocosms ranged from 400μatm, to 1500μatm. Then, the phytoplankton community bloomed (Phase II) until nutrient depletion. At this point, the Post-bloom (Phase III) began. All the mesocosms showed similar trends in Chlorophyll\_a (Chl<sub>a</sub>) and Biomass (B) during these three phases (Taucher et al. (2018)). In the mesocosms with elevated pCO<sub>2</sub>, three changes were noted: (1) cell abundances of picocyanobacteria, *Synechococcus* sp. increased; (2) interspecific shifts favouring the large diatom species, *Guinardia strata* and the acrylic-acid producing Prymmnesiophyceae, *Phaeocystis globosa*; and (3) the harmful algal species (HAS), *Vicicitus globosus* proliferated (Taucher et al. (2018); Hernández-Hernández et al. (2018); Riebesell et al. (2018)). *V. globosus* disrupted the plankton community in these mesocosms, inhibiting the growth of micro and mesozooplankton, and increasing the Chl<sub>a</sub> retention time in the water column (Taucher et al. (2018); Riebesell et al. (2018); Algueró-Muñiz et al. (2019)).

Here, we assessed how ocean acidification impacted the oligotrophic nano and microplankton respiratory metabolism during the simulated deep-water upwelling event. During the different stages of the phytoplankton bloom, shifts in the community, ranging from an autotroph-dominated community to a heterotroph-dominated one, was detected from the changes in the ratio between Potential Respiration rate ( $\Phi$ ) and Chl<sub>a</sub>. Accordingly, we focused on the effects of simulated upwelling on microplankton metabolism and its relationship with the shifts in the microplankton composition.

As  $\Phi$  is a measurement of the plankton catabolic metabolism (see Chapter 1, section 1.3.3), and as Chl<sub>a</sub> is related to photosynthetic capacity in photautotrophs, we expect the  $\Phi/\text{Chl}_a$  ratio to behave as follows:

- Low  $\Phi/\text{Chl}_a$  values should represent photoautotroph dominance, due to a high levels of Chl<sub>a</sub> or low levels of ETS activity, in the plankton community. During the bloom phase, the lowest  $\Phi/\text{Chl}_a$  values in the experiment, were expected.
- High  $\Phi/\text{Chl}_a$  values should represent higher ETS activity and low Chl<sub>a</sub> levels and would indicate a heterotroph-dominated plankton community.
- Finally, intermediate  $\Phi/\text{Chl}_a$  values should reflect a balanced plankton community where photoautotrophic and heterotrophic metabolism coexist.

In addition, the coexistence of both photosynthetic and heterotrophic pathways in the same organism (mixotrophy) that arose in both planktonic algae and protozoa, challenge the classic conception of a pure auto- or heterotroph marine food web (Sanders (1991); Stoecker (1999); Caron (2016); Stoecker et al. (2017)). However, the mixotrophic effects on the food web

structure, on the trophic transfer efficiency, and on the vertical carbon flux ( $F_c$ ) have been omitted in ecosystem models until recently (Sanders (1991); Caron (2016); Ward and Follows (2016)).

In plankton communities, harmful algal species (HAS) were thought to be strictly photoautotrophic until the last decade, when Burkholder et al. (2008) highlighted mixotrophy as the major mode of nutrition among these organisms. In the oligotrophic waters, mixotrophy not only functions well, but is particularly advantageous, as organisms find the potentially limiting nitrogen and phosphate much more concentrated in microbial prey (usually bacteria) than in the water column (Sanders (1991)). Although mixotrophy is thought to impact positively on the food web (Ward and Follows (2016)), when HAS is dominant, the impact may be negative due to the toxic effects on micro and mesozooplankton, and on higher levels of the food chain. These effects may also impact negatively on  $F_c$ .

As said before, in this experiment, elevated CO<sub>2</sub> treatments resulted in proliferation of HAS, *V. globosus*, that disrupted the plankton community (Riebesell et al. (2018)). Even though *V. globosus* has been cultured, strictly as a photoautotroph, in the literature (Chang et al. (2012); Chang (2015)), we hypothesize that mixotrophy is the major form of nutrition for this HAS. The combination of the oligotrophic environment and mixotrophy may be a key factor for the HAS proliferation in these areas and this may lead to a disrupted food-web with low  $F_c$ .

## 5.2 Material and Methods

### 5.2.1 Experimental design

A 55-day in situ mesocosm experiment was carried out in Gando Bay, on the east coast of Grand Canary Island (Spain), during autumn (October–November 2014). Nine 35m<sup>3</sup>-mesocosms, with increasing partial pressure CO<sub>2</sub> (pCO<sub>2</sub>) treatments ranging from ambient levels to on average 1140μatm, were deployed in order to study the physical, ecological, and biogeochemical responses before, during and after a simulated upwelling event. The detailed description of the study site, experimental design, key events, overall sampling and measurements was reported in Taucher et al. (2017).

### 5.2.2 Sampling

Seawater was collected every two days during phase I and III and every day during phase II using the depth-integrated water-sampler (IWS, HYDROBIOS, Kiel) taking in a total volume of 5L uniformly distributed over the

mesocosms depth (0-13m in every mesocosm, and in the Atlantic as a reference level). An average volume of 60-70L per mesocosms was collected.

The water column plankton community between 0.7 and 50 $\mu$ m was sampled for metabolic assays every four days during phase I and III and every two days during phase II, as an aliquot (2-5L) from the IWS sampling per mesocosms. It was filtered through a 50 $\mu$ m net before passing through a GF/F glass fiber filter (Whatman 0.7 $\mu$ m nominal pore size) until the filter was saturated (always before 1h of filtration). Samples were frozen in liquid nitrogen and stored at -80°C until their treatment in the laboratory. There, samples were homogenized at 0-4°C in 0.1M phosphate buffer, by a Teflon® 2mL-pestle PYREX® Potter-Elvehjem tissue grinder (homogenizer) at 2600rpm for 2 min. Crude homogenates were centrifuged (0-4°C) at 4000rpm for 10 min.

### **5.2.3 Nutrients, Chlorophyll *a* (Chl<sub>*a*</sub>) and Water-column community composition**

The overall data related to the nutrient variability, Chl<sub>*a*</sub>, and the water-column community composition (both phytoplankton and microzooplankton) used in the interpretation and discussion of the results of this study were reported by Taucher et al. (2017); Hernández-Hernández et al. (2018) and Algueró-Muñiz et al. (2019).

Chl<sub>*a*</sub> in the IWS was measured from HPLC analyses. Subsamples were homogenized in acetone using glass beads in a cell mill. They were centrifuged at 5200rpm for 10min at 4°C. The supernadant was filtered (0.2 $\mu$ m PTFE filters, VWR International) and used to determine the phytoplankton pigment concentration in a Thermo Scientific HPLC Ultimate 3000 with an Eclipse XDB-C8 3.5 u 4.6 x 150 column. CHEMTAX software (Mackey et al. (1996)) was used to classify phytoplankton on the base of taxon-specific pigment ratios. Every 8 days, subsamples from the IWS were used to determine the microzooplankton abundances. Samples were immediately fixed with acidic Lugol (1-2%) solution and stored in the dark in 250mL brown glass bottles. To identify and count the microzooplankton, the Utermöhl (1931) technique was applied using an inverted microscope (Axiovert 25, Carl Zeiss) (Algueró-Muñiz et al. (2019)).

### **5.2.4 Potential respiration**

Potential respiration ( $\Phi$ ) was determined from ETS activity (Packard and Williams (1981); Packard and Christensen (2004)) incorporating the modifications described in Kenner and Ahmed (1975) and Gómez et al. (1996). The original assay (Packard (1971)) is descended from a classical succinate-INT reductase assay (Nachlas et al. (1960)) in which the maximum velocity

is assured by saturating the reaction with the reactants. The ETS measurements were made on the centrifuged supernatant from the metabolic assay samples (see section 5.2.2).

The absorbance increase of the INT-formazan at 490nm was continuously monitored spectrophotometrically at 18°C in 1cm path-length cuvettes for 8min. The regression line of absorption versus time was used to calculate the ETS activity and potential respiration rates ( $\Phi$ ) after Packard and Christensen (2004). The tetrazolium salt, INT (p-Iodonitrotetrazolium Violet, Sigma #I8377) was reduced by the respiratory ETS enzymes, replacing O<sub>2</sub> as the electron acceptor (Lester and Smith (1961); Smith and McFeters (1997)). INT accepts two electrons whereas O<sub>2</sub> would accept four. Thus, the INT-formazan production rate ( $\delta\text{INT}/\delta t$ ) is stoichiometrically related by a factor of 2 to ETS activity ( $\delta e^-/\delta t$ ) and by a factor of 0.5 to  $\Phi$ . In other words,  $\delta e^-/\delta t=2 \cdot \delta\text{INT}/\delta t$  and  $\Phi=0.5 \cdot \delta\text{INT}/\delta t$  were  $\Phi$  is in units of  $\mu\text{mol O}_2$ , if INT is in  $\mu\text{mols}$ . Blanks were run without the ETS substrates in order to subtract the contribution of the background non-enzymatic and enzymatic reduction of the INT (Maldonado et al. (2012)).  $\Phi$  was reported in  $\mu\text{mol O}_2 \text{ min}^{-1}(\text{mL of homogenate})^{-1}$  units. To convert to  $\mu\text{LO}_2 \text{ min}^{-1}(\text{mL of homogenate})^{-1}$ , multiply by 22.4  $\mu\text{L } \mu\text{mol}^{-1}$ .

### 5.2.5 Biomass and Net Productivity

Biomass content (B) was determined from samples kept for enzyme activities by the protein method of Lowry et al. (1951) modified by Rutter (1967) using bovine serum albumin (BSA) as the standard. Biomass was reported in mg of protein L<sup>-1</sup>.

Net productivity<sup>1</sup> ( $P_N$ ) was determined by calculating the first derivative of the B time-series. Thus,  $P_N=\delta B^-/\delta t$ . Positive values reflected a net protein formation (formation processes exceeded the combination of "losses" caused by remineralization, grazing, and sinking). On the other hand, negative values of  $P_N$  indicated a net loss of proteinaceous biomass.

### 5.2.6 Data Analysis and Statistics

The Mann-Whitney test was used to compare  $\Phi$ , B and Chl<sub>a</sub> distributions between the different mesocosmos (Hollander et al. (2013)).

Integrative analysis per phase and mesocosm was applied to the area below the data curves by trapezoidal integration using the trapz function of the pracma R-Cran package (Borchers (2017)).

<sup>1</sup>We use the term "productivity" instead of "production" because the temporal scale of these data are lower than a year. If these results were extrapolated to higher scales, we should express the results in terms of "Net production"

## 5.3 Results

### 5.3.1 Responses to CO<sub>2</sub>-treatments under low-nutrient conditions.

During the oligotrophic phase, all the mesocosms behaved in the same way. The Φ/Chl<sub>a</sub> ratio reflected a mesocosm shift towards a weaker photoautotroph community inside the mesocosms. Whereas average values between 2.0 and 4.7  $\mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  were obtained in the Atlantic samples, mesocosms resulted in higher values, ranging between 3.4 and 8  $\mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  (Fig.5.1, I).

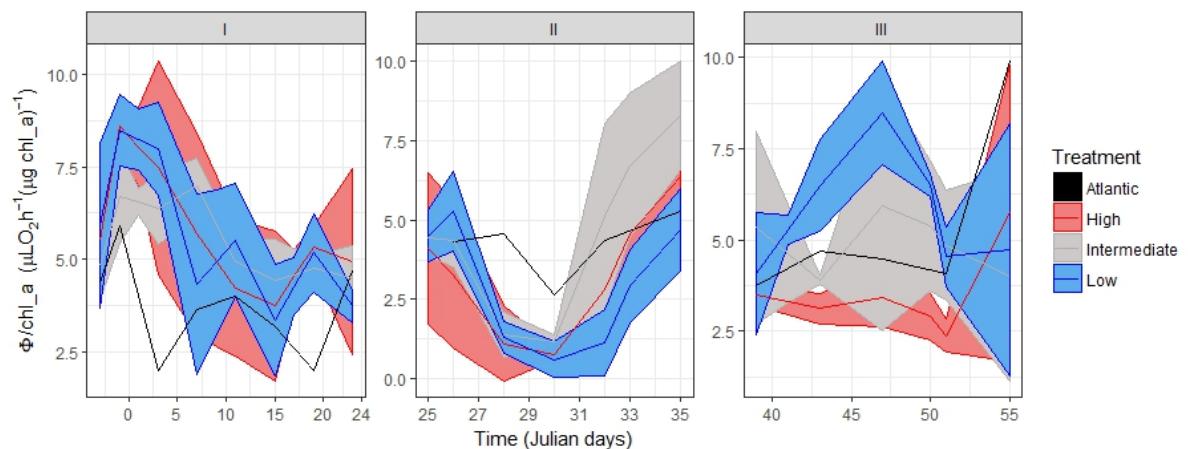


Figure 5.1: Potential respiration per Chl<sub>a</sub> during Phase I: Oligotrophic phase; Phase II: Bloom phase after deep-water addition; and Phase III: Post-bloom phase after nutrient-depletion.

According to Hernández-Hernández et al. (2018) the phytoplankton community evolved from a cyanobacteria-dominated community to a mixed one in all the mesocosms: diatoms, prymnesiophytes, cryptophytes and cyanobacteria. Besides, microplankton (mainly aloricate ciliates and dinoflagellates) remained in low concentrations (Algueró-Muñiz et al. (2019)). Nanoplankton dominated the biomass (50%) but not the Chl<sub>a</sub> (30%) (Hernández-Hernández et al. (2018)), which led to a more heterotrophic community.

Throughout the entire phase,  $P_N$  remained around 0 mg of proteins L<sup>-1</sup>d<sup>-1</sup> until the last two days. Then, the community was fuelled by a natural nutrient addition from a Sahara-dust deposit event (“Calima”) (Fig.5.2, I). Considering the integral analysis of the mesocosms’ biomass and metabolic capacity ( $\int_{t=0}^{t=23} \text{Chl}_a, \int_{t=0}^{t=23} B, \int_{t=0}^{t=23} \Phi$ ), no significant CO<sub>2</sub>-related effects were observed during this phase (Fig.5.3, I).

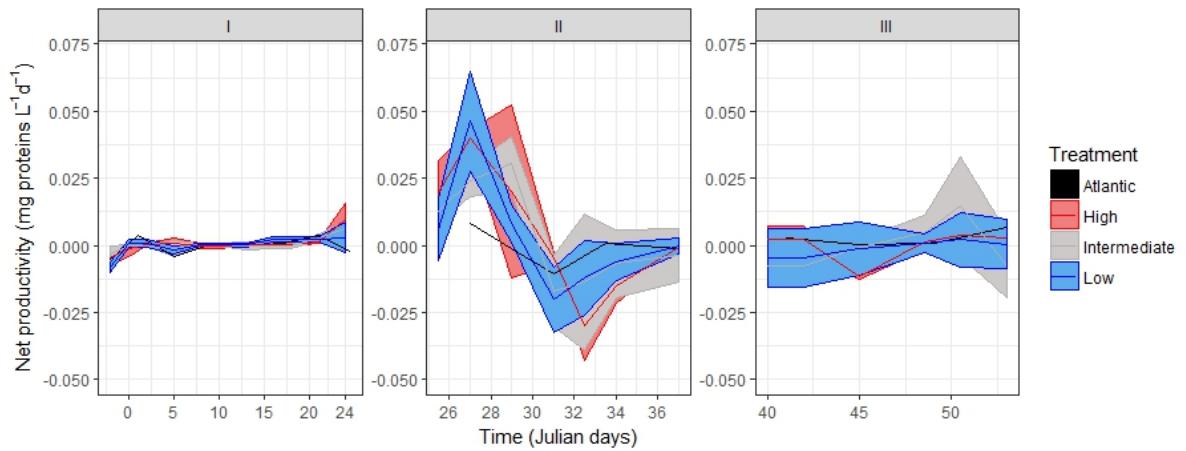


Figure 5.2:  $P_N$  of the microplankton community, as the first derivative of biomass, during Phase I: Oligotrophic phase; Phase II: Bloom phase after deep-water addition; and Phase III: Post-bloom phase after nutrient-depletion. Note that negative values reflects the net loss of biomass, when the combination of sinking, remineralization and grazing outnumbers the biomass production by both autotrophs and heterotrophs

### 5.3.2 Responses during the deep-water addition event.

The deep-water addition on T24 (Day 24) resulted on a phytoplankton bloom between T28 (Day 28) and T30 (Day 30) in all the mesocosms (Taucher et al. (2017); Hernández-Hernández et al. (2018); Taucher et al. (2018)). During the bloom, the average  $\Phi/\text{Chl}_a$  ratio drop to the lowest values in the whole experiment ( $<1.7\mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  in all the mesocosms) due to the photoautotroph-dominated community. The lowest average values occurred during the diatom bloom in T30, ranging from 0.6 to  $1.2\mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  among the mesocosms (Fig.5.1, II).

Even though all the mesocosms behaved in a similar way during the bloom, afterwards, until nutrient depletion, the plankton communities evolved differently. The integral analysis of  $\text{Chl}_a$ ,  $\Phi$  and B (as protein) in Phase II ( $\int_{t=25}^{t=37} \Phi, \int_{t=25}^{t=37} \text{Chl}_a, \int_{t=25}^{t=37} B$ ) in the communities dominated by the micro-phytoplankton, showed a positive trend with increasing  $\text{pCO}_2$  levels (Fig.5.3, II). However only the relationship between  $\Phi$  and  $\text{pCO}_2$  was found to be clearly positive, being particularly strong after extracting the data of M2 and M8, the most acidified mesocosms. Both M2 and M8 developed a harmful algal species (HAS), *Vicicitus globosus* at the end of Phase II that disrupted the community. These two mesocosms reached the lowest micro- and mesozooplankton levels observed during the entire whole experiment (Riebesell et al. (2018); Algueró-Muñiz et al. (2019)) because zooplankton was outcompeted by the presence of *V. globosus* (Riebesell et al. (2018)).

In contrast, microzooplankton, dominated by dinoflagellates, proliferated in the intermediate mesocosms (M3,M4 and M7) at the end of phase II, where the HAS did not grow (Algueró-Muñiz et al. (2019)).

Surprisingly, even though the microzooplankton did not develop in mesocosms M2 and M8,  $\Phi$  reached as high values in M2 and M8 ( $5.3 \pm 0.7 \mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$ ) at the end of Phase II as in mesocosms M3, M4 and M7 ( $4.7 \pm 1.1 \mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$ ), where microzooplankton bloomed (Fig.5.3, II; Appendix D).

$P_N$  was maximum between one and two days before the Chl<sub>a</sub> peak in all the mesocosms ( $0.039 \pm 0.013 \text{ mg proteins L}^{-1}\text{d}^{-1}$ ) (Fig.5.2,II). On the other hand, during the 24h after the peak (Fig.5.2), the combination of remineralization, grazing, and sinking processes resulted in an average biomass net-loss of  $-0.023 \pm 0.008 \text{ mg proteins L}^{-1}\text{d}^{-1}$ . Note that, if Primary Productivity<sup>2</sup> (PP) data were available, we could combine both measurements (expressed in terms of  $\text{mgC L}^{-1}\text{d}^{-1}$ ) in order to calculate the Biomass Loss (BL) in the system ( $P_N = \text{PP-BL}$ ).

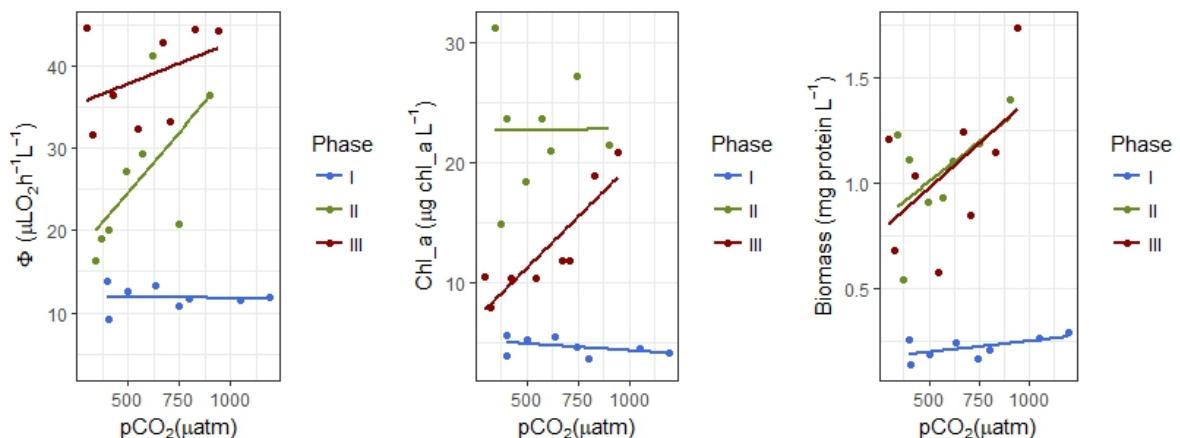


Figure 5.3: Integral analysis of Potential respiration, Chl<sub>a</sub> and Biomass by phase of experiment and treatment. Note that every point symbolizes the integral of each parameter by the related phase in the mesocosm with the CO<sub>2</sub>-treatment reflected in the x-axis. Blue colour represents Phase I, green represents Phase II, and Phase III is represented in red.

<sup>2</sup>We use the term "productivity" instead of "production" because the temporal scale of these data are lower than a year. If these results were extrapolated to higher scales, we should express the results in terms of "Primary production"

### 5.3.3 Responses to the CO<sub>2</sub>-treatments during the nutrient depletion phase.

After the nutrient depletion, the communities differed among the treatments. The Φ/Chl<sub>a</sub> ratio resulted in maximum values when the microzooplankton, dominated by dinoflagellates, peaked in both low pCO<sub>2</sub> (M1, M5 and M9) and intermediate pCO<sub>2</sub> mesocosms (M3, M4 and M7) (Fig.5.1, III; Algueró-Muñiz et al. (2019)). Here, values rose sharply up to 10 μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup>. In addition, as we have seen before, in M2 and M8, high pCO<sub>2</sub> levels fuelled the Chrysophyceae community dominated by *V. globosus* (Taucher et al. (2018); Riebesell et al. (2018)). This HAS had a significantly negative effect on the dinoflagellates but not on the diatoms. They had a longer retention time in biomass, probably due to the lack of grazing (Taucher et al. (2018)). The Φ/Chl<sub>a</sub> ratio (4.1±1.5μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup><sup>3</sup>) remained between 2.5 and 7μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup>, reflecting a weaker heterotrophic community. However, although these values are lower than those related to dinoflagellate-dominated communities, they are not as low as expected for a photoautotroph dominated community (<1.7μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup> during the phytoplankton bloom in Phase II). The values (4.1±1.5μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup>) are closer to those of the mixed community from the surrounding Atlantic water (4.3±1.73 μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup>), or the former community in Phase I from all the mesocosms (between 3.4 and 8 μLO<sub>2</sub>h<sup>-1</sup> μg Chl<sub>a</sub><sup>-1</sup>).

Besides, changes in the communities did not impact P<sub>N</sub> during the Post-Bloom phase. Although it was more variable than in Phase I, P<sub>N</sub> trends were to stay close to the minimum in all the mesocosms (Fig.5.2, III). This suggests that, after the bloom and the nutrient depletion, the community tended to reach a steady state close to the one in Phase I.

Finally, the integral analysis in B and Φ ( $\int_{t=39}^{t=55} B, \int_{t=39}^{t=55} \Phi$ ) did not find a significant relationship with pCO<sub>2</sub> (Fig.5.3, III), particularly after extracting the M2 and M8 data linked to the HAS-disrupted community. The integral analysis in Chl<sub>a</sub> ( $\int_{t=39}^{t=55} Chl_a$ ) illustrated a clearly positive relationship with pCO<sub>2</sub> during this phase. However, this result is linked to the disrupted communities by the *V. globosus* presence in M2 and M8. Once these data were extracted, just a slightly positive pCO<sub>2</sub> effect on phytoplankton biomass was found.

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<sup>3</sup>Averaged value from Day 35 (T35, nutrient depletion (Taucher et al. (2017)) to Day 47 (T47, the *V. globosus* community dropped (Riebesell et al. (2018)))

## 5.4 Discussion

### 5.4.1 The use of the $\Phi/\text{Chl}_a$ ratio as an index of autotrophic, mixotrophic or heterotrophic dominance in the microplankton community.

The electron transport system (ETS) is an ubiquitous catabolic pathway used to generate energy in all cells, in the form of adenosine triphosphate (ATP). Even though the autotrophs use other pathways to obtain energy from inorganic resources or light throughout anabolic pathways, they also use this type of catabolism when those energy resources are limited. Thus, we have considered the ETS as a proxy for the catabolic metabolism in plankton communities. In contrast, the anabolic pathway of photosynthesis relays on the Chl<sub>a</sub> to capture solar energy. As photoautotrophs are the dominant autotrophs in the euphotic zone, where this experiment was carried out, we used Chl<sub>a</sub> as a proxy of the anabolic metabolism in this community. Here, we had the chance to follow the  $\Phi/\text{Chl}_a$  relationship through the community changes related to the simulated deep-water upwelling and the subsequent diatom bloom and nutrient depletion that occurred in all the mesocosms.

The Atlantic water microplankton community is composed of *Prochlorococcus* and *Synechococcus*-type cyanobacteria, picoeukaryotes, nanoeukaryotes, diatoms (Hernández-Hernández et al. (2018); Taucher et al. (2018)), ciliates (aloricate and loricate) and dinoflagellates (athecate and thecate) (Algueró-Muñiz et al. (2019)). The reference  $\Phi/\text{Chl}_a$  ratio from the mixed Atlantic waters community varied around  $4.3 \pm 1.7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  throughout the 55 days experiment. However, this ratio was higher in all 9 mesocosms ( $5.6 \pm 2 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ) from the very beginning of the experiment. This mesocosm-enhanced effect was related to the shift in the microplankton towards a more heterotrophic dinoflagellate community. These  $\Phi/\text{Chl}_a$  results in a mixed community were consistent with those obtained from the  $\Phi$  and Chl<sub>a</sub> measurements found in the literature when the ETS activities were converted to ETS activity as it is currently measured (Christensen and Packard (1979)). The upwelled waters of Baja California were characterized by a  $\Phi/\text{Chl}_a$  ratio of  $3.28 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  (Packard et al. (1974)). The upwelled waters off Northwest African were characterized by a  $\Phi/\text{Chl}_a$  ratio of  $2.5\text{--}4.1 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  (Packard (1979))).

In the mesocosms, this ratio also reflected the shift towards a phytoplankton bloom between T28 (Day 28) and T30 (Day 30), dropping to values lower than  $1.7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  in all the mesocosms ( $1.1 \pm 0.6 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ). These results were consistent with those reported at the Chl<sub>a</sub> maximum of the Western Mediterranean Sea ( $0.3 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  (Martinez et al. (1990))) and in the Barents Sea ( $0.7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  (Mar-

tinez (1991))). They are slightly higher than those for the euphotic-zone community in eutrophic waters of the Taiwan Strait ( $0.02\text{-}1.6 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  (Bangqin et al. (2005))) and those in the Gulf of Maine ( $0.5\text{-}0.9 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  (Packard and Williams (1981))). They are also similar to those from different bloom conditions ( $0.3\text{-}1.9 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ) in the phytoplankton from the Nova Scotian shelf (Packard et al. (2000)). From this evidence, we can consider that  $\Phi/\text{Chl}_a$  ratios lower than  $2.5 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  to reflect a mainly photoautotrophic microplankton community.

In the plankton community of our mesocosms, during the nutrient depletion and the post-bloom phases, there was great variability in the community development and in the evolution of the  $\Phi/\text{Chl}_a$  ratio. However, similar  $\Phi/\text{Chl}_a$  ratios were found in those mesocosms where the microzooplankton bloomed (M3, M4, M7, the intermediate-pCO<sub>2</sub> mesocosms; and M1, M5 and M9, the low-pCO<sub>2</sub> mesocosms). First, the microzooplankton, dominated by small thecate and large athecate dinoflagellates (Algueró-Muñiz et al. (2019)), bloomed in the intermediate mesocosms at the end of Phase II (T32 (Day 32) and T35 (Day 35) in M3 and M4, and during T35 (Day 35) and T37 (Day 37) in M7), when values of  $7.6\pm1.5 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  were attained. This dinoflagellate community also bloomed in the weakly acidified mesocosms during Phase III (T43 (Day 43) and T47 (Day 47) in M1, M5 and M9), when  $\Phi/\text{Chl}_a$  ratio was high ( $7.5\pm1.6 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ). From the benchmark, we consider that  $\Phi/\text{Chl}_a$  ratios higher than  $7 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  reflected dominance of a heterotrophic microplankton community.

Lower  $\Phi/\text{Chl}_a$  ratios of around  $4.1\pm1.5 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$  were found in the high-pCO<sub>2</sub> mesocosms (M2 and M8) at the end of Phase II and during the main part of Phase III. At that time, the presence of *V. globosus* disrupted the community. However, even though the microzooplankton did not proliferate, and the Chl<sub>a</sub> retention time was higher in these mesocosms, the  $\Phi/\text{Chl}_a$  ratio was not as low as expected for a photoautotroph-dominated community. Indeed, it reflected a mixed community (or a mixotroph-dominated one), closer to the reference values in the Atlantic surrounded waters ( $4.3\pm1.7 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ). Note that, during the last days of the experiment, the loricate ciliates bloomed in M2. This made the  $\Phi/\text{Chl}_a$  ratio soar up to  $8.6 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ , indicating a heterotroph-dominated community.

#### 5.4.2 Acidification, mixotrophy and the development of *Vicicitus globosus*

As said before, one of the main effects of acidification in the high pCO<sub>2</sub> mesocosms (M2 and M8) was the proliferation of *V. globosus*. This was clear after nutrient depletion until T47 (Day 47). In the literature, this HAS has been

assessed and cultured as a pure autotrophic organism. However, if *V. globosus* is considered a pure autotroph HAS, that negatively impacts microzooplankton, a low  $\Phi/\text{Chl}_a$  would be expected ( $<2.5 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ). However, this was not observed. During nutrient depletion until the disappearance of *V. globosus*, higher  $\Phi/\text{Chl}_a$  were noted ( $4.1 \pm 1.5 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ). These values are similar to those attained in the mixed oligotrophic community of the surrounded Atlantic waters ( $4.3 \pm 1.7 \mu\text{LO}_2\text{h}^{-1}\mu\text{g Chl}_a^{-1}$ ), suggesting that the community in M2 and M8 is a mixed or a mixotroph-dominated one. As said in the introduction, in the oligotrophic waters, mixotrophy is particularly advantageous, as organisms find the potentially limiting nutrient much more concentrated in the prey than in the water column (Sanders (1991)). So, could *V. globosus* be a mixotroph?

First of all, *V. globosus* was able to maintain its population during nutrient depletion, when other pure autotrophs, such as diatoms, dropped to their minimum because of nutrient deficiency. In contrast, *V. globosus* was able to maintain its population for at least 12 more days. This Dycythochales is the only one of its class that lacks pyrenoids. In addition, its haploid form has less pigment than other Dycythochales (Chang et al. (2012)). The pyrenoid is a sub-cellular body used in photoautotrophs to optimize the performance of RuBisCo by concentrating CO<sub>2</sub> in the chloroplast (Badger et al. (1998)). When pyrenoids are absent, RuBisCo is forced to work barely at a quarter of its maximum velocity because of normally low environmental CO<sub>2</sub> concentrations. In comparison with other phytoplankton species, *V. globosus* has an inefficient photosynthetic system under ambient pCO<sub>2</sub>. To survive these normal conditions, it has developed heterotrophic pathways to obtain a variety of organic compounds and energetic molecules to supplement its retarded photosynthesis. In addition, when pCO<sub>2</sub> levels rise, *V. globosus*'s photosynthetic system would be less inefficient, facilitating its proliferation and outcompeting other phytoplankton species. This is our hypothesis why *V. globosus* had a positive relationship with increasing pCO<sub>2</sub> levels.

Furthermore, *V. globosus* is a shape-shifter. It is able to change from a flagellated-shape to an amoeboid-shape and back in minutes. As it was considered an autotroph, this shape-shifting was reported in relation to the reproduction pathways (Chang et al. (2012)). However, we suggest that one of the *V. globosus*' nutritive pathways may be bacterivory. In addition, *V. globosus* has a cytotoxic effect on the membranes of both diatoms and microzooplankton (Chang (2015)). It destroys the cells and releases dissolved organic matter to the seawater. For the bacterial community, this DOC excretion may impact positively on the bacterial community. Consequently, *V. globosus* could feed on bacteria and maintain its community even under low-nutrients conditions.

To sum up, *V. globosus* is more a mixotroph than a photoautotroph. This is suggested from the  $\Phi/\text{Chl}_a$  results in the *V. globosus*-disrupted commu-

nities, considering *V. globosus*'s normally inefficient photosynthetic system and its shape-shifting ability.

## 5.5 Conclusion

We used the  $\Phi/\text{Chl}_a$  ratio to monitor the metabolic changes in the microplankton community during the effects of a simulated upwelling of enriched deep-water. Where its range fell indicated three different conditions.

- For  $\Phi/\text{Chl}_a$  values between 0 and  $2.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , the plankton community was dominated by photoautotrophs. The lowest ratios ( $<1.7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ) were attained during the phytoplankton bloom. These values had also been reported in the open ocean at the depths of the  $\text{Ch}_a$  maximum, or in the euphotic zone of highly eutrophic regions.
- For  $\Phi/\text{Chl}_a$  values between  $2.5$  and  $7.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , the plankton community appeared mixed or was a community where mixotrophs dominated.
- For  $\Phi/\text{Chl}_a$  values higher than  $7.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , the plankton community indicated microzooplankton proliferation. When heterotrophic dinoflagellates dominated the community,  $\Phi/\text{Chl}_a$  soared to  $10 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ .

In addition, to strengthen our analysis of the KOSMOS timecourses, we measured the Leibnizian first-derivative of the biomass per time  $P_N$ . This derivative advanced the quantification and monitoring of the plankton dynamics when  $^{14}\text{C}$  fixation and biomass loss measurements were not available. During the oligotrophic phase,  $P_N$  was minimum. However, in the bloom phase, between 24-48h before the  $\text{Chl}_a$ -peak,  $P_N$  reached a maximum. During the post-bloom phase,  $P_N$  tended towards minimum values similar to those in the original oligotrophic phase. When  $P_N$  negative values were observed, they facilitated the calculation of biomass-loss processes in the ecosystem. The highest net loss of biomass in the mesocosms occurred during the 24h after the  $\text{Chl}_a$ -peak.

In contrast, the Newtonian integral analysis of  $\text{Chl}_a$ ,  $\Phi$ , and  $B$ , for the different phases and mesocosms, allowed the assessment of the accumulated acidification effects on the community. A clearly positive relationship was found between  $\Phi$  and  $\text{pCO}_2$  during the bloom phase. This correlation was stronger when HAS were absent. On the contrary, HAS presence led to a positive relationship between  $\text{Chl}_a$  and  $\text{pCO}_2$  during the post-bloom phase. Apart from the reduction of grazing on the phytoplankton community due to the presence of *V.globosus*, we suggest that *V.globosus*' photosynthetic system, whose lack of pyrenoid made it inefficient in low  $\text{pCO}_2$  environments,

was stimulated in the high pCO<sub>2</sub> mesocosms, leading to acidification's positive impact on Chl<sub>a</sub>.

Finally, the Φ/Chl<sub>a</sub> results in the communities were *V. globosus* bloomed; *V. globosus*' inefficient photosynthetic system and its shape-changing ability suggest that it is a mixotroph more than a photoautotroph.

To sum up, the development of Φ/Chl<sub>a</sub> ratios with the use of Liebnizian differential and Newtonian integral analysis advances our knowledge and understanding of the plankton community dynamics at both the ecosystem and the species scale.

## Chapter 6

# Chapter 6: Conclusions

*The pursuit of curiosity about the basic facts of nature has proven, with few exceptions throughout the history of medical science, to be the route by which the successful drugs and devices of modern medicine were discovered.*

Arthur Kornberg

### 6.1 Developing the assay for NADP-IDH activity as a tool to detect CO<sub>2</sub> production in cells

An enzyme assay for isocitrate dehydrogenase (NADP-IDH) in zooplankton and microplankton, based on the reduction of NADP<sup>+</sup>, was developed. The rates of NADPH formation ( $\mu\text{mol NADPH min}^{-1} \text{mL}^{-1}$ ) were obtained kinetically at 340nm. Activities in  $\mu\text{mol}$  of NADPH were based on a molar extinction coefficient ( $\epsilon$ ) of 5.42  $\text{mL } \mu\text{mol}^{-1} \text{cm}^{-1}$ . In our hands, the maximum NADP-IDH activity ( $V_{max}$ ) in marine plankton samples was attained in 0.1M phosphate buffer at pH 8.2 ( $pK_a_1=7.6$  and  $pK_a_2=8.8$ ), by adding 6mM MgCl<sub>2</sub>, 0.3mM NADP<sup>+</sup> and 2mM DL-trisodium isocitrate.

The apparent Michaelis-Menten  $K_m$  values for the substrates in crude homogenates at pH 8.5 and 18°C were  $271\pm63 \mu\text{M}$  for isocitrate and  $18\pm3 \mu\text{M}$  for NADP<sup>+</sup>. Arrhenius plots for NADP-IDH in mesozooplankton samples and porcine heart yielded similar Energies of activation, 20.4 Kcal mol<sup>-1</sup> and 21.4 Kcal mol<sup>-1</sup>, respectively. Thus, although there seems to be an adaptation to homoeothermic conditions allowing high activity at higher temperatures, a common basic structure seems to have been maintained even after millions of years of evolution.

## 6.2 The assessment of mesozooplankton potential CO<sub>2</sub> production from NADP-IDH measurements.

Considering all the CO<sub>2</sub>-producing enzymes and normalising their activities by NADP-IDH activity, the potential CO<sub>2</sub> production ( $\Psi_{NADP}$ , in  $\mu\text{mol CO}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ) was calculated as 2.94·NADP-IDH in microplankton and mesozooplankton samples. Without considering biosynthesis, the ratio between the purely respiratory CO<sub>2</sub> production (Respiratory  $\Psi_{NADP}$ ) and the potential O<sub>2</sub> consumption is 1.47, indicating that the enzymatic pathways related to the organic carbon oxidation phase of cellular respiration is prepared to exceed the requirements of the ETS. This may serve as an advantage when, eventually, the organism has a high energy requirement. Once this requirement is met, NAD(P)H excess may be redirected to biosynthesis by the IDH substrate cycle of Sazanov and Jackson.

From our calculations,  $\Psi_{NADP}$  always exceeded the physiological  $R_{CO_2}$  rate. This is in harmony with theory because the physiological rate should not exceed the potential. We predict that, from our measurements in the North Eastern Subtropical Atlantic, the  $R_{CO_2}/\Psi_{NADP}$  ratios for open ocean mesozooplankton samples will fall between 0.3 and 0.4, whereas the ratios for a mixed microplankton community will fall between 0.2 and 0.3. However, we recommend the use of physiological  $R_{CO_2}$ , if available, to develop accurate  $R_{CO_2}/\Psi_{NADP}$  ratios for every region, experiment or case.

## 6.3 The assessment of the relationship between $R_{CO_2}$ and NADP-IDH activity

CO<sub>2</sub> production rates ( $R_{CO_2}$ ) were measured in *Leptomysis lingvura* to compare with the potential CO<sub>2</sub> production rates, proxied here for the first time by the NADP<sup>+</sup>-dependent isocitrate dehydrogenase activity (NADP-IDH).

Nutrition prior to sampling, has a significant impact on the respiratory enzymatic activity of *L. lingvura*. Although higher activities normalised per biomass were found in bloom communities than in post-bloom communities, an average  $R_{CO_2}/\text{NADP-IDH}$  ratio of 0.12 was obtained on well-fed mysids from both communities. This result was only slightly higher in nutrient-limited mysids (0.16). This value well characterizes the physiological response of *L. lingvura*.

Sex differences did not significantly alter enzymatic activities nor the ratios of respiratory physiology to enzyme activity. Starvation for longer than 12h decreased the  $R_{O_2}/\Phi$  ratios, whereas it increased the  $R_{CO_2}/\text{NADP-IDH}$  ratios, leading to higher RQs. Our interpretation is that energy production is diminished whereas lipid synthesis and anti-oxidation are strengthened.

Incubation of mysids longer than 12h mixes starvation and circadian effects. We recommend incubations between 4 and 6 hours in future ecophysiological studies. This will reduce the circadian influences.

## 6.4 Assessing variations in the microplankton autotrophy-heterotrophy ratio using enzyme analysis

We used the  $\Phi/\text{Chl}_a$  ratio to monitor the metabolic changes in the microplankton community during the effects of a simulated upwelling of enriched deep-water.

- Values between 0 and  $2.5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  were attained when the community was dominated by photoautotrophs. The lowest results ( $<1.7$ ) were reached during the phytoplankton bloom. These values had also been reported at the depths of the  $\text{Ch}_a$  maximum in the ocean, or at the euphotic zone in highly eutrophic regions.
- Values between  $2.5$  and  $7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  represented a mixed community (or a community where mixotrophs were dominant).
- Values higher than  $7$  were related to the microzooplankton proliferation. When heterotrophic dinoflagellates were dominant in the community,  $\Phi/\text{Chl}_a$  soared up to  $10\mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ .

Net productivity ( $P_N$ ) measurements as the biomass first derivative per time strengthened the quantification and the monitoring of this process when  $^{14}\text{C}$  fixation measurements were not available. Negative values allowed the quantification of the biomass-loss processes in the ecosystem.

The integral analysis per time and mesocosms of  $\text{Chl}_a$ ,  $\Phi$  and  $B$  allowed the assessment of the acidification effects on the community during the oligotrophic, bloom and post-bloom phases. A significant positive relationship was found between  $\Phi$  and  $\text{pCO}_2$  during the bloom phase. In addition, a positive relationship between  $\text{Chl}_a$  and  $\text{pCO}_2$  during the post-bloom phase was also found.

During the post-bloom phase of the experiment, a harmful algal species (HAS, *Vicicitus globosus*) proliferated in the most acidified mesocosms. From the  $\Phi/\text{Chl}_a$  results in these communities (similar to those related to a mixed one); an inefficient photosynthetic system at current  $\text{pCO}_2$  levels (it lacks pyrenoid); and the ability to quickly change from a flagellate-shaped microorganism to an amoeboid-shaped one, we suggest that *V. globosus* is a mixotroph more than a photoautotroph.

To sum up, the development of  $\Phi/\text{Chl}_a$  ratios strengthens the knowledge of the community metabolic state, not only from an ecosystem scale, but also from a species scale.



## Chapter 7

# Chapter 7: Future research.

*One never notices what has been  
done;  
one can only see what remains to be done.*  
Marie Curie

- Independent measurements of cytosolic and mitochondrial NADP-IDH may improve our knowledge about IDH's role in lipid-synthesis and in the cell's ROS scavenging capacity.
- Other CO<sub>2</sub>-producing enzymes in plankton should be investigated in order to elucidate the relationships among their activities, activators and inhibitors and to accurately model the total potential plankton CO<sub>2</sub> production.
- Further research, on the relationship between  $R_{CO_2}$  and NADP-IDH, on how different environmental factors (such as age, pO<sub>2</sub>, T, circadian rhythm, etc.) impact this relationship, and on the in-cell substrates availability, is still needed.
- New models based on measurements of NADP-IDH, isocitrate and NADP<sup>+</sup> to predict  $R_{CO_2}$  should be developed and tested with  $R_{CO_2}$  measurements from plankton incubations.
- Vertical  $R_{CO_2}$  profiles based on NADP-IDH measurements throughout the water column should be made in order to calculate and appraise  $F_C$ . In other words, the calculations should be compared with other  $F_C$  measurements (i.e.  $F_C$  from sediment traps), noting that different time scales must be considered during this comparison.
- In order to develop an average  $R_{CO_2}$ /NADP-IDH ratio (or linear regression) to be used over a large-scale, further  $R_{CO_2}$  and NADP-IDH measurements should be widely made on oceanographic campaigns. Over shorter scales, different measurements should be made under oligotrophic and eutrophic conditions in different regions and in different seasons.

- Measurements of heterotrophic metabolism (both  $\Phi$  and/or NADP-IDH) should be made during oceanographic campaigns on a regular basis in order to promote comparison with autotroph proxies (i.e. Chl\_a) and to assess the community metabolic state and its relationship with environmental factors.

## **Chapter 8**

# **Chapter 8: Resumen en español.**

*Los "midiclorianos" son criaturas microscópicas que se encuentran dentro de todos los seres vivos en simbiosis. Sin "midiclorianos" la vida no podría existir y no conoceríamos la Fuerza. Cuanto más nivel de "midiclorianos" tenga un ser vivo, más aptitud tiene para usar la Fuerza.*

Star Wars Episodio I. George Lucas.

Estraído, en referencia a las mitocondrias, del libro "Power, Sex, Suicide" de Nick Lane.

### **8.1 Introducción**

Debido a las emisiones por la combustión de fuentes de energía fósiles, los niveles de CO<sub>2</sub> atmosféricos han aumentado exponencialmente en los últimos 250 años (Doney and Schimel (2007)). El océano puede considerarse un sumidero de este CO<sub>2</sub>, ya que su concentración aquí es menor que en la atmósfera, y posee una comunidad de organismos autótrofos que fijan CO<sub>2</sub> para producir materia orgánica (producción primaria). Sin embargo, en la columna de agua, esta nueva materia orgánica en forma de carbono orgánico particulado (POC) es consumida por organismos heterótrofos tanto para crecer como para generar energía por medio de la respiración. De este modo, parte del CO<sub>2</sub> previamente fijado, vuelve al océano. Dependiendo de las tasas de respiración en la columna de agua y su magnitud frente a las de producción primaria, esta producción de CO<sub>2</sub> puede reducir el flujo de CO<sub>2</sub>

atmósfera-océano hasta quedar en equilibrio, o incluso ser una fuente de CO<sub>2</sub> en algunas regiones. Por ello, para estimar adecuadamente el papel del océano en este balance, es necesario conocer tanto las tasas de producción primaria como las de respiración.

Desde un punto de vista oceanográfico, la producción primaria y la función de los organismos fotosintéticos en las capas más superficiales del océano ha sido ampliamente estudiada. Por el contrario, la respiración fue durante muchos años derivada de estas tasas de producción primaria. Desde los años 30, la realización de largos períodos de incubación ha permitido la medición directa y el seguimiento de las tasas de consumo de O<sub>2</sub> de organismos vivos ( $R_{O_2}$ ) (Waksman and Carey (1935); Johnson (1936)). Sin embargo, estas técnicas requieren mucho tiempo de incubación, con lo que la tasa de obtención de datos es muy baja para los estudios oceanográficos. Asimismo, requieren que los organismos se encuentren en buen estado fisiológico. Esto no siempre es viable, especialmente en el estudio de aguas profundas donde suelen quedar gravemente dañados durante el muestreo. Además, el propio proceso de incubación hace que los organismos puedan sufrir durante la manipulación de las muestras o que, dentro de las botellas, puedan encontrarse en densidades de población superiores a las existentes en el medio natural (Fernández-Urruzola et al. (2011); Bondyale-Juez et al. (2017)). Por ello, a partir de los años 70 se desarrollaron técnicas basadas en la medición de la actividad de las enzimas responsables de esa respuesta fisiológica: las enzimas del sistema de transporte de electrones (ETS) (Packard (1971)). Estas medidas aportan información sobre la capacidad máxima de consumo de O<sub>2</sub> de los organismos ( $\Phi$ ) (King and Packard (1975)). La utilización de estas técnicas bioquímicas ha permitido aumentar las tasas de obtención de datos, así como obtener medidas del océano profundo. Debido a que son medidas del estado metabólico del organismo, requieren del análisis de su relación con la respuesta fisiológica real en cada momento ( $R_{O_2}/\Phi$ ). Durante décadas se ha analizado y observado la variabilidad del ratio  $R_{O_2}/\Phi$  (Hernández-León and Gómez (1996); Martínez et al. (2010); Herrera et al. (2011b); Osma et al. (2016a,b)). Los mismos factores ambientales y fisiológicos que influyen sobre  $R_{O_2}$ , son los responsables de la variabilidad de  $R_{O_2}/\Phi$  (Hernández-León and Gómez (1996)). De entre todos, la disponibilidad de nutrientes parece tener un papel fundamental.

Lo cierto es que las enzimas, responsables últimas de la respuesta fisiológica, catabolizan las reacciones en función de la disponibilidad de sus sustratos a nivel celular. Esta idea, transformada en ecuación por Michaelis-Menten, es la base de los modelos cinéticos enzimáticos (Enzymatic Kinetic Model, EKM) (Packard et al. (1996b); Packard and Gómez (2008); Aguiar-González et al. (2012); Osma et al. (2016b)). Su aplicación al ETS, midiendo a su vez la concentración de sus sustratos principales (NADH y NADPH) y su afinidad por ellos (mediante las constantes cinéticas o  $K_m$ ) ha permitido el

cálculo del consumo de O<sub>2</sub> de diferentes organismos marinos de la manera más precisa hasta la fecha (Packard and Gómez (2008); Osma et al. (2016b)).

Todos estos avances se han centrado en la respiración desde el punto de vista del consumo de O<sub>2</sub>. Sin embargo, para el adecuado estudio de la función del océano en los flujos de carbono, es necesario conocer la producción de CO<sub>2</sub> asociada a la respiración ( $R_{CO_2}$ ). Debido a la dificultad de la medición directa  $R_{CO_2}$  en la columna de agua, generalmente se han utilizado los denominados Coeficientes Respiratorios (Respiratory Quotient, RQ, moles producidos de CO<sub>2</sub> por mol consumido de O<sub>2</sub>) sobre los valores de  $R_{O_2}$ . Lamentablemente, este coeficiente puede ser muy variable (Berggren et al. (2012); Romero-Kutzner et al. (2015)). Esta variabilidad, sumada a la ya existente en los ratios  $R_{O_2}/\Phi$ , esenciales en el estudio del océano profundo, hace que se requiera una mejora en las técnicas de análisis de  $R_{CO_2}$ .

Al igual que en el caso del consumo de O<sub>2</sub>,  $R_{CO_2}$  puede ser estudiado a partir de la medición una de las enzimas responsables de la producción de CO<sub>2</sub>. Dentro de todas las enzimas implicadas, la Isocitrato Dehidrogenasa (IDH) es la responsable de la producción de la primera molécula de CO<sub>2</sub> dentro del ciclo de Krebs. En base al cofactor que requieren, se pueden clasificar dos isoenzimas: (1) la dependiente de NADP<sup>+</sup> (NADP-IDH) y (2) la dependiente de NAD<sup>+</sup> (NAD-IDH) (Kornberg and Pricer (1951); Gálvez and Gadál (1995)). La NADP-IDH se encuentra tanto en procariotas como en eucariotas, está presente en el citosol con funciones relacionadas con la síntesis lipídica (Koh et al. (2004)), y es mucho más estable que la NAD-IDH para su medición (Plaut (1970, 1962)). Además, se encuentra relacionada con la formación de la especie reducida de glutatióñ, implicada en funciones antioxidantes de la célula (Jo et al. (2001)). Debido a su ubicuidad y a su participación en los distintos procesos catabólicos, hace que estudios en diferentes organismos hayan observado una actividad de NADP-IDH significativamente mayor que la de la NAD-IDH (Colman (1975); Alp et al. (1976); Denton et al. (1978)). Estas características hacen de la NADP-IDH la enzima idónea sobre la que desarrollar un nuevo método para el cálculo de la producción de CO<sub>2</sub> en la columna de agua. Por ello, en los años 90 se llevaron a cabo estudios en bacterias marinas analizando el  $R_{CO_2}/\text{NADP-IDH}$  y desarrollando un EKM específico (Packard et al. (1996a); Roy and Packard (2001)). Sin embargo, no ha sido hasta la presente tesis cuando se ha conseguido adaptar esta metodología a la comunidad planctónica (Tames-Espinosa et al. (2018)) (ver capítulo 2).

Aun siendo una de las enzimas responsables de la producción de CO<sub>2</sub> de la célula, la NADP-IDH no es la única. Otras enzimas relacionadas directa o indirectamente con el ciclo de Krebs (como la piruvato deshidrogenasa (PDH), la alfa-ketoglutarato deshidrogenasa (KGDH), la enzima málica (ME) o la fosfoenolpiruvato carboxiquinasa (PEPCK)) también producen CO<sub>2</sub> en diferentes proporciones cada vez que se activa el ciclo. Asimismo,

otras enzimas relacionadas con el crecimiento o la síntesis de lípidos y ácidos grasos también colaboran en el aumento de la producción de CO<sub>2</sub> en momentos puntuales del desarrollo de los organismos. Debido a ello, parte de la presente tesis se ha centrado en la relación entre la NADP-IDH y la producción total de CO<sub>2</sub> en base a los estudios existentes hasta la fecha (ver capítulo 3).

En conclusión, este estudio se centra en el desarrollo y aplicación de un nuevo método enzimático para el análisis de la producción potencial de CO<sub>2</sub> asociada a la comunidad planctónica. Para ello, se han definido los siguientes objetivos:

- Objetivo 1: La medición de la máxima actividad de una de las enzimas principales en la producción de CO<sub>2</sub> en la comunidad planctónica: la NADP-IDH. Partiendo de la metodología existente, aplicada a otros organismos, se ha optimizado la misma adaptándola al micro y mesozooplancton (ver capítulo 2).
- Objetivo 2: Modelización de la producción máxima de CO<sub>2</sub> de la comunidad planctónica en base a la actividad de la NADP-IDH. Para ello, se han analizado las relaciones entre la NADP-IDH y las actividades de las principales enzimas productoras de CO<sub>2</sub> (ver capítulo 3).
- Objetivo 3: Análisis de la relación  $R_{CO_2}$ /NADP-IDH. La respuesta fisiológica de este proceso, la  $R_{CO_2}$ , es la que finalmente impacta en los ciclos biogeoquímicos. Este estudio se ha llevado a cabo por medio de incubaciones del misidáceo *Leptomyysis lingvura* (ver capítulo 4).

Estos tres primeros objetivos permiten mejorar el conocimiento existente de uno de los procesos esenciales en el flujo de carbono en la columna de agua. Asimismo, la posibilidad de analizar este tipo de parámetros en muestras de gran profundidad puede suponer una nueva fuente de conocimiento directo sobre los procesos metabólicos de los ecosistemas batipelágicos.

Por otra parte, no hay que olvidar que el estudio de la respiración desde un punto de vista enzimático es un indicador de la estructura que la comunidad planctónica posee para la obtención catabólica de energía. Debido a ello, la medición de  $\Phi$  o de la NADP-IDH puede utilizarse, en combinación con otros parámetros relacionados con la producción primaria (por ejemplo la Chl<sub>a</sub>), para la determinación del grado de autotrofía o heterotrofía de una comunidad. Este tipo de estudios complementan la aplicación de las medidas enzimáticas como forma de cuantificar otros procesos que se dan en los ecosistemas, tales como la cuantificación del Flujo de Carbono vertical ( $F_c$ , Packard et al. (2015)), el cálculo de la eficiencia en la regeneración de nutrientes (NRE, Packard et al. (2015)) o la cuantificación de la producción de energía heterotrófica (HEP, Packard et al. (2015)). Por ello, se ha añadido un cuarto objetivo a esta tesis, donde se muestra la posible utilización

de estas técnicas para monitorizar los cambios metabólicos en la comunidad microplanctónica.

- Objetivo 4: Análisis del cambio del grado de autotrofía-heterotrofía existente en una comunidad microplanctónica en condiciones oligotróficas tras la influencia de un afloramiento de agua profunda. Para ello, se ha utilizado  $\Phi$  como la técnica bioquímica relacionada con el metabolismo catabólico y la medición de la Chl<sub>a</sub> como el parámetro oceanográfico relacionado con la fotoautotrofía. De esta manera, se ha analizado el nivel de consumo de O<sub>2</sub> por mg de Chl<sub>a</sub> a lo largo de los 55 días de duración del experimento. Además, se ha estudiado el impacto de la acidificación oceánica en estos resultados y su relación con las variaciones en la comunidad (ver capítulo

## 8.2 Metodología General

### 8.2.1 Muestreo de mesozooplancton y conservación de las muestras.

Las muestras de mesozooplancton (50-200 $\mu\text{m}$  y 200-2000 $\mu\text{m}$ ) fueron recogidas en Sardina del Norte (Gáldar, Gran Canaria) por medio del arrastre de una red WP2 por tres buzos con escafandra autónoma, en una profundidad nunca superior a los 10m y siempre a más de un metro del lecho marino. Tras ser almacenados en contenedores de plástico para su traslado al laboratorio, las muestras se fraccionaron pasándolas por una columna de filtración con mallas de 2000 $\mu\text{m}$ , de 200 $\mu\text{m}$  y 50 $\mu\text{m}$ . Las muestras fueron recogidas con cuidado de las mallas de 200 $\mu\text{m}$  y 50 $\mu\text{m}$  y almacenadas en tubos Microvials<sup>®</sup> de 2mL a -80°C, para los posteriores análisis enzimáticos y de biomasa.

### 8.2.2 Muestreo microplancton y conservación de las muestras.

El microplancton (0,7-50 $\mu\text{m}$ ) se muestreó en las aguas de la playa de las Alcaravaneras (Las Palmas de Gran Canaria). Las muestras de agua se almacenaron en contenedores de 25L para su traslado al laboratorio. Una vez allí, se pasaron por una malla de 50 $\mu\text{m}$  antes de filtrarlas por un GF-F con un poro de 0,7 $\mu\text{m}$  hasta su saturación (aproximadamente 4L por filtro o un máximo de 1h de filtrado). Posteriormente, la muestra recogida en el filtro fue congelada y almacenada a -80°C.

Para el experimento llevado a cabo en el capítulo 5, se utilizó un equipo de muestreo de agua integrada por profundidad (IWS, HYDRO-BIOS, Kiel) que recogía volúmenes de 5L de una mezcla homogénea de la

columna de agua (0-13m) en cada mesocosmos, muestreando entre 60 y 70L por mesocosmos y día. De esos volúmenes, se tomó una alícuota de entre 2 y 5L representativa de cada mesocosmos. Las muestras se pasaron por una malla de 50 $\mu\text{m}$  antes de ser filtradas por un GF-F con un poro de 0,7 $\mu\text{m}$  hasta su saturación (un máximo de una hora de filtración). Posteriormente, fueron congeladas en Nitrógeno líquido y almacenadas a -80°C hasta su análisis en el laboratorio.

### 8.2.3 Muestreo y cultivos de *Leptomysis lingvura*.

Los muestreos de *L. lingvura* para su cultivo se realizaron a finales del otoño de 2017 y a finales de otoño de 2018 en Taliarte (Telde, Gran Canaria). Se realizaron buceos con escafandra autónoma en la zona de contacto entre los fondos arenosos y el veril rocoso a una profundidad de entre 5 y 10m. Se utilizaron redes de mano con una apertura de malla de 500 $\mu\text{m}$  y unos contenedores de plástico para almacenar los individuos asegurando el intercambio de agua con el entorno durante el tiempo de muestreo. Una vez capturados se trasladaron al laboratorio donde se mantuvieron en cultivo conforme a Herrera et al. (2011a). Para ello se utilizaron cuatro tanques de 40L con agua de mar filtrada, interconectados entre sí formando un sistema con un volumen de recirculación de 6L/h. El sistema además incluía un tanque de 70L de filtrado biológico por bacterias nitrificantes (BIO DIGEST, 1 ampolla cada 15 días), así como por minifiltros en cada uno de los tanques. Esto, junto a la renovación del 25% del agua cada semana, aseguró que los niveles de NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup> y NO<sub>3</sub><sup>-</sup> se mantuvieran por debajo de 0,1, 0,02 y 0,2 mg L<sup>-1</sup> (Lussier et al. (1988)). La alimentación se basó en dos tomas diarias de *Artemia sp.* proveniente de cultivos secundarios. Tras un mínimo de una semana de aclimatación de los cultivos, se seleccionaron los individuos que fueron objeto del estudio de tasas respiratorias (ver capítulo 4) en incubaciones durante diferentes períodos de tiempo o bajo distintas condiciones de alimentación. Tras la incubación se recogieron las muestras con cuidado con una malla de 50 $\mu\text{m}$  y se almacenaron en Microvials<sup>®</sup> a -80°C, hasta su análisis.

### 8.2.4 Medición de NADP-IDH.

Una vez en el laboratorio, las muestras de microzooplancton fueron homogeneizadas en 0,1M de buffer fosfato por medio de un sistema con varilla de teflón<sup>®</sup> en un recipiente de 2mL (PYREX<sup>®</sup> Potter-Elvehjem) a 2600rpm durante 2 min. El mesozooplancton y los misidáceos se homogeneizaron por medio de sonicación durante 45sec al 70% de la amplitud en un procesador ultrasónico Vibracell VCX 130 Sonics<sup>®</sup>.

Los homogeneizados resultantes fueron centrifugados a 4000rpm durante 10min entre 0-4°C. Nada más finalizar la centrifugación, se realizaron los ensayos enzimáticos sobre los sobrenadantes obtenidos.

La actividad de la NADP-IDH se midió siguiendo la metodología descrita por Tames-Espinosa et al. (2018), desarrollada en el capítulo 2 de la presente tesis doctoral. La actividad de la NADP-IDH se obtiene siguiendo el aumento de la absorbancia del NADPH (340nm) cinéticamente durante 1200sec, a 18°C en una cubeta de 1cm de anchura transparente a la longitud de onda ultravioleta con un espectrofotómetro Cary 100 UV-Visible. El  $\beta$ -NADP<sup>+</sup> (SIGMA #N0505, de aquí en adelante NADP<sup>+</sup>) se reduce por la NADP-IDH a NADPH mientras se decarboxila el DL-trisodio-isocitrato (SIGMA #I1252; de aquí en adelante isocitrato) a  $\alpha$ -cetoglutarato. Los blancos se realizan sin añadir isocitrato para extraer los valores del resultado asociados a reacciones no enzimáticas o de aquellas enzimas no relacionadas con la NADP-IDH. El aumento de la absorbancia adquiere forma de regresión lineal, cuya pendiente es considerada la actividad de la NADP-IDH. La actividad de NADP-IDH se ha expresado tanto en  $\mu\text{mol NADPH min}^{-1}$  ( $\text{mL de homogeneizado})^{-1}$  como en  $\mu\text{mol CO}_2 \text{ min}^{-1}$  ( $\text{mL de homogeneizado})^{-1}$  (ambas unidades son estequiométricamente equivalentes). Para pasarlo a  $\mu\text{LCO}_2 \text{ min}^{-1}$  ( $\text{mL de homogeneizado})^{-1}$  se debe multiplicar por 22,4 $\mu\text{L } \mu\text{mol}^{-1}$ .

### 8.2.5 Optimización de parámetros para la medición de la $V_{max}$ de NADP-IDH en la comunidad planctónica.

Para asegurar la medición de la velocidad máxima de reacción ( $V_{max}$ ) de NADP-IDH se determinaron las concentraciones óptimas del substrato (isocitrato), del cofactor (NADP<sup>+</sup>) y del cofactor mineral ( $\text{MgCl}_2$ ), así como su pH óptimo. Asimismo, se estudió la dependencia de esta velocidad con la temperatura, y se comprobó la relación lineal existente con la biomasa. Otros aspectos, como el tiempo e intensidad óptimas de centrifugación de las muestras durante la preparación de los homogeneizados, así como la estabilidad de éstos también fueron estudiados para asegurar la reproducibilidad del ensayo y la emisión de la máxima señal de la reacción. En todos los casos se utilizó la metodología definida en la sección 8.2.4 para medir la actividad en muestras de 0,7-50 $\mu\text{m}$ , 50-200 $\mu\text{m}$  y de 200-2000 $\mu\text{m}$ . Las distintas especificaciones para cada parámetro pueden consultarse en la sección 2.2 del capítulo 2). Por otra parte, la sensibilidad al buffer, el coeficiente de extinción, la precisión del ensayo y la interpretación de las diferentes respuestas espectrofotométricas se encuentran recogidas en el Apéndice A de la presente tesis doctoral.

### **8.2.6 Medición del consumo máximo de O<sub>2</sub> ( $\Phi$ ) en la comunidad planctónica.**

La respiración potencial basada en el consumo de O<sub>2</sub> se determinó por medio del análisis de la actividad del sistema de transporte de electrones, ETS, en el sobrenadante de las muestras homogeneizadas y centrifugadas (Packard (1971), incorporando las modificaciones descritas por Kenner and Ahmed (1975) para el fitoplancton y por Owens and King (1975) y Gómez et al. (1996) para el zooplancton). El aumento de la absorbancia asociada al INT-formazán (490nm) fue monitorizado a 18°C en cubetas de plástico de 1cm de anchura, transparentes a esta longitud de onda. Siguiendo la metodología de Packard and Williams (1981) se estimó la actividad potencial del ETS, así como el consumo potencial de O<sub>2</sub>, en base a la recta de regresión del aumento de la absorbancia a 490nm por tiempo. Durante el proceso, la sal de tetrazolium (INT, p-Iodonitrotetrazolium Violet, Sigma #I8377) es reducida por las enzimas respiratorias presentes en el ETS, sustituyendo al O<sub>2</sub> como acceptor de electrones. El INT capta 2 electrones mientras que el O<sub>2</sub> requiere 4 para reducirse. Por ello, la tasa de producción del INT-formazán equivale estequiométricamente a dos veces la actividad del ETS y a 0,5 al consumo potencial de O<sub>2</sub>. El consumo potencial de O<sub>2</sub> se ha expresado en  $\mu\text{molO}_2\text{h}^{-1}(\text{mg de proteína})^{-1}$ . Para pasarlo a  $\mu\text{LO}_2\text{h}^{-1}(\text{mg de proteína})^{-1}$  se debe multiplicar por 22,4  $\mu\text{L }\mu\text{mol}^{-1}$ .

### **8.2.7 Medición del consumo de O<sub>2</sub> fisiológico en base a la aplicación de los ratios $R_{O_2}/\Phi$ a las mediciones de $\Phi$ .**

Cuando no se encuentran disponibles las medidas de consumo de O<sub>2</sub> fisiológico ( $R_{O_2}$ ), es común utilizar los ratios  $R_{O_2}/\Phi$  presentes en la literatura, para obtener  $R_{O_2}$  a partir de las medidas de  $\Phi$  tanto para el micro como para el mesozooplancton (Cleland (1967); Hernández-Leon and Gómez (1996); Packard and Codispoti (2007); Osma et al. (2016c)). Aquí hemos aplicado tres ratios distintos, asociado a tres escenarios diferentes, para aplicarlos sobre las muestras de mesozooplancton. En el caso del microplancton, se ha definido un cuarto escenario específico. La descripción de los cuatro escenarios, así como sus valores, se encuentra recogida en la Tabla 3.1 del capítulo 3. En todos los casos, se obtiene  $R_{O_2}$  con la Eq.8.1:

$$R_{O_2} = \Phi \cdot (R_{O_2}/\Phi) \quad (8.1)$$

Donde  $R_{O_2}$  es el consumo de O<sub>2</sub> fisiológico,  $\Phi$  es el consumo de O<sub>2</sub> potencial, y  $(R_{O_2}/\Phi)$  es el ratio a aplicar. Ambas tasas se expresan en las mismas unidades, bien en base al volumen de agua o en base a la biomasa (bien  $\mu\text{mol O}_2 \text{ h}^{-1} \text{ L}^{-1}$ ; o bien  $\mu\text{mol O}_2 \text{ h}^{-1} (\text{mg de}$

proteína)<sup>-1</sup>), por ello el ratio entre ellas es adimensional, pudiendo publicarse el resultado en tanto por uno o en tanto por ciento para una mejor comprensión.

### **8.2.8 Medición de la producción de CO<sub>2</sub> fisiológica ( $R_{CO_2}$ ) teórica en base a la aplicación de RQ ratios a los resultados en $R_{O_2}$**

A partir de los valores de  $R_{O_2}$  obtenidos tanto por medio de incubaciones como con la aplicación de  $R_{O_2}/\Phi$ , se ha calculado  $R_{CO_2}$  tal y como se expresa en la Eq.8.2. Para ello, se barajaron distintas posibilidades incluyendo la aplicación de ratios de Redfield (Packard and Gómez (2013)) o distintos cocientes respiratorios (RQs) (Mayzaud et al. (2005); Berggren et al. (2012); Romero-Kutzner et al. (2015), de forma que se abarcara un amplio rango de condiciones fisiológicas. La descripción de las tres condiciones, así como sus valores, se encuentra recogida en la Tabla 3.1 del capítulo 3.

$$R_{CO_2} = R_{O_2} \cdot R_{CO_2:O_2} \quad (8.2)$$

Donde  $R_{CO_2:O_2}$  es la conversión aplicada en cada escenario en la Tabla 3.1. Las tasas de respiración se expresan en las mismas unidades, bien en base al volumen de agua, bien en base a la biomasa ( $\mu\text{mol del gas h}^{-1} \text{ L}^{-1}$  o  $\mu\text{mol del gas h}^{-1}$  (mg de proteína)<sup>-1</sup>), de tal forma que el ratio aplicado se medirá en moles de CO<sub>2</sub> por mol de O<sub>2</sub>.

### **8.2.9 Medición del consumo de O<sub>2</sub> ( $R_{O_2}$ ) fisiológico en incubaciones de *L. lingvura*.**

La tasa de consumo de O<sub>2</sub> se midió sobre incubaciones de tres individuos en botellas de 60mL de agua de mar filtrada (por triplicado). Se utilizaron los optodos Presens® 1-Channel Fibox 4 con sensores tipo PSt3. Se seleccionaron cuatro sensores: uno en el interior de cada una de las tres botellas de incubación y un cuarto en el interior de la botella blanco. Se realizó una medida de la concentración de O<sub>2</sub> en cada botella tres veces cada 15 min durante un tiempo máximo de incubación de 120 min (Bondyale-Juez et al. (2017)). La concentración de O<sub>2</sub> obtenida por el optodo se expresa en  $\mu\text{mol O}_2 \text{ L}^{-1}$ . El descenso de la concentración de O<sub>2</sub> en el tiempo, eliminando los datos de los primeros 30 min, puede modelarse con líneas de regresión cuya pendiente es negativa. En nuestro caso la pendiente ( $\alpha$ ) es la variación de  $\mu\text{mol O}_2 \text{ L}^{-1} \text{ min}^{-1}$ . Para obtener la tasa de respiración de los organismos se debe transformar este valor en  $\mu\text{mol O}_2 \text{ h}^{-1}$ , utilizando el volumen de la botella de incubación (V) expresado en L, tal y como se

recoge en la Eq.8.3:

$$R_{O_2} = -(\alpha) \cdot V \cdot 60 \quad (8.3)$$

Siendo  $\alpha$  la pendiente de la línea de regresión obtenida, V el volumen de la botella de incubación en L y 60 los minutos que tiene una hora. Una vez aplicada la Eq.8.3,  $R_{O_2}$  se expresa en  $\mu\text{mol O}_2 \text{ h}^{-1}$ .

### **8.2.10 Medición de la producción de CO<sub>2</sub> ( $R_{CO_2}$ ) fisiológica en incubaciones de *L. lingvura*.**

Debido a los requerimientos de volumen de muestra de la técnica de medición de CO<sub>2</sub> (250mL en cada medida), se ha realizado una medición de punto final sobre incubaciones de diez individuos en botellas de 1L (por triplicado) con una duración de entre 4 y 24h dependiendo del objetivo de cada experimento (ver capítulo 4). La producción de CO<sub>2</sub> por los organismos impacta en el sistema de carbonatos del agua de mar, dando lugar a un incremento del Carbono inorgánico Total (TC). Por ello, y tras comprobar que la alcalinidad no variaba durante el periodo de incubación, se consideró el incremento de TC una medida directa de la producción de CO<sub>2</sub>. Se midió la concentración de TC en el agua de mar filtrada al inicio de la incubación (por triplicado, TC<sub>0</sub>), en botellas con individuos al final de la incubación (por triplicado, TC<sub>e</sub>) y en botellas sin individuos al final de la incubación (blancos, por triplicado, TC<sub>e-blank</sub>). Para la determinación de TC se utilizó el Sistema VINDTA 3C de determinación coloumétrica (Mintrop et al. (2000)) con una precisión de  $\pm 1,0 \mu\text{mol L}^{-1}$  ([www.MARIANDA.com](http://www.MARIANDA.com)). Con los valores de concentración de TC obtenidos, la tasa de producción de CO<sub>2</sub> en  $\mu\text{mol CO}_2 \text{ h}^{-1}$  se obtiene por medio de la Eq.8.4:

$$R_{CO_2} = ((TC_e - TC_0) - (TC_{e-blank} - TC_0)) \cdot V / t \quad (8.4)$$

Siendo V el volumen de incubación en L (en el presente caso, 1L en todas las botellas) y t el tiempo de incubación expresado en h. Con ello,  $R_{CO_2}$  se expresa en  $\mu\text{mol O}_2 \text{ h}^{-1}$ .

### **8.2.11 Medición de la biomasa en términos de contenido proteínico.**

El contenido proteico en biomasa se ha determinado en los mismos sobrenadantes en los que se han realizade los ensayos enzimáticos. Para ello, se ha utilizado el método de Lowry et al. (1951) modificado por Rutter (1967) usando albúmina de serum bovino (BSA) como estándar de calibración. La biomasa se expresa en mg de proteína (mL de homogeneizado)<sup>-1</sup>. Conociendo el volumen total de homogeneizado y

el volumen de agua que representa la muestra (en L) se puede expresar también como mg de proteína L<sup>-1</sup>.

### 8.2.12 Medición de la productividad neta

La productividad neta se ha medido aplicando la primera derivada a la biomasa en las series temporales obtenidas en el experimento KOS-MOS GC 2014. Los resultados positivos de esta primera derivada son indicador del proceso de producción neta de biomasa en el tiempo, esto es, la productividad neta. Sin embargo, los valores negativos implican que los procesos de remineralización, precipitado y pastoreo (grazing) son mayores que la producción de biomasa en el tiempo, resultando en una pérdida neta de biomasa en determinados períodos de tiempo.

## 8.3 Principales resultados y discusión

### 8.3.1 Optimización de la medición de la actividad de la NADP-IDH en la comunidad planctónica.

En el curso de esta tesis doctoral se ha desarrollado una metodología para medir la actividad de la NADP-IDH en el plancton. Para ello se han determinado las concentraciones necesarias del substrato, cofactor y cofactor metálico para asegurar la medición de la velocidad máxima de la reacción (*Vmax*). Los valores óptimos en la cubeta para la comunidad planctónica (de 0,7 a 2000 μm) fueron 2mM para el isocitrato, 0,3mM para el NADP<sup>+</sup> y 6mM para el MgCl<sub>2</sub>. Las constantes de Michaelis-Menten (*Kms*) se obtuvieron tanto por los gráficos de Michaelis-Menten (Fig.2.1) como con la linearización de Hanes-Woolf, resultando en 271±63 μM para el isocitrato y 18±3 μM para el NADP<sup>+</sup>. La *Vmax* se alcanzó con un pH de 8,19 (Fig.2.2), reduciéndose a la mitad cuando éste era menor que 7,62 o mayor que 8,75 (*pKas*). Si bien la temperatura óptima para esta enzima fue de 28°C en el mesozooplanc ton y de 39°C en la NADP-IDH proveniente de corazón de cerdo, ambas necesitaron similares Energías de activación (*Ea*) (20,4 Kcal mol<sup>-1</sup> y 21,4Kcal mol<sup>-1</sup> respectivamente) (Fig.2.3). Esto ha permitido el desarrollo de una ecuación de Arrhenius precisa a la hora de calcular la actividad a diferentes temperaturas (Eq.2.8 en el capítulo 2). Además, el *Q<sub>10</sub>* (entre 15 y 25°C) de la NADP-IDH del mesozooplanc ton fue 1,98.

Por otra parte se comprobó la relación lineal existente entre la actividad enzimática y la biomasa (Fig.2.4), asegurando que la dilución de una muestra no afecta a la actividad de la NADP-IDH normalizada por mg de proteína (con una desviación estándar del 3,5%). El límite de

detección de este nuevo método fue de  $8 \pm 2 \mu\text{g}$  de proteína  $\text{mL}^{-1}$  para el microplancton y de  $10 \pm 2 \mu\text{g}$  de proteína  $\text{mL}^{-1}$  para el mesozooplancton.

El nivel de centrifugación impactó de manera significativa en las muestras filtradas por GF/F (en nuestro caso, el microplancton) (Fig.2.5). En estos casos, una fuerza mayor que 1500g debe mantenerse por más de 2min para asegurar que se alcanza la  $V_{max}$ . Sin embargo, las muestras que fueron sonicadas (mesozooplancton), no sufrieron cambios significativos asociados a la centrifugación. No obstante, ya que los blancos de las muestras sin centrifugar daban valores muy negativos que influían en el resultado final, se recomienda una centrifugación de al menos 375g durante 2min. Una vez centrifugadas, observamos que las muestras sufrían una pérdida del 1,1% de NADP-IDH por cada 10 min (Fig.2.6). Si se analizan durante la primera media hora tras la centrifugación, la pérdida de la actividad no será mayor del 3%.

### **8.3.2 Producción máxima de $\text{CO}_2$ de la comunidad planctónica a partir de la medición de la actividad de la NADP-IDH ( $\Psi_{NADP}$ ).**

La NADP-IDH es una de las principales enzimas productoras de  $\text{CO}_2$ , pero no es la única. Por ejemplo, la producción de  $\text{CO}_2$  en el ciclo de Krebs será la suma del  $\text{CO}_2$  producido por la piruvato deshidrogenasa ( $\text{PDH}_{\text{CO}_2}$ ), por la  $\alpha$ -cetoglutarato deshidrogenasa ( $\text{KGDH}_{\text{CO}_2}$ ), por la enzima mállica ( $\text{ME}_{\text{CO}_2}$ ), por la fosfoenolpiruvato carboxiquinasa ( $\text{PEPCK}_{\text{CO}_2}$ ) y por las isoenzimas isocitrato deshidrogenasas ( $\text{IDH KC}_{\text{CO}_2}$ ) (Fig.1.1). Gracias al trabajo de Walsh and Koshland (1984) en *Escherichia coli* pudimos relacionar la actividad de cada una de ellas con la actividad de la  $\text{IDH KC}_{\text{CO}_2}$ , tal y como se refleja en las ecuaciones Eq.3.4 y Eq.3.5 del capítulo 3. Según esta aproximación, la producción potencial de  $\text{CO}_2$  asociada al ciclo de Krebs, es 3,57 veces la actividad de la  $\text{IDH KC}_{\text{CO}_2}$ .

Sin embargo, la  $\text{IDH KC}_{\text{CO}_2}$  en los organismos eucariotas<sup>1</sup> se encuentra en la mitocondria y está compuesta por dos isoenzimas: una dependiente del  $\text{NAD}^+$  (NAD-IDH) y otra dependiente del  $\text{NADP}^+$  (NADP-IDH) (Eq.8.5). Por otra parte, la medición de la NADP-IDH no sólo incluye la enzima mitocondrial ( $\text{NADP-IDH}_{mit}$ ), sino también una NADP-IDH presente en el citosol ( $\text{NADP-IDH}_{out}$ ) (Eq.8.6). Para poder calcular la producción total de  $\text{CO}_2$  del organismo a partir de nuestras medidas de NADP-IDH debíamos encontrar la forma de relacionar estos

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<sup>1</sup>Los organismos procariotas sólo poseen una isoenzima, la NADP-IDH, careciendo de la NAD-IDH. Además, carecen de mitocondria. Por ello, en su caso, la producción máxima de  $\text{CO}_2$  puede calcularse sustituyendo directamente la  $\text{IDH KC}_{\text{CO}_2}$  por la actividad de la NADP-IDH. La producción máxima de  $\text{CO}_2$  sería 3,57 veces la NADP-IDH.

valores.

$$\text{IDH KC} = \text{NADP-IDH}_{mit} + \text{NAD-IDH}_{mit} \quad (8.5)$$

$$\text{NADP-IDH} = \text{NADP-IDH}_{out} + \text{NADP-IDH}_{mit} \quad (8.6)$$

Para ello, utilizamos los resultados obtenidos por Alp et al. (1976) en crustáceos, para la CS (que se encuentra en la mitocondria), la NAD-IDH (que se encuentra en la mitocondria) y la NADP-IDH (que, al igual que el caso de nuestro método, incluye tanto la mitocondrial como la citosólica). En primer lugar, a partir de los datos aportados por Walsh and Koshland (1984), establecimos la relación entre la CS y la IDH  $KC_{CO_2}$  (Eq.8.7), donde la IDH  $KC_{CO_2}$  es 0,72 veces la actividad de la CS en el ciclo de Krebs.

$$\text{CS} = 1.39 \cdot \text{IDH KC}$$

$$\text{IDH KC} = 0.72 \cdot \text{CS} \quad (8.7)$$

Combinando las tres ecuaciones (Eq.8.5, Eq.8.6, y Eq.8.7) Se puede extraer la fracción de la NADP-IDH que no se encuentra a nivel mitocondrial (Eq.8.8):

$$\text{NADP-IDH}_{out} = \text{NADP-IDH} - (\text{IDH KC} - \text{NAD-IDH}) \quad (8.8)$$

Una vez calculados los valores de cada enzima a partir de los resultados de Alp et al. (1976) (ver Tabla 3.2 en el capítulo 3), procedimos a su normalización por la actividad medida de NADP-IDH (ver Tabla 3.3 en el capítulo 3). Por medio de esta relación se puede calcular la producción de  $CO_2$  de cada enzima a partir de la medición de la NADP-IDH. Teniendo en cuenta todo lo anterior, determinamos que la producción total de  $CO_2$  debe incluir tanto la relacionada con el ciclo de Krebs, como la que sucede en el citosol asociada a la biosíntesis (Eq.8.9).

$$\Psi = KC_{CO_2} + \text{NADP-IDH}_{out} \quad (8.9)$$

Combinando esta ecuación con la Eq.8.5, obtenemos la Eq.8.10.

$$\Psi = 3.57 \cdot (\text{NADP-IDH}_{mit} + \text{NAD-IDH}) + \text{NADP-IDH}_{out} \quad (8.10)$$

Finalmente, incluyendo los ratios medios obtenidos a partir de los datos de Alp et al. (1976) en la Tabla 3.3, podemos relacionar estas actividades a la NADP-IDH (Eq.8.11).

$$\Psi_{NADP} = 3.57 \cdot (0.65 \cdot \text{NADP-IDH} + 0.07 \cdot \text{NADP-IDH}) + 0.38 \cdot \text{NADP-IDH}$$

$$\Psi_{NADP} = 2.94 \cdot \text{NADP-IDH} \quad (8.11)$$

En base a la aplicación de este modelo lineal a las mediciones de NADP-IDH en muestras de mesozooplancton y microzooplancton, la producción potencial de CO<sub>2</sub> fue de  $42,1 \pm 10,9 \text{ } 10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> y  $53,5 \pm 9,5 \text{ } 10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> respectivamente, hasta 1.7 veces el valor de  $\Phi$  en las mismas muestras ( $26,6 \pm 4,8 \text{ } 10^{-3} \mu\text{mol O}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el mesozooplancton y  $42,7 \pm 9,1 \text{ } 10^{-3} \mu\text{mol O}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el microplancton). Si sólo se tiene en cuenta la producción de CO<sub>2</sub> asociada al ciclo de Krebs (Respiratory  $\Psi_{NADP}$ ), ésta es 1.47 veces superior a  $\Phi$ , lo que indica que la estructura enzimática asociada con la oxidación de las moléculas de carbono orgánico está preparada para superar las demandas del ETS.

### 8.3.3 Desarrollo de los ratios teóricos $R_{CO_2}/\Psi_{NADP}$ .

Una vez determinado el valor de  $\Psi_{NADP}$  en nuestras muestras a partir de la medición de la actividad de NADP-IDH, necesitamos conocer la  $R_{CO_2}$  para el estudio de los ratios  $R_{CO_2}/\Psi_{NADP}$ .

Para ello, medimos  $\Phi$  en las mismas muestras en las que medimos  $\Psi_{NADP}$ . En la literatura pudimos encontrar distintos ratios  $R_{O_2}/\Phi$  asociados a diferentes escenarios (ver Tabla 3.1 en capítulo 3). De esta manera determinamos el consumo de O<sub>2</sub> ( $R_{O_2}$ ) que tendría esa comunidad en los diferentes escenarios (ver Fig.3.3a). Estos valores oscilaron entre  $11$  y  $25 \text{ } 10^{-3} \mu\text{mol O}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el mesozooplancton, siendo  $11,1 \pm 2,4 \text{ } 10^{-3} \mu\text{mol O}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el microplancton (Fig.3.3a).

A partir de la aplicación de distintos cocientes respiratorios (RQs) sobre los valores obtenidos de  $R_{O_2}$  (ver Tabla 3.1), obtuvimos los valores de  $R_{CO_2}$  en 12 condiciones diferentes (9 para el mesozooplancton y 3 el microplancton; ver Fig.3.3b). Los valores de  $R_{CO_2}$  obtenidos oscilaron entre  $8$  y  $30 \text{ } 10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el mesozooplancton y entre  $6$  y  $18 \text{ } 10^{-3} \mu\text{mol CO}_2 \text{ min}^{-1}$  (mg protein)<sup>-1</sup> en el microplancton.

En base a estos valores y a  $\Psi_{NADP}$  analizamos los ratios  $R_{CO_2}/\Psi_{NADP}$ , que variaron entre  $0,17$  y  $0,86$  para el mesozooplancton y entre  $0,12$  y  $0,32$  para el microplancton (Tabla 3.4). Teniendo en cuenta las condiciones asociadas a una comunidad muestreada en las Islas Canarias durante la primavera, estimamos que el ratio a utilizar para nuestras muestras debe ser de entre  $0,3$  y  $0,4$  para el mesozooplancton y entre  $0,2$  y  $0,3$  para el microplancton. No obstante, siempre que sea posible, se debe determinar el ratio específico en cada región, campaña, o comunidad, combinando la medición fisiológica de  $R_{CO_2}$  y la medición de  $\Psi_{NADP}$ . De esta manera el ratio utilizado será más representativo del objeto de estudio en cada momento.

### 8.3.4 Estudio de los efectos de la inanición, las diferencias sexuales y el estado trófico de la comunidad en la producción de CO<sub>2</sub> del misidáceo *L. lingvura*

Siguiendo la recomendación del punto anterior, medimos las tasas de producción de CO<sub>2</sub> ( $R_{CO_2}$ ) en distintas incubaciones de *L. lingvura* y las comparamos con la producción máxima de CO<sub>2</sub> de la NADP-IDH. Para analizar su variabilidad en relación a diferentes factores biológicos y ambientales, se llevaron a cabo comparaciones entre: 1) una comunidad en estado de bloom y una en estado post-bloom; 2) entre machos y hembras bien alimentados; y 3) entre hembras bien y mal alimentadas.

#### Ratios $R_{CO_2}$ /IDH en incubaciones de machos y hembras de *L. lingvura*, así como en distintos estadios de inanición.

Durante los experimentos realizados no se evidenciaron diferencias significativas entre machos y hembras en ninguno de los parámetros estudiados: ni en la  $\Phi$  (con un valor medio de  $0,6 \pm 0,12 \mu\text{mol O}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$ ) o ni en la NADP-IDH (con un valor medio de  $0,97 \pm 0,17 \mu\text{mol CO}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$ ), ni en los ratios  $R_{O_2}/\Phi$  (en torno a 0,35 para los organismos bien alimentados) o  $R_{CO_2}/\text{NADP-IDH}$  (en torno a 0,12).

Dentro de los experimentos realizados sobre la comunidad bloom, observamos las variaciones del ratio  $R_{CO_2}/\text{NADP-IDH}$  durante 24h tras haber sido alimentados. Este ratio se mantuvo en una media de  $0,13 \pm 0,04$  durante las primeras 12h, pero se redujo a  $0,08 \pm 0,01$  en las incubaciones más largas (Fig.4.8). Debido a que la actividad de *L. lingvura* sigue un ritmo circadiano (Herrera et al. (2011b)), consideramos que éste influye en los resultados observados en las incubaciones más largas. Por ello, recomendamos no realizar incubaciones superiores a las 6 horas en estudios con *L. lingvura*.

Por otra parte, el grado de inanición sí que afectó a la respuesta fisiológica (ver Fig.4.6). Los valores de  $R_{O_2}/\Phi$  descendieron a  $0,27 \pm 0,08$  entre las hembras que se encontraban tras 12h de inanición, mientras que el  $R_{CO_2}/\text{NADP-IDH}$  ascendió a  $0,16 \pm 0,06$ . Esta diferencia en la respuesta fisiológica hizo que el RQ casi se dobrara durante la inanición, pasando de valores en torno a 0,6 a valores superiores a 1 mol CO<sub>2</sub> mol O<sub>2</sub><sup>-1</sup> (ver Fig.4.7). Este incremento en el RQ durante las primeras horas de la inanición ha sido reportado en otras ocasiones en bacterias (Romero-Kutzner et al. (2015)), en eufausiáceos (Mayzaud et al. (2005)), o en lepidópteros (Kozhantshikov (1938)). Este aumento es debido a que: 1) Durante la inanición se mantiene el metabolismo asociado a la síntesis de lípidos estructurales Jansen et al. (1966); Aster

and Moon (1981); Zara et al. (2000), reduciéndose el consumo de O<sub>2</sub> mientras que se mantiene o incluso se refuerza la producción de CO<sub>2</sub> por biosíntesis. 2) Durante la síntesis lipídica a partir de hidratos de carbono se produce O<sub>2</sub>, reduciéndose aún más el consumo neto de O<sub>2</sub> (Kozhantshikov (1938)).

En base a lo anterior, nosotros hipotetizamos que durante la inanición los nutrientes se utilizan principalmente para el mantenimiento de la estructura celular, en detrimento a la generación de energía. Esta suposición es coherente con el estudio realizado por Teruya et al. (2019) en seres humanos. En él observaron que durante la inanición, se veía una regulación positiva de algunos marcadores metabólicos relacionados con la gluconeogénesis y la protección ante moléculas oxidantes<sup>2</sup>.

### Efectos del estado trófico de la comunidad durante el muestreo sobre las actividades enzimáticas y el ratio $R_{CO_2}/NADP-IDH$

Las condiciones tróficas en las que se encontraba la comunidad de *L. lingvura* durante el muestreo tuvieron un impacto significativo sobre las actividades enzimáticas relacionadas con la respiración. Las actividades de Φ y de la NADP-IDH fueron significativamente mayores en las hembras provenientes de la comunidad en "bloom" que en aquellas provenientes de la comunidad "post-bloom" ( $1,04 \pm 0,14 \mu\text{mol O}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$  y  $1,96 \pm 0,09 \mu\text{mol CO}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$  respectivamente en la comunidad "bloom"; y  $0,58 \pm 0,12 \mu\text{mol O}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$  y  $0,97 \pm 0,19 \mu\text{mol CO}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$  en la comunidad "post-bloom") (Fig.4.1 en el Capítulo 4). Los resultados de Φ obtenidos son coherentes con los encontrados en la literatura, al encontrarse dentro de los valores publicados por Osma et al. (2016c) para el mesozooplancton de Canarias ( $1,47 \pm 0,76 \mu\text{mol O}_2 \text{ h}^{-1} (\text{mg protein})^{-1}$ ). A su vez, son mayores que los publicados por Bondyale-Juez et al. (2017) para una comunidad de *L. lingvura* muestreada en el momento más cálido y oligotrófico del año 2015 ( $0,25 \pm 0,06 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ), y menores que los publicados por Herrera et al. (2011b) para una comunidad de *L. lingvura* muestreada al final de la primavera ( $1,53 \pm 0,43 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg of protein}^{-1}$ ).

El ratio NADP-IDH/Φ, con valores de entre 1,5 y 2,2 mol CO<sub>2</sub> mol O<sub>2</sub><sup>-1</sup> y sin diferencias significativas entre ambas comunidades, muestra que el proceso de descarboxilación está preparado para exceder ampliamente la demanda del sistema de transporte de electrones en el caso de que el misidáceo requiera un aporte extra de energía.

<sup>2</sup>Tal y como se comenta en la introducción, se ha demostrado que la NADP-IDH está involucrada en la producción del NADPH necesario para la producción de glutatión reducido, esencial en la protección celular ante especies de oxígeno reactivo (ROS) (Jo et al. (2001))

A su vez, el ratio  $R_{CO_2}/NADP-IDH$  se mantuvo en torno a 0,12 en ambas comunidades. Este valor puede resultar muy bajo si lo comparamos con los ratios  $R_{O_2}/\Phi$  (0,81 en Bondyale-Juez et al. (2017), entre 0,66 y 1 en Herrera et al. (2011b)). Sin embargo, es similar a los obtenidos en otros procesos asociados a la remineralización, como el ratio de excreción de  $NH_4^+$  frente a la actividad de la GDH ( $0,11 \pm 0,05$  en Fernández-Urruzola et al. (2011)).

Por otra parte, nótese que este ratio está basado en la comparación directa entre  $R_{CO_2}$  y NADP-IDH, no entre  $R_{CO_2}$  y la producción total de  $CO_2$  tal y como se calculó en la sección anterior ( $\Psi_{NADP}$ ). Al ser  $\Psi_{NADP}$  2,94 veces NADP-IDH, los ratios  $R_{CO_2}/\Psi_{NADP}$  obtenidos en estas comunidades de *L.lingvura* fueron aproximadamente 0,04. Este valor es sensiblemente inferior a los ratios teóricos obtenidos en el capítulo 3. Esto es debido a que, por un lado, la respuesta fisiológica obtenida durante estos experimentos, también para el caso del consumo de  $O_2$ , fue menor de lo esperado ( $R_{O_2}/\Phi$  entre 0,27 y 0,35). Si tenemos en cuenta que los RQ obtenidos sí son coherentes con los valores de la literatura, las causas que implicaron la reducción de la respuesta fisiológica en términos de consumo de  $O_2$ , también afectaron a la producción de  $CO_2$ . Por otra parte, los ratios teóricos fueron calculados sobre una comunidad mixta de mesozooplancton, dominada por copépodos. En ellos la proporción  $\Psi_{NADP}/\Phi$  fue de 1,7 mol  $CO_2$  mol  $O_2^{-1}$ . Por el contrario, en *L.lingvura* observamos una proporción media de  $\Psi_{NADP}/\Phi$  de 4,8 mol  $CO_2$  mol  $O_2^{-1}$  ( $NADP-IDH/\Phi = 1,63 \pm 0,4$  mol  $CO_2$  mol  $O_2^{-1}$ ).

El descenso de la respuesta fisiológica junto con una actividad de NADP-IDH superior a la esperado son los responsables de unos valores de  $R_{CO_2}/\Psi_{NADP}$  más bajos de lo esperado inicialmente. Por todo ello se recomienda, siempre que sea posible, medir de manera combinada estos parámetros en cada estudio (al menos  $R_{CO_2}$  y NADP-IDH, siendo recomendable su comparación con  $R_{O_2}$  y  $\Phi$ ). De esta manera, se podrá aplicar ratios específicos de cada campaña, región o comunidad, sobre resto de las muestras de cada estudio. Asimismo, se generará una base de datos que permitirá ahondar en la relación entre estos parámetros y el correcto desarrollo de modelos cinéticos bisustrato (EKM).

### 8.3.5 Variaciones del valor del ratio $\Phi/Chl\_a$ y su relación con la composición de la comunidad microplanctónica en términos de dominancia de fotoautótrofos, heterótrofos o mixótrofos.

Gracias al experimento KOSMOS GC 2014 (ver capítulo 5 desarrollado en la Bahía de Gando (Gran Canaria) en el otoño de 2014 pudi-

mos observar las variaciones del ratio  $\Phi/\text{Chl}_a$  asociado a la comunidad microplanctónica, tanto en condiciones oligotróficas, como bajo la influencia de un afloramiento de agua profunda rica en nutrientes y en condiciones posteriores, monitorizando los cambios en la comunidad.

Durante los primeros 24 días se mantuvieron condiciones oligotróficas, y pudimos observar un cambio en la comunidad asociado a la propia instalación de los mesocosmos (Fig.5.1, I). Los valores de  $\Phi/\text{Chl}_a$  obtenidos en la comunidad de las aguas circundantes fueron de entre 2 y  $4,7 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ , mientras que en los mesocosmos ascendieron a valores de entre 3,4 y  $8 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ . Este cambio está asociado a la transformación de una comunidad dominada por cianobacterias a una mixta, donde el nanoplankton heterótrofo y mixótrofo domina la biomasa Hernández-Hernández et al. (2018). La productividad neta se mantuvo alrededor de  $0 \text{ mg proteína L}^{-1} \text{ d}^{-1}$  (Fig.5.2, I). La integral de  $\Phi$ ,  $\text{Chl}_a$  y  $B$  para cada mesocosmos a lo largo de toda esta fase no mostró ningún efecto asociado al aumento de  $\text{pCO}_2$  (Fig.5.3, I).

Tras la simulación del afloramiento de agua profunda se produjo un bloom de fitoplancton en todos los mesocosmos entre los días 28 y 30 (Taucher et al. (2017); Hernández-Hernández et al. (2018); Taucher et al. (2018)). Esto dio lugar a los valores más bajos de  $\Phi/\text{Chl}_a$  de todo el experimento (Fig.5.1, II), debido a la dominancia de la comunidad fotoautótrofa en el microplankton ( $1,1 \pm 0,6 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ). Durante esta fase, la productividad neta fue máxima entre uno y dos días antes del pico de  $\text{Chl}_a$  ( $0,039 \pm 0,013 \text{ mg proteína L}^{-1} \text{ d}^{-1}$ ) (Fig.5.2, II). Sin embargo, 24 horas después del pico de  $\text{Chl}_a$ , los procesos de remineralización, pastoreo ("grazing") y sedimentación dieron lugar a una pérdida neta de biomasa de  $-0,023 \pm 0,008 \text{ mg proteína L}^{-1} \text{ d}^{-1}$  (Fig.5.2, II). En esta fase, a diferencia de la anterior, se empezaron a observar ciertas tendencias relacionadas con los distintos niveles de  $\text{pCO}_2$ . Si bien la  $\text{Chl}_a$ ,  $\Phi$  y  $B$  aumentaron a mayor  $\text{pCO}_2$ , sólo se puede considerar significativa la relación positiva entre  $\Phi$  y  $\text{pCO}_2$  (Fig.5.3, II). Esta relación es especialmente clara tras extraer los datos asociados a aquellos mesocosmos en los que la proliferación de un alga tóxica (harmful algal species, HAS), *Vicicitus globosus*, impidió el desarrollo del microzooplankton (y del mesozooplankton) (Riebesell et al. (2018); Algueró-Muñiz et al. (2019)).

Una vez que se agotaron los nutrientes, la comunidad microplanctónica evolucionó de manera diferente. Se alcanzaron los valores máximos de  $\Phi/\text{Chl}_a$  cuando el microzooplankton, dominado por dinoflagelados, proliferó en los mesocosmos con tratamientos de  $\text{pCO}_2$  bajos o intermedios (hasta  $10 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ). Sin embargo, en los meso-

cosmos más ácidos, la proliferación de *V.globosus* hizo que  $\Phi/\text{Chl}_a$  se mantuviera más bajo (entre 2,5 y 7  $\mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ ) (Fig.5.1, III). Sin embargo estas variaciones entre los distintos mesocosmos no afectaron a su productividad neta. Tras los efectos del bloom de diatomeas y el agotamiento posterior de los nutrientes, la productividad neta tendió a un estado de equilibrio similar al de la fase oligotrófica en todos los casos (Fig.5.2, III), solo que con mayor variabilidad en esta fase.

El análisis integral de  $\text{Chl}_a^{-1}$ ,  $\Phi$  y B en esta última fase no muestra una relación significativa con el aumento pCO<sub>2</sub>, especialmente si se extraen los valores asociados a los mesocosmos en los que *V.globosus* proliferó (Fig.5.3, III). Sin embargo, considerando estos valores,  $\text{Chl}_a^{-1}$  sí tiene una relación positiva con la acidificación. La proliferación de *V.globosus* en condiciones de alta pCO<sub>2</sub> puede estar asociada al refuerzo de su actividad fotosintética asociado a una mayor concentración de CO<sub>2</sub> en el entorno. Al carecer de pirenoide, su sistema fotosintético es ineficiente en condiciones normales, sin embargo queda reforzado en condiciones de alta pCO<sub>2</sub>. En océanos donde HAS similares a *V.globosus* formen parte de la comunidad, el aumento de pCO<sub>2</sub> puede resultar en una relación positiva con la Chl<sub>a</sub> debido al refuerzo de este tipo de sistema fotosintético carente de pirenoide, ineficiente en las condiciones actuales de pCO<sub>2</sub>. Asimismo, la ausencia de microzoooplancton (y mesozooplancton) reduce el grazing sobre cianobacterias y diatomeas, lo que genera una mayor retención de Chl<sub>a</sub> en el medio (Riebesell et al. (2018); Taucher et al. (2018); Algueró-Muñiz et al. (2019)).

## 8.4 Conclusiones

### 8.4.1 El desarrollo de un método para determinar la actividad de la NADP-IDH en el plancton como una de las principales enzimas responsables de la producción de CO<sub>2</sub>

Durante el desarrollo de esta tesis creamos un método enzimático para determinar la actividad de la NADP-IDH en el zooplancton y el microplancton. Este método se basa en monitorizar la reducción de NADP<sup>+</sup> a NADPH durante la reacción de decarboxilación del isocitratro a  $\alpha$ -acetoglutarato, catalizada por esta enzima. Las tasas de producción de NADPH ( $\mu\text{mol NADPH min}^{-1} \text{mL}^{-1}$ ) se obtienen espectrofotométricamente siguiendo el aumento de la absorbancia a 340nm en el tiempo. Para poder expresar la actividad en  $\mu\text{mol}$  de NADPH se utiliza un coeficiente de extinción molar ( $\epsilon$ ) de 5,42 mL  $\mu\text{mol}^{-1}\text{cm}^{-1}$ . En nuestras condiciones de laboratorio, conseguimos alcanzar la velocidad máxima

de reacción ( $V_{max}$ ) en la comunidad planctónica con un buffer fosfato 0,1M a pH 8.2 ( $pK_a_1=7,6$  and  $pK_a_2=8,8$ ), con unas concentraciones en la cubeta de 6mM de MgCl<sub>2</sub>, 0,3mM de NADP<sup>+</sup> y 2mM de isocitrato DL-trisódico.

La afinidad de la NADP-IDH con sus sustratos se obtuvo determinando las constantes aparentes de Michaelis-Menten ( $K_m$ s). A un pH de 8.5 y a 18°C el  $K_m$  del isocitrato fue  $271\pm63 \mu\text{M}$  y el del NADP<sup>+</sup>,  $18\pm3 \mu\text{M}$ .

A partir de la representación de Arrhenius tanto de la actividad de la NADP-IDH proveniente del mesozooplancton como de la proveniente del corazón de cerdo, observamos que la energía de activación ( $E_a$ ) es muy similar (20,4 Kcal mol<sup>-1</sup> y 21,4 Kcal mol<sup>-1</sup> respectivamente). Esto indica que, si bien existen diferencias entre los organismos homeotermos o poiquilotermos en cuanto a su óptimo de temperatura, una estructura básica similar, a nivel celular, parece haberse mantenido a lo largo de la evolución de las distintas especies. Asimismo, permite el desarrollo de una ecuación de Arrhenius precisa para la comparación de actividades medidas a distintas temperaturas.

#### 8.4.2 El estudio de la producción máxima de CO<sub>2</sub> del mesozooplancton a partir de las mediciones de NADP-IDH

Por otra parte, hemos calculado la producción máxima de CO<sub>2</sub> del mesozooplancton teniendo en cuenta otras enzimas productoras de CO<sub>2</sub> y normalizando su actividad por la NADP-IDH. La producción máxima de CO<sub>2</sub> derivada de la NADP-IDH ( $\Psi_{NADP}$ ) es  $2,94 \cdot \text{NADP-IDH}$  para muestras mixtas de mesozooplancton.

Durante el desarrollo del modelo de  $\Psi_{NADP}$  observamos que no sólo el metabolismo respiratorio, sino también la biosíntesis, afectan a la producción de CO<sub>2</sub>. Sin embargo, al analizar el metabolismo respiratorio, observamos que la producción máxima de CO<sub>2</sub> asociada al ciclo de Krebs ( $\Psi_{NADP}$  Respiratorio), excede en 1,47 veces el consumo máximo de O<sub>2</sub> asociado al ETS. Esto indica que la estructura enzimática asociada con la oxidación de las moléculas de carbono orgánicos está preparada para superar las demandas del ETS. Esto puede ser una ventaja cuando el organismo tiene una alta demanda de energía. Una vez que se han alcanzado los niveles de energía requeridos, el exceso de NADPH puede ser redirigido hacia la biosíntesis lipídica.

Cuando comparamos  $\Psi_{NADP}$  con la tasa de producción de  $R_{CO_2}$  obtenida a partir de la medición de  $\Phi$  y su conversión con los distintos ratios publicados en la literatura observamos que en todos los casos  $\Psi_{NADP}$  es mayor que  $R_{CO_2}$ . Este resultado es acorde con el concepto teórico ya que una tasa fisiológica nunca debería superar la capacidad má-

ima del organismo. Por ello  $\Psi_{NADP}$  puede ser considerada la capacidad máxima de producción de CO<sub>2</sub> en la comunidad planctónica marina. De nuestros cálculos en el Atlántico Subtropical Nordeste, estimamos que los ratios  $R_{CO_2}/\Psi_{NADP}$  para el zooplancton oscilarán entre 0,3 y 0,4, siendo entre 0,2 y 0,3 en una comunidad microplanctónica. Sin embargo, recomendamos medir las tasas  $R_{CO_2}$  siempre que sea posible y desarrollar los propios  $R_{CO_2}/\Psi_{NADP}$  específicos para cada región, experimento o situación.

#### 8.4.3 El estudio de la relación entre $R_{CO_2}$ and la actividad de la NADP-IDH

Al desarrollar los ratios  $R_{CO_2}/\text{NADP-IDH}$  en *L. lingvura* observamos que:

- \* El estado nutricional en el que se encuentre la comunidad en el momento de muestreo tiene un impacto significativo en la actividad enzimática respiratoria de *L. lingvura*.
- \* Aunque la actividad de la NADP-IDH fue mayor en la comunidad "bloom" que en la "post-bloom", el ratio  $R_{CO_2}/\text{NADP-IDH}$  tuvo unos valores medios de 0,12 en ambas comunidades para aquellos individuos que se encontraban bien alimentados. Durante el periodo de inanición, este valor ascendía levemente hasta 0,16. Debido a estos valores se mantienen en comunidades diferentes, se puede decir que caracterizan la relación entre la respuesta fisiológica y la actividad metabólica de *L. lingvura*.
- \* Las diferencias entre machos y hembras no reflejaron diferencias significativas ni en las actividades enzimáticas ni en los ratios relacionados con la respuesta fisiológica.
- \* La inanición durante más de 12h da lugar a un descenso en el ratio  $R_{O_2}/\Phi$ , y a un aumento en el ratio  $R_{CO_2}/\text{NADP-IDH}$ . Esto dió lugar a un aumento de los valores del RQ durante la inanición. Nuestra interpretación es que los requerimientos de producción de energía se limitan durante la inanición mientras que la síntesis lipídica y de los sistemas de protección ante moléculas oxidantes se refuerzan.
- \* La incubación de los misidáceos por más de 12h no permite identificar claramente los efectos de la inanición, ya que se mezclan con la influencia del ritmo circadiano en la producción de CO<sub>2</sub>. Para evitarlo, recomendamos incubaciones de entre 4 y 6 horas en los futuros estudios con *L. lingvura*.

#### **8.4.4 La aplicación de una técnica enzimática para el estudio de las variaciones en el estado autotrófico-heterotrófico de la comunidad microplanctónica**

La monitorización del ratio  $\Phi/\text{Chl}_a$  durante los cambios metabólicos de una comunidad debidos a la influencia de un afloramiento de aguas profundas cargadas de nutrientes permite concluir lo siguiente:

- \* Valores de entre 0 y  $2,5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  son los resultantes en una comunidad dominada por fotoautótrofos. Durante el bloom de diatomeas este ratio obtuvo valores inferiores a 1,7. Valores similares se han reportado en la literatura para zonas muy eutróficas o en el máximo de  $\text{Chl}_a$  de la columna de agua.
- \* Valores de entre  $2,5$  y  $7,5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$  son valores asociados bien a una comunidad mixta, bien a una comunidad dominada por mixótrofos.
- \* Valores superiores a  $7,5 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ . Son valores relacionados con la proliferación de microzooplancton. Durante el experimento, en el momento en el que los dinoflagelados heterótrofos dominaron la comunidad, se alcanzaron valores de hasta  $10 \mu\text{LO}_2\text{h}^{-1} \mu\text{g Chl}_a^{-1}$ .

El cálculo de la productividad neta como la primera derivada de la biomasa por tiempo permite identificar este proceso en aquellos casos en los que no existan medidas de fijación de  $^{14}\text{C}$ . Asimismo, los valores negativos de este parámetro, permiten cuantificar el resultado neto de la combinación de los distintos procesos de pérdida de biomasa en el ecosistema.

El uso del análisis integral por fase y mesocosmos de los parámetros oceanográficos  $\text{Chl}_a$ ,  $\Phi$  y  $B$  permite la identificación de los efectos de la acidificación. Se observa una relación positiva claramente significativa entre  $\Phi$  y  $\text{pCO}_2$  en los casos en los que la ausencia de HAS permite la proliferación de la comunidad microzooplantónica. En los casos en los que se produce una proliferación de HAS, la  $\text{Chl}_a$  también recibe un impacto positivo con el aumento de  $\text{pCO}_2$ , bien por la reducción del grazing sobre las diatomeas y cianobacterias, bien por la proliferación del sistema fotosintético asociado a las HAS, carente de pirenoide e ineficiente en condiciones de bajo  $\text{pCO}_2$ .

## **Appendix A**

# **Appendix A. NADP-IDH assay extended information**

### **A.1 Buffer studies**

An enzymatic assay is performed on buffered reaction mixtures containing cofactors and effectors. Here, the optimum combination between buffer and the main cationic effectors ( $Mg^{2+}$  or  $Mn^{2+}$ ) was obtained from ten different experiments that were run at 18°C, pH 8.5, 4mM isocitrate, 0.08mM NADP<sup>+</sup> and 2mM MgCl<sub>2</sub> on five different concentrations of purified NADP-IDH from porcine heart (I2002, Sigma Aldrich) (zooplankton samples were not used here in order to maintain the same sample conditions in every experiment). Preliminary tests were performed to determine the range of purified NADP-IDH concentrations needed to optimize this study. The response in the enzyme activity was analysed for different combinations of buffer, cationic effector (Kornberg and Pricer (1951); Munilla-Moran and Stark (1989)) and the presence of lysozyme in the homogenisation buffer (Berdelet et al. (1995a)): (1) 25mM MOPS, 2mM MgCl<sub>2</sub> and 0.167mg mL<sup>-1</sup> lysozyme (SIGMA L6876); (2) 25mM MOPS and 2mM MgCl<sub>2</sub>; (3) 25mM MOPS, 2mM MnCl<sub>2</sub>·4H<sub>2</sub>O (PANREAC 131410) and 0.167mg mL<sup>-1</sup> lysozyme; (4) 25mM MOPS and 2mM MnCl<sub>2</sub>·4H<sub>2</sub>O; (5) 0.1M phosphate buffer, 2mM MgCl<sub>2</sub> and 0.167mg mL<sup>-1</sup> lysozyme; (6) 0.1M phosphate buffer and 2mM MgCl<sub>2</sub>; (7) 0.1M TRIS 2mM MgCl<sub>2</sub> and 0.167mg mL<sup>-1</sup> lysozyme; (8) 0.1M TRIS and 2mM MgCl<sub>2</sub>; (9) 0.1M TRIS, 2mM MnCl<sub>2</sub>·4H<sub>2</sub>O and 0.167mg mL<sup>-1</sup> lysozyme; (10) 0.1M TRIS and 2mM MnCl<sub>2</sub>·4H<sub>2</sub>O. Results were based on triplicate measurements. NADP-IDH activity was reported in  $\mu$ mol NADPH min<sup>-1</sup>(mL of homogenate)<sup>-1</sup> units.

NADP-IDH activity in phosphate buffer resulted to be 45.88% higher than the next higher activity (in MOPS) for the same dilute enzyme so-

lution in the lowest concentration (the closest to zooplankton samples,  $1 \mu\text{mol NADPH min}^{-1} \text{ mL}^{-1}$  in the X axis of Fig.A.1). No results for  $\text{Mn}^{2+}$ -phosphate buffer were obtained because  $\text{Mn}^{2+}$  reacted with the phosphate to give a  $\text{Mn}_3(\text{PO}_4)_2$  precipitate, resulting in a competing chemical reaction among the reagents that prevented correct readings of the spectrophotometer. Given the reactivity of  $\text{Mn}^{2+}$  with phosphate buffer,  $\text{Mn}^{2+}$  was eliminated and  $\text{Mg}^{2+}$  was selected as the metal cation for this assay.

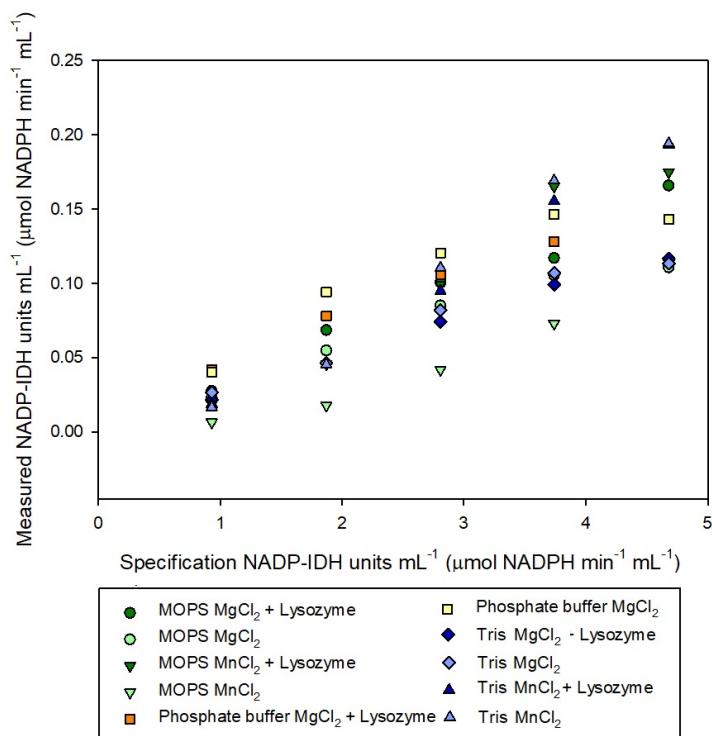


Figure A.1: Buffer effects on NADP-IDH activity

## A.2 Molar extinction coefficient ( $\varepsilon$ ) studies

This assay is based on spectrophotometry, the Beer-Lambert law, and the absorbance at 340nm ( $A_{340}$ ), all of which go to determine the concentration of NADPH (Section 3.2.2). To accomplish this we needed to determine the molar extinction coefficient ( $\varepsilon$ ) for the NADPH in our specific reactant solution. The molar extinction coefficient ( $\varepsilon$ ) is a measurement of the attenuation of light, at a specific wavelength, due

to a molar concentration of the target chemical species. It quantifies the relationship between the light absorbance and the concentration of the chemical species. Thus,  $\epsilon$  for the NADPH was analysed at pH 8.5 and 18°C by measuring absorbance at 340nm ( $A_{340}$ ) under simulated reaction conditions and simulated blank conditions. Ten different concentrations of NADPH, ranging from 0.08 to 40 $\mu$ M, were prepared with three solutions: (1) phosphate buffer 0.1M, 2mM isocitrate and 3.3mM MgCl<sub>2</sub>; (2) phosphate buffer 0.1M and 3.3mM MgCl<sub>2</sub>; and (3) only phosphate buffer 0.1M. Results were based on triplicate measurements. The molar extinction coefficient was reported in absorptivity units mL  $\mu$ mol<sup>-1</sup> cm<sup>-1</sup>.

Reported NADPH  $\epsilon$  in the literature is 6.2 L mmol<sup>-1</sup> cm<sup>-1</sup> (Morris and Redfearn (1969)), which was measured at pH 7.5 in a solution different from the buffer in our NADP-IDH assay. In our hands, Fig.A.2, the increase of absorbance per mmol L<sup>-1</sup> of NADPH was similar for the three experiments (5.21 L mmol<sup>-1</sup> cm<sup>-1</sup> for the phosphate buffer, 5.35 L mmol<sup>-1</sup> cm<sup>-1</sup> for the phosphate buffer with MgCl<sub>2</sub> and 5.42 L mmol<sup>-1</sup> cm<sup>-1</sup>, for the phosphate buffer with MgCl<sub>2</sub> and isocitrate). Our reaction solution is similar to the third buffer, so we will use an  $\epsilon$  of 5.42L mmol<sup>-1</sup> cm<sup>-1</sup> in our calculations. If we were to use an  $\epsilon$  from the literature (6.2 L mmol<sup>-1</sup> cm<sup>-1</sup>) we would introduce an error of 12.6%.

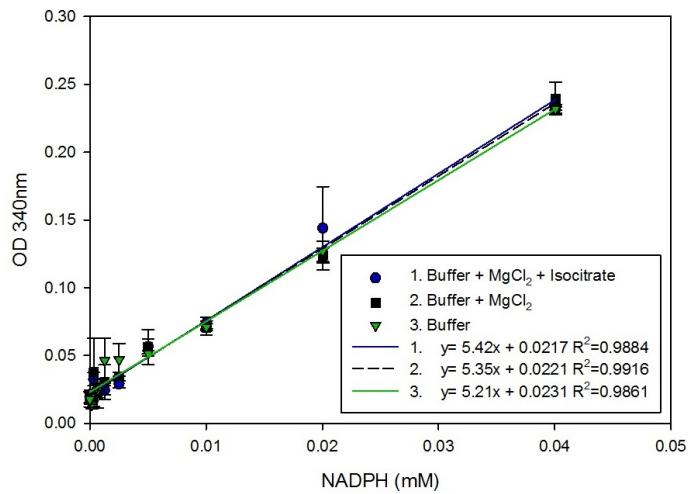


Figure A.2: NADPH molar extinction coefficient, in three different buffer combinations.

Note that the  $\epsilon$  increased with the complexity of the buffer solution. So, testing the  $\epsilon$  related to product measurements on the solution where re-

action is happening, would be a good practice to reduce error in future measurements.

### A.3 Kinetics

The NADP-IDH activity related to the substrate concentration showed a rectangular hyperbolic relationship for isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub>, following the expression represented in Eq.A.1.

$$\%V_{max} = a \cdot [S] / (y_0 + [S]) \quad (\text{A.1})$$

The values of “a” which represents the V<sub>max</sub> (in this case, the % of V<sub>max</sub>); and “y<sub>0</sub>”, which represents the K<sub>m</sub>, are shown in Table A.1.

Table A.1: Michaelis-Menten hyperbolic curves for isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub> affinity in marine plankton samples. K<sub>m</sub> is in mM units. For isocitrate, V<sub>max</sub>=0.038 in mesozooplankton; 0.031 in microzooplankton; and 0.002 in microplankton. For NADP<sup>+</sup>, V<sub>max</sub>=0.029 in mesozooplankton; 0.019 in microzooplankton; and 0.001 in microplankton. For MgCl<sub>2</sub>, V<sub>max</sub>=0.010 in mesozooplankton; 0.009 in microzooplankton; and 0.00002 in microplankton. V<sub>max</sub> is in μmol NADPH min<sup>-1</sup> (mL homogenate)<sup>-1</sup> units.

Size-fractionated sample / Substrate	a (%V <sub>max</sub> )	y <sub>0</sub> (K <sub>m</sub> )	r <sup>2</sup>	p-value
Mesozooplankton / isocitrate	109.34	0.225	0.987	<0.0001
Microzooplankton / isocitrate	109.25	0.270	0.985	<0.0001
Microplankton / isocitrate	96.99	0.203	0.962	<0.0001
Average values for isocitrate	105.20	0.232±0.034	NA	NA
Mesozooplankton / NADP <sup>+</sup>	102.75	0.024	0.926	<0.0001
Microzooplankton / NADP <sup>+</sup>	93.45	0.023	0.930	0.0001
Microplankton / NADP <sup>+</sup>	96.83	0.020	0.981	<0.0001
Average values for NADP <sup>+</sup>	97.68	0.022±0.007	NA	NA
Mesozooplankton / MgCl <sub>2</sub>	142.94	2.189	0.976	0.0135
Microzooplankton / MgCl <sub>2</sub>	133.70	2.198	0.968	0.0243
Microplankton / MgCl <sub>2</sub>	106.11	2.542	0.930	0.0395
Average values for MgCl <sub>2</sub>	127.58	2.310±0.201	NA	NA

NADP-IDH dependences on isocitrate were not significantly different among these samples (F test, F<sub>(4,63)</sub>=1.73, p=0.01). On the other hand, dependence on NADP<sup>+</sup> showed significant differences among the three size-fractions (F test, F<sub>(4,69)</sub>=6.85, p=0.01). Even though the regressions were different, the average K<sub>m</sub> and optimum values in both experiments can be properly used in our assay in order to obtain the V<sub>max</sub> for the whole plankton community. Furthermore, although there are significant differences in the NADP-IDH and MgCl<sub>2</sub> affinity among

the samples ( $F$  test,  $F_{(4,114)}=2.75$ ,  $p=0.01$ ), the optimal value obtained by inspection of Fig.2.1, can be properly used for our assay in order to obtain the  $V_{max}$  for the whole plankton community (6mM MgCl<sub>2</sub> for the whole community).

We applied three main linearisation techniques to evaluate the  $K_m$ s of isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub>. In the Hanes-Woolf linearisation, substrate concentration [S] was plotted against concentration per enzymatic activity ([S]/V). The  $V_{max}$  was obtained from the inverse of the slope (a) ( $V_{max} = 1/a$ ), whereas  $K_m$ s were equal to the ratio of the Y intercept ( $y_0$ ) and a ( $K_m = y_0/a$ ) (Suelter et al. (1985)). In the case of the Lineweaver-Burk method, the inverse of the substrate (1/[S]) was plotted against the inverse of the activity (1/V), so the  $V_{max}$  was the inverse of  $y_0$  ( $V_{max}=1/y_0$ ) and  $K_m = a/y_0$ . Finally, using the Eadie-Hofstee methodology we plotted V/[S] against V, where the  $V_{max} = y_0$  and  $K_m = -y_0/a$ .  $K_m$  was reported in mmol of substrate, and  $V_{max}$  in  $\mu\text{mol NADPH min}^{-1}(\text{mL of homogenate})^{-1}$  units (Suelter et al. (1985)).

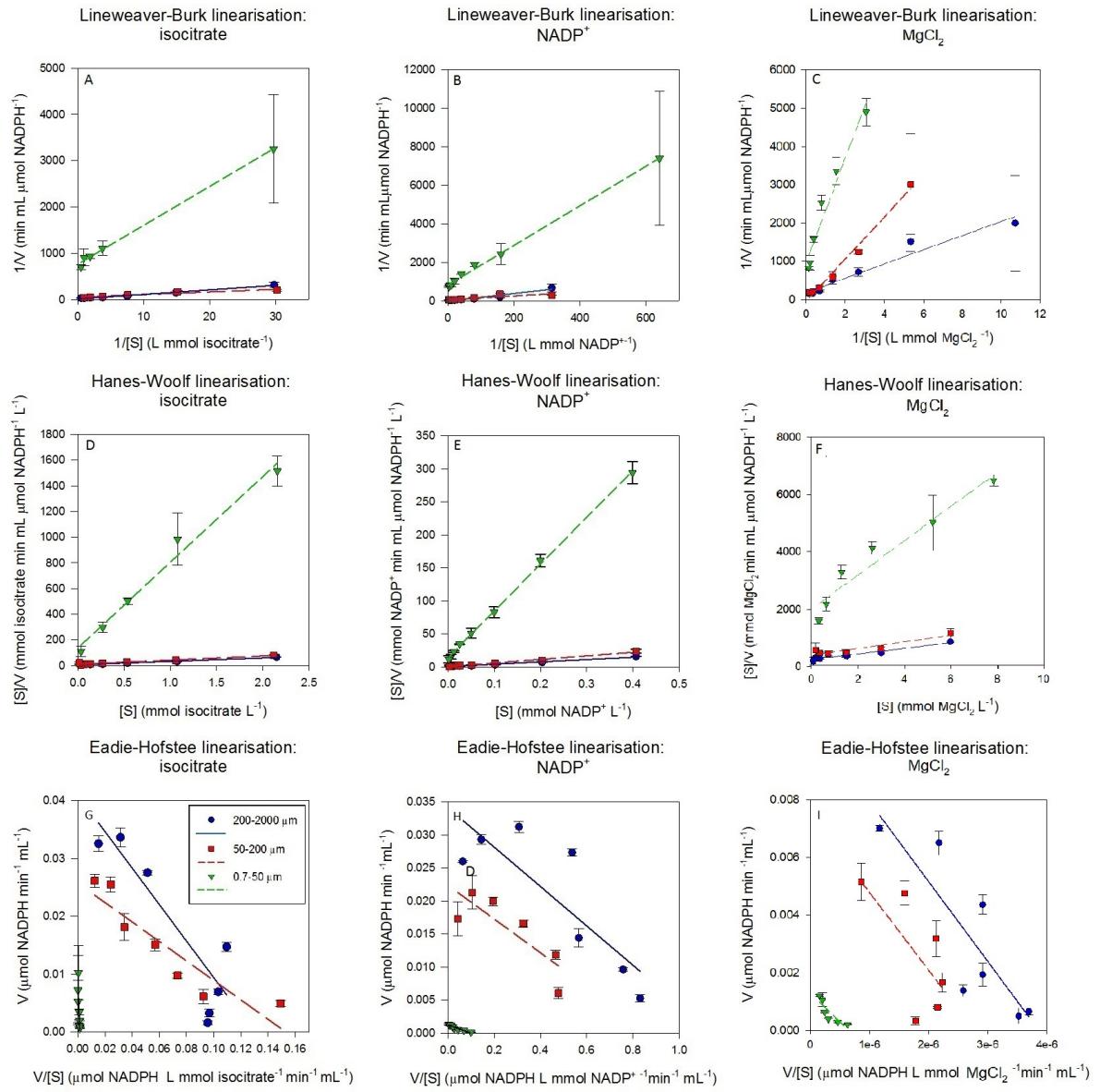


Figure A.3: Hanes-Woolf, Lineweaver-Burk and Eadie-Hofstee linearisations to determine the affinity ( $K_m$ ) of NADP-IDH for isocitrate, NADP $^{+}$  and MgCl $_2$  in marine-plankton samples. The symbols in the key of panel G apply to all six panels.

Finally, Hanes-Woolf linearisation was the most accurate to our data base for both isocitrate, NADP<sup>+</sup> and MgCl<sub>2</sub> ( $p<0.01$ ). The apparent  $K_m$ s of  $271\pm63\mu\text{M}$  of isocitrate,  $18\pm3\mu\text{M}$  of NADP<sup>+</sup> and  $3060\pm643\mu\text{M}$  of MgCl<sub>2</sub> are consistent with those obtained by Michaelis-Menten hyperbolic regressions.

## A.4 pH effects

Michaelis-Menten constants are affected by pH levels that impact ionisation around the enzyme active site. Thus, the conformations of the active site, the binding of substrate to the enzyme, and/or the transformation from substrates to products may be altered. This ionisation is sensitive to the chemical solution, shifting the  $pK_a$  values in relation to the nature of the buffer solution (Whitaker (1993)). Although we have not checked the  $K_m$ s for every pH condition, during the pH experiments we maintained concentrations of isocitrate and NADP<sup>+</sup> high enough to broadly saturate the enzyme at pH 8.5, assuming the  $V_{max}$  was reached at every experimental pH level. Finally, a pH of 8.2 (Fig.2.2) resulted to be the marine NADP-IDH optimum, with  $pK_{as}$  at 7.62 and 8.75. For Fig.2.2, results were normalised by  $V_{max}$  and reported as a percentage.

## A.5 Temperature effects: Development of the Arrhenius equation

Another reason for determining the Arrhenius relationship between NADP-IDH activity and T, is to correct measurements when environmental T is different from the measurement T. If plankton samples were assayed at the in-situ T from a low temperature environment, there is a risk that the NADP-IDH signal would fall below the limit of detection. Accordingly, enzyme activities are measured at T above the in-situ one (Packard (1971)) and converted back to the activity at in-situ T using the Arrhenius equation (Fig.2.3) (Packard (1971)). This equation is equivalent to Van't Hoff equation (Atkin and de Paula (2006)), as expressed on Eq.A.2, A.3 and A.4:

$$\ln(K_2/K_1) = (Ea/R) \cdot ((1/T_1) - (1/T_2)) \quad (\text{A.2})$$

Taking anti-logarithms we obtained the ratio of the rate constants:

$$K_2/K_1 = \exp((Ea/R) \cdot ((1/T_1) - (1/T_2))) \quad (\text{A.3})$$

Then, we transposed to solve for the in-situ rate:

$$K_2 = K_1 \cdot \exp((Ea/R) \cdot ((1/T_1) - (1/T_2))) \quad (\text{A.4})$$

where:  $K_2$  is the in-situ reaction rate and  $K_1$  is the measured reaction rate at  $T_2$  (in-situ T) and  $T_1$  (measurement T) respectively, in  $\mu\text{mol NADPH min}^{-1} \text{ mL}^{-1}$ ;  $Ea$  is the energy of activation in  $\text{Kcal mol}^{-1}$ ;  $R$  is the molar gas constant in  $\text{Kcal mol}^{-1} \text{ }^\circ\text{K}^{-1}$ ; and  $T_1$  and  $T_2$  are temperatures in  ${}^\circ\text{K}$ .

In the case of the marine plankton NADP-IDH, with an  $Ea$  of 20.39  $\text{Kcal mol}^{-1}$  NADP-IDH activity for the in-situ T ( $T_i$ ) is obtained by applying Eq.A.5:

$$\text{NADP-IDH}(T_i) = \text{NADP-IDH}(T_m) \cdot \exp((20.39/R) \cdot (1/T_m - 1/T_i)) \quad (\text{A.5})$$

where:  $\text{NADP-IDH}(T_i)$  is NADP-IDH activity at in-situ temperature;  $\text{NADP-IDH}(T_m)$  is NADP-IDH activity at temperature in the spectrophotometric cuvette during measurement;  $R$  is  $1.9872 \cdot 10^{-3} \text{ Kcal mol}^{-1} \text{ }^\circ\text{K}^{-1}$ ;  $T_m$  is temperature in the spectrophotometric cuvette during measurement in  ${}^\circ\text{K}$ ; and  $T_i$  in-situ temperature in  ${}^\circ\text{K}$ .

## A.6 Precision studies

The precision of this assay was tested on four different solutions of porcine heart NADP-IDH (I2002, Sigma Aldrich). These solutions were made by diluting the original solution (93.6 units of NADP-IDH  $\text{mL}^{-1}$ ), resulting in four solutions which ranged from 0.0004 to 0.25 units of NADP-IDH  $\text{mL}^{-1}$  which were reported in specification units (*UsIDH*, i.e.,  $\mu\text{mol NADPH min}^{-1} (\text{mL of homogenate})^{-1}$ ). Ten duplicates were used for every solution. Note that one unit (U) of enzyme activity was defined as the amount of sample able to catalyse the production of  $1\mu\text{mol NADPH in 1 min}$ .

Precision was scaled by using the standard deviation with the optimum pH and ligands from previous experiments (pH 8.2, 6mM  $\text{MgCl}_2$ , 2.1mM isocitrate and 0.2mM  $\text{NADP}^+$ ). The measurement was made at  $37.5^\circ\text{C}$  for comparison with the theoretical characteristics of pure porcine heart NADP-IDH (specification sheet I2002 by Sigma Aldrich). Results were tested in ten samples for every solution and reported as measured units (*UmIDH*), in  $\mu\text{mol NADPH min}^{-1} (\text{mL of homogenate})^{-1}$ .

The average standard deviation was  $1.2\text{nmol NADPH min}^{-1}$ , ranging from  $0.6\text{nmol}$  in the two lower values, to  $2\text{nmol NADPH min}^{-1}$  in the most concentrated ones (Fig.A.4). On a percentage basis, except for the lowest value, where it represented 25% of the sample value, the standard deviation was less than 3.5% for the samples with more than  $25\text{nmol NADPH min}^{-1}$ , dropping to 1.73% for the highest concentration (Fig.A.4).

Using our new NADP-IDH assay we also measured the activity of the Sigma-Aldrich porcine-heart NADP-IDH. We used the same pH and T as those specified in the Sigma-Aldrich specification sheet (pH=7.4; T= $37.5^\circ\text{C}$ ), and found 39.3% of the theoretical activity (Fig.A.4). When we measured the activity at pH 8.2 (optimum value in Fig.2.2) we found 48.2% (Fig.A.4) of the theoretical activity. The use of pure-porcine NADP-IDH activity to normalise the method is useful to compare with other samples, organisms and methodologies. In our hands, one unit ( $\mu\text{mol NADPH min}^{-1}$ ) of NADP-IDH activity in marine plankton homogenates, was equivalent to 2.08 pure porcine-heart NADP-IDH units, by applying the equation obtained from Fig.A.4 and Eq.A.6. This equation pertains to the optimum conditions of our new NADP-IDH method:

$$UmIDH = 0.48 \cdot UsIDH + 0.0029 \quad (\text{A.6})$$

where *UmIDH* refers to NADP-IDH activity units as measured by the new method, and *UsIDH* refers to the specification NADP-IDH activity

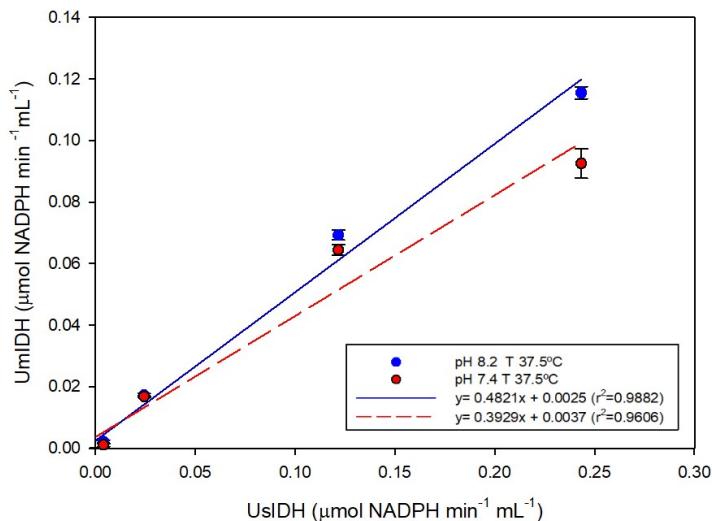


Figure A.4: Porcine-heart NADP-IDH activity measured by the marine-plankton NADP-IDH assay at pH:8.2 and pH:7.4

units as calculated for the pure porcine-heart NADP-IDH solutions.

In the linear range of Fig.A.4, which describes the increase in NADP-IDH activity as a function of enzyme concentration, it is clear that the substrate concentrations are sufficient to maintain maximum activity until 0.25  $UsIDH$  units of NADP-IDH (which represents a measured value of 0.12  $UmIDH$  with our methodology). Samples from the Canary Islands plankton never reached such a high value, so they were always in the linear range. For future plankton samples, if a pilot test were to show activity above 0.12  $UmIDH$   $\mu\text{mol NADPH min}^{-1} \text{mL}^{-1}$ , the homogenate would need dilution.

## A.7 Understanding changes in the NADP-IDH spectrophotometrical response

A linear  $A_{340}$  increasing (Fig.A.5(A)) is expected when NADP-IDH is catalysing the reaction at  $V_{max}$ . However, in marine plankton samples, some conditions may change this linearity as follows:

- 1 Low Biomass samples (Fig.A.5(B)). Under these conditions, the rate of NADPH formation is low, and the signal was not detectable for the first 500s. The NADP-IDH activity, in this case, is under the limit of detection. If more biomass is not available, sensitivity can be obtained by raising the reaction temperature to 30°C and

applying the Arrhenius equation to correct the activity to that of the in-situ T (Eq. 17).

- 2 T effects (Fig.A.5(C)). Depending on the difference between the laboratory T and the spectrophotometer read-out T, the reaction temperature will lag until thermal equilibrium is reached in the cuvette. We monitored T in the cuvette over time and for 8 to 46°C range, and found that steady-state was reached after 450s. We suggest that enzyme activity measurements should be made only after this lag-period (7.5 min). If the plateau zone is reached within the first 500s, the cuvette might not have reached the equilibrium T. In this case, we suggest repeating the measurement with a diluted sample. Given the importance of T and all the problems with maintaining it in the cuvette, we also suggest measuring T in the cuvette at the end of the reaction.
- 3 High Biomass samples (Fig.A.5(D)). The optimum values for NADP-IDH substrates reported on this study have proved to work well for the biomass levels found in plankton samples around the Canary Islands archipelago. Eventually, the reagents are completely consumed when the  $A_{340}$  reaches 0.8. After that point, the reaction becomes non-linear, underestimating the  $V_{max}$ .
- 4 Blank effects (Fig.A.5(E)). Working with natural homogenates, instead of pure enzyme extracts, rises interference from other enzymes oxidizing the produced NADPH. Although this effect is not usually significant because these enzymes are not supplied with their other substrates, eventually, some samples will produce negative blanks that could significantly increase the results. Particle sinking in the cuvette due to low centrifugation may also affect the measurements in this way (Section 3.5). To minimise the sedimentation effect we recommend centrifugation (at least 2 min at 1000rpm for zooplankton samples, or 10min at 4000rpm for microplankton samples collected on 0.7GF/F filters). If the condition persists, even after the centrifugation, and the blank remains negative, a background reaction is likely real and has to be taken into account.

During the T and pH experiments, great differences in the slopes (the velocity of the reaction) at different levels of T or pH were observed (Fig. A.6) Higher slopes were observed near the optimum values. Temperatures shown in Fig. A.6 represent final T in the cuvette. When this temperature was close to the temperature of the lab (around 21°C), we obtained the same slope for 1500s, throughout the measurement. However, when the reaction temperatures were far from the laboratory T, time was needed for the cuvette to reach thermal equilibrium. As said before, we have observed that 500s ensures a good measurement at the

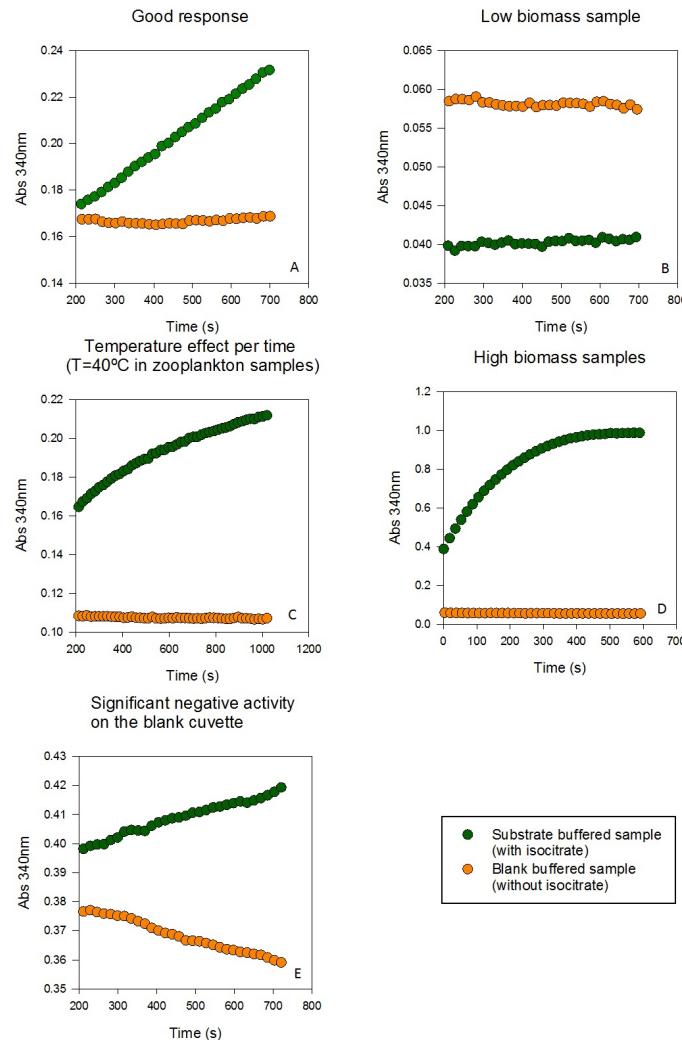
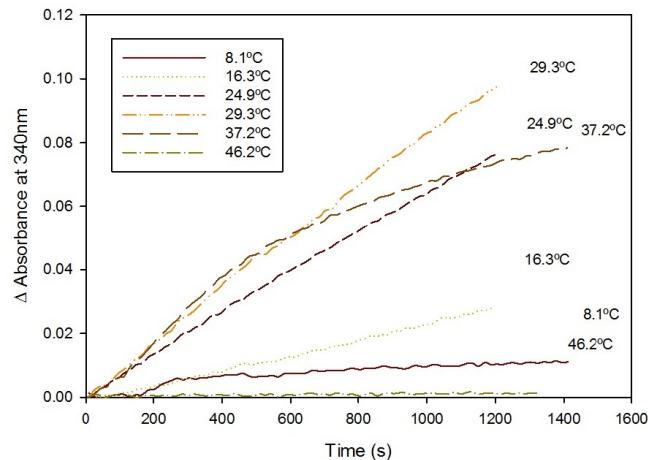


Figure A.5: Spectrophotometrical response of NADP-IDH activity, following the  $A_{340}$  per time of the substrate and blank buffered cuvettes, under conditions that affect the signal during the measurement

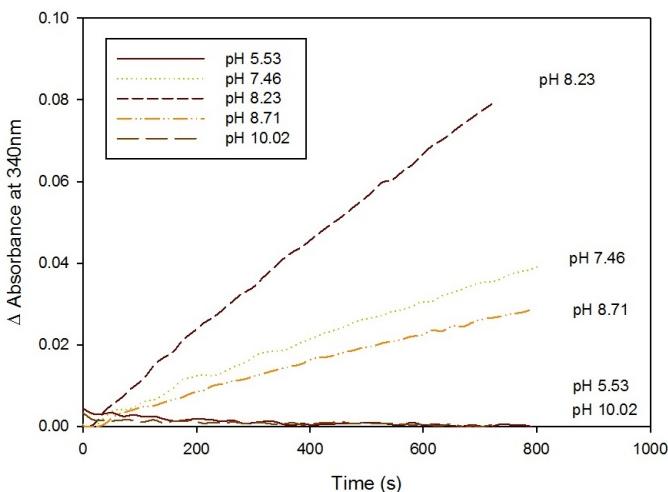
selected temperature. Furthermore, when  $T$  higher than  $30^{\circ}\text{C}$  is used, denaturation of the enzyme becomes a factor that needs to be considered (Fig.A.6(a)).

Thus, cuvette temperature should be measured before the assay is initiated, to make sure that the target temperature has been reached after the cuvette has been placed into the thermostatted cell holder of the

spectrophotometer.



[a]



[b]

Figure A.6: Kinetic responses of NADP-IDH to different levels of T and pH. (a) Temperature effects on NADP-IDH activity by the kinetic assay. (b) pH effects on NADP-IDH activity by the kinetic assay.



## Appendix B

# Appendix B. NADP-IDH thermodynamic factors

The thermodynamic parameters were calculated applying the equations of Arrhenius, Eyring and Low *et al.* (1973) Arrhenius (1889); Laidler (1984); Eyring (1935); Dixon and Webb (1979); Low et al. (1973). The Arrhenius equation describes the dependence of the rate of a chemical reaction on the absolute temperature, as expressed in Eq.B.1 Arrhenius (1889):

$$k_{IDH}=A \cdot \exp(-E_a/(R \cdot T)) \quad (\text{B.1})$$

where:  $k_{IDH}$  is the NADP-IDH activity related to the temperature in mol NADPH s<sup>-1</sup>(kg of protein)<sup>-1</sup>; A is the preexponential frequency factor, a unique value, in mol NADPH s<sup>-1</sup>(kg of protein)<sup>-1</sup>, for each chemical reaction defined by the rate due to collision frequency when reactants are in the correct orientation;  $E_a$  is the energy of activation in Kcal mol<sup>-1</sup>; T is the absolute temperature in °K and R is the universal molar gas constant in Kcal mol<sup>-1</sup> °K<sup>-1</sup> (1.9872 · 10<sup>-3</sup> Kcal mol<sup>-1</sup> °K<sup>-1</sup>). The  $E_a$  and A were obtained from the Arrhenius plot (Fig. 2.3), where the  $\ln(k_{IDH})$  is plotted against  $1/T$  and T is in 10<sup>-3</sup> °K units. Eq.B.2 describes the linear regression from this Arrhenius plot ( $y=a \cdot x + y_0$ ), where x is  $1/T$  and  $y_0$  is  $\ln(k_{IDH})$ :

$$\ln(k_{IDH})=\ln(A)-(E_a/R) \cdot (1/T) \quad (\text{B.2})$$

$E_a$  is related to the slope (a) of this linear regression, so  $a=-E_a/R$ . A is obtained from the intercept on the y axis ( $y_0=\ln(A)$ ). The enthalpy of activation ( $\Delta H^*$ ), in Kcal mol<sup>-1</sup>, was calculated from the  $E_a$  as represented on Eq.B.3 Low et al. (1973):

$$\Delta H^*=E_a-R \cdot T \quad (\text{B.3})$$

Here, R is 1.9872 · 10<sup>-3</sup> Kcal mol<sup>-1</sup> °K<sup>-1</sup>, as above, and T is the stan-

dard temperature of 25°C, expressed in °K (298°K).

$\Delta H^*$  also was obtained by combining the definition of Gibbs free energy ( $\Delta G^*$ ) (Eq.B.4) and the Eyring equation Eyring (1935); Dixon and Webb (1979) (Eq.B.5), resulting in Eq.B.6 and B.7:

$$\Delta G^* = \Delta H^* - T \cdot \Delta S^* \quad (\text{B.4})$$

$$k_{IDH} = (N_A \cdot k_B \cdot T / N_A \cdot h) \cdot \exp(-\Delta G^* / (R \cdot T)) \quad (\text{B.5})$$

$$k_{IDH}/T = (k_B/h) \cdot \exp(-\Delta H^* / (R \cdot T)) \cdot \exp(\Delta S^* / R) \quad (\text{B.6})$$

$$\ln(k_{IDH}/T) = \ln(k_B/h) + (\Delta S^* / R) - (\Delta H^* / R) \cdot (1/T) \quad (\text{B.7})$$

where:  $k_{IDH}$  is the NADP-IDH activity related to the temperature in mol NADPH s<sup>-1</sup>(kg of protein)<sup>-1</sup>; T in °K;  $k_B$  is the Boltzmann constant in cal °K<sup>-1</sup> (3.29 · 10<sup>-27</sup> cal °K<sup>-1</sup>); h is the Planck constant in cal s (1.58 · 10<sup>-37</sup> cal s). Note that these two constants are atomic constants. Thus, in order to transform them into the molar scale, both have been multiplied by Avogadro's number ( $N_A$ ) in Eq.B.5.  $\Delta G^*$  is the Gibbs free energy at standard conditions in Kcal mol<sup>-1</sup>,  $\Delta H^*$  is the enthalpy of activation at standard conditions in Kcal mol<sup>-1</sup> and  $\Delta S^*$  is the entropy of activation at standard conditions in Kcal mol<sup>-1</sup> °K<sup>-1</sup>.  $\Delta H^*$  is related to the slope (a) of this linear regression, so  $a = -\Delta H^* / R$ , and  $\Delta S^*$  is obtained from the intercept on the y axis ( $y_0 = \ln(k_B/h) + \Delta S^* / R$ ).

For comparison,  $\Delta G^*$ ,  $\Delta H^*$  and  $\Delta S^*$  were also calculated from the expressions reported by Low *et al.* (1973) represented in Eq.B.3, B.4 and B.8 Low et al. (1973). As the molecular weight of the enzyme NADP-IDH was needed for this method, these expressions were only applied to pure NADP-IDH from porcine heart (I2002, Sigma Aldrich), with a molecular weight (MW) of 57800 g mol<sup>-1</sup> (60KDa Dixon and Webb (1979)). We calculated the  $\Delta G^*$  from Eq.B.4. To do that, we also obtained  $\Delta H^*$  from Eq.B.3, and  $\Delta S^*$ , from Eq.B.8:

$$\Delta S^* = 4.576 \cdot (\log(k_{IDH}) - 10.753 - \log(T) + Ea / (4.576 \cdot T)) \quad (\text{B.8})$$

For this expression  $k_{IDH} = NADP-IDH^* MW$ , where *NADP-IDH* is the NADP-IDH activity in mol NADPH s<sup>-1</sup>(kg of protein)<sup>-1</sup> units, and MW is the NADP-IDH molecular weight in kg mol<sup>-1</sup> units, so  $k_{IDH}$  is in s<sup>-1</sup>.

Note that the results related to energy are reported in cal. To convert to Joules (J), multiply the result by 4.184 J cal<sup>-1</sup>. Furthermore, all the constants used here are in molar units instead of atomic units, in order to be consistent with most of the chemical, biochemical and biological research since Arrhenius (1915) Arrhenius et al. (1915). Furthermore, molar units are more appropriate for all scientific fields other

than atomic physics Feynman et al. (2011). This is because all these fields deal with large numbers of atoms or molecules in the range of Avogadro's number ( $6.02 \cdot 10^{23}$ ), i.e. molar quantities, whose chemical characteristics reflect their means described by Boltzmann distribution function Eggers Jr et al. (1964); Moore (1972). We chose not to use the atomic units of electron-volts because all of our measurements are made on a molar-scale.

The thermodynamic factors related to the NADP-IDH activity were calculated for both zooplankton and porcine heart NADP-IDH. The energy of activation ( $E_a$ ), the frequency factor ( $A$ ), the enthalpy of activation at  $25^\circ\text{C}$  ( $\Delta H^*$ ), the entropy of activation at  $25^\circ\text{C}$  ( $\Delta S^*$ ) and the Gibbs free energy ( $\Delta G^*$ ), calculated by the Arrhenius, Eyring, and Low *et. al* methodology when possible Arrhenius et al. (1915); Eyring (1935); Low et al. (1973), are reported in Table B.1.

Table B.1: NADP-IDH thermodynamic factors.

Sample origin	Marine Plankton	Porcine Heart
$E_a$ Arrhenius (Kcal mol $^{-1}$ )	20.40	21.40
$A$ Arrhenius (mol NADPH s $^{-1}$ (kg of protein) $^{-1}$ )	$2.19 \cdot 10^{11}$	$1.60 \cdot 10^{13}$
$\Delta H^*$ Arrhenius (Kcal mol $^{-1}$ )	19.81	20.76
$\Delta H^*$ Eyring (Kcal mol $^{-1}$ )	19.88	20.76
$\Delta S^*$ Eyring (Kcal mol $^{-1}$ K $^{-1}$ )	$-8.59 \cdot 10^{-3}$	$-8.74 \cdot 10^{-5}$
$\Delta G^*$ Eyring (Kcal mol $^{-1}$ )	22.44	20.79
$\Delta S^*$ Low <i>et al.</i> (Kcal mol $^{-1}$ K $^{-1}$ )	Not Available	$7.97 \pm 0.19$
$\Delta G^*$ Low <i>et al.</i> (Kcal mol $^{-1}$ )	Not Available	$18.39 \pm 0.06$

$F$  test and an independent  $t$  test were conducted to compare the Arrhenius  $E_a$  and  $\Delta H^*$  between poikilothermic (marine plankton) and homoiothermic (porcine heart) organism samples. Although linear regressions were significantly different ( $F$  test,  $F_{(2,37)}=4.64$ ,  $p=0.05$ ), there was not a significant difference between the regression coefficients ( $t$  test,  $t(39)=0.0773$ ,  $p=0.05$ ). These results suggested that, for NADP-IDH, although the frequency factor ( $A$ ) was different, the Energy of activation ( $E_a$ ) and the other thermodynamic factors derived from it ( $\Delta H^*$ ,  $\Delta S^*$ ,  $\Delta G^*$ ) were not significantly different between the

two types of organisms.

## Appendix C

### Appendix C. *Leptomysis lingvura* dataset

Table C.1: Respiratory rates measured in a bloom community of *L. lingvura*. Note that both NADP-IDH and CO<sub>2</sub> production rates ( $R_{CO_2}$ ) are expressed in  $\mu\text{molCO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$  units.

Bloom Community (2017)				$R_{CO_2}$	NADP-IDH
Incubation	Sex	Starvation	TriPLICATE		
4 h	Females	Well-Fed	1	0.160	1.815
4 h	Females	Well-Fed	2	0.404	1.928
4 h	Females	Well-Fed	3	0.308	2.095
8 h	Females	Well-Fed	1	0.165	1.933
8 h	Females	Well-Fed	2	0.301	1.977
8 h	Females	Well-Fed	3	0.235	1.995
12 h	Females	Well-Fed	1	0.340	2.108
12 h	Females	Well-Fed	2	0.244	2.123
12 h	Females	Well-Fed	3	0.247	2.148
16 h	Females	Well-Fed	1	0.205	2.665
16 h	Females	Well-Fed	2		
16 h	Females	Well-Fed	3	0.256	2.830
24 h	Females	Well-Fed	1	0.208	2.813
24 h	Females	Well-Fed	2	0.179	2.147
24 h	Females	Well-Fed	3	0.150	1.992

Table C.2: Respiratory rates measured in a post-bloom community of *L. lingvura*. Note that both NADP-IDH and CO<sub>2</sub> production rates ( $R_{CO_2}$ ) are expressed in  $\mu\text{molCO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$  units; and both  $\Phi$  and O<sub>2</sub> consumption rates ( $R_{O_2}$ ) are expressed in  $\mu\text{molO}_2 \text{ h}^{-1} \text{ mg protein}^{-1}$  units.

Post-Bloom Community (2018)				$R_{CO_2}$	NADP-IDH	$R_{O_2}$	$\Phi$
Incubation	Sex	Starvation	Triplicate				
1-O <sub>2</sub>	Females	Well-Fed	1		0.862	0.187	0.447
1-O <sub>2</sub>	Females	Well-Fed	2		0.972	0.179	0.531
1-O <sub>2</sub>	Females	Well-Fed	3		0.735	0.157	0.486
1-CO <sub>2</sub>	Females	Well-Fed	1	0.073	1.310		0.754
1-CO <sub>2</sub>	Females	Well-Fed	2	0.143	0.996		0.687
1-CO <sub>2</sub>	Females	Well-Fed	3	0.128	0.934		0.604
2-O <sub>2</sub>	Females	Starved	1		0.955	0.159	0.388
2-O <sub>2</sub>	Females	Starved	2		0.979	0.133	0.594
2-O <sub>2</sub>	Females	Starved	3		0.921	0.124	0.624
2-CO <sub>2</sub>	Females	Starved	1	0.132	0.791		0.525
2-CO <sub>2</sub>	Females	Starved	2		1.217		0.534
2-CO <sub>2</sub>	Females	Starved	3	0.135	1.009		0.610
3-O <sub>2</sub>	Males	Well-Fed	1		1.220	0.256	0.571
3-O <sub>2</sub>	Males	Well-Fed	2		0.888	0.224	0.876
3-O <sub>2</sub>	Males	Well-Fed	3		1.085	0.193	0.638
3-CO <sub>2</sub>	Males	Well-Fed	1	0.104	1.078		0.573
3-CO <sub>2</sub>	Males	Well-Fed	2	0.094	0.927		0.774
3-CO <sub>2</sub>	Males	Well-Fed	3	0.160	0.910		0.781
4-O <sub>2</sub>	Males	Non-starved	1		0.734	0.143	0.536
4-O <sub>2</sub>	Males	Non-starved	2		0.543	0.134	0.641
4-O <sub>2</sub>	Males	Non-starved	3		0.915	0.146	0.428
4-CO <sub>2</sub>	Males	Non-starved	1	0.202	0.798		0.596
4-CO <sub>2</sub>	Males	Non-starved	2	0.081	0.960		0.627
4-CO <sub>2</sub>	Males	Non-starved	3	0.171	1.032		0.633

## Appendix D

# Appendix D. KOSMOS GC 2014 Main dataset

Table D.1: Potential respiration ( $\Phi$ ) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the  $\text{pCO}_2$  levels of the treatment. Note that  $\Phi$  is expressed in  $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$  units. To convert to “per protein” units, use the data on table D.4.

Treatment	Mesocosm	I: Oligotrophic phase (day)						
		-3	-1	3	7	11	15	19
Low	M1	0.595	0.477	0.641	0.644	1.023	0.543	0.660
High	M2	0.460	0.369	0.852	0.339	0.362	0.656	0.780
Intermediate	M3	0.347	0.513	0.467	0.678	0.680	0.599	0.774
Intermediate	M4	0.438	0.470	0.452	0.589	0.509	0.526	0.720
Low	M5	0.381	0.440	0.716	0.640	0.706	0.328	0.780
Intermediate	M7	0.318	0.302	0.486	0.598	0.486	0.328	0.674
High	M8	0.427	0.486	0.396	0.736	0.650	0.273	0.734
Low	M9	0.440	0.601	0.418	0.186	0.497	0.340	0.768
Atlantic	Atlantic	0.350	0.302	0.440	0.272	0.273	0.215	0.267
								0.314

Table D.2: Potential respiration ( $\Phi$ ) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that  $\Phi$  is expressed in  $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$  units. To convert to “per protein” units, use the data on table D.5.

<b>Treatment</b>	<b>Mesocosm</b>	<b>II: Bloom Phase (day)</b>				
		<b>26</b>	<b>28</b>	<b>30</b>	<b>32</b>	<b>35</b>
<b>Low</b>	<b>M1</b>	2.483	3.529	1.285	1.496	2.167
<b>High</b>	<b>M2</b>	2.392	1.043	1.490	2.254	5.174
<b>Intermediate</b>	<b>M3</b>	2.766	2.107	1.622	4.708	3.515
<b>Intermediate</b>	<b>M4</b>	2.190	5.044	2.398	6.194	5.845
<b>Low</b>	<b>M5</b>	2.724	4.254	0.855	1.229	2.683
<b>Intermediate</b>	<b>M7</b>	2.373	4.652	2.197	2.779	4.220
<b>High</b>	<b>M8</b>	2.631	4.494	1.257	5.136	6.321
<b>Low</b>	<b>M9</b>	2.477	3.733	0.152	0.441	3.277
<b>Atlantic</b>	<b>Atlantic</b>	0.467	0.388	0.234	0.465	0.565

Table D.3: Potential respiration ( $\Phi$ ) during the Post-bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that  $\Phi$  is expressed in  $\mu\text{LO}_2\text{h}^{-1}\text{L}^{-1}$  units. To convert to “per protein” units, use the data on table D.6.

<b>Treatment</b>	<b>Mesocosm</b>	<b>III: Post-bloom Phase (day)</b>				
		<b>39</b>	<b>43</b>	<b>47</b>	<b>51</b>	<b>55</b>
<b>Low</b>	<b>M1</b>	1.691	1.995	2.214	1.659	2.419
<b>High</b>	<b>M2</b>	4.573	3.371	2.032	1.524	3.837
<b>Intermediate</b>	<b>M3</b>	1.692	1.647	1.409	2.658	3.071
<b>Intermediate</b>	<b>M4</b>	4.189	1.657	1.811	2.048	1.443
<b>Low</b>	<b>M5</b>	2.548	2.146	2.339	3.146	0.408
<b>Intermediate</b>	<b>M7</b>	4.080	1.952	2.793	3.100	1.736
<b>High</b>	<b>M8</b>	4.672	3.331	2.309	1.926	2.377
<b>Low</b>	<b>M9</b>	4.001	2.548	2.606	2.257	3.506
<b>Atlantic</b>	<b>Atlantic</b>	0.328	0.459	0.711	0.631	1.978

Table D.4: Biomass (B) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L<sup>-1</sup> units.

<b>B</b> Treatment	Mesocosm	I: Oligotrophic phase (day)						
		<b>-3</b>	<b>-1</b>	<b>3</b>	<b>7</b>	<b>11</b>	<b>15</b>	<b>19</b>
Low	M1	0.027	0.012	0.018	0.006	0.005	0.007	0.021
High	M2	0.022	0.009	0.011	0.008	0.010	0.013	0.017
Intermediate	M3	0.025			0.008	0.009	0.015	0.013
Intermediate	M4	0.021	0.013		0.009	0.014	0.010	0.005
Low	M5	0.026	0.005	0.012	0.005		0.006	0.013
Intermediate	M7	0.014	0.004	0.002	0.009		0.006	0.008
High	M8	0.022		0.009	0.018	0.014		0.013
Low	M9	0.019	0.006	0.003		0.006	0.003	0.012
Atlantic	Atlantic	0.018	0.007	0.022	0.005	0.006	0.005	0.009
								0.023

Table D.5: Biomass (B) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L<sup>-1</sup> units.

<b>B</b> Treatment	Mesocosm	II: Bloom Phase (day)				
		<b>26</b>	<b>28</b>	<b>30</b>	<b>32</b>	<b>35</b>
Low	M1	0.029	0.079	0.079	0.047	0.050
High	M2	0.079	0.163	0.158	0.146	0.067
Intermediate	M3	0.054	0.096	0.134	0.103	0.091
Intermediate	M4	0.040	0.101	0.170	0.161	0.078
Low	M5	0.024	0.140	0.162	0.141	0.089
Intermediate	M7	0.042	0.082	0.159	0.096	0.111
High	M8	0.040	0.116	0.201	0.195	0.154
Low	M9	0.054	0.165	0.192	0.124	0.099
Atlantic	Atlantic	0.016	0.032	0.030	0.009	0.010

Table D.6: Biomass (B) during the Post-bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that B is expressed in mg protein L<sup>-1</sup> units.

B		III: Post-bloom Phase (day)				
Treatment	Mesocosm	39	43	47	51	55
Low	M1	0.038	0.063	0.034	0.030	0.049
High	M2	0.071	0.094	0.044	0.066	0.095
Intermediate	M3	0.042	0.035	0.041	0.026	0.043
Intermediate	M4	0.107	0.041	0.023	0.072	0.045
Low	M5	0.087	0.026	0.067	0.095	0.055
Intermediate	M7	0.085	0.064	0.066	0.119	0.039
High	M8	0.144	0.137	0.095	0.089	0.081
Low	M9	0.110	0.089	0.065	0.053	0.080
Atlantic	Atlantic	0.005	0.015	0.017	0.020	0.047

Table D.7: Chlorophyll *a* (Chl<sub>*a*</sub>) during the Oligotrophic phase (Phase I) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that Chl<sub>*a*</sub> is expressed in  $\mu\text{g L}^{-1}$  units. To convert to “per protein” units, use the data on table D.4.

Total Chl_α		I: Oligotrophic phase (day)							
Treatment	Mesocosm	-3	-1	3	7	11	15	19	23
Low	M1	0.071	0.053	0.069	0.103	0.161	0.107	0.140	0.130
High	M2	0.081	0.046	0.089	0.091	0.124	0.127	0.159	0.213
Intermediate	M3	0.068	0.044	0.072	0.087	0.122	0.132	0.154	0.192
Intermediate	M4	0.078	0.058	0.084	0.088	0.119	0.096	0.149	0.125
Low	M5	0.075	0.060	0.091	0.124	0.110	0.127	0.122	0.169
Intermediate	M7	0.084	0.049	0.066	0.094	0.099	0.100	0.156	0.161
High	M8	0.081	0.053	0.073	0.097	0.117	0.119	0.128	0.113
Low	M9	0.105	0.067	0.061	0.119	0.132	0.143	0.173	0.185
Atlantic	Atlantic	0.080	0.051	0.220	0.075	0.068	0.068	0.136	0.067

Table D.8: Chlorophyll *a* (Chl<sub>*a*</sub>) during the Bloom phase (Phase II) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that Chl<sub>*a*</sub> is expressed in  $\mu\text{g L}^{-1}$  units. To convert to “per protein” units, use the data on table D.5.

Treatment	Mesocosm	II: Bloom phase (day)				
		26	28	30	32	35
Low	M1	0.3838	1.9067	1.0872	0.6553	0.3506
High	M2	1.4529	4.0160	1.6453	1.3349	0.8758
Intermediate	M3	0.7153	3.5005	1.2707	0.8083	0.4436
Intermediate	M4	0.5618	3.2992	1.8301	0.8133	0.5730
Low	M5	0.5048	3.4288	1.5563	1.3006	0.6894
Intermediate	M7	0.4477	2.3366	2.3752	1.4558	0.6201
High	M8	0.5371	2.2904	2.0110	1.2962	0.9296
Low	M9	0.6218	4.5137	2.5883	2.2726	0.8154
Atlantic	Atlantic	0.1081	0.0847	0.0892	0.1065	0.1067

Table D.9: Chlorophyll *a* (Chl<sub>*a*</sub>) during the Post-Bloom phase (Phase III) in the KOSMOS GC 2014 experiment. Mesocosms have been classified as Low, Intermediate or High depending on the pCO<sub>2</sub> levels of the treatment. Note that Chl<sub>*a*</sub> is expressed in  $\mu\text{g L}^{-1}$  units. To convert to “per protein” units, use the data on table D.6.

Treatment	Mesocosm	III:Postbloom Phase (day)				
		39	43	47	51	55
Low	M1	0.7304	0.2513	0.2182	0.4631	0.3497
High	M2	1.2395	0.9873	0.7085	0.7447	0.4449
Intermediate	M3	0.7393	0.4303	0.4046	0.4195	0.4210
Intermediate	M4	0.6270	0.4098	0.4024	0.6174	0.5299
Low	M5	0.5920	0.3764	0.3075	0.6406	0.5428
Intermediate	M7	0.5786	0.5112	0.2835	0.6345	0.8797
High	M8	1.4283	1.1776	0.5760	0.7102	0.8263
Low	M9	0.7103	0.4357	0.3379	0.4422	0.5378
Atlantic	Atlantic	0.0871	0.0977	0.1593	0.1553	0.1992



## Appendix E

# Appendix E. Abbreviations

Table E.1: Summary of the symbols used to identify the parameters (1/2).

Parameter	Definition
<b>A</b>	Preeponential frequency factor
<b><math>A_{340}</math></b>	Absorbance at 340nm
<b>Acon</b>	Aconitase
<b>ATP</b>	Adenosine triphosphate
<b>B</b>	Biomass
<b><math>B_{340}</math></b>	$A_{340}$ increase per time of the absorbance at 340nm in the blank assay
<b>Biosynthetic <math>\Psi_{NADP}</math></b>	Potential CO <sub>2</sub> production related to biosynthesis
<b>Chl_α</b>	Chlorophyll α
<b>CS</b>	Citrate synthase
<b>χ</b>	Empirical relationship between physiological CO <sub>2</sub> production and potential O <sub>2</sub> consumption
<b><math>\Delta_{340}</math></b>	$A_{340}$ increase per time
<b><math>\Delta H^*</math></b>	Enthalpy of activation at standard conditions
<b><math>\Delta_{NADPH}</math></b>	[NADPH] increase per time
<b>DOC</b>	Dissolved organic carbon
<b>Ea</b>	Energy of activation
<b>EKM</b>	Enzyme kinetic model
<b>ETS</b>	Electron transport system
<b>ε</b>	Extinction coefficient
<b>Fc</b>	Carbon flux
<b>Fum</b>	Fumarase
<b>Φ</b>	Potential O <sub>2</sub> consumption based on ETS
<b>HEP</b>	Heterotrophic energy production
<b>IDH<sub>KC</sub></b>	IDH isoenzymes linked to the Krebs cycle
<b>IDH<sub>KC<sub>CO2</sub></sub></b>	CO <sub>2</sub> production related to IDH isoenzymes linked to the Krebs cycle
<b>INT</b>	2-(p-iodophenyl)-3-(p-nitrophenyl)-5-phenyl tetrazolium chloride (tetrazolium salt)
<b>K<sub>1</sub></b>	Velocity constant at T1
<b>K<sub>2</sub></b>	Velocity constant at T2
<b>KC</b>	Krebs cycle
<b>KC<sub>CO2</sub></b>	CO <sub>2</sub> production related to the Krebs cycle
<b>KGDH</b>	α-ketoglutarate dehydrogenase
<b>KGDH<sub>CO2</sub></b>	CO <sub>2</sub> production related to α-ketoglutarate dehydrogenase
<b>k<sub>IDH</sub></b>	NADP-IDH activity related to temperature
<b>K<sub>m</sub></b>	Michaelis-Menten constant

Table E.2: Summary of the symbols used to identify the parameters (2/2).

Parameter	Definition
<b>MDH</b>	Malate dehydrogenase
<b>ME</b>	Malic enzyme
<b>MTE</b>	Metabolical theory of ecology
<b>ME<sub>CO<sub>2</sub></sub></b>	CO <sub>2</sub> production related to the malic enzyme
<b>NAD-IDH</b>	NAD <sup>+</sup> -dependent isocitrate dehydrogenase
<b>NADP-IDH</b>	NADP <sup>+</sup> -dependent isocitrate dehydrogenase
<b>NADP-IDH<sub>mit</sub></b>	NADP-IDH isoenzyme linked to the mitochondria
<b>NADP-IDH<sub>out</sub></b>	NADP-IDH isoenzymes outside the mitochondria
<b>NADP-IDH<sub>out</sub><sub>CO<sub>2</sub></sub></b>	CO <sub>2</sub> production related to the NADP-IDH isoenzymes outside the mitochondria
<b>NADP-IDH(Ti)</b>	NADP-IDH activity at in-situ T
<b>NADP-IDH(Tm)</b>	NADP-IDH activity at T during measurement
<b>NRE</b>	Nutrient regeneration efficiency
<b>P<sub>N</sub></b>	Net productivity
<b>PEPCK</b>	Phosphoenolpyruvate carboxykinase
<b>PEPCK<sub>CO<sub>2</sub></sub></b>	CO <sub>2</sub> production related to phosphoenolpyruvate carboxykinase
<b>PDH<sub>CO<sub>2</sub></sub></b>	CO <sub>2</sub> production related to pyruvate dehydrogenase
<b>POC</b>	Particulate organic carbon
<b>Ψ</b>	Potential CO <sub>2</sub> production
<b>Ψ<sub>NADP</sub></b>	Potential CO <sub>2</sub> production based on NADP-IDH
<b>Q<sub>10</sub></b>	Temperature coefficient
<b>R</b>	Universal molar gas constant
<b>Respiratory Ψ<sub>NADP</sub></b>	Potential CO <sub>2</sub> production related to the respiratory metabolism
<b>R<sub>CO<sub>2</sub></sub></b>	Physiological CO <sub>2</sub> production
<b>R<sub>CO<sub>2</sub></sub>/NADP-IDH</b>	Ratio between physiological CO <sub>2</sub> production and potential CO <sub>2</sub> production of the NADP-IDH
<b>R<sub>CO<sub>2</sub></sub>/Ψ<sub>NADP</sub></b>	Ratio between physiological and potential CO <sub>2</sub> production
<b>R<sub>O<sub>2</sub></sub></b>	Physiological O <sub>2</sub> consumption
<b>R<sub>O<sub>2</sub></sub>/Φ</b>	Ratio between physiological and potential O <sub>2</sub> consumption
<b>RQ</b>	Respiratory Quotient
<b>SuccDH</b>	CO <sub>2</sub> Succinate dehydrogenase
<b>[S]</b>	Substrate concentration
<b>T</b>	Temperature
<b>T<sub>i</sub></b>	In-situ temperature
<b>T<sub>m</sub></b>	Inside-cuvette temperature during measurement
<b>V<sub>c</sub></b>	Final volume in the cuvette
<b>V<sub>h</sub></b>	Volume of the sample-homogenate added into the cuvette
<b>V<sub>s</sub></b>	Total volume of the sample-homogenate
<b>V<sub>max</sub></b>	Maximum velocity of an enzyme-catalysed reaction

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# **Declaration**

I herewith declare that I have produced this work without the prohibited assistance of third parties and without making use of aids other than specified; notions taken over directly or indirectly from other sources have been identified as such. This work has not previously been presented in identical or similar form to any examination board. The dissertation work was conducted from 2014-2019 under the supervision of Prof. María M. Gómez Cabrera and Dr. Theodore T. Packard at the University of Las Palmas de Gran Canaria.

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